



# Natural products isolation studies of native Australian fern species

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#### **ABSTRACT**

Natural products isolation studies of 16 native Australian fern species have been undertaken, facilitated by pressurised hot water extraction (PHWE). Fourteen of these fern species have not been the subject of natural products isolation research previously. In total, 14 different compounds were isolated from 12 of these 16 different fern species. This included  $\gamma$ - and  $\delta$ -lactones; flavonoid glycosides, a dihydrobenzofuran neolignan, in addition to hydroxycinnamate/caffeic acid esters. More specifically, the lactones 5,6-dihydro-5-hydroxy-6-methyl-2H-pyran-2-one, 5-(1-hydroxyethyl)-2(5H)-furanone and osmundalin were obtained from Todea barbara, while a dihydrobenzofuran neolignan, (-)-trans-blechnic acid were found in Austroblechnum pennamarina subsp. alpina, and the shikimate ester 5-O-caffeoylshikimic acid was isolated from Parablechnum wattsii. In addition, flavonoids and their glycoside derivatives, kaempferol 3-O-glucopyranoside,  $4\beta$ -carboxymethyl-(-)-epicatechin, (2R)-eriodictyol-7-O- $\beta$ -D-glucopyranoside, naringin, quercitrin, quercetin 3-O-(6"-acetyl)- $\beta$ -D-glucopyranoside, rutin, and tiliroside were isolated from seven other fern species.

**Keywords:** ferns, flavonoid, glycoside, natural products, natural products isolation, neolignan, *Polystichum*, *Todea*.

#### Introduction

Ferns are a group of vascular plants bearing complex leaves called megaphylls. These plants do not produce flowers or seeds and reproduce via spores. There are more than 12 000 species of ferns that are widely distributed across the globe, with the greatest diversity typically found in the tropics.<sup>[1]</sup> Ferns represent the phylogenetic bridge between the lower and higher plants in the plant kingdom. For centuries, ferns have been used in many different contexts: as food, medicines, and ornaments. The fiddleheads (or croziers) of many fern species often feature in Asian cuisine. Indeed, it is reported that 52 species feature in Chinese food and it is estimated that the actual number of edible ferns may extend to 144 species. [2] In Japan, ostrich (Matteuccia struthiopteris (L.) Tod.), bracken (Pteridium aquilinum (L.) Kuhn), and royal ferns (Osmunda japonica Thunb.) are the most popular edible ferns that are harvested. [3] In Australia, the sporocarps of the small freshwater fern, nardoo (Marselia drummondii A. Braun) are consumed as baked cakes by first nations people following proper and extensive preparation.<sup>[4]</sup> In addition to their use in cuisine, many fern species feature in traditional pharmacopoeias and are used to treat an array of ailments. [5-8] In this context, relative to other species of vascular plants, ferns and lycophytes are poorly represented.[9]

Ferns and lycophytes, like angiosperms, are a rich source of phytochemicals with interesting biological properties. Natural products isolation studies reveal that they contain flavonoids, terpenoids (including steroids), and polyphenols (Fig. 1). [5,10] They also contain more distinctive alkaloid secondary metabolites. For example, lycopodium alkaloids such as lycopodine (1), lycodine (2), fawcettimine (3), and phlegmarine (4) skeletons have been isolated from Lycopodiaceae and Huperziaceae. [10–12] Flavonoids are commonly isolated from numerous fern species. For example, species of

Fig. 1. Examples of secondary metabolites isolated from ferns and lycophytes: alkaloids (1–4), complex flavonoids (5–28), sesquiterpenoids (29–32) and miscellaneous natural products (33–37); \*stereogenic centres were not assigned in these molecules

the genus *Pteris* are rich in flavonoids with mainly  $\alpha$ - and  $\beta$ -glucosides, galactosides, rhamnosides or arabinosides present. Distinctive flavonoids that have been isolated include neoflavonoids, calomelanols A–J (5–14) from

farinose of *Pityrogramma calomelanos* (L.) Link, <sup>[15,16]</sup> bioflavonoids such as hinokiflavone (**15**), 7″-O-methylhinokiflavone (**16**) amentoflavone (**17**) and 7,7″-di-O-methylamentoflavone (**18**) from *Selaginella tamariscina* 

(P.Beauv.) Spring, [17] involvenflavones A-F (19-24) from S. involven (Sw.) Spring<sup>[18]</sup> and prenylated flavonoids (25–28) from Helminthostachys zeylanica (L.) Hook. [19] Many sesquiterpenoid compounds with indane or cadinene skeletons are found in ferns.<sup>[10]</sup> Sesquiterpenyl indanones, known as pterosins, and their glycosides (pterosides) have been isolated from bracken fern species and polypodiaceous ferns. [20] For example, multifidosides A-C (29-31) have been isolated from *Pteris multifida* Poir. [14] and the carcinogenic pteroside, ptaquiloside (32) has been isolated from Pteridium aquilinium (L.) Kuhn. [21] The terpenoids obtained from ferns are typically *ent*-kaurane-, *ent*-atisane- and *ent*-primarane-type diterpenoids, which are present in Pteris species. [22] Labdaneand clerodane-type diterpenoids, diterpenoid glycosides and triterpenoids are major constituents in the Gleicheniaceae family, [23-26] and ecdysteroids have been isolated from species of genera Microsorum and Diploptervgium in the family Polypodiaceae. [27–31] Phenolic compounds are another class of secondary metabolites widely distributed in ferns. Commonly isolated molecules of this type include caffeic (33), chlorogenic (34) and vanillic (35) acids. [10,32,33] The glycosylated phenolic acid, 7-O-caffeoylhydroxymaltol-3-β-Dglucopyranoside (36) was isolated from Pteris ensiformis Burm., [33-35] in addition to a chalcone derivative, licoagrochalcone D (37) from Pteris multifida Poir. [36]

Natural products research concerning ferns native to Australia are mainly restricted to toxicity studies. For example, the sporocarps of the freshwater fern nardoo, Marsilea drummondii A. Braun, which are used for food by Australian Aborigines,<sup>[4]</sup> are reportedly toxic to humans, cattle, and sheep. [37] Studies on this waterfern has revealed that its toxicity derives from high levels of the enzyme thiaminase which breaks down thiamine (vitamin B1).[38] Similarly, bracken ferns Pteridium aquillinum (L.) Kuhn and P. esculentum (G.Forst.) Nakai have been the subject of many phytochemical and pharmacological studies in order to elucidate the mechanism of toxicity involving the carcinogenic norsesquiterpene glucoside, ptaquiloside (32). [39–42] Consequently, many sesquiterpenoid compounds have been isolated from P. aquillinum (L.) Kuhn. [20,43-45] The presence of ptaquiloside has been reported in fern species from the genera Pteris, Microlepia, Hypolepis and Chelianthes. [43,44,46] Beyond their toxicity, very limited information regarding the phytochemistry of ferns found in Australia exists. Nevertheless, due to their wide geographic distribution, many native Australian ferns are also found in Asia and South America and natural products isolation studies of species found in these locations have been undertaken. For example, Helminthostachys zeylanica (L.) Hook which is found widely distributed in tropical parts of Asia, the Pacific region, and Australia, [1] contains prenylated flavonoids, ugonins, [47,48] cyclised geranyl stilbenes, and ugonstiblenes. [49] Similarly, Salvinia species have a global distribution and the species, S. auriculata Aubl. and a hybrid, S.xmolesta D.S.Mitch. are found in Australia. [1] Bioactivity-guided

phytochemical investigation of these two Australian species allowed for the isolation of more than sixty different secondary metabolites that include diterpenes, polyphenols, fatty acids, triterpene, apocarotenoids, acyclic sesquiterpenoids, monoterpenes, jasmonates, steroids and coumarins. [50–52] However, a large number of native Australian ferns, particularly endemic species, have not been the subject of natural products isolation studies.

In this report, a total of 16 Australian native fern species formed the basis of natural products isolation studies. Specifically, our research concerned Todea barbara (L.) T. Moore, Alsophila australis (R.Br.) Domin, Dicksonia antarctica Labill., Calochlaena dubia (R.Br.) M. D. Turner & R. A. White, Polystichum proliferum (R.Br.) C. Presl, P. vestitum (G. Forst.) C. Presl, Pellaea falcata (R.Br.) Fée, Lecanopteris pustulata subsp. pustulata (G. Forst.) Testo & A. R. Field, Oceaniopteris cartilaginea (Sw.) Gasper & Salino, Lomaria nuda (Labill.) Willd., Doodia australis (Parris) Parris, Austroblechnum penna-marina (Poir.) Gasper & V. A. O. Dittrich subsp. alpina (R.Br.) S. Jess. & L. Lehm., Parablechnum wattsii (Tindale) Gasper & Salino, Gleichenia alpina R.Br., Histiopteris incisa (Thunb.) J.Sm. and Pteridium esculentum (G. Forst.) Nakai subsp. esculentum. Two of these species, Polystichum vestitum (G. Forst.) C. Presl and Gleichenia alpina R.Br., were previously thought to be endemic to Tasmania, [53] however, these species were recently found in New Zealand. [1] We isolated a total of 14 different compounds (38–51) from 12 native Australian fern species, including  $\gamma$ - and  $\delta$ -lactones; flavonoid glycosides, a dihydrobenzofuran neolignan, in addition to hydroxycinnamate/ caffeic acid esters (Fig. 2).

#### Results and discussion

For each of the 16 native Australian fern species that we investigated, PHWE of leaf material, followed by liquid–liquid extraction of the aqueous PHWE extract with ethyl acetate provided a crude organic extract after concentration under reduced pressure. In each case, the remaining aqueous phase was concentrated under reduced pressure to afford a crude aqueous extract. The combined yield (% w/w) of the respective crude extracts thus obtained is shown in Table 1. With the exception of *P. vestitum* (0.38% w/w), yields of crude extracts were >0.5% w/w in all cases. The crude extracts were then subjected to various standard flash column chromatography and preparative thin layer chromatography procedures.

Dicksonia antarctica and Pteridium esculentum are the only 2 of these 16 fern species that have been the subject of previous natural products isolation studies. Previously, the phenolic compounds (5S,6S,9S,10S)-15-hydroxycadina-3,11-dien-2-one and p-hydroxystyrene β-vicianoside, in addition to p-hydroxystyrene β-D-glucoside, kaempferol 3-O-β-D-glucoside, kaempferol 3-O-(2-O-β-D-xylosyl)-β-D-glucoside, kaempferol

Fig. 2. Secondary metabolites 38–51 isolated from 12 of the 16 native Australian fern species investigated in this study.

Table 1. Yields (% w/w) of the respective crude extracts obtained following PHWE of 16 native Australian fern species in this study.

Entry	Fern species	Leaf material (g)	Crude extract (mg)	Yield (% w/w)
1	Todea barbara	29	646	2.23
2	Alsophila australis	30	268	0.89
3	Calochalaena dubia	30	575	1.92
4	Polystichum vestitum	30	114	0.38
5	Polystichum proliferum	30	240	0.80
6	Pellaea falcata	30	1000	3.33
7	Austroblechnum penna-marina subsp. alpina	29	409	1.41
8	Lecanopteris pustulata subsp. pustulata	30	154	0.51
9	Lomaria nuda	30	254	0.85
10	Parablechnum wattsii	30	523	1.74
11	Gleichenia alpina	29	793	2.73
12	Dicksonia antarctica	30.5	298	0.98
13	Doodia australis	31	617	1.99
14	Oceaniopteris cartilaginea	33	1500	4.55
15	Histiopteris incisa	35	877	2.51
16	Pteridium esculentum	150	2700	1.80

3-O-(6-p-coumaroyl)-β-D-glucoside and chlorogenic acid have been isolated from P. esculentum fronds (1.2 kg of dry plant material).<sup>[54]</sup> 4-O-Caffeovlshikimic acid and 4-O-(pcoumaroyl) shikimic acid were isolated from croziers of D. antarctica (2.4 kg of fresh plant material) in 1997. [55,56] In this present study, secondary metabolites were not isolated from D. antarctica, Doodia australis, Oceaniopteris cartilaginea and Histiopteris species. However, 14 different compounds were isolated from the remaining 12 fern species we investigated. A combination of standard <sup>1</sup>H, <sup>13</sup>C and 2D (COSY, HMBC and HSOC) NMR spectroscopic techniques were employed to elucidate the structures of these compounds. In each case, these data were consistent with equivalent data reported in the literature. Specifically, we obtained 5,6-dihydro-5-hydroxy-6-methyl-2*H*-pyran-2-one (**38**), 5-(1-hydroxyethyl)-2(5H)-furanone (39), osmundalin (40), astragalin (41),  $4\beta$ -carboxymethyl-(-)-epicatechin (42), (2R)-eriodictyol-7- $O-\beta$ -D-glucopyranoside (43), naringin (44), (-)-trans-blechnic acid (45), (p-hydroxybenzyl)malonic acid (46), quercitrin (47), quercetin 3-O-(6"-acetyl-β-D-glucopyranoside) (48), 5-Ocaffeoylshikimic acid (49), rutin (50), and tiliroside (51) (Fig. 2).

