

*Supplementary Material*

**New Insights into the Origin of the *cis*-Configuration Preferences in 1,2-Dihaloethenes: The Importance of the Bonding Orbital Deviations**

Leila Tavanaei and Davood Nori-Shargh\*

*Chemistry Department, Science Faculty, Arak Branch, Islamic Azad University, Arak, Iran*

\*Corresponding author. Email: [D-norishargh@iau-arak.ac.ir](mailto:D-norishargh@iau-arak.ac.ir); nori\_ir@yahoo.com

**Table S1.** G3MP2 calculated zero point energies, corrected electronic energies [ $E_o=E_{el}+ZPE$ ], thermodynamic functions [ $H, S, G$ ] and their corresponding differences [ $\Delta ZPE, \Delta H, \Delta G$  (in kcal mol<sup>-1</sup>) and  $\Delta S$  (in cal mol<sup>-1</sup>K<sup>-1</sup>)] at 300 K for the *cis*- and *trans*-configurations of compounds **1-3**.

	G3MP2										Exp.
	<i>H</i>	<i>S</i>	<i>G</i>	<i>ZPE</i>	<i>E<sub>o</sub></i>	$\Delta H$	$\Delta S$	$\Delta G$	$\Delta ZPE$	$\Delta E_o$	$\Delta E_o$
<b>1-cis</b>	-276.740885	64.128	-276.771379	0.036108	-276.745722	0.00	0.000	0.00	0.00	0.00	0.00
<b>1-trans</b>	-276.739635	63.947	-276.770043	0.035757	-276.744577	0.78	-0.181	0.84	-0.22	0.72	(1.08 ± 0.12) <sup>a</sup>
<b>2-cis</b>	-996.748379	69.184	-996.781277	0.033277	-996.753622	0.00	0.000	0.00	0.00	0.00	0.0
<b>2-trans</b>	-996.747338	69.392	-996.780335	0.032957	-996.752839	0.91	0.208	0.59	-0.20	0.49	(0.72 ± 0.16) <sup>a</sup>
<b>3-cis</b>	-5222.897277	74.742	-5222.932818	0.031971	-5222.902856	0.00	0.000	0.000	0.00	0.00	0.00
<b>3-trans</b>	-5222.896787	75.074	-5222.932486	0.031696	-5222.902684	0.31	0.332	0.21	-0.17	0.11	(0.25 ± 0.33) <sup>a</sup>

<sup>a</sup>From gas chromatography, Ref. [4].

**Table S2.** CCSD(T)/6-311+G\*\* calculated zero point energies, corrected electronic energies [ $E_o=E_{el}+ZPE$ ], thermodynamic functions [ $H, S, G$ ] and their corresponding differences [ $\Delta ZPE, \Delta H, \Delta G$  (in kcal mol<sup>-1</sup>) and  $\Delta S$  (in cal mol<sup>-1</sup>K<sup>-1</sup>)] for the *cis*- and *trans*-configurations of compounds **1-3**.

	CCSD(T)/6-311+G**										Exp.
	<i>H</i>	<i>S</i>	<i>G</i>	<i>ZPE</i>	<i>E<sub>o</sub></i>	$\Delta H$	$\Delta S$	$\Delta G$	$\Delta ZPE$	$\Delta E_o$	$\Delta E_o$
<b>1-cis</b>	-276.491846	64.252	-276.522374	0.036782	-276.496690	0.00	0.000	0.00	0.005276	0.00	0.00
<b>1-trans</b>	-276.491076	64.141	-276.521552	0.036431	-276.496033	0.48	-0.111	0.52	0.004925	0.41	(1.08 ± 0.12) <sup>a</sup>
<b>2-cis</b>	-996.479827	69.263	-996.512736	0.034036	-996.485077	0.00	0.000	0.00	0.004174	0.00	0.0
<b>2-trans</b>	-996.479470	69.483	-996.512483	0.033770	-996.484963	0.22	0.220	0.16	0.003908	0.07	(0.72 ± 0.16) <sup>a</sup>
<b>3-cis</b>	-5222.171049	74.807	-5222.206592	0.032601	-5222.176635	0.00	0.000	0.00	0.004151	0.00	0.00
<b>3-trans</b>	-5222.171898	75.100	-5222.207580	0.032381	-5222.177780	-0.53	0.293	-0.62	0.003931	-0.72	(0.25 ± 0.33) <sup>a</sup>

<sup>a</sup>From gas chromatography, Ref. [4].

