

Supplementary Material

Cocrystals and Salts of 3,5-bis(pyridinylmethylene)piperidin-4-one with Aromatic Poly-carboxylates and Resorcinols: Influence of Stacking Interactions on Solid-state Luminescence Properties

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CCDC numbers for crystal structures: 1875811-1875817.

Contents

- | | | |
|----|--|----------|
| 1) | IR spectra | |
| 2) | XRPD patterns | |
| 3) | Geometrical parameters of the hydrogen bonds in the Complexes | Table S1 |
| 4) | pK _a of ketones, coformers and ΔpK _a values of the complexes | Table S2 |

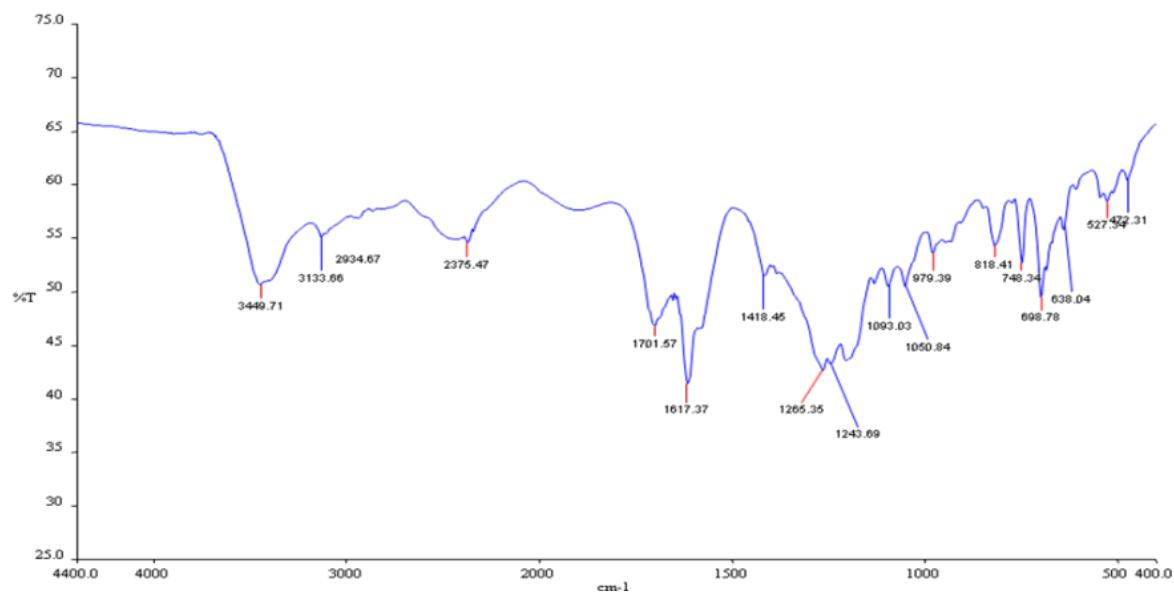


Figure S1: IR spectrum of **4**.