5,6-Dihydro-5-hydroxy-6-methyl-2*H*-pyran-2-one (**38**) and 5-(1-hydroxyethyl)-2(5*H*)-furanone (**39**) were obtained as a mixture from *Todea barbara* in a ~2:1 ratio, as judged by NMR spectroscopic and GC-MS analysis (see Supplementary Material). Osmundalin (**40**) was also isolated from *T. barbara*. Lactones **38**–**40** have been isolated from *Osmunda japonica*, a common Japanese fern species, and are reported to exhibit antifeedant properties against the larvae of yellow butterfly, *Eurema hecabe mandarina*. [57,58] In addition, all three natural products have also been isolated from *Angiopteris caudatiformis*, a fern species used in Chinese folk medicine for the treatment of a broad range of ailments. [59] Natural products **38** and **39** have been isolated from the fern species *A. esculenta* [60] and angiopteroside, an epimer of osmundalin (**40**), was isolated from *Angiopteris evecta*. [61]

 $4\beta$ -Carboxymethyl-(-)-epicatechin (42), [62] was isolated from Polystichum vestitum. This secondary metabolite has been isolated from Davallia divaricata, [62] D. solida [63] Dryopteris crassirhizoma. [64] We also obtained molecule 42 from Polystichum proliferum and Lecanopteris pustulata subsp. pustulata. All of these fern species are members in the order Polypodiales. Dihydrobenzofuran neolignane, ( – )-transblechnic acid (45) was isolated from Austroblechnum pennamarina subsp. alpina; a species formerly classified within the genus *Blechnum*. (–)-*trans*-Blechnic acid (45) and its epimer, epiblechnic acid, represent characteristic constituents of the family Blechnaceae. [65] Blechnic acid has been isolated from various fern species, including Blechnopsis orientalis, Spicantopsis amabilis, S. niponica, Woodwardia orientalis, prolifera, Brainea insignis<sup>[65]</sup> and Struthiopteris spicant. [66] We isolated 5-O-caffeoylshikimic acid (49) (0.5% w/w yield) from Parablechnum wattsii (also formerly in the genus Blechnum). Compound 49 is a major secondary

metabolite present in this fern and a known enzymatic browning agent present in dates, *Phoenix dactylifera*. [55,67] It is a major phytochemical and an anti-thiamine factor isolated from Pteridium aquillinum var. latiusculum and reportedly causes depression of leucocytes and thrombocytes in calves. [68] 5-O-Caffeoylshikimic acid is found widely distributed in Equisetaceae family and in ferns from the families Adiantaceae, Dryopteridaceae, Athyriaceae, Dennstaedtiaceae, Osmundaceae and Thelypteridaceae. [69] Interestingly, we isolated p-hydroxybenzylmalonic acid (46) and  $4\beta$ -carboxymethyl-(–)-epicatechin (42) from Lecanopteris pustulata subsp. pustulata. Natural product 46 has been isolated from liquorice previously.<sup>[70]</sup> Liquorice primarily derives from three species, Glycyrrhiza glabra, Glycyrrhiza uralensis and Glycyrrhiza inflata and the presence of p-hydroxybenzylmalonic acid (46) has been reported from all three. [71]

Flavonoid glycosides 41, 43, 44, 47, 48, 50, and 51 were also isolated in our study. Specifically, we obtained the common flavonoid glucoside astragalin (41) from Alsophila australis, Calochalaena dubia and Pteridium esculentum. It has been isolated from many plant species including from the bracken fern P. aquilinum. [72-76] We isolated (2R)eriodictyol-7-*O-β*-D-glucopyranoside (**43**) from *P. vestitum*. This natural product has been isolated from a wide range of flowering plants. [77-85] In ferns, its presence has been identified in species of *Pyrrosia*. [86] Molecule **43** is a reported Nrf2 activator and confers protection against cisplatin-induced toxicity and cerebral ischemic injury. [84,87] Naringin (44) and rutin (50) were isolated from Pellaea falcata and Gleichenia alpina, respectively. Both are commonly reported flavonoid glycosides found in citrus, and exhibit a broad range of pharmacological activity. [88–95] Compounds **44** and **50** have also been found in fern species. [96-99] Trace amounts of quercitrin (47) and quercetin 3-O-(6"-acetyl)-glucoside (48) were isolated from Lomaria nuda. Both molecules are present in an array of plant species and possess wide ranging biological properties. Tiliroside (51), a kaempferol flavonoid glucoside with a coumaroyl moiety, was isolated from Pteridium esculentum. Its presence has been identified in various plant species, including P. aquilinum; and compound 51 is a reported carcinogen found in bracken fern. [100-103] However, tiliroside also exhibits profound anti-hyperglycemic, anti-hyperlipidemic and antioxidant effects, and has potential therapeutic applications for the treatment of diabetes. [100,104–106]

Among the 14 secondary metabolites isolated in this study, osmundalin (40),  $4\beta$ -carboxymethyl-(-)-epicatechin (42) and *trans*-blechnic acid (45) have only been isolated from fern sources to date (Table 2). This reveals that various Australian fern species investigated in our study contain secondary metabolites that are consistent with other members in the genera, families or orders that are found distributed beyond Australia. For example, *Todea barbara* is exclusively a southern hemisphere species and both *Osmunda japonica* and *Angiopteris caudatiformis* are species found in eastern Asia

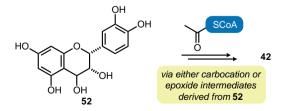
Table 2. Distribution of isolated secondary metabolites 38-51 in ferns species.

Natural product	Ferns	Exclusive to fern species
38: osmundalactone	Todea barbara; Osmunda japonica	_
	Angiopteris caudatiformis	
<b>39</b> : 5-(I-hydroxyethyl)-2(5 <i>H</i> )-furanone	Todea barbara; Osmunda japonica	-
	Angiopteris caudatiformis	
40: osmundalin	Todea barbara; Osmunda japonica	Yes
	Angiopteris caudatiformis	
41: astragalin	Alsophila australis; Calochalaena dubia	-
	Pteridium esculentum; Pteridium aquilinum	
<b>42</b> : 4β-carboxymethyl-(–)-epicatechin	Davallia divaricate; Davallia solida	Yes
	Dryopteris crassirhizoma	
	Polystichum vestitum; Polystichum proliferum	
	Lecanopteris pustulata subsp. pustulata	
<b>43</b> : $(2R)$ -eriodictyol-7- $O$ - $\beta$ -D-glucopyranoside	Polystichum vestitum; Pyrrosia (genus)	-
44: naringin	Pellaea falcata; Elaphoglossum spathulatum	-
	Ceterach officinarum; Drynaria fortune	
45: trans-blechnic acid	Austroblechnum penna-marina subsp. alþina	Yes
	Blechnopsis orientalis; Spicantopsis amabilis	
	Spicantopsis niponica; Struthiopteris spicant	
	Woodwardia orientalis	
	Woodwardia prolifera; Brainea insignis	
<b>46</b> : <i>p</i> -hydroxybenzylmalonic acid	Lecanopteris pustulata subsp. pustulata Diplazium esculentum	-
47: quecertin-3-O-(6'-O-acetyl)glucoside	Lomaria nuda	-
48: quercitrin	Lomaria nuda	-
49: 5-O-caffeoylshikimic acid	Parablechnum wattsii; Phoenix dactylifera	_
	Dicksonia antarctica	
	Pteridium aquilinum var. latiusculum	
50: rutin	Gleichenia alpina	-
	Sphaerostephanos arbusculus	
51: tiliroside	Pteridium esculentum; Pteridium aquilinum	_

regions. *trans*-Blechnic acid (45) has only been obtained from fern species of the family Blechnaceae. The remaining natural products are regularly isolated from angiosperms, including various fern species.  $4\beta$ -Carboxymethyl-(-)-epicatechin (42) is a particularly rare natural product and its isolation solely from ferns suggests that its formation might derive from a biosynthetic pathway unique to certain fern species (Scheme 1).

# **Conclusions**

Our natural products isolation studies provide further evidence that fern species represent a source of structurally



**Scheme 1.** Overview of possible biosynthetic pathway leading to  $4\beta$ -carboxymethyl-(-)-epicatechin (42).

diverse phytochemicals with interesting biological properties. Phytochemical screening of 16 native Australian ferns enabled the isolation of 14 previously reported compounds.

The structures of these compounds were primarily elucidated via 1D ( $^{1}$ H and  $^{13}$ C) and 2D (COSY, HSQC and HMBC) NMR spectroscopy. In each case, the characterisation data were consistent with equivalent data reported in the literature. These isolated compounds included flavonoid glycosides, caffeic acid esters, lactones, and a dihydrobenzofuran neolignan. Lactone molecules, 5,6-dihydro-5-hydroxy-6-methyl-2*H*-pyran-2-one, 5-(1-hydroxyethyl)-2(5*H*)-furanone and osmundalin were isolated from *Todea barbara*. 4 $\beta$ -Carboxymethyl-(-)-epicatechin was isolated from both species of *Polystichum* investigated in this study. Our isolation of (-)-*trans*-blechnic acid from *Austroblechnum penna-marina* subsp. *alpina* is consistent with this natural product representing a characteristic compound in the family Blechnaceae.

# **Experimental**

#### Plant material

Leaf material from 13 ferns Todea barbara, Alsophila australis, Dicksonia antarctica, Calochlaena dubia, Polystichum proliferum, P. vestitum, Pellaea falcata, Lecanopteris pustulata subsp. pustulata, Oceaniopteris cartilaginea, Lomaria nuda, Doodia australis, Austroblechnum penna-marina subp. alpina and Parablechnum wattsii were collected from the Royal Botanical Gardens of Tasmania in Hobart during February 2020. Aerial parts of Gleichenia alpina was collected from Wombat Moor at Mt Field National Park (42.6829°S, 146.6174°E; 1073 m above sea level) in February 2021. Histiopteris incisa was collected from Mt Field National Park (42.7806°S, 146.5844°E; 461 m above sea level) in February 2021. Aerial parts of *Pteridium esculentum* were collected from five healthy plants growing at Marion Bay (42.8218°S, 147.8671°E) in October 2021. Voucher specimens have been provided to the Tasmanian Herbarium, Tasmanian Museum and Art Gallery (no. HO607603-HO607618). With the exception of *Pteridium esculentum*, all plant material was air-dried for 2 weeks and stored prior to extraction. P. esculentum plant material was dried in an oven at 45°C for 3 days prior to extraction.

# **General**

Solvents used in all experiments were of analytical grade or purified by standard laboratory procedures. Plant material was ground using a Sunbeam spice grinder. Pressurised hot water extraction (PHWE) was performed using Breville Expresso Machine Model 800ES. This PHWE method is a well-established natural products extraction technique. [107–109] The extracted organic solvents were dried using anhydrous MgSO<sub>4</sub> and Na<sub>2</sub>SO<sub>4</sub>. Solvents were removed under reduced pressure on a rotary evaporator. Flash column chromatography was performed using flash grade silica gel (Kieselgel 60). Automated flash chromatography was performed using a Grave Reveleris X2 flash column chromatography system or

a Büchi Flash Pure system with 40 µm silica gel cartridges. TLC analysis was performed using Merck silica gel 60-F<sub>254</sub> plates. NMR spectroscopy was performed on a Bruker Avance III NMR spectrometer operating at 400 MHz (1H) and 100 MHz (<sup>13</sup>C) or Bruker Ascend<sup>TM</sup> 600 NMR spectrometer operating at 600 MHz for (<sup>1</sup>H) and 150 MHz (<sup>13</sup>C). The deuterated solvents used were D<sub>2</sub>O, CDCl<sub>3</sub>, acetone-d<sub>6</sub>, CD<sub>3</sub>OD and DMSO- $d_6$ . Spectra were calibrated by assignment of the residual solvent peak to  $\delta_H$  7.26 and  $\delta_C$  77.16 for CDCl<sub>3</sub>;  $\delta_{\rm H}$  2.50 and  $\delta_{\rm C}$  39.52 for DMSO- $d_6$ ;  $\delta_{\rm H}$  3.31 and  $\delta_{\rm C}$  49.00 for CD<sub>3</sub>OD;  $\delta_{\rm H}$  4.79 for D<sub>2</sub>O; and  $\delta_{\rm H}$  2.05 and  $\delta_{\rm C}$  29.84 for acetone-d<sub>6</sub>. [110] Infrared spectroscopy was performed using a Shimadzu FTIR 8400s spectrometer, with samples prepared as thin films on NaCl plates. Gas chromatography mass spectrometry (GC-MS) experiments were performed on Agilent 6850 GC and Agilent 5975C mass spectrometers. HRESIMS analyses were conducted on a Thermo LTQ Orbitrap XL mass spectrometer (Thermo Fisher Scientific).