**Table S3.** LC- $\omega$ PBE/6-311+G\*\* calculated zero point energies, corrected electronic energies [ $E_o=E_{el}+ZPE$ ], thermodynamic functions [ $H, S, G$ ] and their corresponding differences [ $\Delta ZPE, \Delta H, \Delta G$  (in kcal mol<sup>-1</sup>) and  $\Delta S$  (in cal mol<sup>-1</sup>K<sup>-1</sup>)] at 300 K for the *cis*- and *trans*-configurations of compounds **1-3**.

	LC- $\omega$ PBE/6-311+G**										Exp.
	<i>ZPE</i>	<i>E<sub>o</sub></i>	<i>H</i>	<i>S</i>	<i>G</i>	$\Delta ZPE$	$\Delta E_o$	$\Delta H$	$\Delta S$	$\Delta G$	$\Delta E_o$
<b>1-cis</b>	0.037673	-276.935966	-276.931192	63.999	-276.961600	0.00	0.00	0.00	0.000	0.00	0.00
<b>1-trans</b>	0.037268	-276.935195	-276.930302	63.886	-276.960656	-0.25	0.48	0.56	-0.113	0.59	(1.08 ± 0.12) <sup>a</sup>
<b>2-cis</b>	0.035165	-997.480930	-997.475803	68.791	-997.508488	0.00	0.00	0.00	0.000	0.00	0.0
<b>2-trans</b>	0.034794	-997.480659	-997.475279	68.969	-997.508048	-0.23	0.17	0.33	0.178	0.28	(0.72 ± 0.16) <sup>a</sup>
<b>3-cis</b>	0.033892	-5224.838383	-5224.832951	74.193	-5224.868203	0.00	0.00	0.00	0.000	0.00	0.00
<b>3-trans</b>	0.033528	-5224.839033	-5224.833275	74.534	-5224.868688	-0.23	-0.41	-0.20	0.341	-0.30	(0.25 ± 0.33) <sup>a</sup>

<sup>a</sup>From gas chromatography, Ref. [4].

**Table S4.** NBO-LC- $\omega$ PBE/6-311+G\*\* calculated orbital occupancies ( $e$ ) and orbital energies ( $\varepsilon$ , in a.u.) for the *cis*- and *trans*-configurations of compounds **1-3**.

	<b>1</b>		<b>2</b>		<b>3</b>	
	<i>cis</i>	<i>trans</i>	<i>cis</i>	<i>trans</i>	<i>cis</i>	<i>trans</i>
$\varepsilon$						
$\sigma_{\text{C-H}}$	-0.70496	-0.70646	-0.69573	-0.70128	-0.69328	-0.70008
$\sigma_{\text{C-X}}$	-1.16131	-1.15108	-0.88813	-0.88037	-0.81937	-0.81086
LP <sub>3</sub> X	-0.54573	-0.54849	-0.44486	-0.44927	-0.41582	-0.42043
$\sigma^*_{\text{C-X}}$	0.38302	0.37549	0.27491	0.26798	0.21290	0.20532
$\sigma^*_{\text{C-H}}$	0.48360	0.48338	0.50191	0.50268	0.51111	0.51216
$\pi^*_{\text{C=C}}$	0.09133	0.09277	0.07644	0.07787	0.07821	0.07787
$\Delta[\varepsilon(\pi^*_{\text{C=C}})-\varepsilon(\text{LP}_3\text{X})]$	0.63706	0.64126	0.52130	0.52714	0.49403	0.4983
$\Delta[\varepsilon(\sigma^*_{\text{C-X}})-\varepsilon(\sigma_{\text{C-H}})]$	1.08798	1.08195	0.97064	0.96926	0.90618	0.9054
$\Delta[\varepsilon(\sigma^*_{\text{C-X}})-\varepsilon(\sigma_{\text{C-X}})]$	1.54433	1.52657	1.16304	1.14835	1.03227	1.01618
$e$						
$\sigma_{\text{C-H}}$	1.98285	1.98655	1.97488	1.98278	1.97217	1.98203
$\sigma_{\text{C-X}}$	1.99652	1.99489	1.99158	1.98638	1.98872	1.98032
LP <sub>3</sub> X	1.92942	1.93537	1.92259	1.93194	1.92961	1.93929
$\sigma^*_{\text{C-X}}$	0.01407	0.01156	0.01922	0.01807	0.02339	0.02329
$\sigma^*_{\text{C-H}}$	0.01966	0.02073	0.02334	0.02359	0.02205	0.02206
$\pi^*_{\text{C=C}}$	0.13383	0.12329	0.14841	0.13088	0.13523	0.11713

**Table S5.** LC-wPBE/6-311+G\*\* calculated energies (in hartree) of HOMO ( $\varepsilon_{\text{HOMO}}$ ), LUMO ( $\varepsilon_{\text{LUMO}}$ ),  $\varepsilon_{\text{LUMO}} - \varepsilon_{\text{HOMO}}$ , ionization potential ( $IP$ ), electron affinity ( $EA$ ), global hardness ( $\eta$ ), global electronegativity ( $\chi$ ) for the *cis*- and *trans*-configurations of compounds **1-3**.