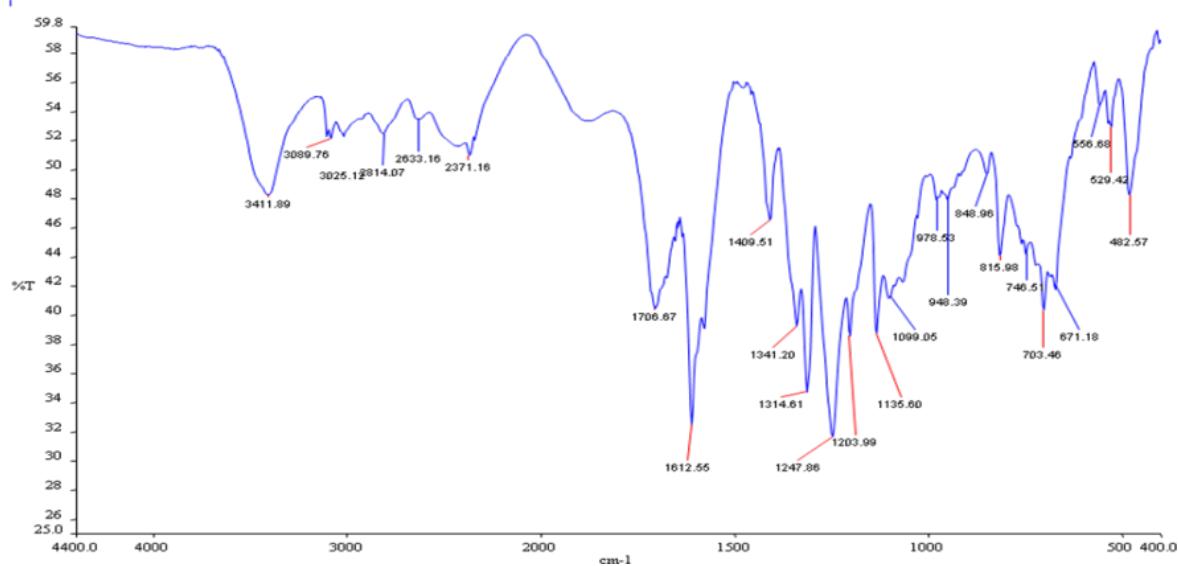


Figure S2: IR spectrum of **5**.

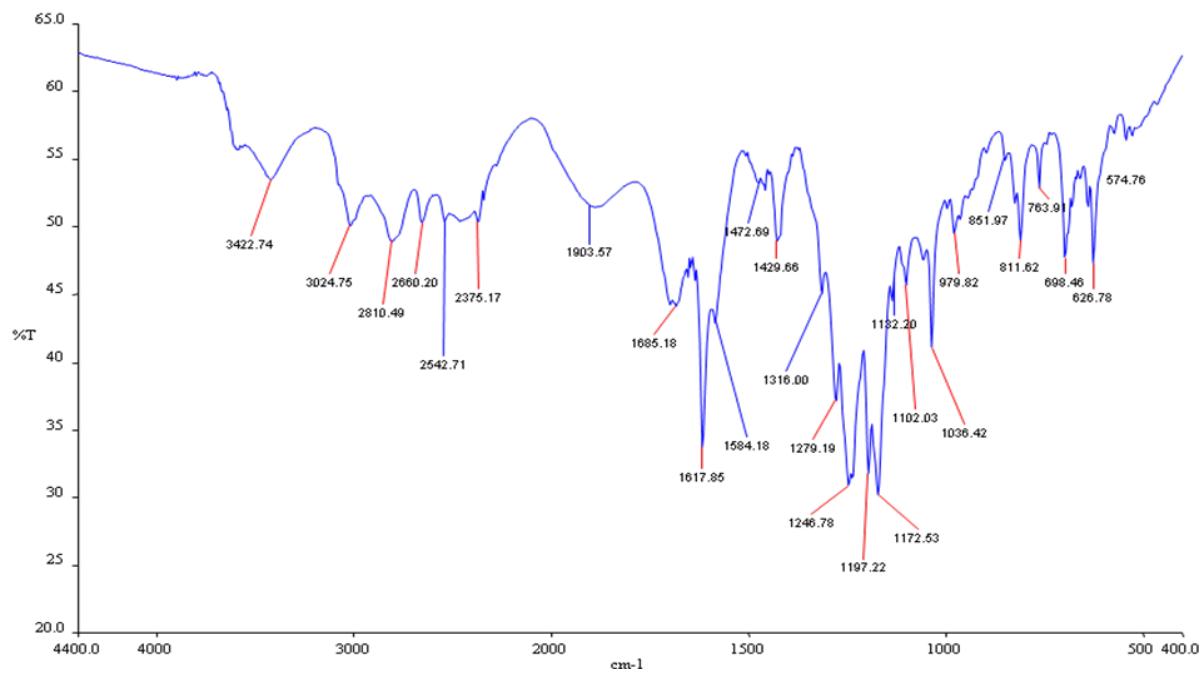


Figure S3: IR spectrum of **6**.