#### **Extraction and isolation**

#### PHWE of Todea barbara

T. barbara dried leaflets (29 g) were finely ground using a spice grinder, mixed with sand (~4g), and extracted via PHWE (35% EtOH/H<sub>2</sub>O). This provided a hot extract (200 mL) that was cooled immediately in an ice bath. This process was repeated a further two times to provide a combined extract (600 mL). The extract was concentrated under reduced pressure (35°C) to remove EtOH and extracted with EtOAc (150 mL and  $2 \times 100$  mL). The combined organic phase was dried (MgSO<sub>4</sub>), filtered, and concentrated under reduced pressure to provide extract A (646 mg). The remaining aqueous phase was concentrated under reduced pressure (50°C) to yield a dark brown extract B (10 g). Extract A: Extract A (646 mg) was redissolved in EtOAc, adsorbed onto silica/Celite® (1:1 mixture by mass), and subjected to flash column chromatography (silica; hexanes (50 mL), 10% acetone/hexanes (50 mL), 20% acetone/hexanes (100 mL), 30% acetone/hexanes (150 mL), 40% acetone/hexanes (100 mL), 50% acetone/hexanes (50 mL), 60% acetone/hexanes (50 mL) and acetone (150 mL)} to provide a mixture of compounds 38 and 39 (23 mg, 0.8% w/w yield in a  $\sim$ 2:1 ratio) as colourless crystalline solids. This mixture was analysed by GC-MS. Extract B: Extract B (10 g) was soaked in MeOH (~200 mL) for 1 h, repeated five times, combined, and concentrated to provide extract B.1 (4g). Extract B.1 was then soaked in acetone (~200 mL) for 1 h, repeated 5 times, combined, and concentrated to provide extract B.2  $(\sim 2 g)$ . Extract B.2  $(\sim 1 g)$  was redissolved in acetone, absorbed onto silica/Celite® (1:1 mixture by mass), and subjected to flash column chromatography {silica; hexanes (50 mL), 10% acetone/hexanes (100 mL), 50% acetone/ hexanes (50 mL), acetone (50 mL) and 50% MeOH/acetone (50 mL). Following TLC and <sup>1</sup>H NMR spectroscopic analysis, the resulting fractions were combined to afford six fractions,

F.1-6. Fraction F.5 (709 mg) was again adsorbed onto silica/Celite® (1:1 mixture by mass) and subjected to flash column chromatography {silica; hexanes (50 mL), 10% acetone/hexanes (50 mL), 30% acetone/hexanes (200 mL), 50% acetone/hexanes (200 mL), and acetone (30 mL)}, which provided compound 40 (570 mg, 2.0% w/w yield) as an off-white solid. Extract B.1 (1.5 g) was absorbed onto silica/ Celite<sup>®</sup> (1:1 mixture by mass) and subjected to flash column chromatography {silica; CH<sub>2</sub>Cl<sub>2</sub> (75 mL), 13% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (150 mL), 27% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (75 mL), 53% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (150 mL) and MeOH (100 mL)}. Following TLC and <sup>1</sup>H NMR spectroscopic analysis the resulting fractions were combined to afford four fractions, F.1-4. Fraction F.3 (641 mg) was redissolved in MeOH, absorbed onto silica/Celite® (1:1 mixture by mass) and subjected to flash column chromatography {silica; CH<sub>2</sub>Cl<sub>2</sub> (50 mL), 25% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (100 mL), 66% MeOH/ CH<sub>2</sub>Cl<sub>2</sub> (75 mL) and MeOH (50 mL)}, which provided compound 40 (30 mg, 2.0% w/w yield) as an off-white solid.

Osmundalactone (38). [57] Colourless crystalline solid.  $^{1}$ H NMR (600 MHz, CDCl<sub>3</sub>)  $\delta$  6.78 (1H, dd, J = 9.9 and 2.3 Hz; H-4), 5.92 (1H, dd, J = 9.9 and 1.9 Hz; H-3), 4.31 (1H, m; H-6), 4.18 (1H, d, J = 8.7 Hz; H-5), 1.42 (3H, d, J = 6.4 Hz; 7-CH<sub>3</sub>) ppm;  $^{13}$ C NMR (150 MHz, CDCl<sub>3</sub>)  $\delta$  163.2 (C-2), 148.5 (C-4), 120.7 (C-3), 79.0 (C-6), 67.7 (C-5), 18.2 (7-CH<sub>3</sub>) ppm.

5-(1-Hydroxyethyl)-2(5H)-furanone (39). [57,58] Colourless crystalline solid.  $^1$ H NMR (600 MHz, CDCl $_3$ ) δ 4.35 (1H, m; H-5), 4.08 (1H, m; H-6), 2.53 (1H, m; H-3a), 2.48 (1H, m; H-3b), 2.19 (1H, m; H-4a), 2.12 (1H, m; H-4b), 1.13 (3H, d, J = 6.5 Hz; 7-CH $_3$ ) ppm;  $^{13}$ C NMR (150 MHz, CDCl $_3$ ) δ 177.5 (C-2), 83.5 (C-5), 67.4 (C-6), 28.6 (C-3), 20.9 (C-4), 17.7 (7-CH $_3$ ) ppm.

Osmundalin (40) (CAS# 54835-71-1). [57] Off-white solid. [α] -65.1° (c 0.075, MeOH), lit. [α] -107° (c 1.0, MeOH). [58] <sup>1</sup>H NMR 600 MHz, CD<sub>3</sub>OD) δ 7.04 (1H, dd, J = 9.9 and 2.9 Hz; H-3), 5.98 (1H, dd, J = 9.9 and 1.5 Hz; H-2), 4.48 (1H, quintet, J = 7.4 Hz; H-5), 4.45 (1H, d, J = 7.8 Hz; H-1'), 4.43 (1H, ddd, J = 7.6, 2.8 and 1.5 Hz; H-4), 3.85 (1H, dd, J = 11.8 and 5.6 Hz; H-6'b), 3.79 (1H, dd, J = 11.8 and 1.9 Hz; H-6'a), 3.33 (1H, t, J = 8.9 Hz; H-3'), 3.26 (2H, m; H-4',5'), 3.16 (1H, t, J = 9.1 Hz; H-2'), 1.42 (3H, d, J = 6.5 Hz; 6-CH<sub>3</sub>) ppm; <sup>13</sup>C NMR (150 MHz, CD<sub>3</sub>OD) δ 165.14 (C-1), 147.71 (C-3), 121.55 (C-2), 102.81 (C-1'), 79.29 (C-5), 78.16 (C-5'), 77.95 (C-3'), 74.82 (C-2'), 73.36 (C-4), 71.49 (C-4'), 62.72 (C-6'), 18.57 (C-6) ppm.

### PHWE of Alsophila australis

A. australis dried leaflets (30 g) were finely ground using a spice grinder, mixed with sand ( $\sim$ 6 g), and extracted via PHWE (35% EtOH/H<sub>2</sub>O). This provided a hot extract (200 mL) that was cooled immediately in an ice bath. This process was repeated a further two times to provide a combined extract (600 mL). The extract was concentrated under reduced pressure (35°C) to remove EtOH and extracted

with EtOAc ( $2 \times 200 \, \text{mL}$  and 150 mL). The combined organic phase was dried (MgSO<sub>4</sub>), filtered, and concentrated under reduced pressure to provide extract A (268 mg). The remaining aqueous phase was concentrated under reduced pressure (50°C) to yield a dark brown extract B (7 g). Extract A: Extract A (268 mg) was redissolved in EtOAc and MeOH, absorbed onto silica/Celite® (1:1 mixture by mass), and subjected to flash column chromatography {silica; hexanes (100 mL), 50% EtOAc/hexanes (100 mL), EtOAc (100 mL), 10% MeOH/EtOAc (100 mL) and 20% MeOH/EtOAc (100 mL)}. Following TLC and <sup>1</sup>H NMR spectroscopic analysis, the resulting fractions were combined to afford five fractions, F.1-5. Fraction F.3 (146 mg) was redissolved in EtOAc and MeOH, adsorbed onto absorbed onto silica/Celite® (1:1 mixture by mass), and subjected to flash column chromatography {silica; CH<sub>2</sub>Cl<sub>2</sub> (150 mL), 20% EtOAc/CH<sub>2</sub>Cl<sub>2</sub> (100 mL), 40% EtOAc/CH<sub>2</sub>Cl<sub>2</sub> (100 mL), 60% EtOAc/CH<sub>2</sub>Cl<sub>2</sub> (100 mL), 80% EtOAc/CH<sub>2</sub>Cl<sub>2</sub> (100 mL), EtOAc (100 mL) and 20% MeOH/EtOAc (100 mL)}, which afforded compound 41 (12 mg, 0.04% w/w) as a pale-yellow solid.

Astragalin (41) (CAS# 480-10-4). [73,75] Pale-yellow solid. <sup>1</sup>H NMR (600 MHz, CD<sub>3</sub>OD) δ 8.06 (2H, d, J = 8.9 Hz; H-2′,6′), 6.90 (2H, d, J = 8.9 and 1.9 Hz; H-3′, 5′), 6.41 (1H, d, J = 2.1 Hz; H-8), 6.21 (1H, d, J = 2.1 Hz; H-6), 5.26 (1H, d, J = 7.4 Hz; H-1″), 3.69 (1H, dd, J = 11.9 and 2.3 Hz; H-6a″), 3.53 (1H, dd, J = 11.9 and 5.6 Hz; H-6b″), 3.40–3.46 (2H, m; H-2″,3″), 3.32 (1H, m; H-4″), 3.21 (1H, m; H-5″) ppm; <sup>13</sup>C NMR (150 MHz, CD<sub>3</sub>OD) δ 179.54 (C-4), 165.98 (C-7), 163.10 (C-5), 161.58 (C-4′), 159.09 (C-2), 158.52 (C-9), 135.45 (C-3), 132.28 (C-2′,6′), 122.80 (C-1′), 116.07 (C-3′,5′), 105.75 (C-10), 104.04 (C-1″), 99.87 (C-6), 94.73 (C-8), 78.43 (C-5″), 78.04 (C-3″), 75.73 (C-2″), 71.36 (C-4″), 62.62 (C-6″) ppm.

#### PHWE of Calochlaena dubia

C. dubia leaflets (30 g) were finely ground using a spice grinder, mixed with sand (~6 g), and extracted via PHWE (35% EtOH/H<sub>2</sub>O). This provided a hot extract (200 mL) that was cooled immediately in an ice bath. This process was repeated a further two times to provide a combined extract (600 mL). The extract was concentrated under reduced pressure (35°C) to remove EtOH and extracted with EtOAc (3 × 200 mL). The combined organic phase was dried (MgSO<sub>4</sub>), filtered, and concentrated under reduced pressure to provide dark green extract A (575 mg). The remaining aqueous phase was concentrated under reduced pressure (50°C) to yield a dark brown extract B (10 g). Extract A: Extract A (315 mg) was redissolved in MeOH and EtOAc, absorbed onto silica/Celite® (1:1 mixture by mass), and subjected to automated flash chromatography {silica cartridge (24 g); 0-100% EtOAc/hexanes and 0-50% MeOH/ EtOAc, for 12 min with flow rate of 28 mL/min}. Following TLC and <sup>1</sup>H NMR spectroscopic analysis, the resulting fractions were combined to give nine fractions, F.1-9. Fraction F.8 (41 mg) was subjected to flash column chromatography

{silica;  $CH_2Cl_2$  (50 mL), 10% MeOH/ $CH_2Cl_2$  (200 mL) and 50% MeOH/ $CH_2Cl_2$  (20 mL)}, which provided compound 41 (4 mg, 0.014% w/w).

# PHWE of Polystichum vestitum

P. vestitum dried leaflets (30 g) were finely ground using a spice grinder, mixed with sand ( $\sim$ 6 g), and extracted via PHWE (35% EtOH/H<sub>2</sub>O). This provided a hot extract (200 mL) that was cooled immediately in an ice bath. This process was repeated a further two times to provide a combined extract (600 mL). The extract was concentrated under reduced pressure (35°C) to remove EtOH and extracted with EtOAc (150 mL and 2 × 100 mL). The combined organic phase was dried (Na<sub>2</sub>SO<sub>4</sub>), filtered, and concentrated under reduced pressure to provide extract A (114 mg). The remaining aqueous phase was concentrated under reduced pressure (50°C) to yield a dark brown extract B (7 g). Extract A: Extract A (114 mg) was redissolved in EtOAc and MeOH, adsorbed onto silica/Celite® (1:1 mixture by mass), and subjected to automated flash column chromatography {silica cartridge (4 g); 0-100% EtOAc/hexanes and 0-50% MeOH/EtOAc, for 26 min with flow rate 12 mL/min)} and afforded compound 42 (30 mg, 0.1% w/w) as a light brown solid. Following TLC and <sup>1</sup>H NMR spectroscopic analysis, the fractions were combined to give six fractions, F.1-6. Fractions F.3-5 (27 mg) were combined and subjected to flash column chromatography {silica; CH<sub>2</sub>Cl<sub>2</sub> (50 mL), 5% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (50 mL), 15% MeOH/ CH<sub>2</sub>Cl<sub>2</sub> (50 mL) and 40% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (30 mL)} to provide compound 43 (4 mg) as a yellow solid. Extract B: Extract B (4g) was redissolved in MeOH and water, adsorbed onto silica/Celite® (1:1 mixture by mass), and fractionated through a silica plug (150 mL of EtOAc, 10% MeOH/ EtOAc, 20% MeOH/EtOAc, 30% MeOH/EtOAc, 40% MeOH/EtOAc and 50% MeOH/EtOAc) which afforded six fractions, F.1–6. Following <sup>1</sup>H NMR spectroscopic analysis, fractions F.1 and 2 (145 mg) were recombined, absorbed onto silica and subjected to flash column chromatography {silica; CH<sub>2</sub>Cl<sub>2</sub> (100 mL), 2.5% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (100 mL), 5% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (100 mL), 10% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (100 mL), 15% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (100 mL), 20% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (100 mL) and 35% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (100 mL)} to provide compound 42 (10 mg, 0.035% w/w).

