

Supplementary Material

Bismuth(III) Thiophosphinates: Understanding How Small Atomic Change Influences Antibacterial Activity and Mammalian Cell Viability

Dimuthu C. Senevirathna,^A Rebekah N. Duffin,^{A,B} Liam J. Stephens,^A Megan E. Herdman,^A Melissa V. Werrett,^{A,C} and Philip C. Andrews^{A,C}

^ASchool of Chemistry, Monash University, Clayton, Vic. 3800, Australia.

^BCurrent address: Drug Delivery, Disposition and Dynamics, Monash Institute of Pharmaceutical Sciences, Parkville, Vic. 3052, Australia.

^CCorresponding authors. Email: melissa.werrett@monash.edu; phil.andrews@monash.edu

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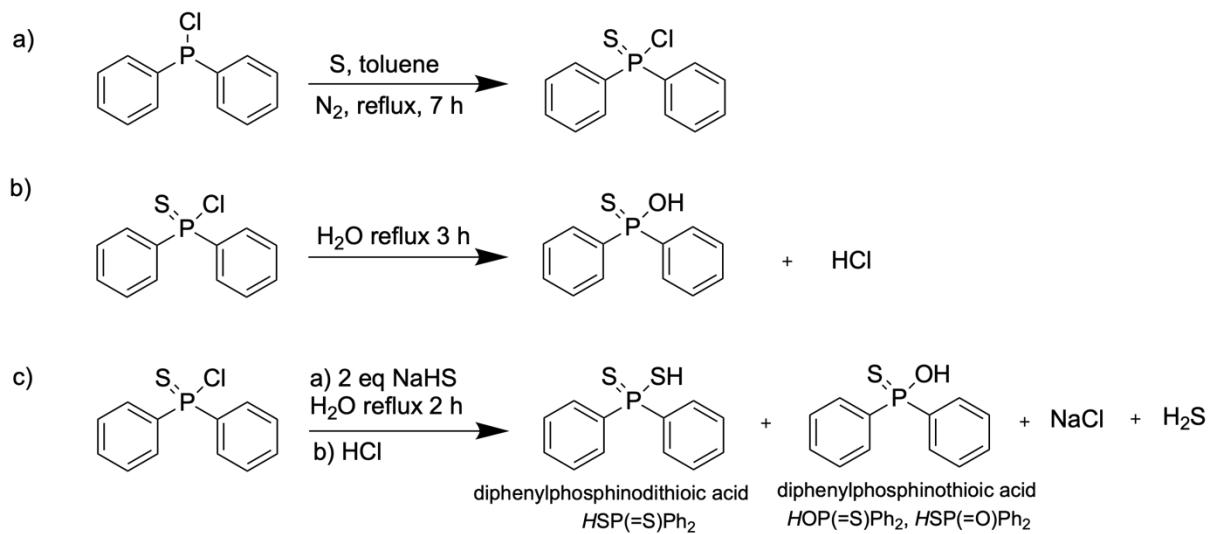
³¹P NMR of complexes **1-4**

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Scheme S1. Synthesis of diphenylphosphinothioic acid, $HOP(=S)Ph_2/HSP(=O)Ph_2$ and diphenylphosphinodithioic acid, $HSP(=S)Ph_2$.