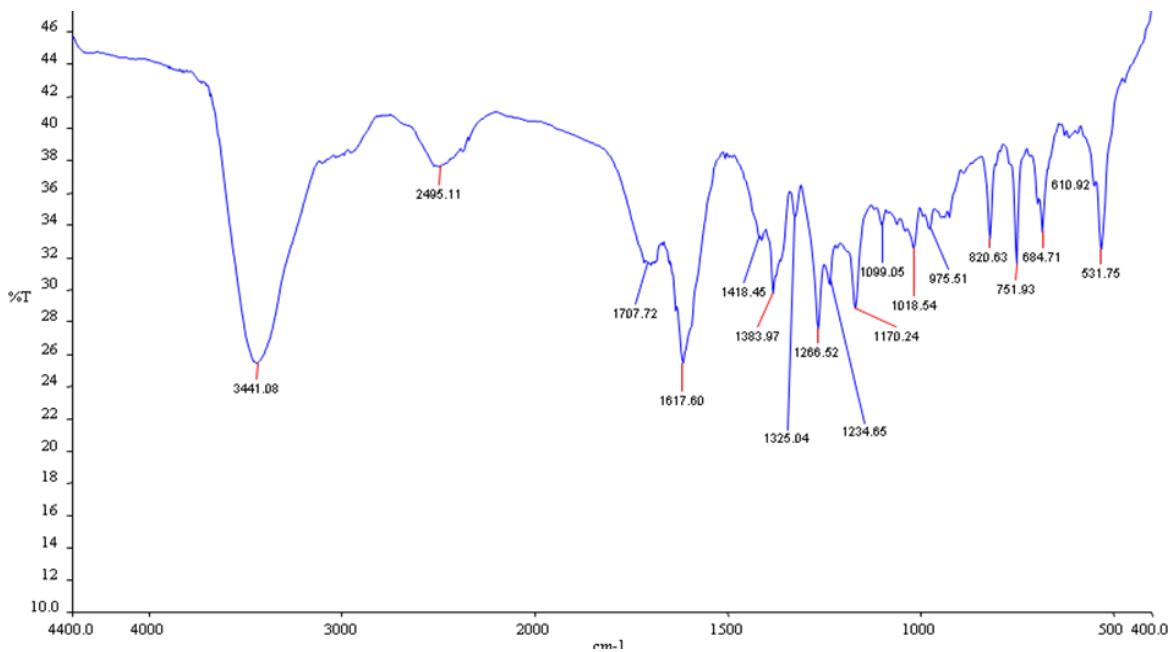


Figure S4: IR spectrum of **7**.