4β-Carboxymethyl-(–)-epicatechin (42) (CAS# 146805-53-0). [62] Light-brown solid.  $^1$ H NMR (600 MHz, acetone-d<sub>6</sub>) δ 7.08 (1H, d, J=1.7 Hz; H-2′), 6.87 (1H, dd, J=8.2 and 1.8 Hz; H-6′), 6.81(1H, d, J=2.2 Hz; H-5′), 6.04 (1H, d, J=2.2 Hz; H-8), 5.94 (1H, d, J=2.2 Hz; H-6), 4.93 (1H, s; H-2), 4.01 (1H, s; H-3), 3.46 (1H, d, J=6.2 Hz; H-4), 3.05 (1H, dd, J=16.4 and 3.5 Hz; H-1b″), 2.45 (1H, dd, J=16.4 and 11.2 Hz; H-1a″) ppm.  $^{13}$ C NMR (150 MHz, acetone-d<sub>6</sub>) δ 173.9 (2″-COOH), 157.9 (C-7,9), 156.3 (C-5), 145.5 (C-4′), 145.3 (C-3′), 132.2 (C-1′), 119.3 (C-6′), 115.6 (C-5′), 115.3 (C-2′), 102.7 (C-10), 96.6 (C-8), 95.8 (C-6), 75.4 (C-2), 70.2 (C-3), 39.1 (C-1″), 36.0 (C-4) ppm.

(2R)-Eriodictyol-7-O-β-D-glucopyranoside (43) (CAS# 38965-51-4). [80] Yellow solid.  $^1$ H NMR (600 MHz, CD<sub>3</sub>OD) δ 6.82 (1H, s, H-2'), 6.69 (2H, t, J=10.9 Hz, H-5',6'), 6.11 (1H, s, H-6), 6.09 (1H, s, H-8), 5.23 (1H, d, J=12.5 Hz, H-2), 4.87 (1H, t, J=6.6 Hz, H-1"), 3.88 (1H, d, J=12.1 Hz, H-6a"), 3.69 (1H, m, H-6b"), 3.42–3.47 (3H, m, H-2",3",5"), 3.39 (1H, m, H-4"), 3.13 (1H, dd, J=17.5 Hz, H-3a), 2.75 (1H, d, J=17.0 Hz, H-3b) ppm.  $^{13}$ C NMR (150 MHz, CD<sub>3</sub>OD) δ 198.5 (C-4), 167.0 (C-7), 164.9 (C-9), 164.6 (C-5), 146.9 (C-3'), 146.5 (C-4'), 131.5 (C-1'), 119.3 (C-6'), 116.3 (C-5'), 114.8 (C-2'), 104.9 (C-10), 101.2 (C-1"), 97.9 (C-8), 96.9 (C-6), 80.7 (C-2), 78.3 (C-3"), 77.8 (C-5"), 74.7 (C-2"), 71.2 (C-4"), 62.3 (C-6"), 44.1 (C-3) ppm.

## PHWE of Polystichum proliferum

P. proliferum leaves (30 g) were finely ground using a spice grinder, mixed with sand (7 g), and extracted via PHWE (35% EtOH/H<sub>2</sub>O). This provided a hot extract (200 mL) that was cooled immediately in an ice bath. This process was repeated a further two times to provide a combined extract (600 mL). The extract was concentrated under reduced pressure (35°C) to remove EtOH and extracted with EtOAc (150 mL and  $2 \times 100$  mL). The combined organic phase was dried (Na<sub>2</sub>SO<sub>4</sub>), filtered, and concentrated under reduced pressure to provide extract A (240 mg). The remaining aqueous phase was concentrated under reduced pressure (50°C) to afford a dark brown extract B (4g). Extract A: Extract A (240 mg) was redissolved in CH<sub>2</sub>Cl<sub>2</sub> and MeOH, adsorbed onto silica/Celite® (1:1 mixture by mass), and subjected to flash column chromatography {silica; CH<sub>2</sub>Cl<sub>2</sub> (200 mL), 10% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (100 mL), 20% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (100 mL) and 40% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (200 mL)} which provided compound 42 (16 mg, 0.06% w/w).

#### PHWE of Pellaea falcata

P. falcata dried leaflets (30 g) were finely ground using a spice grinder, mixed with sand (~8g), and extracted via PHWE (35% EtOH/H<sub>2</sub>O). This provided a hot extract (200 mL) that was cooled immediately in an ice bath. This process was repeated a further two times to provide a combined extract (600 mL). The extract was concentrated under reduced pressure (35°C) to remove EtOH and extracted with EtOAc (150 mL and  $3 \times 100$  mL). The combined organic phase was dried (MgSO<sub>4</sub>), filtered, and concentrated under reduced pressure to provide extract A (1.0 g). The remaining aqueous phase was concentrated under reduced pressure (50°C) to yield a dark brown extract B (8g). Extract A: Extract A (1.0 g) was redissolved in EtOAc and MeOH, adsorbed onto silica/Celite® (1:1 mixture by mass) and subjected to automated flash chromatography {silica cartridge (12 g); (0-100% EtOAc/hexanes and 0-50% MeOH/EtOAc for 21 min with a flow rate of 28 mL/min)}. Following TLC and <sup>1</sup>H NMR spectroscopic analysis, the resulting fractions were combined to afford seven larger fractions, F.1-7.

Fraction F.5 was redissolved in MeOH, adsorbed onto silica/Celite<sup>®</sup> (1:1 mixture by mass) subjected to flash column chromatography {silica;  $CH_2Cl_2$  (50 mL), 10% MeOH/ $CH_2Cl_2$  (100 mL), 20% MeOH/ $CH_2Cl_2$  (50 mL) and 40% MeOH/ $CH_2Cl_2$  (50 mL)} which provided compound 44 (21 mg, 0.7% w/w) as a yellow solid.

Naringin (44) (CAS# 10236-47-2). [111,112] Yellow solid. <sup>1</sup>H NMR (600 MHz, CD<sub>3</sub>OD)  $\delta$  7.32 (2H, d, J = 8.5 Hz; H-2',6'), 6.82 (2H, d,  $J = 8.5 \,\text{Hz}$ ; H-3',5'), 6.18 (1H, d, J = 2.0 Hz; H-8), 6.16 (1H, d, J = 2.1 Hz; H-6), 5.37 (1H, dd, J = 12.9 and 2.7 Hz; H-2), 5.25 (1H, s; H-1"), 5.09 (1H, d, J = 7.6 Hz; H-1"), 3.93 (1H, s; H-2"), 3.85–3.90 (2H, m; H-6a",5""), 3.62-3.69 (2H, m; H-3",6b"), 3.57-3.60 (2H, m; H-2",5"), 3.43-3.46 (1H, m; H-3"), 3.39 (2H, t, J = 9.5 Hz; H-4",4""), 3.16 (1H, dd, J = 17.2 and 12.9 Hz; H-3a), 2.75 (1H, dd, J = 17.2 and 2.8 Hz; H-3b), 1.28 (3H, d,  $J = 6.2 \,\text{Hz}; 6'''\text{-CH}_3) \text{ ppm.}^{13} \text{C NMR } (150 \,\text{MHz}, \,\text{CD}_3 \,\text{OD})$  $\delta$  198.5 (C-4), 166.6 (C-7), 164.9 (C-5), 164.6 (C-9), 159.1 (C-4'), 130.8 (C-1'), 129.1 (C-2',6'), 116.3 (C-3',5'), 104.9 (C-10), 102.5 (C-1"), 99.4 (C-1"), 97.8 (C-6), 96.7 (C-8), 80.7 (C-2), 79.0 (C-2"), 78.9 (C-3"), 78.1 (C-5"), 73.9 (C-4"'), 72.2 (C-3"'), 71.2 (C-2"'), 71.2 (C-4"), 69.9 (C-5"'), 62.3 (C-6"), 44.1 (C-3), 18.2 (6"'-CH<sub>3</sub>) ppm.

# PHWE of Austroblechnum penna-marina subsp. alpina

A. penna-marina subsp. alpina dried leaflets (29 g) were finely ground using a spice grinder, mixed with sand ( $\sim$ 7.5 g), and extracted via PHWE (35% EtOH/ $H_2O$ ). This provided a hot extract (200 mL) that was cooled immediately in an ice bath. This process was repeated a further three times to provide a combined extract (800 mL). The extract was concentrated under reduced pressure (35°C) to remove EtOH and extracted with EtOAc (150 mL and  $2 \times 100$  mL). The combined organic phase was dried (MgSO<sub>4</sub>), filtered, and concentrated under reduced pressure to provide extract A (409 mg). The remaining aqueous phase was concentrated under reduced pressure (50°C) to yield a dark brown extract B (6g). Extract A. Extract A (400 mg) was redissolved in MeOH and EtOAc, adsorbed onto silica/Celite® (1:1 mixture by mass) and subjected to flash column chromatography (silica; hexanes (100 mL), 20% EtOAc/hexanes (200 mL), 40% EtOAc/hexanes (100 mL), 60% EtOAc/hexanes (100 mL), 80% EtOAc/hexanes (100 mL), EtOAc (50 mL), 10% MeOH/EtOAc (100 mL) and MeOH/EtOAc (100 mL)} that provided compound 45 (8.9 mg, 0.3% w/w) as a pale-green solid.

Blechnic acid (45) (CAS# 146805-53-0). Palegreen solid. [α]  $-23.2^{\circ}$  (c 0.0345, MeOH), lit. [α]  $-28^{\circ}$  (c = 1.0, MeOH). HNMR (600 MHz, CD<sub>3</sub>OD) δ 7.47 (d, J = 15.9 Hz; H-7'), 7.03 (d, J = 8.5 Hz; H-6'), 6.86 (d, J = 2.0 Hz; H-2), 6.74 (dd, J = 8.2, 1.9 Hz; H-6), 6.71 (d, J = 8.4 Hz; H-5), 6.65 (d, J = 8.2 Hz; H-5'), 6.16 (d, J = 15.9 Hz; H-8'), 5.83 (d, J = 9.3 Hz; H-7), 4.50 (d, J = 9.3 Hz; H-8) ppm. CNMR (150 MHz, CD<sub>3</sub>OD) δ 173.6 (C-9), 170.6 (C-9'), 149.6 (C-3'), 146.5 (C-4), 146.0 (C-3),

145.1 (C-4′), 143.3 (C-7′), 129.3 (C-1), 129.1 (C-2′), 124.5 (C-1′), 122.6 (C-6′), 119.7 (C-6), 118.0 (C-5′), 117.8 (C-8′), 115.9 (C-5), 115.1 (C-2), 88.4 (C-7), 55.3 (C-8) ppm. HRESIMS m/z calcd for  $C_{18}H_{14}O_8Na$  [M + Na]  $^+$  381.0586; found 381.0581.

#### PHWE Lecanopteris pustulata subsp. pustulata

L. pustulata subsp. pustulata dried leaflets (30 g) were finely ground using a spice grinder, mixed with sand  $(\sim 7 \text{ g})$ , and extracted via PHWE (35% EtOH/H<sub>2</sub>O). This provided a hot extract (200 mL) that was cooled immediately in an ice bath. This process was repeated a further two times to provide a combined extract (600 mL) and extracted with EtOAc (150 mL and  $2 \times 100$  mL). The combined organic phase was dried (MgSO<sub>4</sub>), filtered, and concentrated under reduced pressure to provide extract A (154 mg). The remaining aqueous phase was concentrated under reduced pressure (50°C) to yield a dark brown extract B (5.6 g). Extract A. Extract A (118 mg) was redissolved in MeOH, adsorbed onto silica/Celite® (1:1 mixture by mass) and subjected to flash column chromatography {silica; hexanes (30 mL), 30% EtOAc/hexanes (50 mL), 60% EtOAc/hexanes (100 mL), 80% EtOAc/hexanes (50 mL), EtOAc (50 mL), 20% MeOH/EtOAc (50 mL) and 50% MeOH/EtOAc (25 mL)}. Following TLC and <sup>1</sup>H NMR spectroscopic analysis, the resulting fractions were combined to afford six larger fractions, F.1-6. Fractions F.2 (19.5 mg) was redissolved in CH<sub>2</sub>Cl<sub>2</sub> and MeOH, adsorbed onto silica/Celite<sup>®</sup> (1:1 mixture by mass) and subjected to flash column chromatography {silica; CH<sub>2</sub>Cl<sub>2</sub> (25 mL), 5% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (25 mL) and 20% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (25 mL)}. Following TLC analysis, the resulting fractions were combined to afford three fractions F.1-3; and F.2 provided compound **42** (12 mg, 0.04% w/w). Extract B. Extract B (5.6 g) was redissolved in MeOH and water, adsorbed onto silica/Celite® (1:1 mixture by mass) and subjected to flash chromatography {silica; EtOAc (200 mL), 10% MeOH/EtOAc (200 mL), 20% MeOH/EtOAc (200 mL) and 30% MeOH/EtOAc (200 mL)} that provided five fractions, F.1-5. Fraction F.1 (127 mg) was redissolved in MeOH and CH<sub>2</sub>Cl<sub>2</sub>, adsorbed onto silica/Celite<sup>®</sup> (1:1 mixture by mass) and subjected to flash column chromatography {silica; CH<sub>2</sub>Cl<sub>2</sub> (50 mL), 5% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (100 mL), 10% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (50 mL), 30% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (100 mL) and MeOH (30 mL)} that provided compound 46 (4.8 mg, 0.016% w/w) as an off-white solid.

