# Crystal Structure of Triclinic Colchiceine Hemihydrate* 

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## Abstract

Triclinic crystals of colchiceine hemihydrate, $\mathrm{C}_{21} \mathrm{H}_{23} \mathrm{NO}_{6} \cdot \frac{1}{2} \mathrm{H}_{2} \mathrm{O}$, belong to the space group $P 1$ with $a 8.211(1), b 8.361(1), c 16 \cdot 898(2) \AA, a 92.35(1)^{\circ}, \beta 93.93(1)^{\circ}, \gamma 121.35(1)^{\circ}$ and $Z 2$. The structure was solved by direct methods and successive difference syntheses with diffractometer data measured with $\mathrm{Cu} \mathrm{K} \alpha$ radiation. Refinement converged at $R 0.053$ for 2798 observed reflections. The two tautomers, one of the isocolchiceine and the other of the colchiceine form, exist as independent entities in the crystal; the angles between the normals to the plane of the benzene and troponoid rings are $44.5(4)^{\circ}$ and $51.5(5)^{\circ}$ respectively. The hydroxyl and N -acetyl groups are oriented to form channels along the $a$-axis which accommodate the water molecules.

## 1. Introduction

In view of the antimitotic properties of colchicine derivatives (Rösner et al. 1981) their molecular geometry is of considerable interest. For colchiceine, in which the methoxy substituent on the troponoid ring in cochicine has been replaced by a hydroxyl group, two tautomeric forms are possible. These are represented as (I) and (II),

(I)

(II)

[^0]and correspond to the isocolchicine and colchicine forms respectively. In crystals of colchiceine ethylacetate water solvate (Silverton 1979) and trimethylcolchinic acid (Margulis 1982) the generally accepted colchicine form (II) was found to exist. However, in the solid state, Iorio et al. (1978) found 7-oxodeacetamidocolchiceine to exist as an unusual 2:3 mixture of the tautomeric forms. More recently, Rius et al. (1984) have reported the crystal structure of 7 -deacetamidocolchiceine in which the moleules are in the isocolchicine form. From their results and those reported earlier by other authors, they suggested that the absence of any substituent at postion 7 in colchicine derivatives may favour the isocolchicine form. We now report the crystal structure of colchiceine hemihydrate in which there is a $1: 1$ mixture of the tautomers.

## 2. Experimental

Pale yellow needle-like crystals of colchiceine hemihydrate are triclinic and belong to the space group $P 1$. Accurate unit cell parameters were determined by the leastsquares method from $2 \theta$ values for 25 reflections ( $2 \theta$ range $35^{\circ}$ to $56^{\circ}$ ) measured on a diffractometer at $289(1) \mathrm{K}$ with $\mathrm{Cu} \mathrm{K} \alpha$ radiation ( $\bar{\lambda} 1.5418 \AA$ ). The crystal density was determined by flotation. Although the cell parameters are in reasonable agreement with those of Morrison (1951), his reported density $1.24 \mathrm{~g} \mathrm{~cm}^{-3}$ is too low.

## Crystal Data

$\mathrm{C}_{21} \mathrm{H}_{23} \mathrm{NO}_{6} \cdot \frac{1}{2} \mathrm{H}_{2} \mathrm{O}, M 394 \cdot 4$, triclinic, space group $P 1$, $a 8 \cdot 211(1), b 8 \cdot 361(1)$, $c$ 16.898(2) $\AA$, $\alpha 92.35(1)^{\circ}, \beta 93.93(1)^{\circ}, \gamma 121.35(1)^{\circ}, U 984.4(2) \AA^{3}, D_{\mathrm{m}} 1.33(1), D_{\mathrm{c}} 1.33 \mathrm{~g} \mathrm{~cm}^{-3}, Z 2$, $\boldsymbol{F}(000) 418, \mu 7.04 \mathrm{~cm}^{-1}(\mathrm{CuK} \mathrm{\alpha})$.

A crystal of approximate dimensions $0.30 \times 0 \cdot 14 \times 0.095 \mathrm{~mm}$ with faces ( 010 ), ( $0 \overline{1} 0$ ), ( 001 ), ( $00 \overline{1}$ ), (100) and ( $\overline{2} 10$ ) was used for the data collection. Intensity data were measured at 289(1) K on a Rigaku-AFC four-circle diffractometer with $\mathrm{Cu} \mathrm{K} \alpha$ radiation (graphite-crystal monochromator, $\bar{\lambda} 1.5418 \AA$ ). Intensities were recorded by an $\omega-2 \theta$ scan, with scan rate $2^{\circ} \mathrm{min}^{-1}$, scan range $\Delta \omega$ of $1 \cdot 2^{\circ}+1 \cdot 5^{\circ} \tan \theta$ and 10 s stationary background counts. Three reference reflections monitored every 50 reflections showed no significant variation in intensity during data collection. Data to a $2 \theta(\max )$ of $130^{\circ}(h-9$ to $9, k-9$ to $9, l 0$ to 19 ) yielded 3256 terms, and the 2798 for which $I \geqslant \sigma I_{0}$ were used for the structure refinement. The integrated intensities were corrected for Lorentz and polarization effects and analytical absorption corrections were made (transmission factors ranged from 0.902 to 0.941 ).