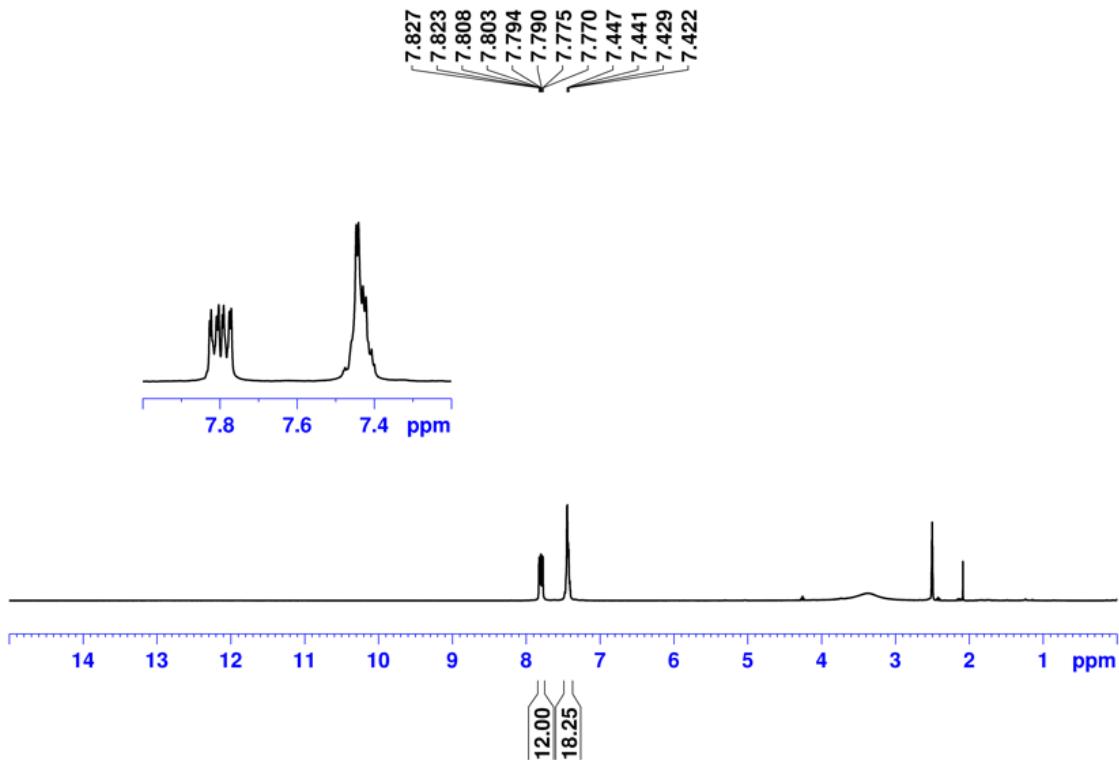


Fig. S1. ^1H NMR of $[\text{Bi}(\text{SP}(=\text{O})\text{Ph}_2)_3]$ (**1**) in DMSO-d_6 .

