Accessory Publication for

Aromaticity in Heterocyclic and Inorganic Benzene Analogs

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Figure S1. Schematic MO interaction diagram of benzene (1) constructed from two $(CH)_3^{9^{\bullet}}$ fragments in their decet valence configuration, based on Kohn-Sham MO analyses at BP86/TZ2P. There are 3 electrons in each of the two fragments, which have mutually opposite spin. The effect on orbital energies of the localization mode defined in Scheme 1 and represented here with curved arrows is indicated by + (stabilization) and – (destabilization).

Figure S2. Alternative bond energy decomposition of **4** and **5**, each constructed from two rigid fragments, as function of the distortion mode (in deg) from delocalized to localized structure as defined for **4** and **5** in Scheme 2. The diagram shows the *changes* relative to $\theta = 60^{\circ} (\Delta \Delta E)$ in the indicated terms $\Delta E_{int} = (\Delta E_{Pauli} + \Delta E_{\sigma}) + [\Delta E_{\pi}] + \Delta V_{elstat} = (total \sigma) + [total] + \Delta V_{elstat}$ computed at BP86/TZ2P.

Table S1. Cartesian coordinates, point-group symmetry, total energy relative to ADF's basic atoms and number of imaginary frequencies (NIMAG) of species occurring in this study, optimized at BP86/TZ2P.



Figure S1. Schematic MO interaction diagram of benzene (1) constructed from two $(CH)_3^{9^{\bullet}}$ fragments in their decet valence configuration, based on Kohn-Sham MO analyses at BP86/TZ2P. There are 3 electrons in each of the two fragments, which have mutually opposite spin. The effect on orbital energies of the localization mode defined in Scheme 1 and represented here with curved arrows is indicated by + (stabilization) and – (destabilization).



Figure S2. Alternative bond energy decomposition (in kcal/mol) of **4** and **5**, each constructed from two rigid *neutral* fragments (see text), as function of the distortion mode (in deg) from delocalized to localized structure as defined in Scheme 2. The diagram shows the *changes* relative to $\theta = 60^{\circ} (\Delta \Delta E)$ in the indicated terms $\Delta E_{int} = (\Delta E_{Pauli} + \Delta E_{\sigma}) + [\Delta E_{\pi}] + \Delta V_{elstat} = (total \sigma) + [total] + \Delta V_{elstat}$ computed at BP86/TZ2P.

Table S1. Cartesian coordinates (in Å), point-group symmetry, total energy relative to ADF's basic atoms, and number of imaginary frequencies (NIMAG) of species occurring in this study, optimized at BP86/TZ2P.

1	$C_{6}H_{6}(D_{6h})$ -1728.98	8 kcal/mol		NIMAG = 0
С	0.00000	1.397664	0.00000	
Η	0.00000	2.487887	0.00000	
С	0.00000	-1.397664	0.00000	
Η	0.00000	-2.487887	0.00000	
С	1.210412	0.698832	0.00000	
Η	2.154573	1.243944	0.00000	
С	1.210412	-0.698832	0.00000	
Η	2.154573	-1.243944	0.00000	
С	-1.210412	-0.698832	0.00000	
Η	-2.154573	-1.243944	0.00000	
С	-1.210412	0.698832	0.00000	
Η	-2.154573	1.243944	0.00000	
2	$C_{3}H_{3}N_{3}(D_{3b}) = -1409.0'$	7 kcal/mol		NIMAG = 0
C	-1 296848	0 00000	0 00000	
н	-2 389197	0 000000	0 000000	
C	0.648424	1,123103	0.000000	
н	1,194598	2.069105	0.000000	
C	0.648424	-1.123103	0.000000	
H	1.194598	-2.069105	0.000000	
N	1.378496	0.000000	0.000000	
N	-0.689248	1.193812	0.000000	
Ν	-0.689248	-1.193812	0.000000	
3	N ₆ (D _{6h}) -974.41	kcal/mol	NIMAG = 1	$3 (B_{2n}: i322 \text{ cm}^{-1}, E_2: i390 \text{ cm}^{-1})$
N	1.328875	0.00000	0.00000	
N	-1.328875	0.000000	0.000000	
N	0.664438	1.150840	0.000000	
N	-0.664438	1.150840	0.000000	
N	-0.664438	-1.150840	0.000000	
Ν	0.664438	-1.150840	0.00000	
4	$B_{2}N_{2}H_{2}(D_{2}) = -1637.3$	9 kcal/mol		NIMAG = 0
	$\frac{1}{1} \frac{1}{1} \frac{1}$		0 00000	NIMAO – 0
D		0.000000	0.000000	
п	-2.054775	1 260507	0.000000	
D	1 227200	2 200102	0.000000	
п	1.32/300	2.299103	0.000000	
D	1 227200	-1.200597	0.000000	
гі NT	1 /10075	-2.2991U3		
ц IN	1,4100/5 0 /000E1			
гі NT		0.000000 1 201161		
П IN	-U./USUS/ _1 011006	1.221101 2 000117		
гі NT	-1.211920 _0 705027	∠.∪୬୬⊥⊥/ _1 ጋጋ1161		
Ц	-0.705037 -1 211026	-1.221101 -2 N99117		
11	1.211920	2.079111	0.00000	

