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# Thiocyanate biodegradation: harnessing microbial metabolism for mine remediation



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Thiocyanate (SCN<sup>-</sup>) forms in the reaction between cyanide (CN<sup>-</sup>) and reduced sulfur species, e.g. in gold ore processing and coal-coking wastewater streams, where it is present at millimolar (mM) concentrations<sup>1</sup>. Thiocyanate is also present naturally at nM to  $\mu$ M concentrations

in uncontaminated aquatic environments<sup>2</sup>. Although less toxic than its precursor CN<sup>-</sup>, SCN<sup>-</sup> can harm plants and animals at higher concentrations<sup>3</sup>, and thus needs to be removed from wastewater streams prior to disposal or reuse. Fortunately, SCN<sup>-</sup> can be biodegraded by

microorganisms as a supply of reduced sulfur and nitrogen for energy sources, in addition to nutrients for growth<sup>4</sup>. Research into how we can best harness the ability of microbes to degrade SCN<sup>-</sup> may offer newer, more costeffective and environmentally sustainable treatment solutions<sup>5</sup>. By studying biodegradation pathways of SCN<sup>-</sup> in laboratory and field treatment bioreactor systems, we can also gain fundamental insights into connections across the natural biogeochemical cycles of carbon, sulfur and nitrogen<sup>6</sup>.

#### Thiocyanate: a common wastewater contaminant

Thiocyanate (SCN) is a common contaminant associated with a range of industries, and is typically found at its highest concentrations in the wastewater of gold and silver mines as a result of the use of CN<sup>-</sup> as a lixiviant during ore processing<sup>7</sup>. Thiocyanate is also commonly a component of coal coking wastewater alongside phenol and  $CN^{-8}$ . These typically voluminous waste streams pose a serious environmental hazard due to their persistence and toxicity. Although a number of chemical SCN<sup>-</sup> degradation techniques exist, they are often inefficient and expensive<sup>9</sup>. Alternatively, many mines choose to store contaminated tailings indefinitely in dam structures, and re-use the SCN- contaminated water during ore processing. However, the presence of SCN<sup>-</sup> in this re-used water is known to impact gold extraction efficiency negatively<sup>10</sup>, as well as to impede the metabolism of biomining microorganisms<sup>11</sup>. Improved SCN<sup>-</sup> treatment systems, therefore, offer an opportunity improve the sustainability of mining processes globally.

# Diversity of thiocyanate-degrading microorganisms

Thiocyanate offers a rich source of growth nutrients and energy to microorganisms, in the form of reduced sulfur and nitrogen, and a number of microbial species are known to be capable of SCNdegradation<sup>4</sup>. Importantly, these species do not belong to a distinct phylogenetic group, and the presence/absence of SCN-degrading potential is often even strain specific. This complicates their identification using phylogenetic markers, such as the 16S rRNA gene. Much of what is currently known has, therefore, been achieved through culturing experiments. These studies have revealed diverse metabolic traits associated with SCN<sup>-</sup> degradation, where chemolithotrophs utilise the reduced sulfur as an energy source<sup>12-</sup>  $^{-14}$ , and heterotrophs utilise the nitrogen as a growth nutrient  $^{15-17}$ . Despite SCN<sup>-</sup> biodegradation being widely regarded as an aerobic process, one bacterium (Thialkalivibrio thiocyanodenitrificans) was found to be able to couple this process to nitrate reduction $^{14}$ , opening up the possibility of anaerobic SCN<sup>-</sup> degradation.

Experimentation on culturable strains has also allowed the elucidation of two distinct pathways of SCN<sup>-</sup> degradation. These pathways proceed via two intermediates, carbonyl sulfide (COS) or cyanate (CNO<sup>-</sup>), catalysed by one of two distinct types of SCN<sup>-</sup> hydrolases (SCNase)<sup>18,19</sup>, or by an SCN<sup>-</sup> dehydrogenase (TcDH) enzyme<sup>20,21</sup>, respectively. The resulting COS and CNO<sup>-</sup> intermediate chemical species are then available for degradation by enzymes associated with  $\beta$ -carbonic hydrase<sup>22</sup> or cyanate anhydrase<sup>23</sup>. Both of these SCN<sup>-</sup> biodegradation pathways result in the release of reduced sulfur (S<sup>2-</sup> or S<sup>0</sup>), NH<sub>4</sub><sup>+</sup> and CO<sub>2</sub>.