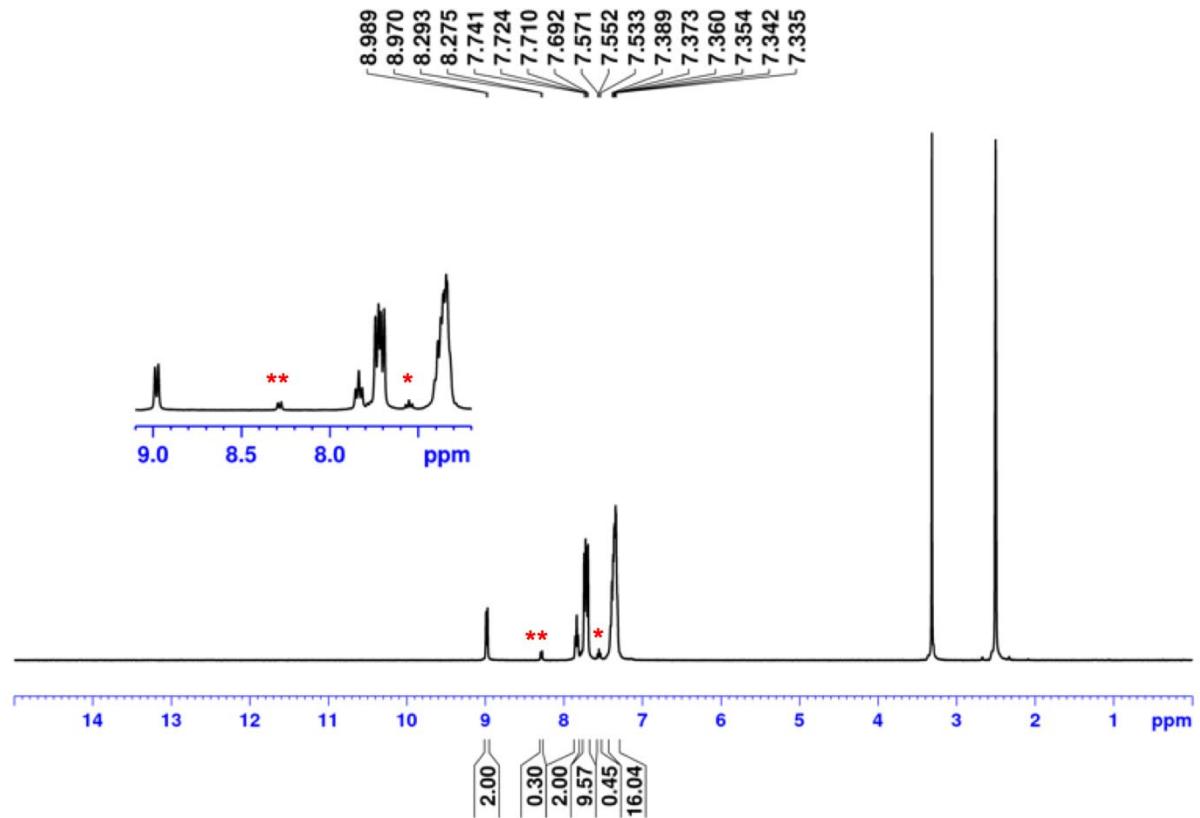


Fig. S2. ^1H NMR of $[\text{BiPh}(\text{SP}(=\text{O})\text{Ph}_2)_2]$, (**2**) in DMSO-d_6 . **ortho and *meta Bi-Ph protons from complex **3**.