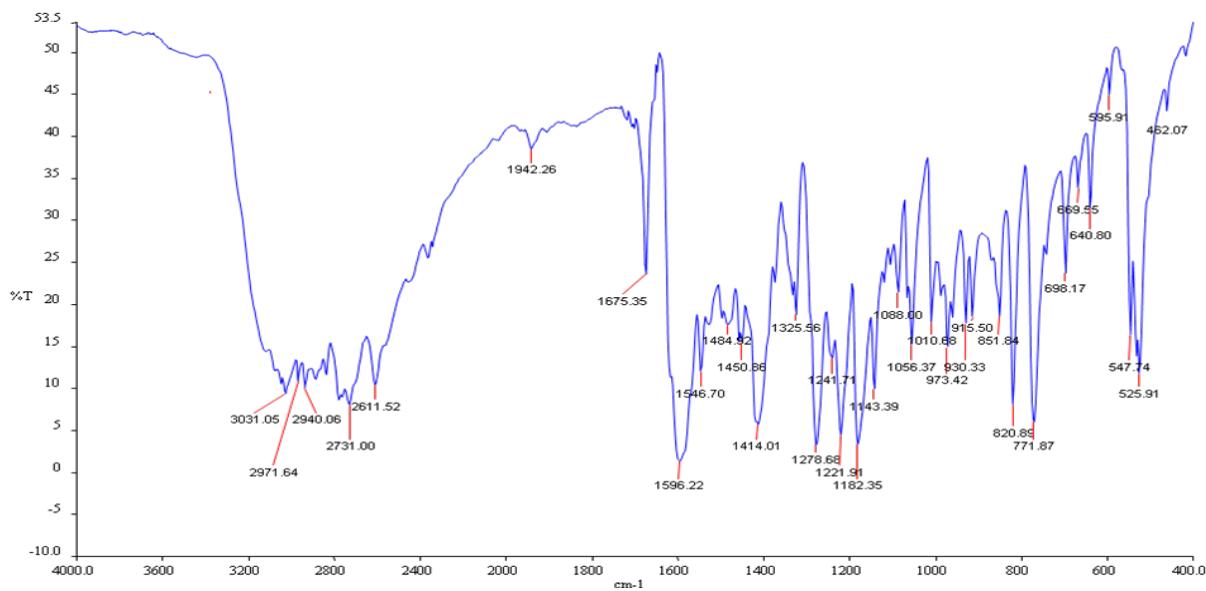


Figure S5: IR spectrum of **8**.

