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Nitrification (DMPP) and urease (NBPT) inhibitors had no effect on pasture yield, nitrous oxide emissions, or nitrate leaching under irrigation in a hot-dry climate

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Abstract. Modern dairy farming in Australia relies on substantial inputs of fertiliser nitrogen (N) to underpin economic production. However, N lost from dairy systems represents an opportunity cost and can pose several environmental risks. N-cycle inhibitors can be co-applied with N fertilisers to slow the conversion of urea to ammonium to reduce losses via volatilisation, and slow the conversion of ammonium to nitrate to minimise leaching of nitrate and gaseous losses via nitrification and denitrification. In a field campaign in a high input ryegrass–kikuyu pasture system we compared the soil N pools, losses and pasture production between (a) urea coated with the nitrification inhibitor 3,4-dimethyl pyrazole phosphate (b) urea coated with the urease inhibitor N-(*n*-butyl) thiophosphoric triamide and (c) standard urea. There was no treatment effect (P > 0.05) on soil mineral N, pasture yield, nitrous oxide flux or leaching of nitrate compared to standard urea. We hypothesise that at our site, because gaseous losses were highly episodic (rainfall was erratic and displayed no seasonal rainfall nor soil wetting pattern) that there was a lack of coincidence of N application and conditions conducive to gaseous losses, thus the effectiveness of the inhibitor products was minimal and did not result in an increase in pasture yield. There remains a paucity of knowledge on N-cycle inhibitors in relation to their effective use in field system to increase N use efficiency. Further research is required to define under what field conditions inhibitor products are effective in order to be able to provide accurate advice to managers of N in production systems.

Additional keywords: autochamber, dairy, diurnal, kikuyu, ryegrass, temporal.

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Introduction

Nitrogen (N) is a critical input in Australian dairy farming enterprises, and there is a very strong relationship between N input and milk production (Gourley et al. 2012). The major source of N input for most Australian farms is as inorganic fertiliser, and the median contribution to total farm N inputs is 53% (Gourley et al. 2012). Typically dairy farmers apply $40-50 \text{ kg N ha}^{-1}$ as urea per application, with the number of applications per year being highly variable. The 2012 'Dairying for Tomorrow NRM Survey' (Watson and Watson 2012) found that the national average N application rate was 155 kg N ha⁻¹, suggesting that on average Australian dairy farmers are making 3-5 applications of N fertiliser per annum. There is often a substantial quantity of applied N in dairy systems that is not recovered in pasture (Rowlings et al. 2016) nor exported in milk (Gourley et al. 2012). It is generally considered that the greater the excess of N in soil-plant systems, the greater its loss to the environment (Jarvis et al.

2011). Increasing intensification of livestock production systems in the future is likely to further exacerbate this situation unless remedial actions are implemented (Gourley and Weaver 2012).

Losses of N represent a potential economic and environmental cost. Dissipation pathways include ammonia (NH₃) volatilisation, gaseous losses of nitric oxide, nitrous oxide (N₂O) and dinitrogen via nitrification and denitrification, and leaching of nitrate (NO₃⁻). In New Zealand dairying systems, it has been estimated that 30–40% of N is lost from the farm via leaching, runoff, NH₃ volatilisation and denitrification (Edmeades 2004). Similar losses were reported by Prasertsak *et al.* (2001) for southern Queensland in Australia. The agricultural sector is the dominant source of anthropogenic N₂O, accounting for 78.6% of the net national N₂O emissions (Commonwealth of Australia 2014). Leaching losses are also of interest because of their potential impact on surface and ground water quality (Thorburn *et al.* 2003). Soils under dairy pastures typically have relatively high soil organic carbon (C) contents of 3-6% (Dougherty 2007), compared with 1-2% in soils used for cropping. Because of the intensive nature of dairy production, farms are typically located either in high rainfall zones or rely on irrigation. The combination of these two factors (high organic C and high soil moisture) with substantial applications of N predisposes dairy soils to potentially high N₂O emissions (Burford and Bremner 1975; Phillips *et al.* 2007).