(*p*-Hydroxybenzyl)malonic acid (46). [70] Off-white solid. <sup>1</sup>H NMR (600 MHz, acetone- $d_6$ ) δ 7.05 (2H, d, J = 8.3 Hz; H-2′,6′), 6.68 (2H, d, J = 8.3 Hz; H-3′,5′), 3.53 (1H, t, J = 7.6 Hz; H-2), 3.06 (2H, d, J = 7.6 Hz; H-3); <sup>1</sup>H NMR (600 MHz, DMSO- $d_6$ ) δ 6.99 (2H, d, J = 8.4 Hz; H-2′,6′), 6.63 (2H, d, J = 8.4 Hz; H-3′,5′), 3.39 (1H, m, H-2), 2.92 (2H, d, J = 7.5 Hz; H-3) ppm. <sup>13</sup>C NMR (150 MHz, acetone- $d_6$ ) δ 172.9 (C-1), 157.1 (C-4′), 130.9 (C-2′,6′), 130.9 (C-1′), 116.2 (C-3′,5′), 55.4 (C-2), 35.2 (C-3); <sup>13</sup>C NMR (150 MHz, DMSO- $d_6$ ) δ 170.7 (C-1), 155.7 (C-4′), 129.6 (C-2′,6′),

128.7(C-1'), 114.9 (C-3',5'), 53.5 (C-2), 33.5 (C-3) ppm. HRESIMS m/z calcd for  $C_{10}H_{10}O_5Na$  [M + Na]  $^+$  233.0426; found 233.0423.

# PHWE of Lomaria nuda

L. nuda fronds (30 g) were finely ground using a spice grinder, mixed with sand (~6 g), and extracted via PHWE (35% EtOH/H<sub>2</sub>O). This provided a hot extract (200 mL) that was cooled immediately in an ice bath. This process was repeated a further two times to provide a combined extract (600 mL). The extract was concentrated under reduced pressure (35°C) to remove EtOH and extracted with EtOAc (150 mL and  $2 \times 100$  mL). The combined organic phase was dried (MgSO<sub>4</sub>), filtered, and concentrated under reduced pressure to provide extract A (254 mg). The remaining aqueous phase was concentrated under reduced pressure (50°C) to yield a dark brown extract B (5g). Extract A. Extract A (254 mg) was redissolved in MeOH, adsorbed onto silica/ Celite® (1:1 mixture by mass) and subjected to automated flash chromatography {silica cartridge (4 g); 0-40% MeOH/EtOAc for 17 min}. Following TLC and <sup>1</sup>H NMR spectroscopic analysis, the resulting fractions were combined into seven large fractions, F.1-7. Fraction F.3 (14.5 mg) was redissolved in MeOH, adsorbed onto silica/Celite<sup>®</sup> (1:1 mixture by mass) and subjected to flash column chromatography {silica; CH<sub>2</sub>Cl<sub>2</sub> (50 mL), 5% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (50 mL), 10% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (50 mL), 20% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (50 mL) and MeOH (20 mL)} which provided compound 47 (2.5 mg, 0.01% w/w) as a yellow solid and compound 48 (5 mg, 0.016% w/w) as a yellow solid.

Quercetin-3-O-(6'-O-acetyl)glucoside (47) (CAS# 54542-51-7). [114] Yellow solid. <sup>1</sup>H NMR (600 MHz, CD<sub>3</sub>OD)  $\delta$  7.81 (1H, d, J = 2.2 Hz; H-6'), 7.63 (1H, dd, J = 8.5 and 2.2 Hz;H-5'), 6.88 (1H, d,  $J = 8.5 \,\text{Hz}$ ; H-2'), 6.44 (1H, d, J = 2.0 Hz; H-8), 6.23 (1H, d, J = 2.0 Hz; H-6), 5.08 (1H, d, J = 7.9 Hz; H-1"), 4.18 (1H, dd, J = 11.4 and 7.8 Hz; H-6a"), 4.07 (1H, dd, J = 11.4 and 4.5 Hz; H-6b"), 3.83 (1H, dd, J = 9.6 and 7.4 Hz; H-2"), 3.80 (1H, d,  $J = 3.3 \,\text{Hz}$ ; H-3"), 3.70 (1H, dd, J = 7.7 and 4.6 Hz; H-5"), 3.58 (1H, dd, J = 9.4 and 3.1 Hz; H-4"), 1.83 (3H, s; 2""-COCH<sub>3</sub>) ppm.  $^{13}$ C NMR (150 MHz, CD<sub>3</sub>OD)  $\delta$  179.4 (C-4), 172.5 (1"'-CO), 167.3 (C-7), 162.9 (C-5), 158.6 (C-9), 157.8 (C-2), 150.0 (C-4'), 145.8 (C-3'), 135.7 (C-3), 123.1 (C-6'), 122.8 (C-1'), 117.6 (C-5'), 116.0 (C-2'), 105.8 (C-1"), 105.1 (C-10), 100.4 (C-6), 95.1 (C-8), 74.9 (C-3"), 74.5 (C-5"), 72.9 (C-2"), 70.2 (C-4"), 64.5 (C-6"), 20.4 (2"'-COCH<sub>3</sub>) ppm.

*Quercitrin* (48) (CAS# 522-12-3). [115] Yellow solid.  $^{1}$ H NMR (600 MHz, CD<sub>3</sub>OD)  $\delta$  7.33 (1H, d, J = 2.1 Hz; H-2′), 7.30 (1H, dd, J = 8.3 and 2.1 Hz; H-6′), 6.91 (1H, d, J = 8.3 Hz; H-5′), 6.34 (1H, d, J = 2.1 Hz; H-6), 6.18 (1H, d, J = 2.1 Hz; H-8), 5.35 (1H, s; H-1″), 4.22 (1H, m; H-2″), 3.75 (1H, m; H-3″), 3.42 (1H, m; H-5″), 3.35 (1H, m; H-4″), 0.95 (3H, d, J = 6.2 Hz; 6″-CH<sub>3</sub>) ppm.  $^{13}$ C NMR (150 MHz, CD<sub>3</sub>OD)  $\delta$  179.5 (C-4), 167.3 (C-7), 163.1 (C-5), 159.1 (C-2), 158.6 (C-9), 149.9 (C-4′), 146.5 (C-3′), 136.1 (C-3), 122.9

(C-6'), 122.8 (C-1'), 116.9 (C-5'), 116.4 (C-2'), 105.5 (C-10), 103.5 (C-1"), 100.3 (C-6), 95.0 (C-8), 73.3 (C-4"), 72.1 (C-3"), 72.0 (C-5"), 71.9 (C-2'), 17.6 (6'-CH<sub>3</sub>) ppm.

#### PHWE of Parablechnum wattsii

P. wattsii dried leaf material (30 g) was finely ground using a spice grinder, mixed with sand ( $\sim$ 6 g), and extracted via PHWE (35% EtOH/H2O). This provided a hot extract (200 mL) that was cooled immediately in an ice bath. This process was repeated further two times to provide a combined extract (600 mL). The extract was concentrated under reduced pressure (35°C) to remove EtOH and extracted with EtOAc (150 mL and  $2 \times 120$  mL). The combined organic phase was dried (MgSO<sub>4</sub>), filtered, and concentrated under reduced pressure to provide extract A (523 mg). The remaining aqueous phase was concentrated under reduced pressure (50°C) to yield a dark-brown extract B (7 g). Extract A. Extract A (523 mg) was redissolved in MeOH, adsorbed onto silica and subjected to automated flash chromatography {silica cartridge (12 g); 0-100% EtOAc/hexanes and 0-40% MeOH/CH2Cl2 for 13 min at a flow rate of 25 mL/min} which provided compound 49 (143 mg, 0.5% w/w) as a dark-brown solid. Approximately 60 mg of compound 49 was absorbed onto silica and further purified by flash column chromatography {silica, 20% MeOH/CH2Cl2 (150 mL) then 40% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (100 mL)} which provided compound 49 as a pale-yellow/-green solid (20 mg).

5-O-Caffeoylshikimic acid (49) (CAS# 73263-62-4). Pale-yellow/-green solid. H NMR (600 MHz, CD<sub>3</sub>OD)  $\delta$  7.56 (1H, d, J = 15.9 Hz; H-7′), 7.04 (1H, d, J = 1.7 Hz; H-2′), 6.95 (1H, dd, J = 8.2 and 1.6 Hz; H-6′), 6.83 (br s; H-2), 6.78 (1H, d, J = 8.2 Hz; H-5′), 6.28 (1H, dd, J = 15.9 Hz; H-8′), 5.25 (1H, dd, J = 13.2 and 5.6 Hz; H-5), 4.40 (1H, s; H-3), 3.90 (1H, dd, J = 7.9 and 4.1 Hz; H-4), 2.87 (1H, dd, J = 18.8 and 4.9 Hz; H-6a), 2.32 (1H, dd, J = 18.4 and 5.1 Hz; H-6b) ppm.  $^{13}$ C NMR (150 MHz, CD<sub>3</sub>OD)  $\delta$  170.3 (C-7), 168.6 (C-9′), 149.6 (C-4′), 147.2 (C-7′), 146.8 (C-3′), 138.2 (C-2), 131.1 (C-1), 127.7 (C-1′), 123.0 (C-6′), 116.5 (C-5′), 115.2 (C-2′), 115.1 (C-8′), 71.4 (C-5), 70.1 (C-4), 67.4 (C-3), 29.4 (C-6) ppm. HRESIMS m/z calcd for  $C_{16}H_{16}O_8Na$  [M + Na]  $^+$  359.0845; found 359.0738.

# PHWE of Gleichenia alpina

*G. alpina* dried aerial parts (29 g) were finely ground using a spice grinder, mixed with sand ( $\sim$ 4 g), and extracted via PHWE (35% EtOH/H<sub>2</sub>O). This provided a hot extract (200 mL) that was cooled immediately in an ice bath. This process was repeated further two times to provide a combined extract (600 mL). The extract was concentrated under reduced pressure (35°C) to remove EtOH and extracted with EtOAc (200 mL and 2  $\times$  150 mL). The combined organic phase was dried (MgSO<sub>4</sub>), filtered, and concentrated under reduced pressure to provide extract A (646 mg). The remaining aqueous phase was concentrated under reduced pressure (50°C) to yield a dark-brown extract B (10 g). *Extract A*.

Extract A (793 mg) was adsorbed onto silica/Celite (1:1 mixture by mass) and subjected to flash column chromatography {silica; EtOAc (300 mL), 10% MeOH/EtOAc (150 mL), 20% MeOH/EtOAc (150 mL), 30% MeOH/EtOAc (150 mL), 40% MeOH/EtOAc (150 mL) and MeOH (100 mL)}, to afford six fractions, F.1–6. Fraction F.3 (202 mg) was and adsorbed onto silica/Celite (1:1 mixture by mass) and subjected to flash column chromatography {silica;  $CH_2Cl_2$  (50 mL), 10% MeOH/ $CH_2Cl_2$  (50 mL), 20% MeOH/ $CH_2Cl_2$  (100 mL), 20% MeOH/ $CH_2Cl_2$  (150 mL) and MeOH (50 mL)} which provided compound 50 (6.9 mg, 0.02% w/w) as a yellow solid that contained minor impurities.

Rutin (50) (CAS# 153-18-4).[116] Yellow solid. <sup>1</sup>H NMR (600 MHz, CD<sub>3</sub>OD)  $\delta$  7.67 (1H, d, J = 2.2 Hz; H-2'), 7.63 (1H, dd, J = 8.4 and 2.2 Hz; H-6'), 6.88 (1H, d, J = 8.4 Hz;H-5'), 6.41 (1H, d, J = 2.1 Hz; H-8), 6.21 (1H, d, J = 2.1 Hz; H-6), 5.11 (1H, d, J = 7.7 Hz; H-1"), 4.52 (1H, d, J = 1.4 Hz; H-1"'), 3.80 (1H, dd, J = 11.2 and 1.5 Hz; H-6a"), 3.63 (1H, dd, J = 3.3 and 1.6 Hz; H-2"'), 3.54 (1H, dd, J = 9.5 and 3.5 Hz; H-3"), 3.38-3.49 (4H, m; H-2",3",5",6b"), 3.25-3.34 (3H, m; H-4",5",4""), 1.12 (3H, d, J = 6.2 Hz; 6""-CH<sub>3</sub>) ppm. <sup>13</sup>C NMR (150 MHz, CD<sub>3</sub>OD)  $\delta$  179.4 (C-4), 166.1 (C-7), 162.9 (C-5), 159.3 (C-2), 158.5 (9), 149.8 (C-4'), 145.8 (3'), 135.6 (C-3), 123.5 (C-6'), 123.1 (C-1'), 117.7 (C-2'), 116.1 (C-5'), 105.6 (C-10), 104.7 (C-1"), 102.4 (C-1""), 99.9 (C-6), 94.9 (C-8), 78.2 (C-3"), 77.2 (C-5"), 75.7 (C-2"), 73.9 (C-4"'), 72.2 (C-3"'), 72.1 (C-2"'), 71.4 (C-4"), 69.7 (C-5"'), 68.5 (C-6"), 17.9 (C-6"") ppm.