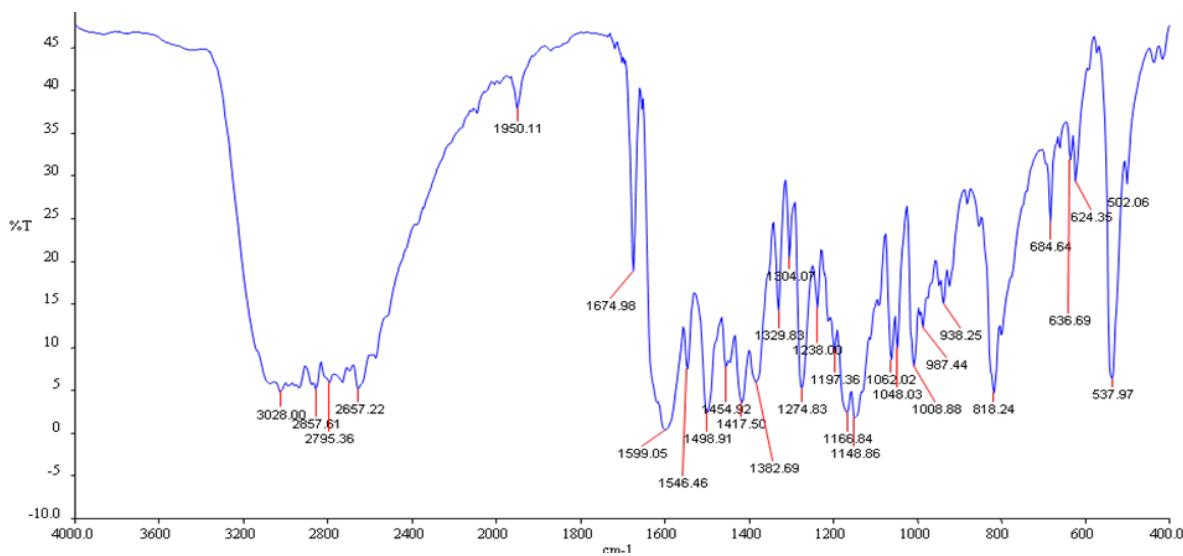


Figure S6: IR spectrum of **9**

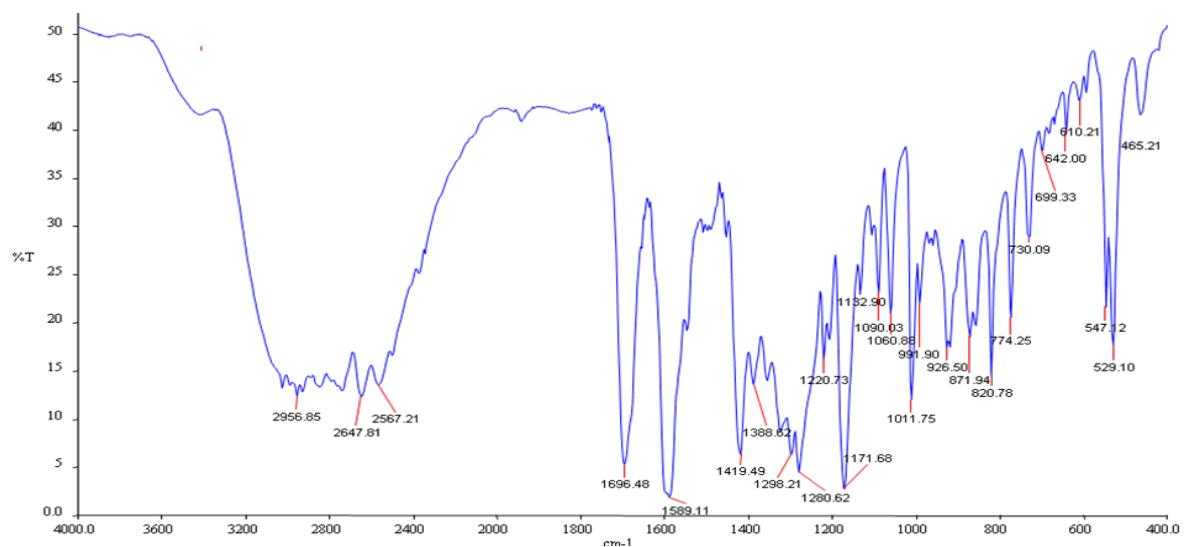


Figure S7: IR spectrum of **10**

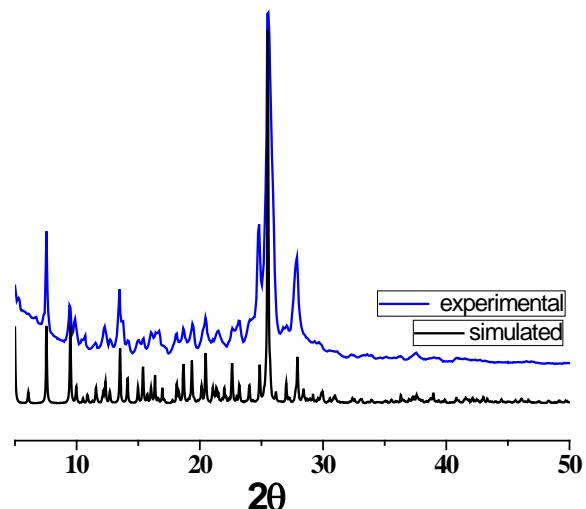


Figure S8: PXRD patterns of **4**.

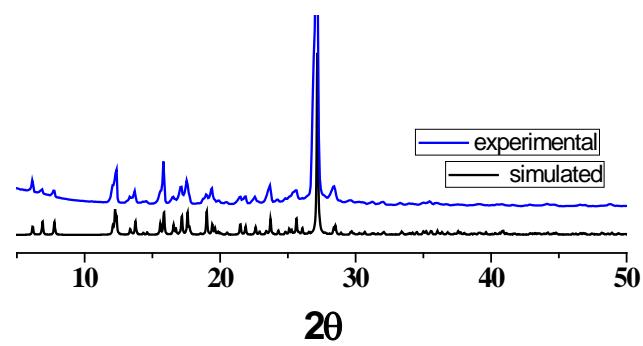


Figure S9: PXRD patterns of **5**.