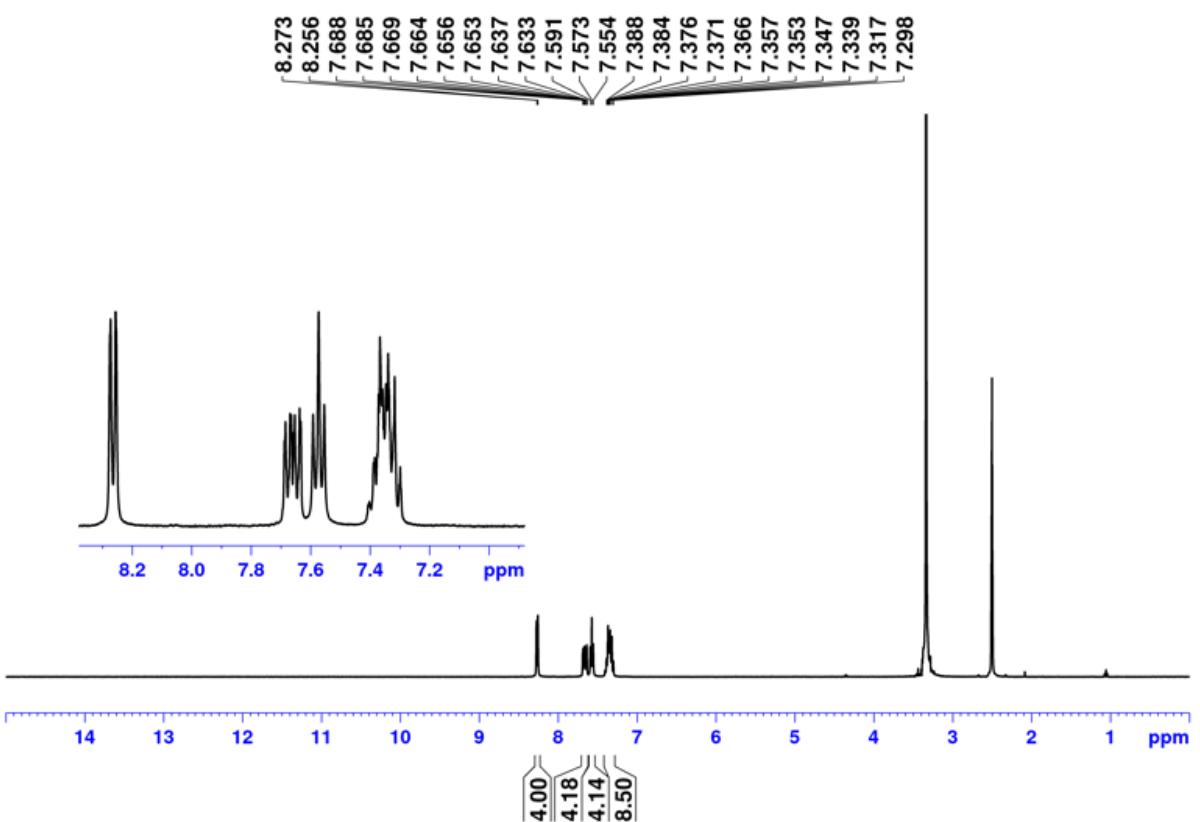


Fig. S3. ^1H NMR of $[\text{BiPh}_2(\text{SP}(=\text{O})\text{Ph}_2)]$ (**3**) in DMSO-d_6 .

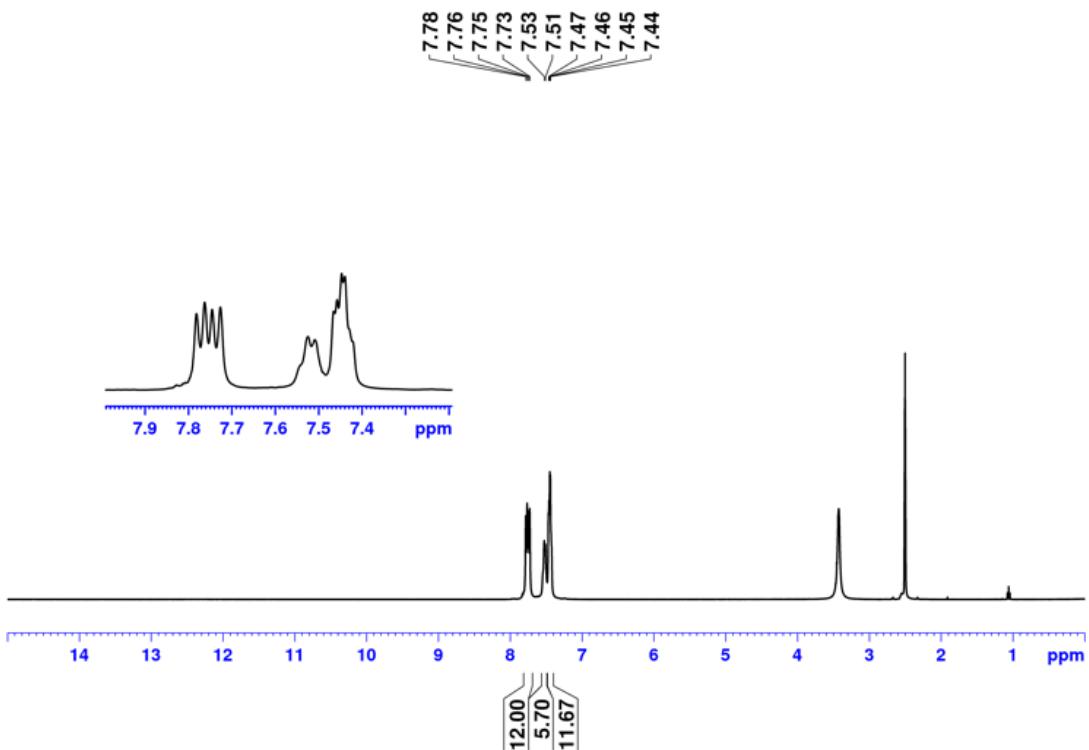


Fig. S4. ^1H NMR of $[\text{Bi}(\text{SP(=S)}\text{Ph}_2)_3]$ (**4**) in DMSO-d_6 .

