

Supporting Information

Two-dimensional layered metal-organic frameworks (MOFs) of lanthanum(III) pyridine-2,6-dicarboxylate

Yong-Ru Liu, Tao Yang, Lei Li, Jun-Min Liu, Cheng-Yong Su

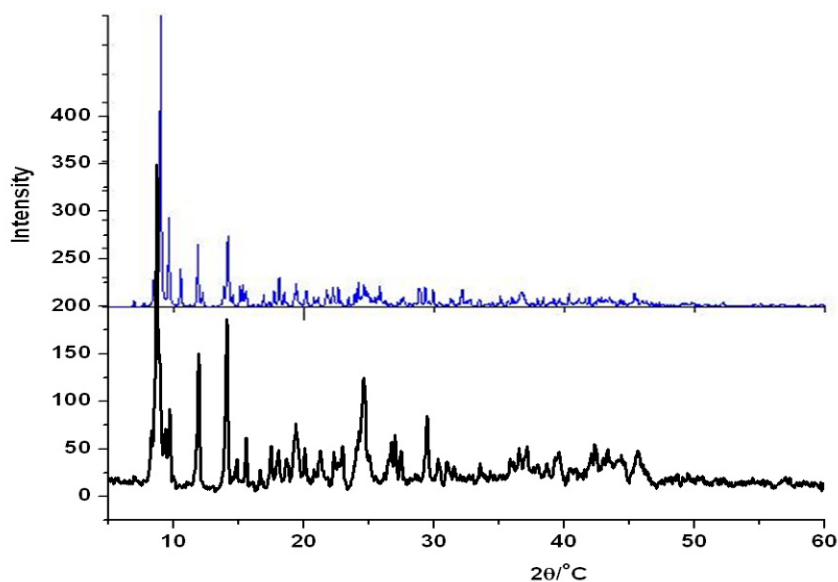


Fig. S1. The simulated X-ray powder diffraction patterns (upper) and the measured one (lower) of complex 1.

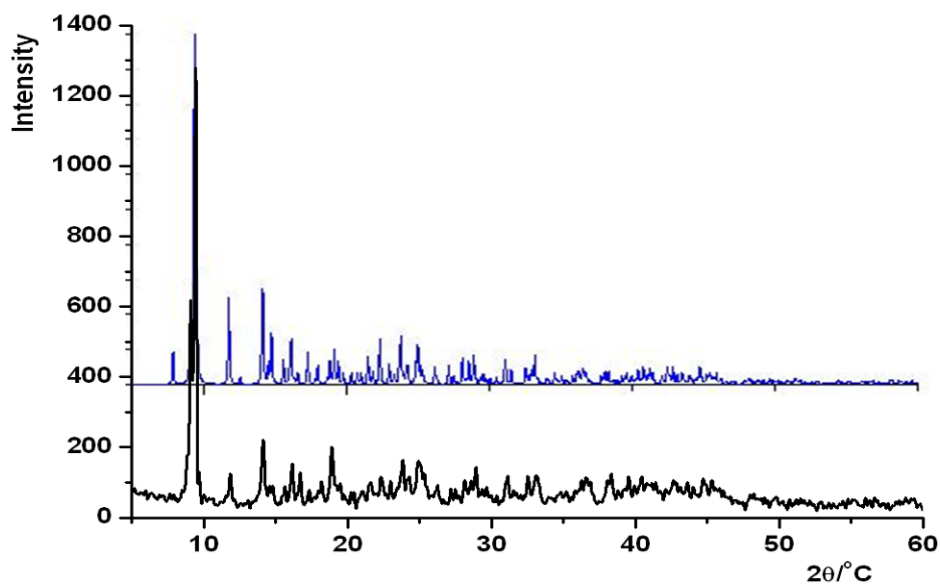


Fig. S2. The simulated X-ray powder diffraction patterns (upper) and the measured one (lower) of complex 2.