The advent of high-throughput sequencing, and the rise in genome/metagenome and proteome/metaproteome sequences, has vielded vast databases of protein and DNA sequences. Significantly, protein sequences for the three known SCN-degrading enzymes are available, which allow a deeper look into the distribution of SCN<sup>-</sup>degrading enzymes. The three-subunit SCNase enzyme, originally isolated in *Thiobacillus thioparus* THI115<sup>18</sup>, is the most widely identified enzyme in protein databases, with 416 sequences identified as the y-subunit in the NCBI nr database. These sequences primarily belong to the Actinobacteria, due to a bias towards full genome sequences of the medically significant Mycobacterium genus. They also contain sequences belonging to a number of thiobacilli, sulfur-oxidising  $\gamma$ -proteobacteria (including a number of Chromatiales) and  $\alpha$ -proteobacteria (*Methylobacter*ium spp. and Sphingomonas spp.). The alternative SCNase and the TcDH have far fewer closely related sequences in the NCBI non-redundant protein sequence database. This SCNase has representatives encoded in a number of thiobacilli, Afipia spp. and sulfuroxidising γ-proteobacteria. The TcDH, originally purified from two Thioalkalivibrio spp.<sup>21</sup>, appears to be closely related to other Thioakalivibrio, and more distantly related to proteins of unknown function in Thioploca ingrica, Nitrospirae and Hydrogenobacter thermophilus. The comparatively few metaproteome and metagenome studies, targeting SCN--contaminated systems, limit our understanding of the true scope of the distribution of these enzymes.

# Thiocyanate biodegradation triggers complex microbial community interactions

Although SCN<sup>-</sup> degradation is undertaken by a limited number of bacterial species (or strains), wider implications can result for the whole microbial community due to the roles its constituent elements can play as biological energy sources or growth nutrients. As a result, interesting and potentially useful symbiotic or dependent relationships can develop between SCN<sup>-</sup> degraders and non-SCN<sup>-</sup> degraders (Figure 1).



Figure 1. Thiocyanate biodegradation can trigger a complex series of interactions within a microbial community revolving around the cycling of carbon, nitrogen and sulfur. The initial degradation is primarily an aerobic process mediated by sulfur-oxidizing chemolithotrophs capable of  $CO_2$  fixation. These primary producers are able to supply recycled organic carbon ( $C_{org}$ ) to a diverse population of heterotrophic bacteria. The released  $NH_4^+$  is available as a growth nutrient to the community or an energy source to nitrifying bacteria.

Thiocyanate-degrading bioreactors are often dominated by SCN-degrading and sulfur-oxidising strains of the thiobacilli<sup>24,25</sup>. Although autotrophic, these bacteria have also been found to dominate in systems fed with a labile carbon source<sup>26,27</sup>. Their dominance is a testament to the strong selective pressure that SCN<sup>-</sup> exerts on a microbial community. The presence in these systems of a number of sulfur-oxidising taxa that do not genetically encode for known SCN-degrading enzymes suggests a fraction of the reduced sulfur is shared within the community. The eventual liberation of NH4<sup>+</sup> from SCN<sup>-</sup> degradation also opens up potential cross-feeding relationships. Several SCN-treating systems were found to sustain populations of nitrifying bacteria, utilising NH<sub>4</sub><sup>+</sup> as an energy source and lacking the ability to degrade  $SCN^{-24,26}$ . The subsequent nitrate can also supply an electron acceptor for denitrification in anoxic regions of these systems<sup>26</sup>. The supply of  $NH_4^+$  as a nitrogen-containing nutrient likely provides the primary source of nitrogen for growth of microbial communities inhabiting these systems<sup>25</sup>. This nitrogen can also support phytoplankton growth under phototrophic conditions<sup>28</sup>. These symbiotic relationships have the potential to be harnessed for the effective removal of nitrogen from wastewater following SCN<sup>-</sup> degradation.