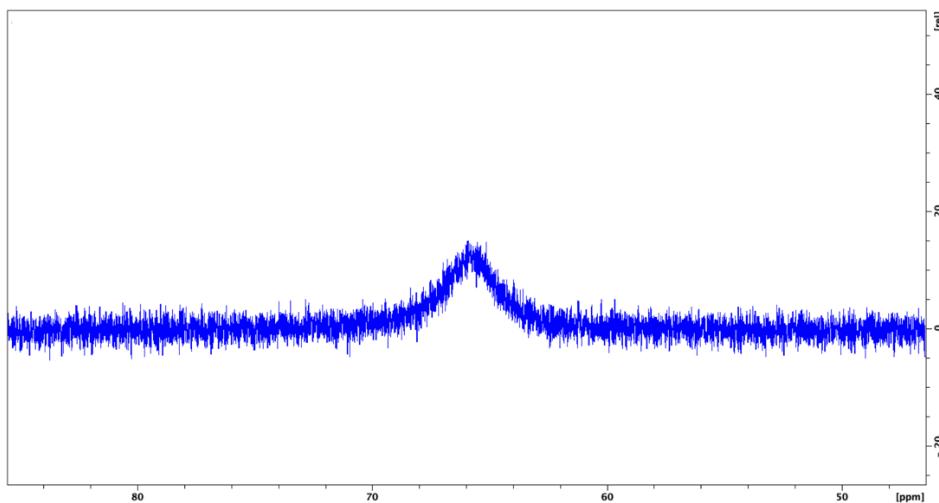


Fig. S5. ^{31}P NMR of $[\text{Bi}(\text{SP}(=\text{O})\text{Ph}_2)_3]$ (**1**) in DMSO-d_6 .

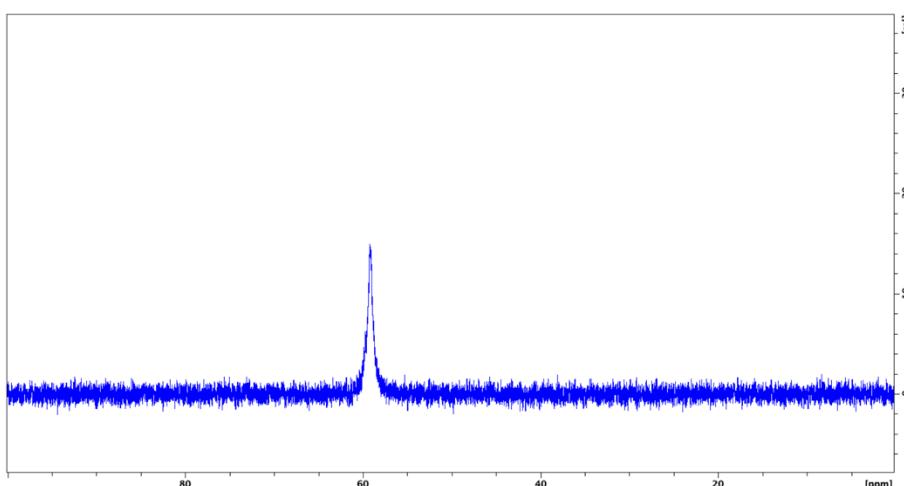


Fig. S6. ^{31}P NMR of $[\text{BiPh}(\text{SP}(=\text{O})\text{Ph}_2)_2]$ (**2**) in DMSO-d_6 .