#### PHWE of Dicksonia antarctica

*D. antarctica* dried leaflets (30.5 g) were finely ground using a spice grinder, mixed with sand ( $\sim$ 7 g), and extracted via PHWE (35% EtOH/H<sub>2</sub>O). This provided a hot extract (200 mL) that was cooled immediately in an ice bath. This process was repeated a further three times to provide a combined extract (800 mL). The extract was concentrated under reduced pressure (35°C) to remove EtOH and extracted with EtOAc (150 mL and 2  $\times$  100 mL). The combined organic phase was dried (MgSO<sub>4</sub>), filtered, and concentrated under reduced pressure to provide extract A (298 g). The remaining aqueous phase was concentrated under reduced pressure (50°C) to yield a dark-brown extract B (5 g). No compounds could be isolated by flash column chromatography.

# PHWE of Doodia australis

*D. australis* dried leaflets (31 g) were finely ground using a spice grinder, mixed with sand ( $\sim$ 7 g), and extracted via PHWE (35% EtOH/H<sub>2</sub>O). This provided a hot extract (200 mL) that was cooled immediately in an ice bath. This process was repeated a further two times to provide a combined extract (600 mL). The aqueous extract was extracted with EtOAc (150 and 2  $\times$  100 mL). The combined organic phase was dried (MgSO<sub>4</sub>), filtered, and concentrated under reduced pressure to provide extract A (617 mg). The remaining aqueous phase was concentrated under reduced pressure (50°C) to yield a

dark-brown extract B (3.5 g). No compounds could be isolated by flash column chromatography.

# PHWE of Oceaniopteris cartilaginea

*O. cartilaginea* dried leaflets (33 g) were finely ground using a spice grinder, mixed with sand ( $\sim$ 8 g), and extracted via PHWE (35% EtOH/H<sub>2</sub>O). This provided a hot extract (200 mL) that was cooled immediately in an ice bath. This process was repeated a further two times to provide a combined extract (600 mL) and extracted with EtOAc (400 mL and 2  $\times$  200 mL). The combined organic phase was dried (MgSO<sub>4</sub>), filtered, and concentrated under reduced pressure to provide extract A (1.5 g). The remaining aqueous phase was concentrated under reduced pressure (50°C) to yield a dark-brown extract B ( $\sim$ 2 g). No compounds could be isolated by flash column chromatography.

#### PHWE of Histiopteris incisa

*H. incisa* dried leaflets (35 g) were finely ground using a spice grinder, mixed with sand ( $\sim$ 16 g), and extracted via PHWE (35% EtOH/H<sub>2</sub>O). This provided a hot extract (200 mL) that was cooled immediately in an ice bath. This process was repeated a further three times to provide a combined extract (800 mL) and extracted with EtOAc (3 × 250 mL). The combined organic phase was dried (MgSO<sub>4</sub>), filtered, and concentrated under reduced pressure to provide extract A (877 mg). The remaining aqueous phase was concentrated under reduced pressure (50°C) to yield a dark-brown extract B (3 g). No compounds could be isolated by flash column chromatography.

#### PHWE of Pteridium esculentum

Aerial parts of P. esculentum (150 g) were finely ground using a spice grinder, mixed with sand (~75 g), and extracted via PHWE (35% EtOH/H2O). This provided a hot extract (200 mL) that was cooled immediately in an ice bath. This process was repeated a further nine times to provide a combined extract (2 L). The extract was concentrated under reduced pressure (35°C) to remove EtOH and extracted with EtOAc (3 × 500 mL). The combined organic phase was dried (MgSO<sub>4</sub>), filtered, and concentrated under reduced pressure to provide extract A (2.7 g). The remaining aqueous phase was concentrated under reduced pressure (50°C) to afford brown extract B (32 g). Extract A. Extract A (1.5 g) was redissolved in MeOH, adsorbed onto silica/Celite® (1:1 mixture by mass) and subjected to flash column chromatography {silica; CH<sub>2</sub>Cl<sub>2</sub> (250 mL), 10% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (250 mL), 20% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (250 mL), 30% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (250 mL), 40% MeOH/CH<sub>2</sub>Cl<sub>2</sub> (250 mL) and MeOH (200 mL)}. Following TLC and <sup>1</sup>H NMR spectroscopic analysis, the resulting fractions were combined to provide seven fractions, F.1-7. Fraction F.4 (150 mg) was redissolved in 20% MeCN/H2O solution and subjected to automated flash chromatography (12 g C18 cartridge; 0–100% MeCN/H<sub>2</sub>O for 17 min with a flow rate of 28 mL/min). Following TLC and <sup>1</sup>H NMR spectroscopic analysis, the

resulting fractions were combined to afford four fractions, F.1–4. Fraction F.2 (19 mg) was redissolved in MeOH/CH $_2$ Cl $_2$  solution and subjected to flash column chromatography {silica; 10% MeOH/CH $_2$ Cl $_2$  (10 mL)} to provide compound **41** (12 mg, 0.006% w/w) as a pale-yellow solid. The fraction F.4 (42 mg) was subjected to preparative TLC which provided compound **51** (25 mg, 0.02% w/w) as a yellow solid.

Tiliroside (51) (CAS# 20316-62-5).[117,118] Yellow solid. <sup>1</sup>H NMR (600 MHz, CD<sub>3</sub>OD)  $\delta$  7.98 (2H, d, J = 8.8 Hz; H-2',6'), 7.40 (1H, d, J = 15.9 Hz; H-7"'), 7.30 (2H, d, J = 8.5 Hz; H-2"',6"'), 6.82 (2H, d, J = 8.8 Hz; H-3',5'), 6.79 (2H, d, J = 8.6 Hz; H-3"',5"'), 6.30 (1H, d, J = 1.9 Hz; H-8), 6.13 (1H, d,  $J = 2.0 \,\text{Hz}$ ; H-6), 6.07 (1H, d,  $J = 15.9 \,\text{Hz}$ ; H-8"), 5.23 (1H, d,  $J = 7.3 \,\text{Hz}$ ; H-1"), 4.31 (1H, dd, J = 11.8 and 1.9 Hz; H-6a''), 4.20 (1H, m; H-6b''),3.41–3.50 (3H, m; H-2",3",5"), 3.31 (1H, m; H-4") ppm. <sup>13</sup>C NMR (150 MHz, CD<sub>3</sub>OD)  $\delta$  179.4 (C-4), 168.6 (9"'-CO), 166.3 (C-7), 162.9 (C-5), 161.5 (C-4'), 161.2 (C-4"), 159.3 (C-2), 158.5 (C-9), 146.6 (C-7"), 135.2 (C-3), 132.2 (C-2',6'), 131.2 (C-2"',6"'), 127.1 (C-1"'), 122.7 (C-1'), 116.8 (C-3",5"), 116.1 (C-3',5'), 114.7 (C-8"), 105.5 (C-10), 104.1 (C-1"), 100.1 (C-6), 94.9 (C-8), 78.0 (C-3"), 75.7 (C-2",5"), 71.7 (C-4"), 64.3 (C-6") ppm.

# Supplementary material

Supplementary material is available online.

#### References

- Hassler M. World ferns. Synonymic checklist and distribution of ferns and lycophytes of the world. 2022. Available at www. worldplants.de [cited 19 January 2022]
- [2] Liu Y, Wujisguleng W, Long C. Food uses of ferns in China: a review. Acta Soc Bot Pol 2012; 81(4): 263-70. doi:10.5586/ asbp.2012.046
- [3] Matsuura T, Sugimura K, Miyamoto A, Tanaka N. Knowledge-based estimation of edible fern harvesting sites in mountainous communities of northeastern Japan. *Sustainability* 2014; 6(1): 175–92. doi:10.3390/su6010175
- [4] Wallis LA, Stephenson B. A nardoo processing grinding stone from a rockshelter in the Pilbara, Western Australia. Aust Archaeol 2020; 86(2): 112–117. doi:10.1080/03122417.2020. 1768627
- [5] Baskaran Xr, Geo Vigila Av, Zhang Sz, Feng Sx, Liao Wb. A review of the use of pteridophytes for treating human ailments. *J Zhejiang Univ Sci B* 2018; 19(2): 85–119. doi:10.1631/jzus. B1600344
- [6] Su GQ, Li HT, Sun H, et al. [Endemic plants for medicine use in China]. Zhongguo Zhongyao Zazhi 2017; 42(22): 4329–4335. doi:10.19540/j.cnki.cjcmm.2017.0179
- [7] Liu B, Guo Z-y, Bussmann R, Li F-f, Li J-q, Hong L-y, *et al.* Ethnobotanical approaches of traditional medicine studies in Southwest China: a literature review. *J Ethnopharmacol* 2016; 186: 343–350. doi:10.1016/j.jep.2016.02.040
- [8] Sureshkumar J, Silambarasan R, Bharati KA, Krupa J, Amalraj S, Ayyanar M. A review on ethnomedicinally important pteridophytes of India. *J Ethnopharmacol* 2018; 219: 269–287. doi:10.1016/j.jep.2018.03.024
- [9] Reinaldo RCPDS, Santiago ACP, Medeiros PM, Albuquerque UP. Do ferns and lycophytes function as medicinal plants? A study

- of their low representation in traditional pharmacopoeias. *J Ethnopharmacol* 2015; 175: 39–47. doi:10.1016/j.jep.2015. 09.003
- [10] Cao H, Chai TT, Wang X, Morais-Braga MFB, Yang JH, Wong FC, et al. Phytochemicals from fern species: potential for medicine applications. *Phytochem Rev* 2017; 16: 379–440. doi:10.1007/ s11101-016-9488-7
- [11] Jiang F, Qi B, Ding N, Yang H, Jia F, Luo Y, et al. Lycopodium alkaloids from *Huperzia serrata*. Fitoterapia 2019; 137: 104277. doi:10.1016/j.fitote.2019.104277
- [12] Zhu QF, Zhao QS. Chemical constituents and biological activities of lycophytes and ferns. Chin J Nat Med 2019; 17(12): 887–891. doi:10.1016/S1875-5364(19)30108-6
- [13] Lin L-j, Huang X-b, Lv Z-c. Isolation and identification of flavonoids components from Pteris vittata L. *SpringerPlus* 2016; 5(1): 5–7. doi:10.1186/s40064-016-3308-9
- [14] Hou M, Chen Y, Wang Y, Hao K. Sesquiterpenoids and flavonoids from *Pteris multifida* Poir. *Biochem Syst Ecol* 2021; 98: 104320. doi:10.1016/j.bse.2021.104320
- [15] Asai F, Iinuma M, Tanaka T, Mizuno M. Complex flavonoids in farinose exudate from *Pityrogramma calomelanos*. *Phytochemistry* 1991; 30(9): 3091–3093. doi:10.1016/S0031-9422(00)98259-1
- [16] Iinuma M, Tanaka T, Takenaka M, Mizuno M, Asai F. Five complex flavonoids in the farinose exudate of *Pityrogramma Calomelanos*. *Phytochemistry* 1992; 31: 2487–2490. doi:10.1016/0031-9422(92)83306-J
  (b) Asai F, Iinuma M, Tanaka T, Mizuno M. Two complex flavonoids in the farinose exudate of *Pityrogramma Calomelanos*.
- [17] Shim S-Y, Lee S-g, Lee M. Biflavonoids isolated from Selaginella tamariscina and their anti-inflammatory activities via ERK 1/2 signaling. Molecules 2018; 23(4): 926. doi:10.3390/ molecules23040926

Heterocycles 1992; 33: 229-23. doi:10.3987/COM-91-S21

- [18] Long HP, Zou H, Li FS, Li J, Luo P, Zou ZX, et al. Involvenflavones A-F, six new flavonoids with 3'-aryl substituent from *Selaginella involven*. Fitoterapia 2015; 105: 254–259. doi:10.1016/j.fitote.2015.07.013
- [19] Su LH, Li YP, Li HM, Dai WF, Liu D, Cao L, et al. Anti-inflammatory prenylated flavonoids from Helminthostachys zey-lanica. Chem Pharm Bull 2016; 64(5): 497–501. doi:10.1248/cpb.c15-00661
- [20] Mohammad RH, Nur-E-Alam M, Lahmann M, Parveen I, Tizzard GJ, Coles SJ, et al. Isolation and characterisation of 13 pterosins and pterosides from bracken (Pteridium aquilinum (L.) Kuhn) rhizome. Phytochemistry 2016; 128: 82–94. doi:10.1016/j. phytochem.2016.05.001
- [21] Niwa H, Ojika M, Wakamatsu K, Yamada K, Hirono I, Matsushita K. Ptaquiloside, a novel norsesquiterpene glucoside from bracken, *Pteridium aquilinum* var. *latiusculum*. *Tetrahedron Lett* 1983; 24(38): 4117–4120. doi:10.1016/S0040-4039(00)88276-3
- [22] Ge X, Ye G, Li P, Tang WJ, Gao JL, Zhao WM. Cytotoxic diterpenoids and sesquiterpenoids from *Pteris multifida*. J Nat Prod 2008; 71(2): 227–231. doi:10.1021/np0706421
- [23] Li MM, Wang K, He J, Peng LY, Chen XQ, Cheng X, et al. Four new labdane-type diterpenoid glycosides from *Diplopterygium laevissimum*. *Nat Prod Bioprospect* 2013; 3(2): 38–42. doi:10.1007/s13659-012-0022-3
- [24] Li XL, Yang LM, Zhao Y, Wang RR, Xu G, Zheng YT, et al. Tetranorclerodanes and clerodane-type diterpene glycosides from *Dicranopteris dichotoma*. *J Nat Prod* 2007; 70(2): 265–268. doi:10.1021/np0603166
- [25] Socolsky C, Asakawa Y, Bardón A. Diterpenoid glycosides from the bitter fern *Gleichenia quadripartita*. J Nat Prod 2007; 70(12): 1837–1845. doi:10.1021/np070119m
- [26] Wada H, Shimizu Y, Hakamatsuka T, Tanaka N, Cambie RC, Braggins JE. Two new clerodane glycosides from *Gleichenia microphylla*. Aust J Chem 1998; 51(2): 171–21. doi:10.1071/C97183
- [27] Snogan E, Vahirua-Lechat I, Ho R, Bertho G, Girault JP, Ortiga S, et al. Ecdysteroids from the medicinal fern Microsorum scolopendria (Burm. f.). Phytochem Anal 2007; 18(5): 441–450. doi:10.1002/pca.1000