	$\varepsilon_{\text{HOMO}}$	$\varepsilon_{\text{LUMO}}$	$\varepsilon_{\text{LUMO}} - \varepsilon_{\text{HOMO}}$	$IP$	$EA$	$\eta$	$\chi$	$\Delta\eta$	$\Delta\chi$
<b>1-cis</b>	-0.38700	0.06048	0.44748	0.38700	-0.06048	0.22374	0.16326	0.00000	0.00000
<b>1-trans</b>	-0.38652	0.07354	0.46006	0.38652	-0.07354	0.23003	0.15649	0.00629(3.95) <sup>a</sup>	-0.00677(-4.25) <sup>a</sup>
<b>2-cis</b>	-0.37318	0.06032	0.43350	0.37318	-0.06032	0.21675	0.15643	0.00000	0.00000
<b>2-trans</b>	-0.37258	0.05441	0.42699	0.37258	-0.05441	0.21350	0.15909	-0.00325(-2.04) <sup>a</sup>	0.00266(1.67) <sup>a</sup>
<b>3-cis</b>	-0.36463	0.05276	0.41739	0.36463	-0.05276	0.20870	0.15594	0.00000	0.00000
<b>3-trans</b>	-0.36364	0.04875	0.41239	0.36364	-0.04875	0.20620	0.15745	-0.00250(-1.57)	0.00151(0.95) <sup>a</sup>

<sup>a</sup> Values are in kcal mol<sup>-1</sup>.

**Table S6.**  $\Delta$ -LC-wPBE/6-311+G\*\* calculated ionization potential ( $IP$ ), electron affinity ( $EA$ ), global hardness ( $\eta$ ), global electronegativity ( $\chi$ ) (in hartree) for the *cis*- and *trans*-configurations of compounds **1-3**.

	$IP$	$EA$	$\eta$	$\chi$	$\Delta\eta$	$\Delta\chi$
<b>1-cis</b>	0.386502	-0.082317	0.2344095	0.1520925	0.00000	0.00000
<b>1-trans</b>	0.386492	-0.075502	0.230997	0.155495	-0.003413(-2.14) <sup>a</sup>	0.003403(2.14) <sup>a</sup>
<b>2-cis</b>	0.366496	-0.048189	0.2073425	0.15915	0.00000	0.00000
<b>2-trans</b>	0.366257	-0.041098	0.2036775	0.16258	-0.00367(-2.30) <sup>a</sup>	0.00343(2.15) <sup>a</sup>
<b>3-cis</b>	0.35724	-0.03842	0.19782	0.15941	0.00000	0.00000
<b>3-trans</b>	0.35664	-0.03205	0.19434	0.16230	-0.003478(-2.18) <sup>a</sup>	0.00289(1.81) <sup>a</sup>

<sup>a</sup> Values are in kcal mol<sup>-1</sup>.

**Table S7.**  $\Delta$ -EOM-CCSD/6-311+G\*\*//CCSD/6-311+G\*\* calculated ionization potential ( $IP$ ), electron affinity ( $EA$ ), global hardness ( $\eta$ ), global electronegativity ( $\chi$ ) (in hartree) for the *cis*- and *trans*-configurations of compounds **1-3**.

	$IP$	$EA$	$\eta$	$\chi$	$\Delta\eta$	$\Delta\chi$
<b>1-cis</b>	0.3479042	-0.1051421	0.22652315	0.1213810	0.00000	0.00000
<b>1-trans</b>	0.3472961	-0.0966451	0.2219706	0.1253255	-0.00455(-2.86) <sup>a</sup>	0.00394(2.47) <sup>a</sup>
<b>2-cis</b>	0.3299978	-0.0722365	0.20111715	0.12888065	0.00000	0.00000
<b>2-trans</b>	0.3299993	-0.065353	0.19767615	0.13232315	-0.00344(-2.16) <sup>a</sup>	0.00344(2.16) <sup>a</sup>
<b>3-cis</b>	0.3252479	-0.073224	0.19923595	0.12601195	0.00000	0.00000
<b>3-trans</b>	0.3254915	-0.0548525	0.190172	0.1353195	-0.00906(-5.69) <sup>a</sup>	0.00931(5.84) <sup>a</sup>

<sup>a</sup> Values are in kcal mol<sup>-1</sup>.