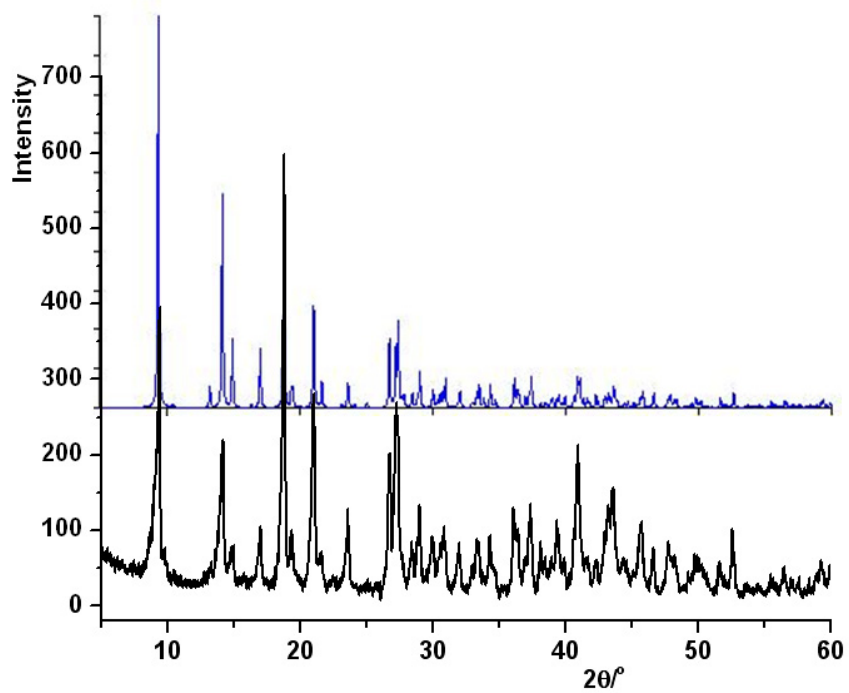


Fig. S3. The simulated X-ray powder diffraction patterns (upper) and the measured one (lower) of complex 3.