Attempts to solve the structure by vector superposition were unsuccessful, and the structure was solved by direct methods with multan 78 (Main et al. 1978) and by successive difference syntheses. The molecular skeletons could not be depicted from any of the $E$ maps although from most, the site of a plausible $\approx 10$-atom molecular fragment could be derived. A number of attempts were made to initiate structure solution, each based on one such fragment. Eventually, from the $E$ map calculated with 324 phased terms with $|E| \geqslant 1 \cdot 20$ (AbSFOM $1 \cdot 96$, PSIZERO $4 \cdot 64$, RESID $15 \cdot 2 \%$ ) peaks were assembled to correspond to sites feasible for a benzene ring and four associated atoms. Using this fragment, with the atoms included as C, structure solution

Table 1. Atomic coordinates for the non-hydrogen atoms and equivalent isotropic temperature factors ( $\mathrm{A}^{\mathbf{2}}$ )
The e.s.d. values for the atomic coordinates are given in parentheses and for $B_{\text {eq }}$ are $0 \cdot 1 \AA^{2}$; $B_{\text {eq }}$ is defined as $\frac{8}{3} \pi^{2} \Sigma_{i} \Sigma_{j} U_{i j} a_{i}^{*} a_{j}^{*} a_{i}, a_{j}$. Atoms in molecule (2) are denoted by the addition of 20 to their atomic numbers

| Atom | $10^{4} x$ | $10^{4} y$ | $10^{4} z$ | $B_{\text {eq }}$ | Atom | $10^{4} x$ | $10^{4} y$ | $10^{4} z$ | $B_{\text {eq }}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| C(1) | 1179(0) | -262(0) | -154(0) | 3.9 | C(21) | 4676(8) | 514(6) | 3112(3) | $4 \cdot 2$ |
| C(2) | -104(8) | -1600(6) | 301(3) | 4.4 | C(22) | 4397(8) | 1736(6) | 2662(3) | $4 \cdot 5$ |
| C(3) | -1710(8) | -1578(6) | 502(3) | 4.7 | C(23) | 2626(8) | 1552(6) | 2561(3) | $4 \cdot 6$ |
| C(4) | -2042(8) | -216(6) | 238(3) | $4 \cdot 9$ | C(24) | 1090(8) | 142(6) | 2909(3) | $4 \cdot 6$ |
| C(4a) | -737(6) | 1150(6) | -212(3) | $4 \cdot 3$ | C(24a) | 1346(6) | -1086(6) | 3359(3) | $4 \cdot 2$ |
| C(5) | -1104(8) | 2652(6) | -486(3) | $4 \cdot 8$ | C(25) | -321(6) | -2655(8) | 3730(3) | $5 \cdot 3$ |
| C(6) | -1687(6) | 2424(6) | -1373(3) | 4.6 | C(26) | -99(6) | -2215(8) | 4638(3) | $5 \cdot 2$ |
| C(7) | -518(6) | 1851(6) | -1855(3) | $3 \cdot 8$ | C(27) | 1966(6) | -1367(6) | 5005(3) | $4 \cdot 3$ |
| C(7a) | 1617(6) | 3070(6) | -1592(3) | $3 \cdot 5$ | C(27a) | 2793(6) | -2537(6) | 4729(3) | $4 \cdot 1$ |
| C(8) | 2710(6) | 4536(6) | -2067(3) | $4 \cdot 2$ | C(28) | 2819(8) | -3765(6) | 5240(3) | 5.1 |
| C(9) | 4681(6) | 5787(6) | -2054(3) | 4.8 | C(29) | 3450(9) | - 5026(6) | 5198(3) | 5.7 |
| C(10) | 6210(6) | 5788(6) | -1580(3) | 4.9 | C(30) | 4396(8) | -5270(6) | 4538(4) | 5.0 |
| C(11) | 5901(6) | 4809(6) | -892(3) | $4 \cdot 8$ | C(31) | 4648(8) | -4461(6) | 3837(3) | $5 \cdot 2$ |
| C(12) | 4219(6) | 3521(6) | -595(3) | $4 \cdot 3$ | C(32) | 4164(6) | -3177(6) | 3564(3) | 4.5 |
| C(12a) | 2297(6) | 2660(5) | -889(3) | 3.6 | C(32a) | 3384(6) | -2274(6) | 3940(3) | 4.0 |
| C(12b) | 875(6) | 1164(6) | -418(3) | $3 \cdot 8$ | C(32b) | 3131(6) | -928(6) | 3461(3) | 3.9 |
| C(13) | -1171(6) | 547(8) | -3243(3) | 5.0 | C(33) | 3588(9) | 398(9) | 6303(3) | 5.0 |
| C(14) | -1570(9) | 788(11) | -4103(3) | $7 \cdot 1$ | C(34) | 3517(10) | 431(11) | 7183(4) | 7.9 |
| C(1M) | 2443(9) | -1753(8) | -904(3) | $6 \cdot 3$ | C(21M) | 7568(9) | 1717(9) | 3929(3) | $6 \cdot 1$ |
| C(2M) | 1219(10) | -2671(8) | 1256(4) | 7.5 | C(22M) | 6205(9) | 2791(9) | 1565(3) | 6.4 |
| C(3M) | -4546(10) | -3043(9) | 1168(4) | 7.9 | C(23M) | 715(10) | 2724(9) | 2037(5) | 8.4 |
| O(1) | 2759(5) | -284(4) | -328(2) | $4 \cdot 8$ | O(21) | 6458(5) | 751(4) | 3192(2) | 4.6 |
| O(2) | 140(5) | -3039(4) | 525(2) | 5.0 | O(22) | 5947(5) | 3206(4) | 2360(2) | 5.4 |
| $\mathrm{O}(3)$ | -2845(6) | -2960(5) | 969(2) | 6.5 | O(23) | 2490(6) | 2801(5) | 2109(2) | 6.4 |
| O(9) | 5267(6) | 7045(6) | - 2577(3) | 7.7 | O(29) | 3264(8) | -6057(6) | 5761(3) | 6.0 |
| O(10) | 7904(5) | 6794(6) | -1779(3) | $7 \cdot 3$ | O(30) | 5040(6) | -6405(5) | 4656(3) | 6.8 |
| $\mathrm{O}(13)$ | -985(5) | -776(6) | -3066(2) | $6 \cdot 2$ | O(33) | 4990(6) | 1581(6) | 5990(2) | $6 \cdot 3$ |
| N(1) | -1027(6) | 1847(6) | -2703(2) | 4.5 | N(21) | 2054(6) | -1086(6) | 5869(3) | 5.0 |
| O(1s) | -1476(6) | 5033(6) | 6664(3) | $8 \cdot 2$ |  |  |  |  |  |