[28] Ho R, Teai T, Loquet D, Bianchini JP, Girault JP, Lafont R, et al. Phytoecdysteroids in the genus Microsorum (Polypodiaceae) of French Polynesia. Nat Prod Commun 2007; 2(8): 1934578X0700200. doi:10.1177/1934578X0700200803

- [29] Watanabe M, Miyashita T, Devkota HP. Phenolic compounds and ecdysteroids of *Diplazium esculentum* (Retz.) Sw. (Athyriaceae) from Japan and their chemotaxonomic significance. *Biochem Syst Ecol* 2021; 94: 104211. doi:10.1016/j.bse.2020.104211
- [30] Hu J, Shi X, Mao X, Li H, Chen J, Shi J. Ecdysteroids from the ethanol extract of *Diplopterygium rufopilosum*. *Phytochem Lett* 2014; 8(1): 73–76. doi:10.1016/j.phytol.2014.02.003
- [31] Savchenko RG, Veskina NA, Odinokov VN, Benkovskaya GV, Parfenova LV. Ecdysteroids: isolation, chemical transformations, and biological activity. *Phytochem Rev* 2022; doi:10.1007/ s11101-021-09792-y
- [32] Jiang J, Tian L, Wang L, Liu Y, Chen Y. Phenolic compounds from the fern Glaphyropteridopsis erubescens (Hook.) Ching. Biochem Syst Ecol 2013; 50: 136–138. doi:10.1016/j.bse.2013. 04.005
- [33] Chen Y-H, Chang F-R, Lin Y-J, Wang L, Chen J-F, Wu Y-C, et al. Identification of phenolic antioxidants from Sword Brake fern (*Pteris ensiformis* Burm.). Food Chem 2007; 105(1): 48–56. doi:10.1016/j.foodchem.2007.03.055
- [34] Chen Y-H, Chang F-R, Lu M-C, Hsieh P-W, Wu M-J, Du Y-C, *et al.* New benzoyl glucosides and cytotoxic pterosin sesquiterpenes from *Pteris ensiformis* Burm. *Molecules* 2008; 13(2): 255–266. doi:10.3390/molecules13020255
- [35] Wei H-A, Lian T-W, Tu Y-C, Hong J-T, Kou M-C, Wu M-J. Inhibition of low-density lipoprotein oxidation and oxidative burst in polymorphonuclear neutrophils by caffeic acid and hispidin derivatives isolated from sword brake fern (*Peris ensiformis* Burm.). *J Agric Food Chem* 2007; 55(26): 10579–10584. doi:10.1021/jf071173b
- [36] Hu H-B, Zheng X-D, Cao H. Xanthone O-glycosides from the roots of *Pteris multifida*. *J Chin Chem Soc* 2006; 53(2): 459–64. doi:10.1002/jccs.200600060
- [37] Finnie JW, Windsor PA, Kessell AE. Neurological diseases of ruminant livestock in Australia. II: Toxic disorders and nutritional deficiencies. *Aust Vet J* 2011; 89(7): 247–253. doi:10.1111/j.1751-0813.2011.00793.x
- [38] McCleary B V, Chick BF. The purification and properties of a thiaminase I enzyme from nardoo (*Marsilea drummondii*). *Phytochemistry* 1977; 16(2): 207–213. doi:10.1016/S0031-9422(00)86787-4
- [39] Yamada K, Ojika M, Kigoshi H. Ptaquiloside, the major toxin of bracken, and related terpene glycosides: Chemistry, biology and ecology. Nat Prod Rep 2007; 24: 798–813. doi:10.1039/b614160a
- [40] Agnew MP, Lauren DR. Determination of ptaquiloside in bracken fern (*Pteridium esculentum*). *J Chromatogr A* 1991; 538(2): 462–468. doi:10.1016/S0021-9673(01)88870-1
- [41] Rasmussen LH, Lauren DR, Smith BL, Hansen HCB. Variation in ptaquiloside content in bracken (*Pteridium esculentum* (Forst. f) Cockayne) in New Zealand. N Z Vet J 2008; 56(6): 304–309. doi:10.1080/00480169.2008.36851
- [42] Fletcher MT, Brock IJ, Reichmann KG, McKenzie RA, Blaney BJ. Norsesquiterpene glycosides in bracken ferns (*Pteridium esculentum* and *Pteridium aquilinum* subsp. wightianum) from eastern Australia: reassessed poisoning risk to animals. *J Agric Food Chem* 2011; 59(9): 5133–5138. doi:10.1021/jf104267c
- [43] Potter DM, Baird MS. Carcinogenic effects of ptaquiloside in bracken fern and related compounds. Br J Cancer 2000; 83(7): 914–920. doi:10.1054/bjoc.2000.1368
- [44] Saito K, Nagao T, Takatsuki S, Koyama K, Natori S. The sesquiterpenoid carcinogen of bracken fern, and some analogues, from the pteridaceae. *Phytochemistry* 1990; 29(5): 1475–1479. doi:10.1016/0031-9422(90)80104-0
- [45] Kovganko N V, Kashkan ZN, Krivenok SN. Bioactive compounds of the flora of Belarus. 4. Pterosins A and B from *Pteridium aquili-num*. *Chem Nat Compd* 2004; 40(3): 227–229. doi:10.1023/ B:CONC.0000039129.46159.ce
- [46] Micheloud JF, Colque-Caro LA, Martinez OG, Gimeno EJ, da Silva Freitas Ribeiro D, Blanco BS. Bovine enzootic haematuria from consumption of *Pteris deflexa* and *Pteris plumula* in northwestern

- Argentina. *Toxicon* 2017; 134: 26–29. doi:10.1016/j.toxicon. 2017.05.023
- [47] Huang YL, Yeh PY, Shen CC, Chen CC. Antioxidant flavonoids from the rhizomes of *Helminthostachys zeylanica*. *Phytochemistry* 2003; 64(7): 1277–1283. doi:10.1016/j.phytochem.2003.09.009
- [48] Huang YL, Shen CC, Shen YC, Chiou WF, Chen CC. Antiinflammatory and antiosteoporosis flavonoids from the rhizomes of *Helminthostachys zeylanica*. *J Nat Prod* 2017; 80(2): 246–253. doi:10.1021/acs.jnatprod.5b01164
- [49] Chen CC, Huang YL, Yeh PY, Ou JC. Cyclized geranyl stilbenes from the rhizomes of *Helminthostachys zeylanica*. *Planta Med* 2003; 69(10): 964–967. doi:10.1055/s-2003-45112
- [50] Li S, Wang P, Deng G, Yuan W, Su Z. Cytotoxic compounds from invasive giant salvinia (*Salvinia molesta*) against human tumor cells. *Bioorg Med Chem Lett* 2013; 23(24): 6682–6687. doi:10.1016/j.bmcl.2013.10.040
- [51] Lima S, Diaz G, Diaz MAN. Antibacterial chemical constituent and antiseptic herbal soap from Salvinia auriculata Aubl. Evid Based Complement Alternat Med 2013; 2013: 480509. doi:10.1155/2013/480509
- [52] Purgato GA, Lima S, Baeta JVPB, Pizziolo VR, de Souza GN, Diaz-Muñoz G, et al. Salvinia auriculata: chemical profile and biological activity against Staphylococcus aureus isolated from bovine mastitis. Braz J Microbiol 2021; 52: 2401–2411. doi:10.1007/s42770-021-00595-z
- [53] Deans BJ, De Salas M, Smith JA, Bissember AC. Natural products isolated from endemic Tasmanian vascular plants. *Aust J Chem* 2018; 71(10): 756–767. doi:10.1071/CH18283
- [54] Tanaka N, Yuhara H, Wada H, Murakami T, Cambie RC, Braggins JE. Phenolic constituents of *Pteridium esculentum*. *Phytochemistry* 1993; 32(4): 1037–1039. doi:10.1016/0031-9422(93)85251-L
- [55] Saito T, Yamane H, Murofushi N, Takahashi N, Phinney BO. 4-O-Caffeoylshikimic and 4-O-(p-coumaroyl)shikimic acids from the dwarf tree fern, Dicksonia antarctica. Biosci Biotechnol Biochem 1997; 61(8): 1397–1398. doi:10.1271/bbb.61.1397
- [56] Deans BJ, Just J, Chhetri J, Burt LK, Smith JN, Kilah NL, et al. Pressurized hot water extraction as a viable bioprospecting tool: isolation of coumarin natural products from previously unexamined *Correa* (Rutaceae) species. *ChemistrySelect* 2017; 2(8): 2439–2443. doi:10.1002/slct.201602006
- [57] Numata A, Hokimoto K, Takemura T, Katsuno T, Yamamoto K. Plant constituents biologically active to insects. V. Antifeedants for the larvae of the yellow butterfly, *Eurema hecabe* mandarina, in *Osmunda japonica*. *Chem Pharm Bull* 1984; 32(7): 2815–2820. doi:10.1248/cpb.32.2815
- [58] Numata A, Takahashi C, Fujiki R, Kitano E, Kitajima A, Takemura T. Plant constituents biologically active to insects. VI. Antifeedants for larvae of the yellow butterfly, *Eurema hecabe* mandarina, in Osmunda japonica. (2). *Chem Pharm Bull* 1990; 38(10): 2862–2865. doi:10.1248/cpb.38.2862
- [59] Yu YM, Yang JS, Peng CZ, Caer V, Cong PZ, Zou ZM, et al. Lactones from Angiopteris caudatiformis. J Nat Prod 2009; 72(5): 921–924. doi:10.1021/np900027m
- [60] Chen Y, Tao Y, Lian X, Wang L, Zhao Y, Jiang J, et al. Chemical constituents of Angiopteris esculenta including two new natural lactones. Food Chem 2010; 122(4): 1173–1175. doi:10.1016/j. foodchem.2010.03.111
- [61] Kamitakahara H, Okayama T, Praptiwi, Agusta A, Tobimatsu Y, Takano T. Two-dimensional NMR analysis of Angiopteris evecta rhizome and improved extraction method for angiopteroside. Phytochem Anal 2019; 30(1): 95–100. doi:10.1002/pca.2794
- [62] Hwang T-H, Kashiwada Y, Nonaka G-i, Nishioka I. 4-Carboxymethyl flavan-3-ols and procyanidins from *Davallia divaricata*. *Phytochemistry* 1990; 29(1): 279–282. doi:10.1016/0031-9422(90)89050-J
- [63] Chen Y-H, Chang F-R, Lin Y-J, Hsieh P-W, Wu M-J, Wu Y-C. Identification of antioxidants from rhizome of *Davallia solida*. *Food Chem* 2008; 107(2): 684–691. doi:10.1016/j.foodchem. 2007.08.066
- [64] Chang X, Li W, Koike K, Wu L, Nikaido T. Phenolic constituents from the rhizomes of *Dryopteris crassirhizoma*. *Chem Pharm Bull* 2006; 54(5): 748–750. doi:10.1248/cpb.54.748

[65] Wada H, Kido T, Tanaka N, Murakami T, Saiki Y, Chen C-M. Chemical and chemotaxonomical studies of ferns. LXXXI. Characteristic lignans of Blechnaceous ferns. Chem Pharm Bull 1992; 40(8): 2099–2101. doi:10.1248/cpb.40.2099