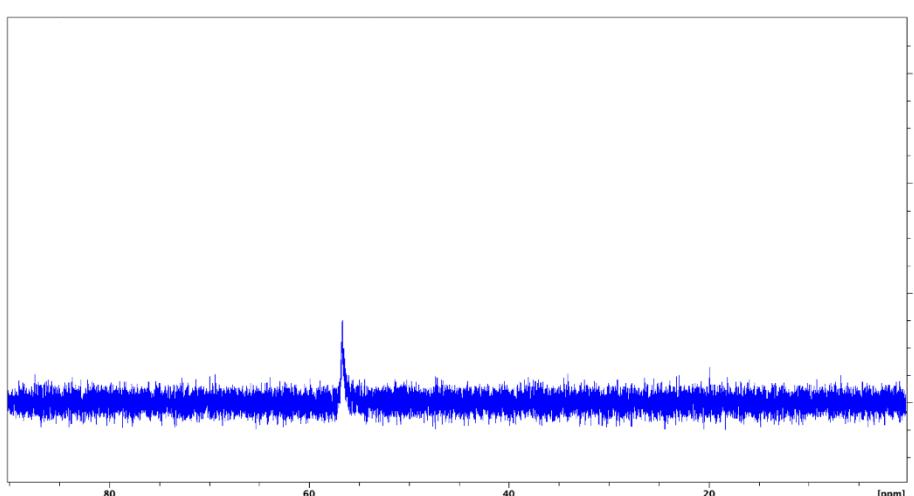


Fig. S7. ^{31}P NMR of $[\text{BiPh}_2(\text{SP}(=\text{O})\text{Ph}_2)]$ (**3**) in DMSO-d_6 .

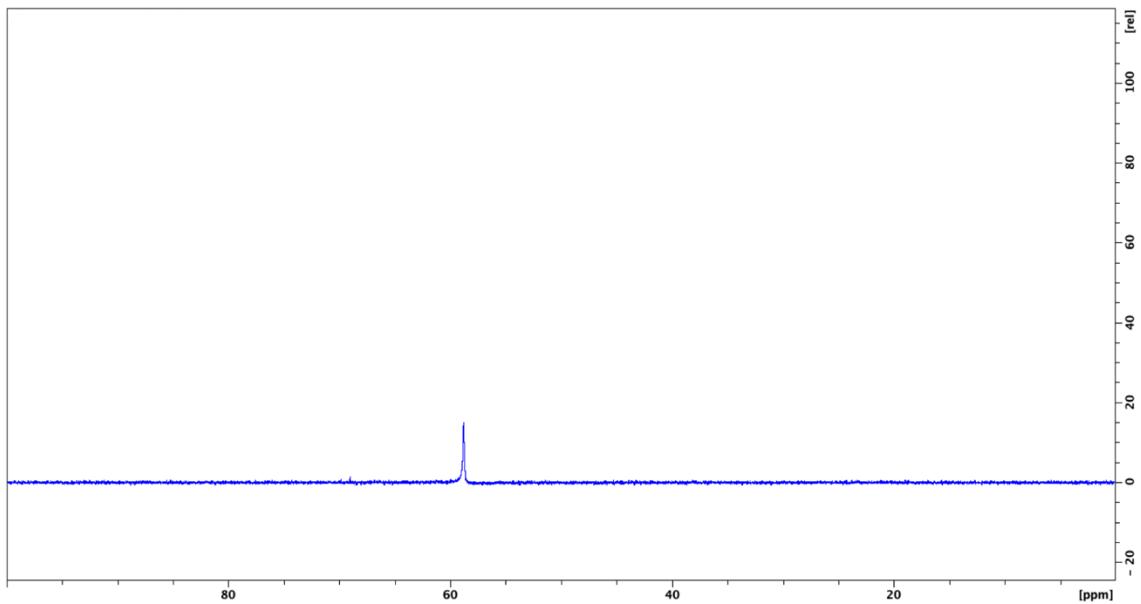


Fig. S8. ^{31}P NMR of $[\text{Bi}(\text{SP}(=\text{S})\text{Ph}_2)_3 \cdot \text{DMSO}]$ (**4**) in DMSO-d_6 .

Table S1. Crystallographic data and structure refinement for complexes **2**, **3** and **4**. CCDC numbers 1947001-1947003.

	2	3	4 ·DMSO
Formula	C ₃₀ H ₂₅ BiO ₂ P ₂ S ₂	C ₂₄ H ₂₀ BiOPS	C ₃₈ H ₃₆ BiOP ₃ S ₇
M _r	752.54	596.41	1034.98
Crystal size [mm]	0.23 x 0.16 x 0.09	0.21 x 0.17 x 0.10	0.24 x 0.16 x 0.10
Crystal system	Monoclinic	Monoclinic	Monoclinic
Space group	P2 ₁ /c	P2 ₁ /n	P2 ₁ /n
a [Å]	10.7908(4)	9.6557(4)	9.29900(10)
b [Å]	11.0293(4)	10.5962(4)	21.7360(2)
c [Å]	23.3960(6)	21.2320(9)	20.4159(2)
α [°]	90	90	90
β [°]	94.434(2)	93.243(3)	100.7700(10)
γ [°]	90	90	90
V [Å ³]	2776.14(16)	2168.85(15)	4053.84(7)
Z	4	4	4
T [K]	123(2)	123(2)	123(2)
ρ _{calcd.} [g cm ⁻¹]	1.801	1.827	1.696
μ [mm ⁻¹]	15.175	8.311	13.269
Independent reflections collected	5740	6685	8349
R _{int}	0.1301	0.0548	0.0473
R1 [<i>I</i> >2σ(<i>I</i>)]	0.0587	0.0261	0.0281
wR2 (all data)	0.1611	0.0486	0.0733
GoF	1.015	1.009	1.095