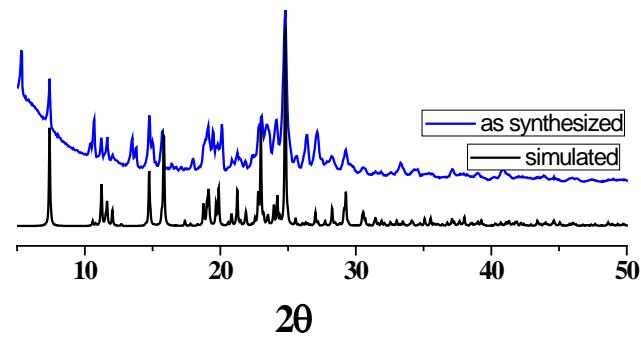


Figure S10: PXRD patterns of **6**.

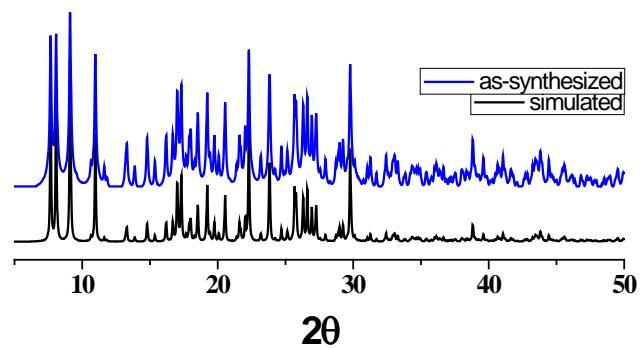


Figure S11: PXRD patterns of **7**.

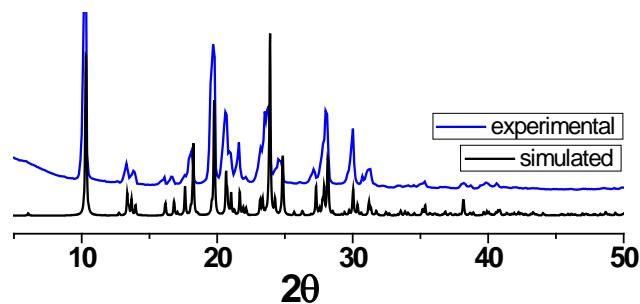


Figure S12: PXRD patterns of **8**.

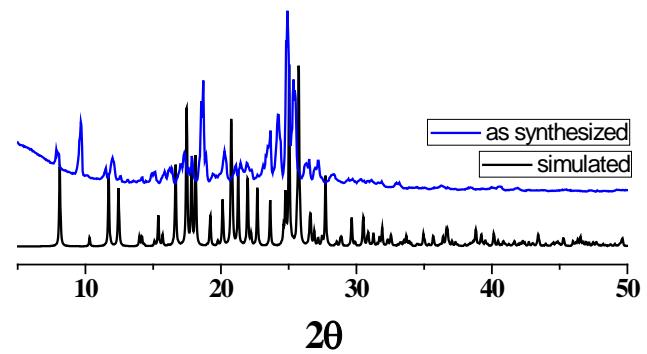


Figure S13: PXRD patterns of **9**.

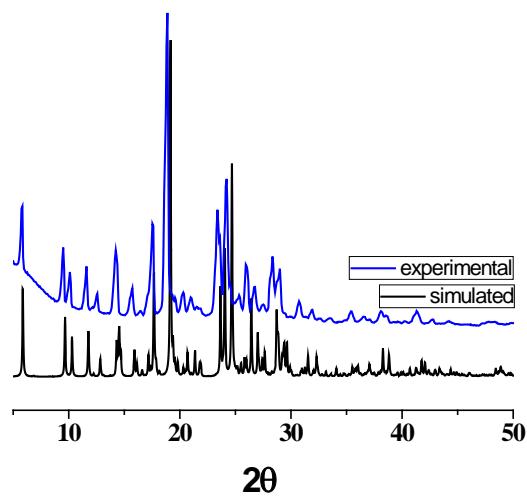


Figure S14: PXRD patterns of **10**.