- [66] Wang CZ, Davin LB, Lewis NG. Stereoselective phenolic coupling in *Blechnum spicant*: formation of 8–2' linked (–)-cisblechnic, (–)-trans-blechnic and (–)-brainic acids. *Chem Commun* 2001; 1: 113–114. doi:10.1039/b0081740
- [67] Maier VP, Metzler DM, Huber AF. 3-O-Caffeoylshikimic acid (Dactyllifric acid) and its isomers, a new class of enzymic browning substrates. *Biochem Biophys Res Commun* 1964; 14(2): 124–128. doi:10.1016/0006-291x(64)90241-4
- [68] Fukuoka M. Chemical and toxicological studies on bracken fern, Pteridium aquilinum var. latiusculum. VI. Isolation of 5-Ocaffeoylshikimic acid as an antithiamine factor. Chem Pharm Bull 1982; 30(9): 3219–3224. doi:10.1248/cpb.30.3219
- [69] Veit M, Weidner C, Strack D, Wray V, Witte L, Czygan FC. The distribution of caffeic acid conjugates in the equisetaceae and some ferns. *Phytochemistry* 1992; 31(10): 3483–3485. doi:10.1016/0031-9422(92)83711-7
- [70] Simmler C, Nikolić D, Lankin DC, Yu Y, Friesen JB, Van Breemen RB, et al. Orthogonal analysis underscores the relevance of primary and secondary metabolites in licorice. J Nat Prod 2014; 77(8): 1806–1816. doi:10.1021/np5001945
- [71] Li G, Nikolic D, van Breemen RB. Identification and chemical standardization of licorice raw materials and dietary supplements using UHPLC-MS/MS. *J Agric Food Chem* 2016; 64(42): 8062–8070. doi:10.1021/acs.jafc.6b02954
- [72] Nakabayashi T. Isolation of astragalin and isoquercitrin from bracken, *Pteridium aquilinum*. J Agric Chem Soc Jpn 1955; 19(2): 104–109.
- [73] Kazuma K, Noda N, Suzuki M. Malonylated flavonol glycosides from the petals of *Clitoria ternatea*. *Phytochemistry* 2003; 62(2): 229–237. doi:10.1016/s0031-9422(02)00486-7
- [74] Oikawa T, Hosoyama K, Hiraga Y, Kurono G, Takemoto T. The constituents of Osmunda spp. II.1) A new flavonoid glycoside of Osmunda asiatica. Chem Pharm Bull 2002; 43: 2091.
- [75] Han JT, Bang MH, Chun OK, Kim DO, Lee CY, Baek NI. Flavonol glycosides from the aerial parts of *Aceriphyllum rossii* and their antioxidant activities. *Arch Pharm Res* 2004; 27(4): 390–395. doi:10.1007/BF02980079
- [76] Deng S, Deng Z, Fan Y, Peng Y, Li J, Xiong D, et al. Isolation and purification of three flavonoid glycosides from the leaves of Nelumbo nucifera (Lotus) by high-speed counter-current chromatography. J Chromatogr B Anal Technol Biomed Life Sci 2009; 877(24): 2487–2492. doi:10.1016/j.jchromb.2009.06.026
- [77] Mandalari G, Tomaino A, Arcoraci T, Martorana M, Turco VL, Cacciola F, et al. Characterization of polyphenols, lipids and dietary fibre from almond skins (Amygdalus communis L.). J Food Compos Anal 2010; 23(2): 166–174. doi:10.1016/j.jfca.2009.08.015
- [78] Yao H, Liao ZX, Wu Q, Lei GQ, Liu ZJ, Chen DF, et al. Antioxidative flavanone glycosides from the branches and leaves of *Viscum coloratum*. Chem Pharm Bull 2006; 54(1): 133–135. doi:10.1248/cpb.54.133
- [79] Malejko J, Nalewajko-Sieliwoniuk E, Nazaruk J, Siniło J, Kojło A. Determination of the total polyphenolic content in *Cirsium palustre* (L.) leaves extracts with manganese (IV) chemiluminescence detection. *Food Chem* 2014; 152: 155–161. doi:10.1016/j.foodchem.2013.11.138
- [80] Mun'im A, Negishi O, Ozawa T. Antioxidative compounds from Crotalaria sessiliflora. Biosci Biotechnol Biochem 2003; 67(2): 410–414. doi:10.1271/bbb.67.410
- [81] Cioffi G, Escobar LM, Braca A, De Tommasi N. Antioxidant chalcone glycosides and flavanones from *Maclura (Chlorophora) tinctoria*. J Nat Prod 2003; 66(8): 1061–1064. doi:10.1021/ np030127c
- [82] Ismaili H, Sosa S, Brkic D, Fkih-Tetouani S, Ilidrissi A, Touati D, et al. Topical anti-inflammatory activity of extracts and compounds from *Thymus broussonettii*. J Pharm Pharmacol 2010; 54(8): 1137–1140. doi:10.1211/002235702320266316
- [83] Jarrett JM, Williams AH. The flavonoid glycosides of Salix purpurea. Phytochemistry 1967; 6(1960): 1585–1586. doi:10.1016/ S0031-9422(00)82955-6

- [84] Hu Q, Zhang DD, Wang L, Lou H, Ren D. Eriodictyol-7-O-glucoside, a novel Nrf2 activator, confers protection against cisplatin-induced toxicity. Food Chem Toxicol 2012; 50(6): 1927–1932. doi:10.1016/j.fct.2012.03.059
- [85] Pan J, Zhang S, Yan L, Tai J, Xiao Q, Zou K, *et al.* Separation of flavanone enantiomers and flavanone glucoside diastereomers from *Balanophora involucrata* Hook. f. by capillary electrophoresis and reversed-phase high-performance liquid chromatography on a C18 column. *J Chromatogr A* 2008; 1185(1): 117–129. doi:10.1016/j.chroma.2008.01.049
- [86] Ma C, Zhou Y, Liu AR. Determination of chlorogenic acid and eriodictyol-7-O-β-D-glucuronide in Pyrrosia by RP-HPLC. Yaoxue Xuebao 2003; 38(4): 286–289.
- [87] Jing X, Ren D, Wei X, Shi H, Zhang X, Perez RG, *et al.* Eriodictyol-7-O-glucoside activates Nrf2 and protects against cerebral ischemic injury. *Toxicol Appl Pharmacol* 2013; 273(3): 672–679. doi:10.1016/j.taap.2013.10.018
- [88] Chen R, Qi QL, Wang MT, Li QY. Therapeutic potential of naringin: an overview. *Pharm Biol* 2016; 54(12): 3203–3210. doi:10.1080/13880209.2016.1216131
- [89] Bharti S, Rani N, Krishnamurthy B, Arya DS. Preclinical evidence for the pharmacological actions of naringin: A review. *Planta Med* 2014; 80(6): 437–451. doi:10.1055/s-0034-1368351
- [90] Abeysinghe DC, Li X, Sun CD, Zhang WS, Zhou CH, Chen KS. Bioactive compounds and antioxidant capacities in different edible tissues of citrus fruit of four species. *Food Chem* 2007; 104(4): 1338–1344. doi:10.1016/j.foodchem.2007.01.047
- [91] Alam MA, Subhan N, Rahman MM, Uddin SJ, Reza HM, Sarker SD. Effect of citrus flavonoids, naringin and naringenin, on metabolic syndrome and their mechanisms of action. Adv Nutr 2014; 5(4): 404–417. doi:10.3945/an.113.005603
- [92] Yusof S, Ghazali HM, King GS. Naringin content in local citrus fruits. Food Chem 1990; 37(2): 113–121. doi:10.1016/0308-8146(90)90085-I
- [93] Jagetia GC, Venkatesha VA, Reddy TK. Naringin, a citrus flavonone, protects against radiation-induced chromosome damage in mouse bone marrow. *Mutagenesis* 2003; 18(4): 337–343. doi:10.1093/mutage/geg001
- [94] Sharma S, Ali A, Ali J, Sahni JK, Baboota S. Rutin: Therapeutic potential and recent advances in drug delivery. Expert Opin Investig Drugs 2013; 22(8): 1063–1079. doi:10.1517/13543784. 2013.805744
- [95] Chua LS. A review on plant-based rutin extraction methods and its pharmacological activities. *J Ethnopharmacol* 2013; 150(3): 805–817. doi:10.1016/j.jep.2013.10.036
- [96] Imperato F. A flavanone glycoside from the fronds of Ceterach officinarum. Phytochemistry 1983; 22(1): 312–313. doi:10.1016/ S0031-9422(00)80121-1
- [97] Socolsky C, Salvatore A, Asakawa Y, Bardón A. Bioactive new bitter-tasting p-hydroxystyrene glycoside and otherconstituents from the fern *Elaphoglossum spathulatum*. *Arkivoc* 2003; 2003: 347–355. doi:10.3998/ark.5550190.0004.a32
- [98] Yang T-H, Lee Y-C, Chung H-N. Constitutents of Drynaria fortunei. Taiwan Yaoxue Zazhi 1966; 18(1): 38–40.
- [99] De Britto J, Manickam VS, Gopalakrishnan S, Ushioda T, Tanaka N. Determination of aglycone chirality in dihydroflavonol 3-O-a-L-Rhamnosdes by 1H-NMR spectroscopy. *Chem Pharm Bull* 1995; 43(2): 338–339. doi:10.1248/cpb.43.338
- [100] Grochowski DM, Locatelli M, Granica S, Cacciagrano F, Tomczyk M. A review on the dietary flavonoid tiliroside. Compr Rev Food Sci Food Saf 2018; 17(5): 1395–1421. doi:10.1111/1541-4337.12389
- [101] Kuroyanagi M, Fukuoka M, Yoshihira K, Natori S, Yamasaki K. Confirmation of the structure of tiliroside, an acylated kaemferol glycoside, by <sup>13</sup>C-nuclear magnetic resonance. *Chem Pharm Bull* 1978; 26(11): 3594–3596. doi:10.1248/cpb.26.3594
- [102] Wang C-Y, Pamukcu AM, Bryan GT. Isolation of fumaric acid, succinic acid, astragalin, isoquercitrin and tiliroside from Pteridium auilinum. Phytochemistry 1973; 12: 2298–2299. doi:10.1016/0031-9422(73)85140-4
- [103] Devi S, Kumar V. Comprehensive structural analysis of cis- and trans-tiliroside and quercetrin from *Malvastrum coromandelia-num* and their antioxidant activities. *Arab J Chem* 2020; 13(1): 1720–1730. doi:10.1016/j.arabjc.2018.01.009

- [104] Qiao W, Zhao C, Qin N, Zhai HY, Duan HQ. Identification of trans-tiliroside as active principle with anti-hyperglycemic, antihyperlipidemic and antioxidant effects from *Potentilla chinesis*. *J Ethnopharmacol* 2011; 135(2): 515–521. doi:10.1016/j.jep. 2011.03.062
- [105] Velagapudi R, Aderogba M, Olajide OA. Tiliroside, a dietary glycosidic flavonoid, inhibits TRAF-6/NF-κB/p38-mediated neuroinflammation in activated BV2 microglia. *Biochim Biophys Acta - Gen Subj* 2014; 1840(12): 3311–3319. doi:10.1016/j.bbagen.2014.08.008
- [106] Zhu Y, Zhang Y, Liu Y, Chu H, Duan H. Synthesis and biological activity of trans-tiliroside derivatives as potent anti-diabetic agents. *Molecules* 2010; 15(12): 9174–9183. doi:10.3390/ molecules15129174
- [107] Just J, Deans BJ, Olivier WJ, Paull B, Bissember AC, Smith JA. New method for the rapid extraction of natural products: efficient isolation of shikimic acid from star anise. *Org Lett* 2015; 17(10): 2428–2430. doi:10.1021/acs.orglett.5b00936
- [108] Ho CC, Deans BJ, Just J, Warr GG, Wilkinson S, Smith JA, et al. Employing pressurized hot water extraction (PHWE) to explore natural products chemistry in the undergraduate laboratory. J Vis Exp 2018; 141: 1–7. doi:10.3791/58195
- [109] Deans BJ, Just J, Smith JA, Bissember AC. Development and applications of water-based extraction methods in natural products isolation chemistry. *Asian J Org Chem* 2020; 9(8): 1144–1153. doi:10.1002/ajoc.202000210
- [110] Fulmer GR, Miller AJM, Sherden NH, Gottlieb HE, Nudelman A, Stoltz BM, *et al.* NMR chemical shifts of trace impurities: Common laboratory solvents, organics, and gases in deuterated solvents relevant to the organometallic chemist. *Organometallics* 2010; 29(9): 2176–2179. doi:10.1021/om100106e

- [111] Chang EJ, Lee WJ, Cho SH, Choi SW. Proliferative effects of flavan-3-ols and propelargonidins from rhizomes of *Drynaria fortunei* on MCF-7 and osteoblastic cells. *Arch Pharm Res* 2003; 26(8): 620–630. doi:10.1007/BF02976711
- [112] Akiyama T, Yamada M, Yamada T, Maitani T. Naringin glycosides α-glucosylated on ring B found in the natural food additive, enzymatically modified naringin. *Biosci Biotechnol Biochem* 2000; 64(10): 2246–2249. doi:10.1271/bbb.64.2246
- [113] Davin LB, Wang CZ, Helms GL, Lewis NG. [13C]-Specific labeling of 8-2' linked (-)-cis-blechnic, (-)-trans-blechnic and (-)-brainic acids in the fern *Blechnum spicant*. *Phytochemistry* 2003; 62(3): 501–511. doi:10.1016/s0031-9422(02)00540-x
- [114] Matsuura H, Amano M, Kawabata J, et al. Isolation and measurement of quercetin glucosides in flower buds of japanese butterbur (Petasites japonicus subsp. Gigantea kitam.). Biosci Biotechnol Biochem 2002; 66(7): 1571–1575. doi:10.1271/bbb. 66.1571
- [115] Lee JH, Ku CH, Baek N-I, Kim S-H, Park HW, Kim DK. Phytochemical constituents from *Diodia teres. Arch Pharm Res* 2004; 27(1): 40–43. doi:10.1007/BF02980043
- [116] Zor M, Aydin S, Güner ND, Başaran N, Başaran AA. Antigenotoxic properties of *Paliurus spina-christi* Mill fruits and their active compounds. *BMC Complement Altern Med* 2017; 17(1): 229. doi:10.1186/s12906-017-1732-1
- [117] Kaouadji M, Morand J-M, Garcia J. Further acylated Kaempferol rhamnosides from *Platanus acerifolia* buds. *J Nat Prod* 1993; 56(9): 1618–1621. doi:10.1021/np50099a027
- [118] Calzada F, Lopéz R, Meckes M, Cedillo-Rivera R. Flavonoids of the aerial parts of *Helianthemum glomeratum*. *Pharm Biol* 1995; 33(4): 351–352. doi:10.3109/13880209509065393

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