was initiated. After four successive difference syntheses, the sites of the non-hydrogen atoms in the two independent molecules, (1) and (2), were located. An isolated peak of significant height was assumed to be the site of an O atom of an unsuspected water molecule; this was proved correct on refinement. Although there were maxima in subsequent difference maps at the expected H -atom sites, not all were clearly resolved. The H atoms therefore, apart from those of the hydroxyl groups and water molecule which were omitted from the analysis, were included at idealized positions. For the final refinement, anisotropic temperature factors were given to the $\mathrm{C}, \mathrm{N}$ and O atoms and common isotropic temperature factors were given to the methyl and non-methyl H atoms; the latter refined to the respective $B$ values of $13 \cdot 3(7)$ and $6 \cdot 7(4) \AA^{2}$. The parameters were refined in two blocks, each block containing parameters for one molecule. At convergence, we found $R=0.053$ (defined as $\Sigma \Delta F / \Sigma\left|F_{0}\right|$ where $\left.\Delta F=\left|F_{0}\right|-\left|F_{\mathrm{c}}\right|\right), R_{w}=0.052$ (defined as $\left.\left(\Sigma w(\Delta F)^{2} / \Sigma w\left|F_{0}\right|^{2}\right)^{\frac{1}{2}}\right)$ and $S=1.73$ (defined as $\left\{\Sigma w(\Delta F)^{2} /\left(N_{0}-N_{\mathrm{v}}\right)\right\}^{\frac{1}{2}}$ for 513 variables $\left(N_{\mathrm{v}}\right)$ and 2798 data $\left.\left(N_{0}\right)\right)$. Refinement was made with shelx 76 (Sheldrick 1976), and the function minimized was $\Sigma w\left(\left|F_{0}\right|-\left|F_{\mathrm{c}}\right|\right)^{2}$, with $w=\left(\sigma^{2}\left|F_{0}\right|+0 \cdot 0005\left|F_{0}\right|^{2}\right)^{-1}$. The maximum parameter shift-to-error ratio at convergence was $0.05: 1$, and the largest peaks on the final difference map were of heights +0.24 and $-0.24 e \AA^{-3}$. Atomic scattering factors were taken from Ibers and Hamilton (1974).


Fig. 1. Perspective view of the molecule (1) with thermal ellipsoids at the $40 \%$ probability level. The H atoms are represented by arbitrary spheres.


Fig. 2. Crystal packing showing the hydrogen bonds.

