

## SUPPLEMENTARY MATERIAL

### **A Computational Comparative Study for the Spectroscopic Evaluation of Triazine Derivative Dyes in Implicit Solvation Model Systems Using Semi-Empirical and Time-Dependent Density Functional Theory Approaches**

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## Gaussian 16 reference

Gaussian 16, Revision C.01, M.J. Frisch, G.W. Trucks, H.B. Schlegel, G.E. Scuseria, M.A. Robb, J.R. Cheeseman, G. Scalmani, V. Barone, G.A. Petersson, H. Nakatsuji, X. Li, M. Caricato, A.V. Marenich, J. Bloino, B.G. Janesko, R. Gomperts, B. Mennucci, H.P. Hratchian, J.V. Ortiz, A.F. Izmaylov, J.L. Sonnenberg, D. Williams-Young, F. Ding, F. Lipparini, F. Egidi, J. Goings, B. Peng, A. Petrone, T. Henderson, D. Ranasinghe, V.G. Zakrzewski, J. Gao, N. Rega, G. Zheng, W. Liang, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, K. Throssell, J.A. Montgomery, Jr., J.E. Peralta, F. Ogliaro, M.J. Bearpark, J.J. Heyd, E.N. Brothers, K.N. Kudin, V.N. Staroverov, T.A. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A.P. Rendell, J.C. Burant, S.S. Iyengar, J. Tomasi, M. Cossi, J.M. Millam, M. Klene, C. Adamo, R. Cammi, J.W. Ochterski, R.L. Martin, K. Morokuma, O. Farkas, J.B. Foresman, and D.J. Fox, Gaussian, Inc., Wallingford CT, 2019.

## Parameters used for defining the solvent, 2-methyltetrahydrofuran (2-MeTHF) in Gaussian 16 software

Eps=6.97  
EpsInf=1.98  
HBondAcidity=-1.68  
HBondBasicity=1.34  
SurfaceTensionAtInterface=17.06  
CarbonAromaticity=0  
ElectronegativeHalogenicity=0

### Notes

**Eps** = Dielectric Constant ( $\epsilon$ ) [1]

**EpsInf** = The square of the index of refraction ( $n^2$ ) [2]

**HBond Acidity** = Abraham's hydrogen bond acidity ( $\alpha$ ) [3]

**HBond Basicity** = Abraham's hydrogen bond basicity ( $\beta$ ) [4]

**SurfaceTensionAtInterface** = Macroscopic surface tension at the liquid-air interface (298 K) (dyne/cm) ( $\gamma$ )<sup>5</sup>

**Carbon Aromaticity** = Aromaticity: fraction of non-hydrogenic solvent atoms that are aromatic carbon atoms ( $\Phi$ )

**Electronegative Halogenicity** = Electronegative halogenicity: fraction of non-hydrogenic solvent atoms that are F, Cl, or Br ( $\Psi$ )

Refer to <https://gaussian.com/smdtip/> for more information.

### References:

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