It has been suggested that fertilisers with enhanced efficiency have the potential to reduce losses of N from agricultural systems (Chen et al. 2008). Two commonly used inhibitors in the Australian dairy industry are the nitrification inhibitor 3,4-dimethyl pyrazole phosphate (DMPP) and the urease inhibitor N-(n-butyl) thiophosphoric triamide (NBPT). DMPP acts to slow the microbial conversion of ammonium (NH_4^+) to NO₃⁻ and production of gaseous forms of N via nitrification and denitrification. There is a body of research that indicates that a range of inhibitors can retard this conversion and thus lower N₂O emissions (Chen et al. 2010; Menéndez et al. 2012). The use of NBPT seeks to inhibit the urease enzyme which is particularly abundant in high-C soils under pasture (Chen et al. 2008), thus slowing the conversion of urea to NH_4^+ and minimising losses of N via volatilisation (Watson et al. 1994) but exposing the NH_4^+ to nitrification and denitrification. Urea can be readily purchased with either of these active ingredients applied to it ready for on-farm use.

Inputs of N via fertiliser constitute an important component of the whole farm N-cycle on Australian dairy farms (Gourley et al. 2012) and recovery of this N in pasture even when livestock are absent is only modest (Rowlings et al. 2016). Any attempt to optimise production and environmental outcomes on dairy farms requires an understanding of the effect of inhibitors applied with fertilisers on not only N loss pathways but also on pasture production. Although there is some knowledge on the effect of inhibitor products on N cycling and production under Australian conditions, there is little comprehensive data where both production and N₂O emissions data have been collected simultaneously. The need to increase production to achieve economies of scale on dairy farms will place pressure on farmers to increase N application rates and thus increase the risk of losses to the environment. Thus opportunities to increase fertiliser N use efficiency will be important for both economic and environmental reasons.

The objectives of this research were to evaluate the effect of the nitrification (DMPP) and urease (NBPT) inhibitors on

pasture yield, N_2O emissions and N leaching relative to standard urea.

Materials and methods

The research was undertaken 50 km south-west of Sydney (34.1°S, 150.7°E) on a site with an average annual rainfall of 788 mm (BOM 2015). The soil is a Eutrophic Red Chromosol (Isbell 2002), or Haploxeralf (Soil Survey Staff 1999). The site had been under permanent pasture used for dairy production for over 20 years, and over this period received regular irrigation and fertiliser N inputs. The soil had moderately high soil C and N and contained at least adequate concentrations of other macronutrients. Key physical and chemical characteristics of the soil at the commencement of the trial are described in Table 1. Twelve plots were laid out in a randomised complete block design with four replicates of three treatments (described below). Each plot measured $5 \text{ m} \times 5 \text{ m}$, with the whole plot being fertilised. harvested and irrigated. Measurements were only taken within the central $4 \text{ m} \times 4 \text{ m}$ area of each plot, thus creating a 1-m buffer between each measurement area.

Site management

The pasture at the site was a mixed ryegrass (*Lolium perenne* and *L. multiflorum* L.) and kikuyu (*Pennisetum clandestinum*) system. Kikuyu has summer active growth at the experimental site during approximately December–April at which point it is over-sown with ryegrass.

The N was typically applied immediately after every simulated grazing (harvest) in spring and autumn and every other harvest in summer and winter. N was applied as urea at a rate of 46 kg N ha^{-1} per application (100 kg ha⁻¹ of urea). There were four replicates of each of three treatments: (1) standard urea, (2) urea + nitrification inhibitor (0.16% w/w DMPP) and (3) urea + urease inhibitor (0.045% w/w NBPT). The concentrations of the active ingredients were as per commonly available commercial urea formulations. In the results section, the N application rate is described as average annualised N application rate (kg N ha⁻¹ year⁻¹) because the N rates varied slightly between the years of the experiment.