Final atomic positional coordinates are listed in Table 1. Fig. 1, which includes the numbering scheme, and Fig. 2 have been prepared from the output of ORTEP (Johnson 1965). Anisotropic thermal parameters, H-atom positional coordinates,
short intermolecular approaches and observed and calculated structure amplitudes have been deposited.*

## 3. Discussion of the Structure

The crystal structure of the title compound contains two crystallographically independent colchiceine molecules of the same chirality, (1) and (2), in the asymmetric unit. Their conformation is similar in gross features, and also similar to that observed for colchiceine in the ethylacetate-water solvate structure (EAWS) (Silverton 1979). There are, however, differences in the conformational detail of the two independent molecules (see Tables 2-4). Fig. 1 shows a perspective view of the molecule with the numbering scheme. The absolute skeletons in Figs 1 and 2, and for colchiceine in the EAWS structure, are consistent with the configuration assigned to ( + )-colchicine (Rösner et al. 1981).

Table 2. Bond lengths (A) in the two independent molecules
The e.s.d. values are given in parentheses

| Atoms | Molecule (1) | Molecule (2) | Atoms | Molecule (1) | Molecule (2) |
| :--- | :---: | :--- | :--- | :---: | :---: |
| C(1)-C(2) | $1.381(5)$ | $1.399(8)$ | C(7a)-C(12a) | $1.408(7)$ | $1.436(7)$ |
| C(1)-O(1) | $1.359(4)$ | $1.367(8)$ | C(8)-C(9) | $1.395(8)$ | $1.398(9)$ |
| C(1)-C(12b) | $1.422(5)$ | $1.407(7)$ | C(9)-C(10) | $1.442(9)$ | $1.467(9)$ |
| C(2)-C(3) | $1.393(10)$ | $1.378(10)$ | C(9)-O(9) | $1.314(7)$ | $1.274(7)$ |
| C(2)-O(2) | $1.382(6)$ | $1.380(6)$ | C(10)-C(11) | $1.410(7)$ | $1.363(8)$ |
| C(3)-C(4) | $1.385(8)$ | $1.391(8)$ | C(10)-O(10) | $1.275(7)$ | $1.320(8)$ |
| C(3)-O(3) | $1.372(6)$ | $1.365(7)$ | C(11)-C(12) | $1.382(7)$ | $1.407(8)$ |
| C(4)-C(4a) | $1.393(7)$ | $1.394(8)$ | C(12)-C(12a) | $1.392(8)$ | $1.380(8)$ |
| C(4a)-C(5) | $1.518(8)$ | $1.518(8)$ | C(12a)-C(12b) | $1.500(7)$ | $1.503(7)$ |
| C(4a)-C(12b) | $1.386(9)$ | $1.399(9)$ | C(13)-C(14) | $1.512(7)$ | $1.492(8)$ |
| C(5)-C(6) | $1.515(7)$ | $1.539(7)$ | C(13)-N(1) | $1.340(7)$ | $1.355(8)$ |
| C(6)-C(7) | $1.539(8)$ | $1.527(9)$ | C(13)-O(13) | $1.238(7)$ | $1.235(8)$ |
| C(7)-C(7a) | $1.519(7)$ | $1.527(8)$ | O(1)-C(1M) | $1.435(7)$ | $1.425(7)$ |
| C(7)-N(1) | $1.466(6)$ | $1.460(7)$ | O(2)-C(2M) | $1.393(8)$ | $1.431(7)$ |
| C(7a)-C(8) | $1.410(7)$ | $1.376(7)$ | O(3)-C(3M) | $1.426(10)$ | $1.421(11)$ |

Ring A in both molecules (1) and (2) is planar within experimental error [ $\chi^{2}=$ $\Sigma\left(\delta / \sigma^{2}\right)=5 \cdot 3$ for (1) and $2 \cdot 5$ for (2)]. Ring B adopts the expected boat conformation found in other colchicine derivatives; the sum of the torsional angles of this ring (see Table 4) is $296^{\circ}$ for (1) and $297^{\circ}$ for (2) similar to the values $296^{\circ}$ for EAWS and $292^{\circ}$ for 7 -deacetamidocolchiceine (Rius et al. 1984). In ring C the atoms are less coplanar in molecule (1) [r.m.s.d. $=0.076(5) \AA, \delta_{\max }=0 \cdot 100(5) \AA$ for $\mathrm{C}(10)$ ] than in molecule (2) [r.m.s.d. $=0.027(6) \AA, \delta_{\max }=0.038(6) \AA$ for $C(9)$ ]; the sum of the torsional angles for the ring is $63^{\circ}$ for (1) and $27^{\circ}$ for (2) compared with the value $15^{\circ}$ for EAWS. The angles between the normals to rings A and C are $44 \cdot 5(4)^{\circ}$ for (1) and $51.5(5)^{\circ}$ for (2) [cf. $54.7(3)^{\circ}$ in EAWS]. The N -acetyl group has the conformation preferred by equatorial O-acetyl groups as first discussed by Mathieson (1965); the mean distance between the carbonyl oxygen, $\mathrm{O}(13)$, and the H atom at $\mathrm{C}(7)$ in the two molecules is $2.46 \AA$, and the atoms $\mathrm{C}(7), \mathrm{N}(1), \mathrm{C}(13), \mathrm{O}(13)$ and $\mathrm{C}(14)$ are nearly coplanar $\left[\chi^{2}=74 \cdot 2\right.$, r.m.s.d. $=0.021(5) \AA$ for $(1)$ and $\chi^{2}=32.9$,

