

Supplementary Material

Kinetic and mechanistic study on the reactions of $[Pd(dien)H_2O]^{2+}$ and $[Pt(dien)H_2O]^{2+}$ with L-cysteine and S-methyl-L-cysteine.

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Table S1. Rate constants as a function of temperature for the reactions between $[Pd(dien)H_2O]^{2+}$ and $[Pt(dien)H_2O]^{2+}$ with L-cysteine and S-methyl L-cysteine in 0.1 M NaClO₄ at pH = 1.0.

Complex	Ligand	T K	k_1 $M^{-1}s^{-1}$	k_{-1} s^{-1}
$[Pd(dien)H_2O]^{2+}$	L-cysteine	288.0	$(5.08 \pm 0.08) \cdot 10^2$	$(5.6 \pm 0.5) \cdot 10^{-1}$
		293.0	$(6.79 \pm 0.25) \cdot 10^2$	$(6.7 \pm 2) \cdot 10^{-1}$
		297.9	$(9.11 \pm 0.1) \cdot 10^2$	$(6.9 \pm 0.7) \cdot 10^{-1}$
		303.0	$(1.18 \pm 0.03) \cdot 10^3$	$(8.9 \pm 2) \cdot 10^{-1}$
	S-meth-L-cyst.	293.0	$(1.87 \pm 0.05) \cdot 10^3$	$(0.9 \pm 0.3) \cdot 10^{-1}$
		297.9	$(2.60 \pm 0.1) \cdot 10^3$	$(1.1 \pm 0.8) \cdot 10^{-1}$
		302.9	$(3.38 \pm 0.06) \cdot 10^3$	$(1.2 \pm 0.3) \cdot 10^{-1}$
		307.9	$(4.54 \pm 0.1) \cdot 10^3$	$(2.9 \pm 0.7) \cdot 10^{-1}$
$[Pt(dien)H_2O]^{2+}$	L-cysteine	288.2	$(7.5 \pm 0.8) \cdot 10^{-3}$	$(0.2 \pm 0.1) \cdot 10^{-4}$
		298.1	$(1.28 \pm 0.08) \cdot 10^{-2}$	$(0.3 \pm 0.1) \cdot 10^{-4}$
		307.4	$(2.28 \pm 0.01) \cdot 10^{-2}$	$(1.1 \pm 0.2) \cdot 10^{-4}$
	S-meth-L-cyst.	288.0	$(2.2 \pm 0.1) \cdot 10^{-2}$	$(0.4 \pm 0.1) \cdot 10^{-4}$
		298.1	$(3.87 \pm 0.02) \cdot 10^{-2}$	$(0.6 \pm 0.1) \cdot 10^{-4}$
		307.9	$(6.81 \pm 0.06) \cdot 10^{-2}$	$(6.0 \pm 2) \cdot 10^{-4}$

Table S2. Observed *pseudo*-first order rate constants as a function of ligand concentration at different temperature for the reactions between $[\text{Pd}(\text{dien})\text{H}_2\text{O}]^{2+}$ with L-cysteine and S-methyl L-cysteine at pH = 1.0, in 0.1 M NaClO₄.

L-cysteine			S-methyl-L-cysteine		
T(K)	$10^{-3} \text{ C}_L/\text{M}$	$10^{-4} k_{\text{obsd}} / \text{s}^{-1}$	T(K)	$10^{-3} \text{ C}_L/\text{M}$	$10^{-4} k_{\text{obsd}} / \text{s}^{-1}$
288.0	3.46	2.306(4)*	288.0	2.29	4.52(5)
	5.78	3.54(4)		4.77	8.779(4)
	8.01	4.605(5)		6.68	12.835(6)
	10.12	5.714(6)		9.54	17.551(6)
				11.78	22.41(5)
293.1	3.46	2.995(5)	292.9	2.29	6.08(5)
	5.78	4.749(6)		4.77	12.21(5)
	8.01	6.045(6)		6.68	17.46(6)
	10.12	7.548(4)		9.54	26.02(4)
				11.78	30.08(5)
297.9	1.47	2.011(5)	297.9	2.29	7.65(5)
	2.25	2.656(5)		3.52	12.2(6)
	2.82	3.406(4)		4.77	16.02(5)
	4.22	4.451(6)		6.68	23.14(5)
	5.63	5.812(6)		9.54	32.11(4)
	7.62	7.542(5)			
	8.44	8.46(4)			
	10.20	10.157(5)			
	14.08	13.41(4)			
303.0	3.46	4.869(5)	302.9	2.29	11.0(6)
	5.78	7.842(4)		4.77	21.25(4)
	8.01	10.398(5)		6.68	31.0(4)
	10.12	12.722(6)		9.54	43.61(5)

*Number of runs with parenthesis.

Table S3. Observed *pseudo*-first order rate constants as a function of ligand concentration at different temperature for the reactions between $[\text{Pt}(\text{dien})\text{H}_2\text{O}]^{2+}$ with L-cysteine and S-methyl L-cysteine at pH = 1.0, in 0.1 M NaClO₄.