During the course of our trial, 19 pasture harvests to simulate grazing were performed over two years with fertiliser applied a total of 17 times. Irrigation was applied on an 'as needed' basis determined by a combination of visual inspection of the pasture and soil moisture data. Formal scheduling of irrigation using the soil moisture data was not undertaken because of the

 Table 1. Key chemical and physical characteristics of soil at the experimental site at the commencement of the research

 Samples were analysed using methods described in Rayment and Lyons (2010)

Depth (m)	$\mathrm{pH}_{\mathrm{Ca}}$	EC (dS m ⁻¹)	ECEC $(amol \ lxg^{-1})$	Total C	Total N	Colwell P $(mg l rg^{-1})$	KCl_{40} -S (mg lrg^{-1})	Bulk density (αcm^{-3})	Clay
(111)		(usm)	(chioi _c kg)	(%)		(mg kg)	(mg kg)	(g cm ⁻)	(%)
0-0.1	5.4	0.08	9.5	2.9	0.24	120	13	1.36	22
0.1-0.2	5.1	0.05	6.7	1.2	0.10	31	11	1.56	23
0.2-0.3	5.5	0.05	6.4	0.74	0.067	8	12	1.64	18
0.3-0.4	5.4	0.07	12	0.77	0.076	3	10	1.73	29
0.4-0.6	4.6	0.10	12	0.54	0.065	2	6	1.76	42
0.6-0.8	4.4	0.09	12	0.39	0.065	2	6	1.84	43
0.8 - 1.0	4.6	0.11	11	0.28	0.056	2	6	2.02	47

potential bias it may introduce to the effect of fertiliser formulations.

Pasture yield and quality measurements

Pasture production was measured by collecting and compositing pasture which was cut to a height of 5 cm from four quadrats (50 cm \times 50 cm) per plot at the appropriate 'grazing' time. Pasture harvests were nominally taken at the three-leaf stage for ryegrass and four-leaf stage for kikuyu. The pasture samples were then dried in a forced draught oven at 60°C and weighed to calculate pasture yield as tonnes of dry matter (DM) ha⁻¹. The remainder of each of the plots was mown and all mown pasture was removed off-site. Pasture samples were then analysed for total N and protein estimated by multiplying total N by 6.25 (AOAC International 2012).

N_2O fluxes

Fluxes of N₂O from the soil were determined using an automated gas sampling system as described in detail by Rowlings et al. (2012). This system consisted of pneumatically operated static chambers, linked to an automated sampling system and subsequently an in situ gas chromatograph for analysis of N₂O. The clear acrylic glass chambers covered a surface area of 0.25 m^2 ($0.5 \text{ m} \times 0.5 \text{ m}$), had a height of 0.5 m and a volume of 0.125 m³ and were secured to stainless steel bases inserted permanently into the soil to a depth of 0.1 m. Each plot had two bases and the chambers were alternated between these on a weekly basis to minimise any effects on pasture growth. In the warmer months of October-March, a reflective foil was placed on the north-facing side of the chambers and the lids to reduce heating in the chambers during their closure. A tipping bucket rain gauge connected to the system allowed for automated opening of the lids during rainfall events to ensure all chambers received the same rainfall as the larger plots.

The N₂O concentrations were determined using a gas chromatograph (SRI GC8610, Torrance, CA, USA) equipped with ⁶³Ni Electron Capture Detector (ECD). To minimise interference from moisture vapour and carbon dioxide on N₂O measurement, a pre-column filled with silica coated by sodium hydroxide was installed ahead of the analytical column and changed weekly. A full measurement cycle for flux determination commenced with lid closure, and finished when the lids were opened 60 min later. During this time, each chamber was sequentially sampled for 3 min followed by a known calibration standard (1.5 ppm N₂O). This process was repeated at 20-min intervals, sampling each chamber four times over the closure period. The lids remained open for a further 120 min before the commencement of the next cycle, allowing eight flux measurements for each chamber to be obtained per day.

Hourly N₂O fluxes were calculated from the slope of the linear regression of N₂O concentration vs time during the chamber lid closure, corrected for air temperature, atmospheric pressure and the ratio of chamber volume to surface area as described by Schwenke and Haigh (2016). The raw data were processed using an Auto GHG System Flux Calculator (Flux. net3.3) developed by the Queensland University of Technology (D. W. Rowlings, pers. comm.). The Pearson's correlation coefficient (r) for the linear regression was calculated and

used as a quality check for each regression. Flux rates were set as missing values if $r^2 < 0.8$. Daily N₂O emission for each chamber was calculated by averaging the eight emission measurements for that day.