Table S1. Hydrogen Bonding Parameters in the Crystal Structures of 4-10

Complexes	Interaction	H···A (Å)	D···A (Å)	D-H···A (°)
4	O-H···O	1.80	2.580(7)	159
		2.06	2.623(11)	126
		1.90	2.640(13)	149
		1.85	2.627(5)	158
	O-H···N	1.85	2.624(7)	158
		1.82	2.616(7)	162
	N-H···N	2.40	3.120(7)	138
	N-H···O	2.53	3.202(7)	133
		1.83	2.707(6)	170
		1.95	2.766(6)	151
		2.38	3.028(6)	129
		1.80	2.691(6)	174
		2.55	3.188(6)	129
	C-H···O	2.22	3.127(8)	164
		2.19	3.094(9)	164
		2.27	3.076(7)	144
		2.55	3.418(14)	155
		2.46	3.358(8)	162
		2.44	3.282(7)	146
5	O-H···N	1.74	2.561(3)	176
		1.87	2.690(3)	180
	N-H···O	1.97	2.795(3)	154
		2.02	2.868(3)	158
		2.43	2.899(3)	113
	O-H···O	1.74	2.555(3)	172
	C-H···O	2.29	3.171(3)	159
		2.41	2.748(3)	101
		2.36	2.754(3)	105
		2.54	3.368(3)	149
	C-H···N	2.36	2.754(3)	106
		2.58	3.461(4)	158
		2.62	3.363(4)	137

Table-S1 continued

6	O-H···N	1.80	2.615(3)	170
	O-H···O	1.93	2.742(3)	170
		2.04	2.852(4)	171
	N-H···O	2.50(3)	2.911(3)	107.3(19)
		2.01(3)	2.876(3)	155(2)
	N-H···N	1.84(3)	2.771(3)	167(2)
	C-H···O	2.48	3.373(3)	161
		2.33	2.727(3)	106
		2.38	3.237(3)	153
		2.44	3.364(3)	175
		2.50	3.471(4)	175
		2.54	2.900(3)	103
7	O-H···O	1.24	2.471(2)	180
		1.94	2.735(5)	162
	O-H···N	1.46(2)	2.605(3)	170(3)
	N-H···O	1.77(2)	2.679(2)	162(3)
	C-H···N	2.45	3.348(4)	153
		2.51	3.302(3)	138
		2.57	3.359(5)	138
		2.52	3.468(2)	164
		2.58	3.447(3)	156
		2.40	3.267(3)	156
		2.37	2.737(3)	103
		2.45	3.348(3)	153
		2.33	2.729(3)	106
		2.38	3.241(3)	155
8	O-H···N	1.85(2)	2.869(3)	176.8(11)
		1.85(2)	2.776(3)	173.3(16)
	C-H···N	2.60	3.323(3)	135
	C-H···O	2.33	2.735(3)	106
		2.33	2.733(2)	106
9	O-H···N	1.93	2.748(2)	176
		2.00	2.8155(19)	174
		2.43	3.362(2)	164
	C-H···O	2.34	2.736(2)	105
		2.33	2.733(2)	106
		1.87	2.6917(18)	176
		1.80	2.6212(18)	176
	O-H···N	2.03	2.846(2)	177
		2.02	2.840(2)	173
		1.88	2.687(2)	170
		1.92	2.733(2)	170
		2.41	3.337(2)	161
10	C-H···O	2.53	3.167(2)	123
		2.51	3.360(2)	146
		2.32	2.734(2)	107
		2.35	2.741(2)	105
		2.54	3.239(2)	130
		2.32	2.728(2)	106

Table S2: pK_a and ΔpK_a values

Ketones	Co-formers	pK_a		ΔpK_a	Complexes	Prediction	Complexes obtained
		$pK_a(BH^+)$	$pK_a(A)$				
1	H ₂ TMA	6.36	3.14	+3.22	4	salt	salt
	H ₂ PDC		3.38	+2.98	5	salt	salt
	H ₂ SIP		-2.88	+9.24	6	salt	salt
3	H ₂ TMA	3.93	3.14	+0.79	7	Partial transfer of proton	Salt, exhibited mixed ionic hydrogen bonds
	DHBA		3.61	+0.32	10	Partial transfer of proton	Co-crystal