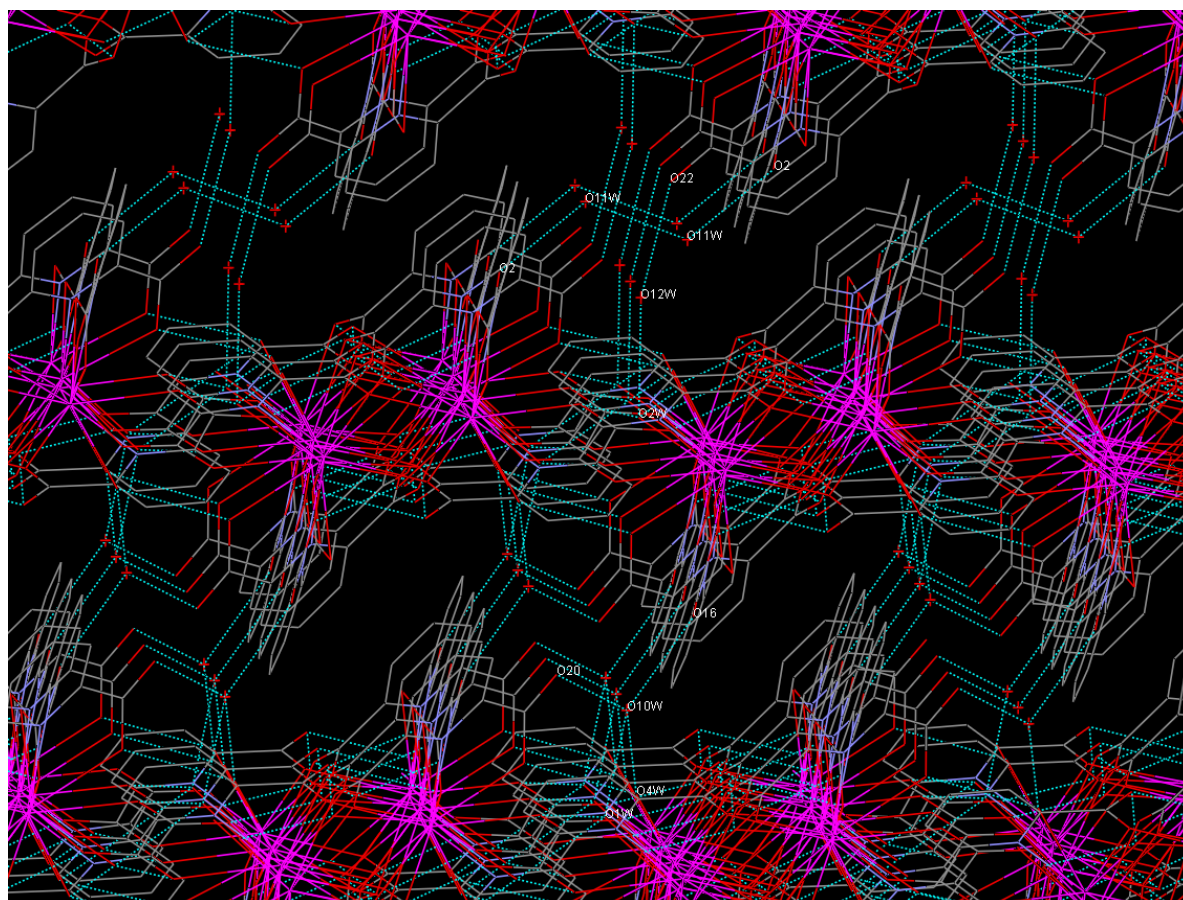


Fig. S4. The O-H...O hydrogen bonds formed between the layers in complex 1.

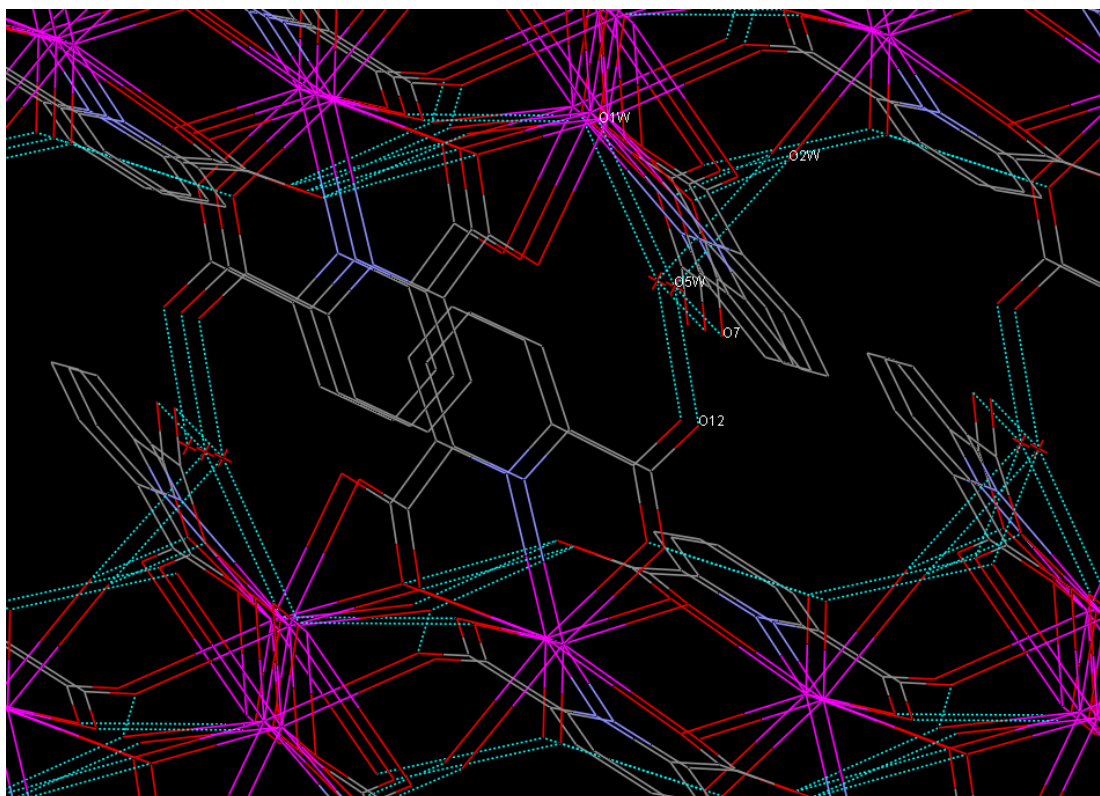


Fig. S5. The O-H...O hydrogen bonds formed between the layers in complex **2**.