[^1]Table 3. Bond angles (degrees) in the two independent molecules
The e.s.d. values are given in parentheses

| Molecule |  |  |  | Molecule |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Atoms | (1) | (2) | Atoms | (1) | (2) |
| $\mathrm{O}(1)-\mathrm{C}(1)-\mathrm{C}(2)$ | 119.7(3) | 118.4(5) | $\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{C}(10)$ | 128.1(5) | 124.9(6) |
| $\mathrm{O}(1)-\mathrm{C}(1)-\mathrm{C}(12 \mathrm{~b})$ | 120.0(3) | 119.8(5) | $\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{O}(9)$ | 117.5(5) | 120.9(6) |
| $\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{C}(12 \mathrm{~b})$ | 120.3(3) | 121.8(5) | $\mathrm{O}(9)-\mathrm{C}(9)-\mathrm{C}(10)$ | 114.2(5) | 114.2(6) |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | 120.7(5) | 120.7(5) | $\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(11)$ | 122.3(5) | 126.6(6) |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{O}(2)$ | 120.0(4) | 119.1(5) | $\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{O}(10)$ | 117.6(5) | 114.7(6) |
| $\mathrm{C}(3)-\mathrm{C}(2)-\mathrm{O}(2)$ | 119.2(5) | 120.1(5) | $\mathrm{O}(10)-\mathrm{C}(10)-\mathrm{C}(11)$ | 119.9(5) | 118.7(6) |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(4)$ | 120.0(5) | 120.2(5) | $\mathrm{C}(10)-\mathrm{C}(11)-\mathrm{C}(12)$ | 130.7(5) | 130.8(5) |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{O}(3)$ | 114.8(5) | 116.4(5) | $\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{C}(12 \mathrm{a})$ | 132.7(5) | 130.8(5) |
| $\mathrm{O}(3)-\mathrm{C}(3)-\mathrm{C}(4)$ | 125.3(5) | 123.4(5) | $\mathrm{C}(7 \mathrm{a})-\mathrm{C}(12 \mathrm{a})-\mathrm{C}(12)$ | 124.9(5) | 125.6(5) |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(4 \mathrm{a})$ | 119.5(5) | 119.8(5) | $\mathrm{C}(12)-\mathrm{C}(12 \mathrm{a})-\mathrm{C}(12 \mathrm{~b})$ | 116.4(4) | 115.6(5) |
| $\mathrm{C}(4)-\mathrm{C}(4 \mathrm{a})-\mathrm{C}(5)$ | 118.9(5) | 120.6(5) | $\mathrm{C}(7 \mathrm{a})-\mathrm{C}(12 \mathrm{a})-\mathrm{C}(12 \mathrm{~b})$ | 118.7(4) | 118.9(5) |
| $\mathrm{C}(4)-\mathrm{C}(4 \mathrm{a})-\mathrm{C}(12 \mathrm{~b})$ | 121.7(5) | 120.7(5) | $\mathrm{C}(1)-\mathrm{C}(12 \mathrm{~b})-\mathrm{C}(4 \mathrm{a})$ | 118.1(4) | 118.8(5) |
| $\mathrm{C}(5)-\mathrm{C}(4 \mathrm{a})-\mathrm{C}(12 \mathrm{~b})$ | 119.4(5) | 118.6(5) | $\mathrm{C}(1)-\mathrm{C}(12 \mathrm{~b})-\mathrm{C}(12 \mathrm{a})$ | 120.3(4) | 120.9(5) |
| $\mathrm{C}(4 \mathrm{a})-\mathrm{C}(5)-\mathrm{C}(6)$ | 113.3(5) | 110.7(5) | $\mathrm{C}(4 \mathrm{a})-\mathrm{C}(12 \mathrm{~b})-\mathrm{C}(12 \mathrm{a})$ | 121.5(5) | 120.2(5) |
| C(5)-C(6)-C(7) | 111.8(5) | 112.2(5) | $\mathrm{O}(13)-\mathrm{C}(13)-\mathrm{C}(14)$ | 120.8(6) | 122.6(6) |
| $\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{C}(7 \mathrm{a})$ | 111.9(4) | 111.2(5) | $\mathrm{O}(13)-\mathrm{C}(13)-\mathrm{N}(1)$ | 123.4(5) | 121.9(6) |
| $\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{N}(1)$ | 108.3(4) | 109.4(5) | $\mathrm{C}(14)-\mathrm{C}(13)-\mathrm{N}(1)$ | 115.8(5) | 115.5(6) |
| $\mathrm{C}(7 \mathrm{a})-\mathrm{C}(7)-\mathrm{N}(1)$ | 114.0(4) | 114.1(5) | $\mathrm{C}(7)-\mathrm{N}(1)-\mathrm{C}(13)$ | 121.3(5) | 121.8(5) |
| $\mathrm{C}(7)-\mathrm{C}(7 \mathrm{a})-\mathrm{C}(8)$ | 116.4(4) | 117.3(5) | $\mathrm{C}(1)-\mathrm{O}(1)-\mathrm{C}(1 \mathrm{M})$ | 116.3(4) | 114.4(4) |
| $\mathrm{C}(7)-\mathrm{C}(7 \mathrm{a})-\mathrm{C}(12 \mathrm{a})$ | 117.0(4) | $115 \cdot 2(5)$ | $\mathrm{C}(2)-\mathrm{O}(2)-\mathrm{C}(2 \mathrm{M})$ | 115.0(4) | 114.2(4) |
| $\mathrm{C}(8)-\mathrm{C}(7 \mathrm{a})-\mathrm{C}(12 \mathrm{a})$ | 126.6(5) | 127.4(5) | $\mathrm{C}(3)-\mathrm{O}(3)-\mathrm{C}(3 \mathrm{M})$ | $117.0(5)$ | 117.7(5) |
| $\mathrm{C}(7 \mathrm{a})-\mathrm{C}(8)-\mathrm{C}(9)$ | 132.1(5) | 133.5(6) |  |  |  |