Soil moisture was monitored continuously at the site using theta (Delta-T, UK) probes connected to a data logger. The probes were calibrated for the soil at this site. Water-filled pore space (WFPS) was calculated as

WFPS = volumetric water content/total soil porosity

where

Soil porosity = 1 - soil bulk density/particle density

The particle density is 2.65 g cm^{-3} and the soil bulk density (0–0.1 m) at the site was determined as 1.36 g cm^{-3} . Rainfall, temperature, humidity and wind speed and direction were all logged using an automated weather station (Measurement Engineering Australia, Magill, Australia).

Soil mineral N

Soil samples (0–0.1 m) were collected from every plot approximately monthly by taking 10 cores from each plot and compositing the cores from each plot. Soils samples were then dried in a forced draught oven at 40°C before grinding to <2 mm. Mineral N was then extracted using 2 M potassium chloride and shaking samples for 30 min on an end-over-end shaker. The extracts were then filtered (<0.45 μ m) and analysed for NO₃⁻ and NH₄⁺ on a flow injection analyser.

N leaching

Intact lysimeter cores were collected for quantifying the volume of drainage and NO₃⁻ contained in this drainage based on the design of Cameron et al. (1992). The cores were 0.3 m in diameter and 0.8 m deep (0.8 m was estimated to approximate the maximum effective base of the rooting zone). A single intact core was taken from each plot and encased in 0.3-m diameter polyvinyl chloride (PVC) tubing. Petroleum jelly was poured down the small gap between the soil and the PVC tube to prevent edge flow effects. The cores were then housed in a lysimeter pit adjacent to the plots with at least 0.5 m of soil surrounding each lysimeter on all sides (to minimise any temperature effects) and sown to pasture as per the plots. The lysimeters were treated in all respects exactly as the plots including the timing and rate of fertiliser, irrigation, pasture harvest and management. Leachate samples were collected and measured to determine volume of leachate, and the concentration of NO3⁻ was determined on a sub-sample as per the method previously described.

Statistical methods

Analysis of data to determine existence of treatment effects was undertaken using mixed model analyses performed in ASReml-R (Butler *et al.* 2009). For the soil mineral N and pasture yield responses, a linear mixed model was fitted with fixed effects of treatment and time (as a factor) and their interaction, and random effects for block, block by time and plot. For these response variates there was little indication of autocorrelation in the data, so a simple equal correlation model was assumed. For N₂O flux, a mixed-model smoothing spline framework was used (Verbyla *et al.* 1999), with fixed effects of treatment and linear time, and random effects of spl(time), ran(time), block and plot and their interactions with linear, spline and random time, and interactions of treatment with spline and factor time. Flux was square-root transformed before analyses. Means in the text are reported with \pm standard error of the mean shown in parentheses.

Results and discussion

 N_2O emissions were monitored continuously for 15 months between November 2012 and January 2014 (inclusive). For the period November 2012 to January 2014, total rainfall was 892 mm compared with the long-term average of 1000 mm. During this period, 710 mm of irrigation was applied and the average annualised N application rate was 442 kg N ha⁻¹ year⁻¹. Other parameters such as pasture production were monitored for 24 months between November 2012 and October 2014. For the two-year period during which pasture production was monitored, the total rainfall was 1425 mm compared with the long-term average of 1764 mm. Over the two-year period, 952 mm of irrigation was applied and the average annualised N application rate was 391 kg N ha⁻¹ year⁻¹.

Soil mineral N

There was no significant effect (P > 0.05) of treatment on total mineral N (NO₃⁻⁺ NH₄⁺), NO₃⁻ and NH₄⁺ nor on their relative proportions, although there was a highly significant effect (P < 0.001) of time on mineral N. Mineral N was highly variable between sampling dates with a range of $32-158 \text{ mg kg}^{-1}$ (Fig. 1). NO₃-N was the dominant form of N, accounting on average for 78% of the mineral N fraction. There was an accumulation of soil mineral N until October 2013 whereupon it declined rapidly. Following this, the concentrations of soil mineral N were more variable and displayed no particular patterns. The lack of an effect of treatment on mineral N contrasts with laboratory incubations and other field experiments in which differences in the forms of mineral N have been reported that are consistent with the mechanism of the respective inhibitor products (Irigoven et al. 2003; Chen et al. 2010; Dawar et al. 2010). Dawar et al. (2010) observed that NBPT resulted in a lower NH₄⁺ concentration for the first 5 days following fertiliser application only, beyond which NH_4^+ concentration did not differ from the urea control. Chen et al. (2010) observed that under moist (60% WFPS) and mild conditions (15°C), the effect of nitrification inhibitor declined substantially after 14 days. In our soils, mean annual temperature at 0.05 m was 17°C. We sampled only approximately monthly and always at least several weeks after fertiliser application and this may have hampered our ability to detect differences in mineral N between treatments. However, our observed lack of treatment effects on mineral N may also have implications for the effectiveness of the inhibitors on N₂O production as discussed below.