PXRD Study

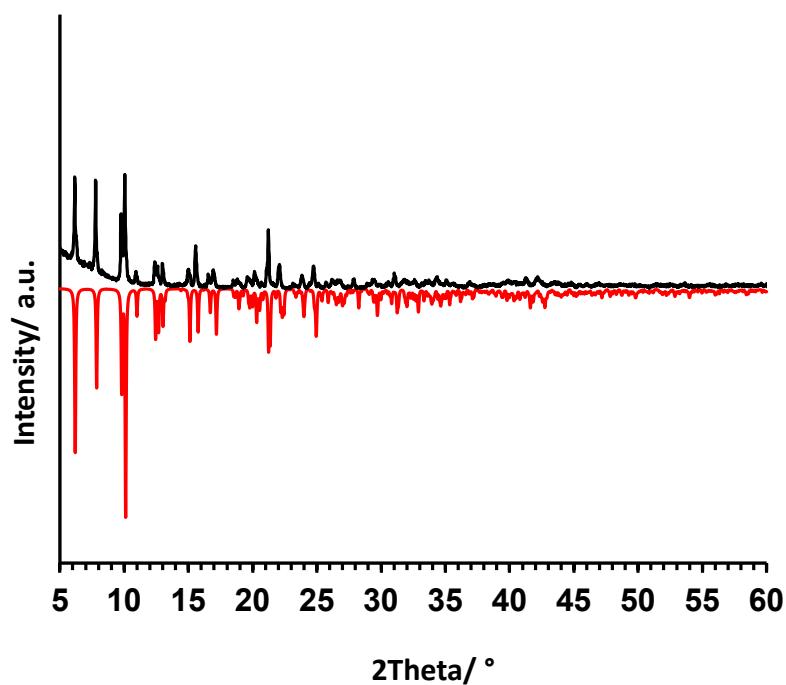


Fig. S9. Comparison of the powder X-ray diffraction pattern of bulk crystalline product (black, top) and the calculated pattern from the single crystal X-ray diffraction studies (red, bottom) of $[\text{BiPh}(\text{SP}(=\text{O})\text{Ph}_2)_2]$ (**2**).