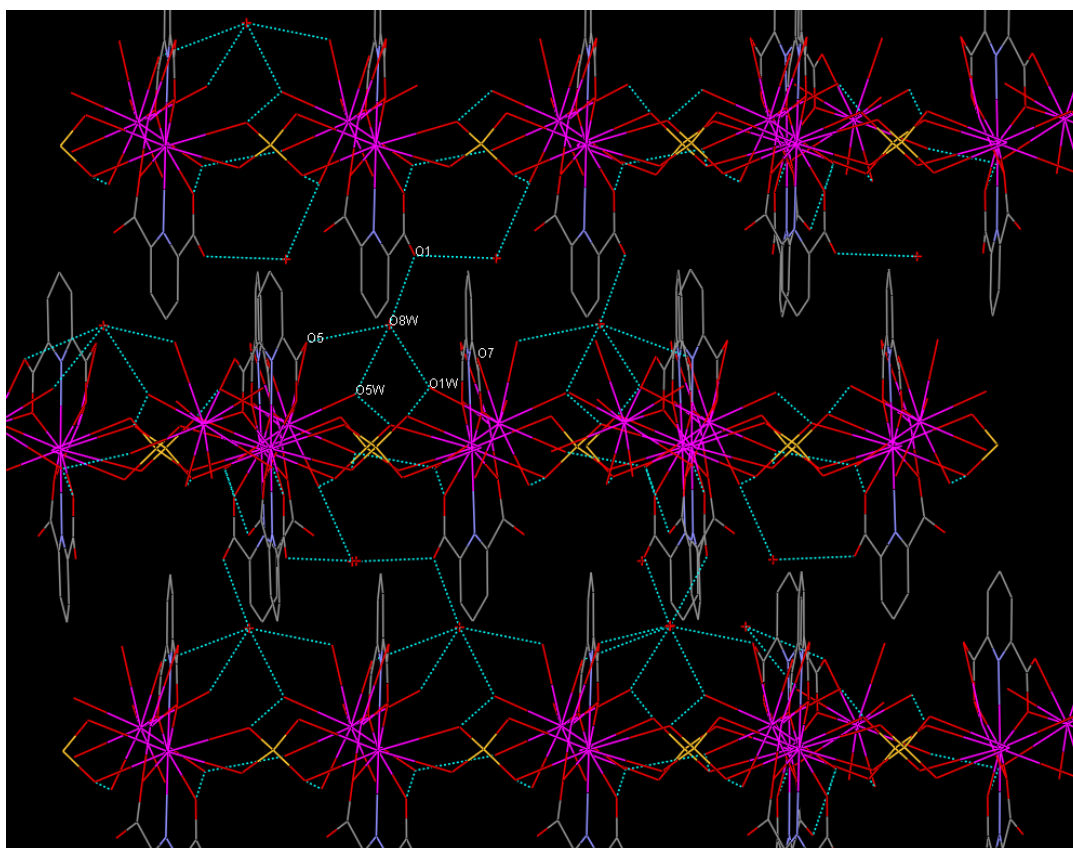


Fig. S6. The O-H...O hydrogen bonds formed between the layers in complex **3**.

Table S1. Summary Ln-pda complexes from CCDC (*version 5.29*).

Composition	Topology	Method	Structure	Ref.
{[Ln ₂ (pda) ₃ (H ₂ O) ₆] \cdot 2H ₂ O} _n	1D	Hydrothermal	Zigzag chain	19,21
{[La(pda)(Hpda)(H ₂ O) ₆] \cdot 2H ₂ O \cdot EtOH} _n	1D	Slow evaporation	Chain	20
{[Pr ₃ (pda) ₄ (Hpda)(H ₂ O) ₈] \cdot 8H ₂ O} _n	2D	Hydrothermal	Grid	22
{[Pr ₂ (pda) ₂ (H ₂ O) ₅ SO ₄] \cdot 2H ₂ O} _n	2D	Hydrothermal	(4,4) grid	22
[Ho(pda) ₂ (Hpda)] _n	2D	Hydrothermal	(4,4) grid	23
[Ce ₂ (pda) ₃ (H ₂ O) ₃] _n	3D	Hydrothermal	Open Framework	24
{[Ln ₂ (pda) ₃ (H ₂ O) ₃] \cdot H ₂ O} _n	3D	Hydrothermal	Open Framework	22,24,25
[Ce ₃ (pda) ₄ (Hpda)(H ₂ O) ₂] _n	3D	Hydrothermal	Open Framework	26
{[Nd ₃ (pda) ₅ (NH ₄)(H ₂ O) ₈] \cdot 5H ₂ O} _n	3D	Slow evaporation	Open Framework	27
{[Pr ₂ (pda) ₃ (H ₂ O) ₃] \cdot H ₂ O} _n	3D	Hydrothermal	Open Framework	22
{[La(pda)(H ₂ O) ₄] \cdot 2Cl} _n	3D	Hydrothermal	Open Framework	19