Pasture yield and N content

Pasture harvests were undertaken on 19 occasions over the twoyear experimental period (Fig. 2). There was no effect (P > 0.05) of treatment on yield or on protein (protein data not shown)



Fig. 1. Average (n=4) potassium chloride (KCl)-extractable soil NH₄⁺-N and NO₃⁻-N for each sampling occasion for urea, urea + DMPP (nitrification inhibitor) and urea + NBPT (urease inhibitor) over 2 years. There was no significant treatment effect (*P* > 0.05).



Fig. 2. Average (n=4) pasture dry matter yield $(t ha^{-1})$ for each sampling occasion for urea, urea + DMPP (nitrification inhibitor) and urea + NBPT (urease inhibitor) over 2 years. There was no significant treatment effect (P > 0.05).

although there was a highly significant effect (P < 0.001) of time on pasture yield. The large differences in yield between harvests reflect differences in growth patterns because of stage of growth, temperature, day length and soil moisture. Small harvest yields typically occurred during periods of pasture species transition when pasture was often being managed to re-establish the new species for the forthcoming season. Total cumulative pasture production was 38.1 (±0.94), 37.4 (±1.29) and 39.3 (±0.58) tha^{-1} for the urea, urea + DMPP and urea + NBPT treatments respectively; and corresponding protein contents on average were 17.0 (± 0.7), 17.2 (± 1.7) and 17.1 (± 0.9)%. There were large differences in protein contents between harvests with low protein content in early summer. The differences are attributable in part to a change from a ryegrass dominated sward to a kikuyu sward, which typically has lower protein content (Fulkerson 2007). The lack of a treatment effect on pasture yields is consistent with production trials in the nearby Hunter Valley where only small non-significant differences in yields were found when the same inhibitor-coated urea products were used on ryegrass-kikuyu pastures (Neil Griffiths, NSW DPI, pers. comm.). Similarly, Suter et al. (2015) reported that neither DMPP- nor NBPTcoated urea resulted in increased pasture yields. However, NBPT has been shown to increase pasture yields in several other studies in New Zealand: Dawar et al. (2010) reported that the use of NBPT increased pasture yield by 20% and Zaman et al. (2008) reported that NBPT increased pasture

yield by 17%. One possible explanation for our observed lack of treatment effects on yields is that maximum yields were already reached at the high rates of N we used and thus any N 'saved' by the use of inhibitors could not translate into higher yields. However, data from a companion N rates trial undertaken at our site (W. J. Dougherty, unpubl. data) indicated that this was not the case. This companion trial showed a linear response $(P < 0.001, r^2 = 0.97)$ to increasing N rates up to a maximum rate of $667 \text{ kg N ha}^{-1} \text{ year}^{-1}$ (application rates of N were 0, 25, 38, 50, 75 and 100 kg N ha^{-1} per application). Furthermore, there were little differences in the key N loss pathways (and they were small) between treatments, suggesting that pasture responses to inhibitors would be unlikely. Although we used only a single concentration of each of the active inhibitor ingredients as dictated by the commonly available commercial urea coated products that we used, it is worth noting that the rate of the active ingredient can affect the quantity of N lost via volatilisation and denitrification (Rawluk et al. 2001; Zerulla et al. 2001).

Currently in the Australian dairy industry, testing for soil mineral N or plant N status is not routinely used for refining N requirements. Emerging technologies to measure plant tissue N remotely and soil mineral N rapidly may be an effective means of refining N inputs to better match plant demand and thus reduce N losses and maximise N use efficiency. These approaches to refining N management require further development and evaluation of their cost effectiveness.