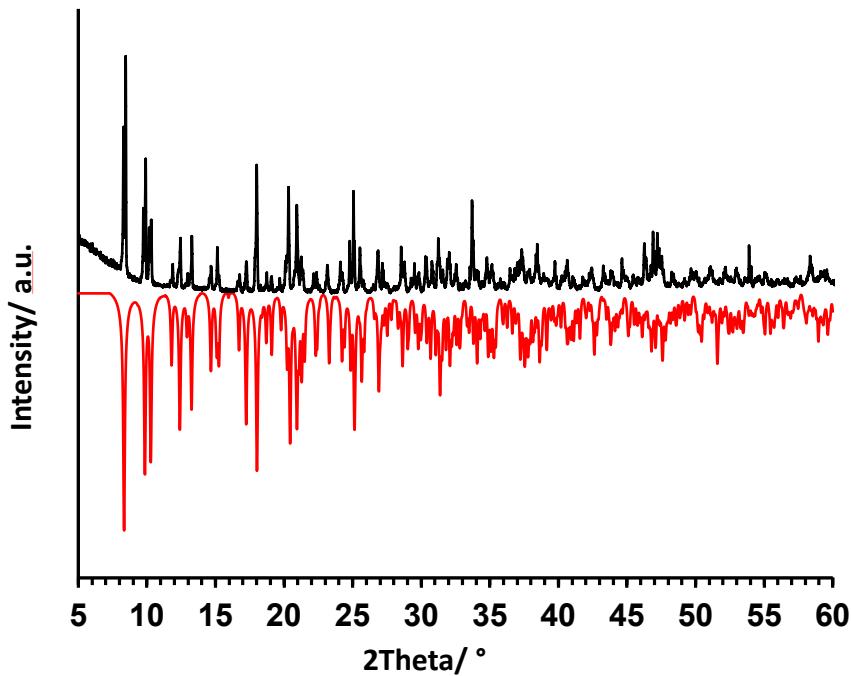


Fig. S10. Comparison of the powder X-ray diffraction pattern of bulk crystalline product (black, top) and the calculated pattern from the single crystal X-ray diffraction studies (red, bottom) of $[\text{BiPh}_2(\text{SP}(=\text{O})\text{Ph}_2)]$ (**3**).

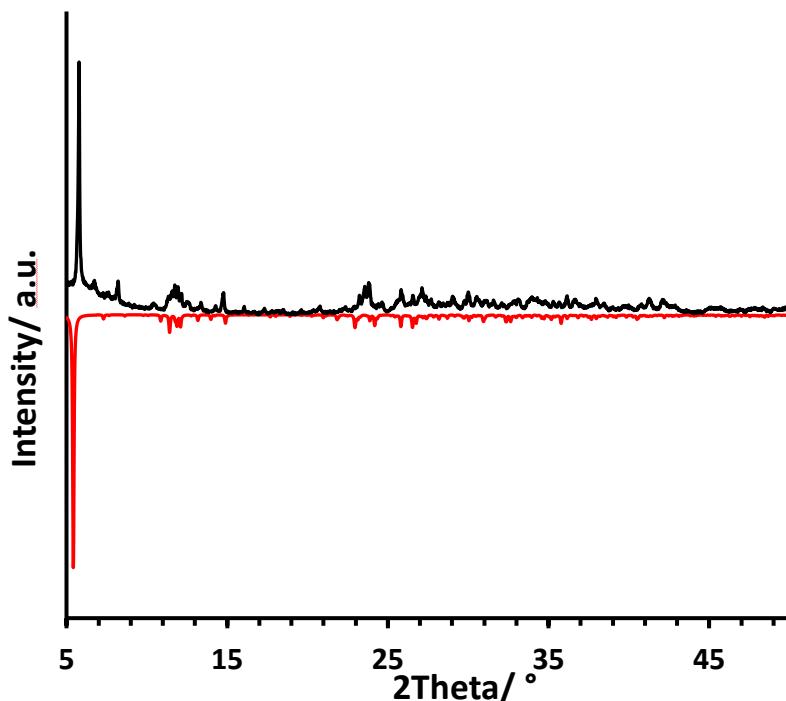


Fig. S11. Comparison of the powder X-ray diffraction pattern of bulk crystalline product (black, top) and the calculated pattern from the single crystal X-ray diffraction studies (red, bottom) of $[\text{Bi}(\text{SP}(=\text{S})\text{Ph}_2)_3 \cdot \text{DMSO}]$ (**4**).

Table S2: Selectivity indices of compounds **1 – 4**, where $\text{SI} = \text{IC}_{50} / \text{MIC}$. IC_{50} of complexes against human fibroblast cells.

Compound	MRSA	VRE	<i>E. coli</i>	<i>P. aeruginosa</i>
1 $[\text{Bi}(\text{SP}(=\text{O})\text{Ph}_2)_3]$	/	/	/	/
2 $[\text{BiPh}(\text{SP}(=\text{O})\text{Ph}_2)_2]$	9.2	9.2	2.3	1.1
3 $[\text{BiPh}_2(\text{SP}(=\text{O})\text{Ph}_2)]$	2.5	2.5	1.3	0.30
4 $[\text{Bi}(\text{SP}(=\text{S})\text{Ph}_2)_3]$	0.49	0.49	/	/