r.m.s.d. $=0.018(6) \AA$ for (2)]. The angles between the normals to this plane and rings A and C are $112 \cdot 1(4)^{\circ}$ and $84 \cdot 6(4)^{\circ}$ respectively for (1) and $110 \cdot 2(5)^{\circ}$ and $82 \cdot 3(5)^{\circ}$ for (2) [cf. values $108.6(3)^{\circ}$ and $87.5(3)^{\circ}$ in EAWS]. As in other colchicine derivatives, two of the methoxy groups, at $\mathrm{C}(1)$ and $\mathrm{C}(2)$, are approximately normal to ring A , pointing in opposite directions, while the methoxy group at $\mathrm{C}(3)$ is almost contained within the plane of the ring $\mathbf{A}$. The angles between the normals to the methoxy planes and ring $A$ for the substituents at $C(1), C(2)$ and $C(3)$ have the respective values $72 \cdot 3(5)^{\circ}, 89 \cdot 4(5)^{\circ}$ and $3 \cdot 0(5)^{\circ}$ for (1) and $100 \cdot 1(6)^{\circ}, 91 \cdot 5(6)^{\circ}$ and $4 \cdot 1(1)^{\circ}$ for (2) [cf. values $96 \cdot 8(4)^{\circ}, 85 \cdot 1(4)^{\circ}$ and $4 \cdot 1(4)^{\circ}$ in EAWS].

Apart from the tropolone systems, the bond lengths and angles (see Tables 2 and 3 ) are as expected and similar in value to those observed in related structures. It is however worth noting that at $C(3)$, where the methoxy substituent lies nearly in the plane of ring A , the exocyclic angles are significantly distorted from the regular trigonal value. The angle enclosed by the methyl is the larger, the mean values for $\mathrm{C}(4)-\mathrm{C}(3)-\mathrm{O}(3)$ and $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{O}(3)$ in (1) and (2) being $124.3(5)^{\circ}$ and $115 \cdot 6(5)^{\circ}$ respectively. This distortion does not occur in the exocyclic angles at $\mathrm{C}(1)$ and $C(2)$, where the methoxy substituents are approximately normal to ring A. The two conformations for the methoxy groups (planar or normal to ring A) have been observed in numerous structures. In addition, the values found in these structures for the comparable exocyclic $\mathrm{C}-\mathrm{C}-\mathrm{O}$ angles are similar to those noted in the colchiceine molecules (see Karle et al. 1984).

The most interesting features of the structure are the differing dimensions of the tropolone systems in the two independent molecules. The $\mathrm{C}-\mathrm{O}$ bonds differ in length

Table 4. Selected torsional angles (degrees) in the two independent molecules
Values for EAWS (Silverton 1979) are included for comparison. The e.s.d. values are about $0.5^{\circ}$