As SCN<sup>-</sup> degradation is able to support dominant populations of chemolithotrophs, it has the potential to exert a large influence on carbon cycling in SCN<sup>-</sup> treatment systems. Indeed, the SCN<sup>-</sup> degrading thiobacilli and *Thioalkalivibrio* are known to utilise the Calvin-Benson-Bassham cycle to fix  $CO_2^{21,26}$ . In the absence of a labile carbon source, the sulfur- and nitrogen-oxidising chemolithotrophs can provide the primary source of carbon in the system. Our own work, for example, found that a SCN<sup>-</sup> treatment system

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operated without labile carbon input, and was dominated by chemolithotrophs, sustaining SCN<sup>-</sup> degradation and a considerable diversity of heterotrophic microorganisms<sup>25</sup>. This study also implicated a role for bacterivorous Amoebazoa in preying on the microbial community and closing the microbial loop in the system.

# Implications for thiocyanate-degrading biotechnology

Microbial communities capable of SCN<sup>-</sup> degradation offer an opportunity to treat large quantities of wastewater effectively. Such systems, adopting various designs, have already been deployed with success at field scale<sup>24,29,30</sup>. Much research is now going into understanding these systems at the molecular scale, and developing more efficient processes. Understanding the aforementioned cross-feeding relationships within SCN<sup>-</sup> treatment systems are vitally important to this process improvement. A good example is the revelation, achieved through a genome-resolved metagenomics approach, that even in organic carbon-fed systems, the community can be dominated by SCN-degrading autotrophic bacteria<sup>26</sup>. Our own work has proven experimentally that SCN-degrading bioreactors can be operated in the absence of organic carbon amendments<sup>25</sup>. This autotrophic SCN<sup>-</sup> degradation approach, combined with nitrification and denitrification, has been deployed at pilot scale to treat contaminated groundwater at Stawell Gold Mines in Victoria (Figure 2). This process is able to completely degrade the influent SCN<sup>-</sup> (300–400 mg L<sup>-1</sup>), resulting in sulfate and N-containing products. By harnessing various metabolic niches the initially released  $NH_4^+$  is nitrified to nitrate and subsequently removed by denitrification. In addition to this



Figure 2. A pilot-scale engineered approach to the biological treatment of SCN<sup>-</sup>-contaminated groundwater at a Victorian gold mine. The bioreactors contain autotrophic and heterotrophic microbial communities capable of degrading SCN<sup>-</sup>, complete nitrification and denitrification. Photo taken by M. Watts.

engineered approach, we reported for the first time the ability to promote *in situ* SCN<sup>-</sup> biodegradation in contaminated gold mine tailings water held in large open air storage facilities<sup>9</sup>. This process was promoted through phosphate nutrient addition alone, and offers a passive approach for the treatment of large quantities of contaminated material. Our future work aims to better resolve the active metabolic interactions within the microbial communities in these *in situ* and *ex situ* approaches. This will help us to promote beneficial symbiotic relationships within the system.

Collectively, the advances in our understanding of the metabolisms underpinning SCN<sup>-</sup> biodegradation allow for better designs and approaches to harnessing this microbial potential. Biodegradation of SCN<sup>-</sup>, therefore, is offering a route to improving the water efficiency of industrial processes such as gold mining on a global scale.

In conclusion, much has been learned at the molecular scale about biodegradation of SCN<sup>-</sup> since this metabolic trait was first reported in *T. thioparus*<sup>12</sup>, including the enzymes responsible and the community wide biogeochemical impacts. The advent of 'multiomics' approaches is allowing us to probe these processes *in situ* in SCN<sup>-</sup>-biodegrading treatment systems. The few studies utilising these techniques have spurred significant process improvements, in addition to revealing fundamental insights into SCN<sup>-</sup> biodegradation and the subsequent metabolic cycling of its breakdown products. These insights, although gained from engineered systems, can help inform the global cycling of SCN<sup>-</sup>, CNO<sup>-</sup> and COS and better constrain the role of microbial communities in the wider carbon, nitrogen and sulfur cycles. Given the limited number of

systems investigated to this depth, more work is needed to appreciate fully the diversity and complexity of microbial communities degrading SCN<sup>-</sup>.

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