Table S2. Geometry Parameters for Hydrogen Bonds in **1^a**

D-H \cdots A	d(D-H)	d(H \cdots A)	d(D \cdots A)	\angle DHA
O(5W)-H(51) \cdots O(17)	0.85	1.87	2.68	156.6
O(7W)-H(71) \cdots O(9)#2	0.85	2.08	2.82	144.4
O(7W)-H(72) \cdots O(6)	0.85	2.52	3.00	115.9
O(5W)-H(52) \cdots O(15)#3	0.85	2.19	2.75	123.0
O(10W)-H(101) \cdots O(20)	0.86	1.94	2.66	139.1
O(10W)-H(102) \cdots O(1W)#1	0.86	2.24	2.75	117.5
O(11W)-H(111) \cdots O(2)#7	0.85	2.02	2.74	141.3
O(12W)-H(122) \cdots O(22)#4	0.86	1.98	2.79	156.3

^aSymmetry codes: # , -1+x,y,z; #2, x,1+y,z; #3, 1+x,1+y,z; #4, 1-x,1-y,-z; #5, -1+x,-1+y,z; #6, -x,1-y,1-z; #7, 2-x,2-y,-z.

Table S3. Geometry Parameters for Hydrogen Bonds in **2^b**

D-H \cdots A	d(D-H)	d(H \cdots A)	d(D \cdots A)	\angle DHA
O(1W)-H(1D) \cdots O(2)	0.94	1.83	2.72	154.8
O(2W)-H(2E) \cdots O(8)	0.94	1.77	2.71	176.5
O(3W)-H(3E) \cdots O(6)#1	0.95	1.96	2.74	137.9
O(4W)-H(4E) \cdots O(1)#2	0.94	1.98	2.87	158.2
O(1W)-H(1E) \cdots O(5W)#1	0.94	1.87	2.81	170.8
O(2W)-H(2D) \cdots O(5W)	0.94	1.99	2.77	139.3
O(3W)-H(3D) \cdots O(11)#3	0.95	1.85	2.74	153.4
O(4W)-H(4D) \cdots O(3)#4	0.94	1.99	2.83	147.2
O(5W)-H(51) \cdots O(7)	0.91	1.74	2.65	171.2
O(5W)-H(52) \cdots O(12)#5	0.86	1.97	2.74	146.1

^bSymmetry codes: #1, 1-x,1-y,1-z; #2, 2-x,1-y,1-z; #3, -1+x,-1+y,-1+z; #4, 1+x,1+y,1+z; #5, 1-x,2-y,1-z.

Table S4. Geometry Parameters for Hydrogen Bonds in **3^c**

D-H...A	d(D-H)	d(H...A)	d(D...A)	∠DHA
O(7W)-H(71)···O(1)#2	0.89	1.92	2.78	161.2
O(7W)-H(72)···O(3W)#5	0.85	1.89	2.74	169.8
O(8W)-H(81)···O(1W)#1	0.85	1.99	2.78	155.1
O(8W)-H(82)···O(5W)	0.86	1.92	2.76	162.0

^cSymmetry codes: #1, $x, -1+y, z$; #2, $1/2-x, 1/2+y, -1/2+z$; #3, $1/2-x, -1/2+y, 1/2+z$; #4, $1/2-x, -1/2+y, -1/2+z$; #5, $-x, 1-y, -1/2+z$.