| Atoms | Molecule (1) | Molecule (2) | EAWS |
| :---: | :---: | :---: | :---: |
| $\mathrm{C}(1 \mathrm{M})-\mathrm{O}(1)-\mathrm{C}(1)-\mathrm{C}(2)$ | 73.1 | $100 \cdot 3$ | 97.2 |
| $\mathrm{C}(2 \mathrm{M})-\mathrm{O}(2)-\mathrm{C}(2)-\mathrm{C}(3)$ | -92.0 | -90.4 | -96.6 |
| $\mathrm{C}(3 \mathrm{M})-\mathrm{O}(3)-\mathrm{C}(3)-\mathrm{C}(4)$ | 2.9 | 4.4 | $3 \cdot 1$ |
| $\mathrm{C}(4)-\mathrm{C}(4 \mathrm{a})-\mathrm{C}(5)-\mathrm{C}(6)$ | -108.8 | -107.5 | $-108.9$ |
| $\mathrm{C}(4 \mathrm{a})-\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(7)$ | -40.6 | -41.2 | -35.6 |
| $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{C}(7 \mathrm{a})$ | -47.9 | -49.8 | -53.7 |
| $\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{C}(7 \mathrm{a})-\mathrm{C}(12 \mathrm{a})$ | 78.4 | 77.8 | $75 \cdot 2$ |
| $\mathrm{C}(7)-\mathrm{C}(7 a)-\mathrm{C}(12 \mathrm{a})-\mathrm{C}(12 \mathrm{~b})$ | -4.4 | -0.6 | $4 \cdot 3$ |
| $\mathrm{C}(7 \mathrm{a})-\mathrm{C}(12 \mathrm{a})-\mathrm{C}(12 \mathrm{~b})-\mathrm{C}(4 \mathrm{a})$ | -49.7 | -52.2 | $-56.5$ |
| $\mathrm{C}(12 \mathrm{a})-\mathrm{C}(12 \mathrm{~b})-\mathrm{C}(4 \mathrm{a})-\mathrm{C}(5)$ | 1.6 | -0.9 | 1.8 |
| $\mathrm{C}(12 \mathrm{~b})-\mathrm{C}(4 \mathrm{a})-\mathrm{C}(5)-\mathrm{C}(6)$ | $70 \cdot 9$ | $74 \cdot 1$ | 68.8 |
| $\mathrm{C}(12)-\mathrm{C}(12 \mathrm{a})-\mathrm{C}(12 \mathrm{~b})-\mathrm{C}(1)$ | -47.1 | $-53.3$ | $-54.9$ |
| $\mathrm{C}(7 \mathrm{a})-\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{C}(10)$ | 7.4 | 3.0 | $-0.3$ |
| $\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(11)$ | $-18.4$ | -6.6 | $-3.8$ |
| $\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{C}(11)-\mathrm{C}(12)$ | $10 \cdot 0$ | $2 \cdot 3$ | $3 \cdot 3$ |
| $\mathrm{C}(10)-\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{C}(12 \mathrm{a})$ | $5 \cdot 2$ | 4.0 | 1.5 |
| $\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{C}(12 \mathrm{a})-\mathrm{C}(7 \mathrm{a})$ | $-5.0$ | $-3.0$ | -3.4 |
| $\mathrm{C}(12)-\mathrm{C}(12 \mathrm{a})-\mathrm{C}(7 \mathrm{a})-\mathrm{C}(8)$ | -7.5 | $-2.7$ | 0.2 |
| $\mathrm{C}(12 \mathrm{a})-\mathrm{C}(7 \mathrm{a})-\mathrm{C}(8)-\mathrm{C}(9)$ | 9.0 | $3 \cdot 2$ | $2 \cdot 6$ |
| $\mathrm{O}(9)-\mathrm{C}(9)-\mathrm{C}(10)-\mathrm{O}(10)$ | -9.0 | -5.1 | $-2 \cdot 2$ |
| $\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{C}(7 \mathrm{a})-\mathrm{C}(8)$ | -99.8 | -99.7 | -103.4 |
| $\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{N}(1)-\mathrm{C}(13)$ | -142.4 | -144.9 | -152.4 |
| $\mathrm{C}(7)-\mathrm{N}(1)-\mathrm{C}(13)-\mathrm{C}(14)$ | -175.9 | -176.5 | $175 \cdot 6$ |
| $\mathrm{C}(7)-\mathrm{N}(1)-\mathrm{C}(13)-\mathrm{O}(13)$ | 3.7 | $-0.3$ | -4.2 |

by about $6 \sigma$; in (1) it is $\mathrm{C}(10)-\mathrm{O}(10) 1 \cdot 275(7) \AA$ which is the shorter with $\mathrm{C}(9)-\mathrm{O}(9)$ $1 \cdot 314(7) \AA$, whilst in (2) it is $\mathrm{C}(9)-\mathrm{O}(9) 1 \cdot 274(7) \AA$ which is the shorter with $\mathrm{C}(10)-$ $\mathrm{O}(10) 1 \cdot 320(8) \AA$. As the lengths reported for the $\mathrm{C}-\mathrm{O}$ double and single bonds in an accurate tropolone structure are $1 \cdot 232(2)$ and $1 \cdot 359(2) \AA$ respectively (Silverton et al. 1977), the values in the title compound indicate that there are two tautomeric forms for each independent molecule in the crystal to give an approximate $1: 1$ mixture. Consequently, molecules (1) have a larger proportion of the isocolchicine form (I) and molelcules (2) have a larger proportion of the colchicine form (II). In molecule (1), apart from the $\mathrm{C}(9)-\mathrm{C}(10)$ bond which has a length $1.442(9) \AA$, the other bonds of ring $C$ have lengths within $2 \sigma$ of the standard aromatic $C-C$ bond length of $1.395 \AA$. This was also noted in 7-oxodeacetamido-colchiceine (Iorio et al. 1978). However, in molecule (2) there is bond alternation in ring C as is observed in EAWS. The carbon atom of the $\mathrm{C}-\mathrm{O}$ bond which has the greater component of double bond character has the maximum deviation from the ring plane in each molecule, that is $\mathrm{C}(10)$ in (1) and $C(9)$ in (2).