5	$B_3H_3O_3(D_{3h}) -1294.$.44 kcal/mol		NIMAG = 0
В	-1.383456	0.00000	0.00000	
Н	-2.576574	0.00000	0.00000	
В	0.691728	1.198108	0.00000	
Η	1.288287	2.231379	0.00000	
В	0.691728	-1.198108	0.00000	
Н	1.288287	-2.231379	0.00000	
0	1.382565	0.00000	0.00000	
0	-0.691282	1.197336	0.00000	
0	-0.691282	-1.197336	0.000000	
				1075 00 1 1/ 1
6	Sl_6H_6 (D_{6h})	1		–1057.09 kcal/mol
	NIMAG = 1 (B_{2g} : <i>i</i> 13	34 cm^{-1})		
Si	2.227832	0.00000	0.00000	
Η	3.721264	0.00000	0.00000	
Si	-2.227832	0.00000	0.00000	
Η	-3.721264	0.00000	0.00000	
Si	1.113916	1.929359	0.00000	
Η	1.860632	3.222709	0.00000	
Si	-1.113916	1.929359	0.00000	
Η	-1.860632	3.222709	0.00000	
Si	-1.113916	-1.929359	0.00000	
Η	-1.860632	-3.222709	0.00000	
Si	1.113916	-1.929359	0.00000	
Η	1.860632	-3.222709	0.00000	
7	P ₆ (D _{6h}) –644.66 kcal/mol			NIMAG = 2 (E_2 : <i>i</i> 93 cm ⁻¹)
Ρ	2.134514	0.00000	0.00000	
Ρ	-2.134514	0.00000	0.00000	
Ρ	1.067257	1.848543	0.00000	
Ρ	-1.067257	1.848543	0.00000	
Ρ	-1.067257	-1.848543	0.00000	
Ρ	1.067257	-1.848543	0.00000	
3'	$N_6(D_2)$ -983.4	3 kcal/mol		NIMAG = 0
Ν	-1.246674	-0.064273	0.000010	
Ν	1.246674	0.064273	0.000010	
Ν	-0.693391	1.096688	0.242723	
Ν	0.576947	1.162169	-0.242735	
Ν	0.693391	-1.096688	0.242723	
Ν	-0.576947	-1.162169	-0.242735	

6' Si₆H₆ (D_{3d})

-1060.11 kcal/mol

6' Si ₆ H ₆ (D _{3d})			–1060.11 kcal/mol
NIM	IAG = 0			
Si	1.911711	1.103727	0.238272	
H	3.188862	1.841090	-0.042527	
Si	-1.911711	-1.103727	-0.238272	
Н	-3.188862	-1.841090	0.042527	
Si	0.000000	2.207453	-0.238272	
Н	0.000000	3.682181	0.042527	
Si	-1.911711	1.103727	0.238272	
H	-3.188862	1.841090	-0.042527	
Si	0.000000	-2.207453	0.238272	
г Н	0.000000	-3.682181	-0.042527	
Si	1,911711	-1.103727	-0.238272	
H	3.188862	-1.841090	0.042527	
7' $P_6(D_2)$	-648.32	kcal/mol		NIMAG = 0
P	-2.009635	0.00000	0.00000	
P	2.009635	0.00000	0.00000	
P	-0.973788	1.794284	0.435079	
Р	0.973788	1.794284	-0.435079	
P	0.973788	-1.794284	0.435079	
P	-0.973788	-1.794284	-0.435079	
$N_2 (D_{\infty h})$				–383.34 kcal/mol
NIM	IAG = 0			
N	0.000000	0.00000	0.550855	
Ν	0.00000	0.00000	-0.550855	
CH_3NH_2 ($C_{\rm s}$) -813.22 kc	al/mol		NIMAG = 0
Ν	0.572854	-0.108718	-0.166118	
Н	0.329265	-1.003324	0.261330	
H	2.183714	0.337383	1.242641	
С	1.255365	0.753443	0.809169	
H	0.567622	0.976517	1.635422	
H	1.505245	1.709717	0.331106	
H	1.196398	-0.325086	-0.945056	
CHANH ((7) _625.67 kc	al/mol		$NIM \Delta G = 0$
	-0.43307 KC	-0 289635	-0 768396	
и П	0.435217	-1 174559	-0.981053	
11 Ц	1 171502	1.174537	0.551747	
п С	1.171502 0.219/16	0.074342	0.551/4/	
Ч	-0.182258	1 355794	0.090992 0 412569	
	0.102230	1.333771	0.112307	
$N_2H_4(C_2)$	-691.33 kc	al/mol		NIMAG = 0
<u> </u>	0.702431	-0.414895	-0.282858	
H	-0.259367	-0.455532	0.066418	
н	0.675632	-0.613716	-1.282598	
N	1,282279	0.895141	-0.123071	
Н	1.715221	0.923584	0.799685	
Н	0.575710	1.635658	-0.156113	