Fig. 3. Temporal pattern of (*a*) N₂O flux for urea, urea + DMPP (nitrification inhibitor) and urea + NBPT (urease inhibitor) and (*b*) water-filled pore space (WFPS) over 15 months. The arrows indicate the occurrence of fertiliser application. There was no significant treatment effect (P > 0.05).

N₂O emissions

The N₂O emissions were monitored for 15 months, N₂O fluxes are shown in Fig. 3 along with average WFPS (0-0.1 m). The effect of treatment on N2O-N emissions was not significant (P > 0.05). The emissions were highly episodic and were often associated with periods of high soil moisture content as would be expected (Weier et al. 1993). Cumulative emissions of N₂O-N (Fig. 4) for the monitoring period were 2976 (\pm 242), 2675 (± 307) and 2999 (± 380) g ha⁻¹ for the urea, urea + DMPP and urea + NBPT treatments respectively (over a 15-month period), during which 598 kg N ha^{-1} was applied. Of the total N₂O emissions, 50% occurred in just 34 days of the total 431 days of monitoring of emissions, illustrating the highly episodic nature of emissions. In a simple regression of WFPS against average daily N₂O emissions for the urea treatment, WFPS explained 33% of the variation in emissions. The relatively low emissions for the last 7 months of monitoring occurred despite high mineral N concentrations (see Fig. 1). We hypothesise that the lack of emissions during this period was the result of WFPS remaining <80% at all times – averaging ~60% WFPS due to low rainfall during this period. Even the application of irrigation only resulted in short periods of high soil moisture and even then WFPS rarely approached 80%. Bulk measures of WFPS (such as those made by moisture probes) fail to accurately estimate microsites of high moisture (and subsequent low oxygen status) and moisture is elevated for

such short periods that oxygen does not become depleted and thus denitrification is limited. This is supported by the conclusion of Rowlings *et al.* (2015) that N₂O emissions were more related to magnitude and duration of rainfall events than WFPS *per se.* Major emissions events occurred after heavy rain resulting in episodic waterlogging, indicating that manipulating irrigation (e.g. smaller applications more frequently) is unlikely to result in a reduction of already small losses of N₂O.

Rowlings *et al.* (2015) reported N₂O emissions of 1.83 kg ha⁻¹ over two years in an unfertilised pasture system. Carran *et al.* (1995) reported annual N₂O-N emissions of 3-5 kg ha⁻¹ year⁻¹, whereas Scheer *et al.* (2011) reported a range of 2.7–3.1 kg ha⁻¹ year⁻¹ in an intensively managed subtropical pasture. The presence of livestock and the associated excretion of high concentrations of N into urine patches may have been responsible for the emissions reported by Rowlings *et al.* (2015) and Carran *et al.* (1995) being higher than would occur if only fertiliser N was added.

As previously noted, we detected no treatment effect on mineral N, although our soil sampling was typically undertaken at least several weeks after fertiliser application, which is consistent with the lack of treatment effects on N_2O emissions. The effectiveness of DMPP can be substantially reduced at high temperatures (Irigoyen *et al.* 2003; Suter *et al.* 2010). Over summer at this site, soil temperatures at 0.1 m



Fig. 4. (a) Average (n=4) cumulative N₂O emissions for urea, urea+DMPP (nitrification inhibitor) and urea+NBPT (urease inhibitor) and (b) cumulative precipitation over 15 months. There was no significant treatment effect (P > 0.05).

reached ~35°C (data not presented) and although not measured, the temperature in the very surface of the soil would reasonably be expected to approach ambient air temperatures which commonly were ~40°C during the day. Irigoyen *et al.* (2003) showed significant declines in the efficacy of DMPP when soil temperatures increased from 10°C to 20 or 30°C. Furthermore, the efficacy of DMPP is affected by soil properties such as clay content that influences its adsorption (Barth *et al.* 2001). The combination of these factors may all contribute to the apparent lack of effect of nitrification inhibitors when monitored in field situations such as ours.

Although there was no significant effect of treatment on N2O emissions, Fig. 4 suggests an apparent lowering in N2O-N emissions from the DMPP treatment of ~10%. The majority of the apparent differences in cumulative N₂O emissions between the urea and the DMPP treatments occurred during two large emission events in February and June 2013. One of these events occurred in summer and the other in winter, suggesting no clear influence of environmental factors on effectiveness. The pattern of rainfall distribution at our site was evenly distributed compared with other dairying locations such as south-east Victoria, which have a strongly winter dominant rainfall resulting in periodic waterlogging. Furthermore, N is mostly used in winter and spring in southeast Victoria and so the use of inhibitor-coated urea may be more easily targeted to key times of the year when emissions would be expected to be high.