The crystal packing is illustrated in Fig. 2. The colchiceine molecules are oriented with their hydroxyl and N -acetyl groups forming channels parallel to the crystallographic $a$-axis which accommodate the water molecules. As the sites of the H atoms of the hydroxyl substituents and water molecules were not located in the analysis, only a brief description of the hydrogen bonding in the crystal can be given. The molecules (1) are linked by $H$ bonds. These involve the hydroxyl substituent at $C(9)$ or $C(10)$, and the acetyl oxygen of an adjacent molecule related by a cell translation along $a$ and $b$. The $\mathrm{O}(9) \ldots \mathrm{O}(13)$ and $\mathrm{O}(10) \ldots \mathrm{O}(13)$ distances are $2 \cdot 848(7)$ and $2 \cdot 897(7) \AA$
respectively. A similar mode of $\mathbf{H}$ bonding links the molecules (2) related by a cell translation along $b$, for which the $\mathrm{O}(9) \ldots \mathrm{O}(13)$ and $\mathrm{O}(10) \ldots \mathrm{O}(13)$ distances have the respective values $2.992(8)$ and $2.858(6) \AA$. In these interactions the acetyl oxygen, $\mathrm{O}(13)$, is the acceptor and the longer $\mathrm{O} . . . \mathrm{O}$ spacings involve the oxygen of the $\mathrm{C}-\mathrm{O}$ bond with the greater double bond component, that is $O(10)$ in (1) and $O(9)$ in (2). This $H$ bonding scheme is also consistent with the presence of the two tautomers in each of the independent molecules (1) and (2). The latter are linked into layers perpendicular to the $c$-axis by H bonds with the water molecules. The water is H bonded to the N atom of molecule (1) by a weak interaction, the $\mathrm{N}(1) \ldots \mathrm{O}(1 \mathrm{~S})$ and $\mathrm{HN}(1) \ldots \mathrm{O}(1 \mathrm{~S})$ distances have the respective values $3.093(7)$ and $2.03 \AA$ and the angle $\mathrm{N}(1)-\mathrm{HN}(1) \ldots \mathrm{O}(1 \mathrm{~S})$ is $172^{\circ}$. It is the acetyl O atom which links molecules (2) to the water, the $\mathrm{O}(13) \ldots \mathrm{O}(1 \mathrm{~S})$ distance being $2.935(7) \AA$. Apart from two short intermolecular contacts, $O(10) \ldots O(1 S) 3 \cdot 160(7) \AA$ and $O(9) \ldots C(13) 3 \cdot 194(8) \AA$, both involving molecules (1), there are no other contacts less than 3.27(1) $\AA$.

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## References

Ibers, J. A., and Hamilton, W. C. (Eds) (1974). 'International Tables for X-ray Crystallography', Vol. 4 (Kynock Press: Birmingham).
Iorio, M. A., Brossi, A., and Silverton, J. V. (1978). Helv. Chim. Acta 61, 1213.
Johnson, C. K. (1965). ORTEP: Thermal ellipsoid plot program. Oak Ridge National Laboratory Rep. No. ORNL-3794.
Karle, I. L., Flippen-Anderson, J. L., Chiang, J. F., and Lowrey, A. H. (1984). Acta Crystallogr. B 40, 500.
Main, P., Hull, S. E., Lessinger, L., Germain, G., Declercq, J. P., and Woolfson, M. M. (1978). MULTAN 78: A sytem of computer programs for the automatic solution of crystal structures from X-ray diffraction data. Univs of York, England, and Louvain, Belgium.
Margulis, T. N. (1982). Cryst. Struct. Commun. 11, 1689.
Mathieson, A. McL. (1965). Tetrahedron Lett. 46, 4137.
Morrison, J. D. (1951). Acta Crystallogr. 4, 69.
Rius, J., Molins, E., Miravitlles, C., and Bladé-Font, A. (1984). Acta Crystallogr. C 40, 839.
Rösner, M., Caparo, H. G., Jacobson, A. E., Atwell, L., Brossi, A., Iorio, M. A., Williams, T. H., Sik, R. H., and Chignell, C. F. (1981). J. Med. Chem. 24, 257.
Sheldrick, G. M. (1976). SHELX 76: Program for crystal structure determination. Univ. of Cambridge, England.
Silverton, J. V. (1979). Acta Crystallogr. B 35, 2800.
Silverton, J. V., Kabuto, C., Buck, K. T., and Cava, M. P. (1977). J. Am. Chem. Soc. 99, 6708.


[^0]:    * Dedicated to Dr A. McL. Mathieson on the occasion of his 65th birthday.

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