$N_2H_2(C_{2h})$	–502.94 kc	al/mol		NIMAG = 0
N	0.924056	-0.299319	-0.119599	
N	1.800796	-0.015292	0.723551	
H	0.222897	-0.892128	0.376291	
H	2.501956	0.577517	0.227661	
BH ₃ NH ₃ (C	_{3v}) –841.23 kc	al/mol		NIMAG = 0
N	-0.565707	-0.399658	-0.925306	
Н	-0.803000	-1.324825	-0.559882	
Н	1.369499	-0.233229	0.294386	
В	0.371288	0.445227	0.148171	
H	-0.307303	0.535596	1.152888	
H	-0.079760	-0.531759	-1.815352	
Н	-1.440985	0.092589	-1.118583	
Н	0.583365	1.512748	-0.393613	
BH ₂ NH ₂ (C	_{2v}) -686.78 kc	al/mol		NIMAG = 0
N	-0.647405	-0.337497	-0.674249	
Н	-0.757593	-1.342407	-0.594645	
H	1.303478	-0.256371	0.411585	
В	0.444015	0.357204	-0.158259	
H	-1.411789	0.110741	-1.167322	
H	0.491203	1.547881	-0.299468	
BH ₃ OH ₂ (C	^L s) –704.16 kc	al/mol		NIMAG = 0
0	-0.777192	-0.249061	-0.613435	
В	0.656692	0.516221	-0.001785	
H	-0.815539	-1.143129	-0.228042	
H	1.533814	-0.250737	-0.338077	
H	0.625244	1.570428	-0.588305	
H	0.413851	0.553440	1.179733	
H	-0.654825	-0.369642	-1.572520	
BH ₂ OH (C.) –565.69 k c	al/mol		NIMAG = 0
0	-0.641331	-0.335650	-0.671091	
B	0.430130	0.327964	-0.161949	
H	-0.606368	-1.293859	-0.516428	
H	1.304240	-0.269843	0.413911	
Н	0.445495	1.516830	-0.320803	
$Si_2H_6(D_{3d})$	–700.14 kc	al/mol		NIMAG = 0
Si	-0.648245	-0.908239	-0.589785	
Si	0.687136	0.858719	0.240575	
Н	-2.087653	-0.686416	-0.239428	
Н	2.126601	0.637078	-0.109701	
Н	0.251907	2.165623	-0.348003	
Н	0.570405	0.947960	1.731311	
H	-0.531596	-0.997390	-2.080533	
H	-0.213177	-2.215253	-0.001310	

$Si_2H_4(C_{2h})$	-497.00 kc	al/mol		NIMAG = 0
Si	-0.322389	-0.945227	-0.661461	
Si	0.361279	0.895718	0.312240	
H	-0.024242	2.201578	-0.308637	
H	0.299575	0.962811	1.805913	
H	-0.260697	-1.012405	-2.155137	
Н	0.063155	-2.251046	-0.040502	
$P_{2}H_{4}\left(C_{2}\right)$	–547.98 ko	al/mol		NIMAG = 0
P	-0.345610	-0.165186	-1.006695	
P	-0.004343	1.563254	0.399251	
Н	0.942852	2.309553	-0.372890	
Н	1.020484	0.891674	1.140494	
H	0.742040	0.051571	-1.912439	
Н	-1.269847	0.554009	-1.830873	
$\mathbf{P}_{2}\mathbf{H}_{2}\left(\mathbf{C}_{2\mathbf{h}}\right)$	–367.76 kcal/mol			NIMAG = 0
P	-0.378135	-0.035759	-0.988514	
P	0.042614	1.440657	0.369171	
Н	1.007051	0.638378	1.078292	
Н	-1.342572	0.766522	-1.697633	