L-cysteine			S-methyl-L-cysteine		
T(K)	$10^{-3} \text{ C}_\text{L}/\text{M}$	$10^{-4} k_\text{obsd} / \text{s}^{-1}$	T(K)	$10^{-3} \text{ C}_\text{L}/\text{M}$	$10^{-4} k_\text{obsd} / \text{s}^{-1}$
288.2	5.28	0.48(5)	288.0	4.17	1.0(5)
	10.55	1.1(6)		10.0	2.7(6)
	21.10	1.7(6)		20.0	4.1(6)
	25.64	2.1(5)		25.4	6.1(5)
298.1	2.67	0.6(5)	298.1	2.67	1.8(5)
	4.27	0.8(5)		5.17	2.8(4)
	5.13	0.9(4)		10.34	4.2(4)
	7.69	1.4(5)		16.03	6.9(5)
	10.25	1.8(6)		20.29	7.9(6)
	15.83	2.3(5)		25.63	11.0(5)
	20.51	2.9(6)			
307.8	2.67	0.5(5)	307.9	2.67	7.0(4)
	5.13	2.4(4)		5.34	10.0(4)
	7.69	2.8(4)		10.68	13.0(5)
	10.25	3.3(5)		16.02	17.0(4)
	15.83	4.8(5)		21.36	20.0(5)

Table S4. Observed *pseudo*-first order rate constants as a function of ligand concentration at different pH value for the reactions between $[\text{Pd}(\text{dien})\text{H}_2\text{O}]^{2+}$ with L-cysteine and S-methyl L-cysteine at 298 K, in 0.1 M NaClO₄.

L-cysteine			S-methyl-L-cysteine		
pH	$10^{-3} \text{ C}_L/\text{M}$	$k_{\text{obsd}} / \text{s}^{-1}$	pH	$10^{-3} \text{ C}_L/\text{M}$	$k_{\text{obsd}} / \text{s}^{-1}$
0.5	1.44	1.575(5)	0.5	2.67	7.76(5)
	2.05	1.979(6)		4.48	12.62(6)
	3.46	3.174(6)		6.28	17.99(6)
	4.50	3.854(5)		7.71	22.04(6)
	5.78	4.902(5)		9.64	27.32(5)
	7.18	5.866(5)		12.56	35.74(6)
	8.09	6.458(6)			
	8.62	6.799(5)			
	10.12	7.784(5)			
	11.06	8.546(6)			
1.0	12.96	10.24(6)			
	1.47	2.011(5)	1.0	2.29	7.65(6)
	2.25	2.656(4)		3.52	12.2(6)
	2.82	3.406(5)		4.77	16.02(5)
	4.22	4.451(5)		6.68	23.14(6)
	5.63	5.812(6)		9.54	32.11(5)
	7.62	7.542(5)			
	8.44	8.460(4)			
	10.20	10.157(5)			
	14.08	13.410(4)			
1.5	1.44	2.50(5)	1.5	2.67	11.18(6)
	3.09	4.40(4)		3.86	16.94(5)
	4.31	6.044(4)		5.78	23.87(5)
	5.76	7.45(5)		7.71	32.66(4)
	7.18	8.904(4)		9.54	40.21(4)
	10.06	12.612(5)			

Table S5. Observed *pseudo*-first order rate constants as a function of ligand concentration at different pH value for the reactions between $[\text{Pt}(\text{dien})\text{H}_2\text{O}]^{2+}$ with L-cysteine and S-methyl L-cysteine at 298 K, in 0.1 M NaClO₄.

L-cysteine			S-methyl-L-cysteine		
pH	$10^{-3} \text{ C}_L/\text{M}$	$10^{-4} k_{\text{obsd}} / \text{s}^{-1}$	pH	$10^{-3} \text{ C}_L/\text{M}$	$10^{-4} k_{\text{obsd}} / \text{s}^{-1}$
0.5	2.64	0.4(5)	0.5	2.63	1.1(6)
	6.33	0.9(6)		6.31	1.7(6)
	9.49	1.3(6)		9.46	2.0(6)
	12.66	1.6(6)		12.62	3.0(5)
	18.99	2.2(5)		18.93	4.1(5)
	25.33	3.0(6)		25.24	5.6(6)
1.0	2.67	0.6(5)	1.0	2.67	1.8(5)
	4.27	0.8(5)		5.17	2.8(4)
	5.13	0.9(4)		10.34	4.2(4)
	7.69	1.4(5)		16.03	6.9(5)
	10.25	1.8(6)		20.29	7.9(6)
	15.83	2.3(5)		25.63	11.0(5)
1.5	2.64	0.9(5)	1.5	4.10	13.0(5)
	6.33	1.6(4)		6.25	16.0(5)
	9.49	2.0(4)		12.5	21.0(4)
	12.66	2.5(5)		18.7	26.0(4)
	18.98	3.8(4)		25.0	33.0(5)
	25.32	4.8(5)			