N loss through NO₃⁻ leaching

There was little or no drainage from the lysimeters until the large rainfall events of late January 2013. Of the total precipitation received of 2200 mm (comprising 1425 mm rainfall and 775 mm irrigation), on average 232 mm year⁻¹ of drainage occurred with no effect of treatment on drainage (P > 0.05). There was no significant effect (P > 0.05) of treatment on the quantities of NO₃⁻ leached. NO₃⁻⁻N was the dominant form of inorganic N leached. The quantities of NO₃⁻ leached are shown in Table 2, and represented only a small proportion of the N applied. Eckard *et al.* (2004) reported NO₃⁻⁻ leaching of $6.2-22 \text{ kg N ha}^{-1}$ in a three-year study applying 200 kg N ha⁻¹ year⁻¹. The losses we measured of <3 kg N ha⁻¹ year⁻¹ are low relative to those reported for New Zealand summarised by Burkitt (2014) and are at the low end of the ranges reported under a zero tension approach – widely

Table 2. The quantity of leachate, flow weighted mean NO₃⁻ concentration and annualised quantity of NO₃⁻ leached

There was no significant (P > 0.05) treatment effect on any of the leaching parameters. The figures in brackets represent standard errors of the means. DMPP, nitrification inhibitor; NBPT, urease inhibitor

Treatment	Urea	Urea + DMPP	Urea+NBPT
$ \begin{array}{c} \mbox{Drainage (mm)}^{\rm A} \\ \mbox{NO}_3^{-} \mbox{N} \ (mg \ L^{-1}) \\ \mbox{NO}_3^{-} \mbox{N} \ (kg \ ha^{-1} \ year^{-1}) \end{array} $	482 (38)	456 (38)	453 (70)
	1.21 (0.31)	0.81 (0.13)	1.10 (0.38)
	2.70 (0.39)	2.86 (0.13)	2.75 (0.38)

^ATotal drainage over 2 years.

recognised to underestimate the volume of drainage - and so our NO_3^{-} leaching fluxes may be underestimated. However, the proportion of precipitation estimated to be leached was substantial (20%) and the estimates of volume of drainage were similar to those reported by Eckard et al. (2004). It should be noted that leaching under urine patches is often the key pathway or source of N leaching from pasture systems (Silva et al. 1999; Di and Cameron 2002), although this was not tested in the current study. The small quantities of N leached most likely reflect the high demand for soil N by plants in our systems – while we only applied 782 kg N ha^{-1} , on average 996 kg N ha^{-1} was removed in harvested pasture over 2 years. Clearly, in this non-grazed system, leaching was not a major loss pathway despite these soils being apparently moderately well drained. Further research is required to examine what effect the rate of N application has on N use efficiency and thus losses via various pathways, with and without other N sources such as urine.

Conclusions

Substantial losses of N₂O can occur when high rates of N are applied, although at our site the losses of N as N₂O were relatively low compared with other studies in pasture systems in Australasia. When soil mineral N concentrations were high, the N₂O emissions were relatively large if soil moisture conditions were conducive to denitrification. We observed no effect of nitrification or urease inhibitor (DMPP and NBPT respectively) coated urea on soil mineral N, pasture yield, N₂O flux or NO₃⁻ leaching. Based on our observations we hypothesise that better matching plant demand with N supply from fertiliser and from mineralisation of organic N in order to minimise soil mineral N, may provide an effective means of minimising the risk of N₂O emissions. On several occasions we observed a large build-up of mineral N in late summer or late autumn. At such times, we propose that N inputs could be reduced without compromising soil fertility and subsequent pasture production. Currently in the dairy industry, soil testing for mineral N is not used for determining N requirements. Emerging technologies to measure plant tissue N remotely and rapidly may be an effective means of refining N inputs to better match plant demand and thus reduce N losses and maximise N use efficiency. These approaches to refining N management require further development and an evaluation of their cost effectiveness.

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