10.1071/CH09166\_AC

© CSIRO 2009

Accessory Publication: Australian Journal of Chemistry, 2009, 62(7), 700-710

#### ACCESSORY PUBLICATION

# $\mathrm{S}_{\mathrm{N}}2$ Substitution Reactions at the Amide Nitrogen in the Anomeric Mutagens, N-Acyloxy-N-alkoxyamides

S<sub>N</sub>2 Substitution Reactions

Katie L. Cavanagh,<sup>A</sup> Stephen A. Glover,<sup>A,B</sup> Helen L. Price,<sup>A</sup> and Rhiannon R. Schumacher<sup>A</sup>

<sup>A</sup>Chemistry, School of Science and Technology, University of New England, Armidale, NSW 2351,

Australia.

<sup>B</sup>Corresponding author. Email: sglover@une.edu.au

#### **Supplementary DFT Computational Data: Structures and Energies**

Calculations carried out using Spartan 04 for macintosh Geometries were optimized at AM1, HF/6-31G\*/3-21G and HF/6-31G(d) levels Frequency calculations at HF/6-31G\* verified minima (all positive eigenvalues) and the transition state (one negative eigenvalue). Energies were calculated by DFT on the stationary points from HF/6-31G\*.

# Cartesian coordinates for transition state for reaction of ammonia with N-acetoxy-N-methoxyacetamide

Η	1.663016	-1.825474	1.577856
С	0.764666	-2.056100	1.015790
Η	0.718879	-3.117995	0.821113
С	0.798194	-1.365299	-0.315469
Η	-0.099206	-1.729176	1.569811
0	0.881775	-1.877846	-1.371699
Ν	0.870391	0.136835	-0.311247
0	1.191372	0.574077	0.848744
Ν	2.447115	0.287169	-1.297553
0	-1.332403	-0.462999	0.167623
С	0.865499	1.968051	1.026162
Η	1.633006	2.565279	0.548757
Η	0.893568	2.117895	2.093377
Η	-0.114601	2.138694	0.605967
С	-2.124447	0.445769	-0.198507
0	-1.854072	1.630631	-0.346356

С	-3.559758	-0.010339	-0.442682		
Η	-3.978838	-0.385233	0.486656		
Η	-3.570239	-0.831400	-1.152064		
Η	-4.170819	0.803552	-0.810092		
Η	2.316276	-0.322545	-2.084947		
Η	3.308754	0.074998	-0.823795		
Η	2.451872	1.241456	-1.607444		
Ima Ene	ginary frequen brgy: HF/ B3L	ncy 446.7cm <sup>-</sup> 6-31G(d) .YP/6-31G(d)	<sup>1</sup> corresponds to I //HF/6-31G(d)	N <sup>1</sup> N <sup>2</sup> O <sup>3</sup> stretch -604.484561 au -607.996238 au	-115.19 kJmol <sup>-1</sup>
Eleo	ctrostatic Grou	ıp Charges:	NH <sub>3</sub> N OCH OAc CH <sub>3</sub>	$\begin{array}{r} +0.4 \\ -0.09 \\ H_3 \\ -0.77 \\ CO \\ +0.3 \end{array}$	

## Cartesian coordinates for N-acetoxy-N-methoxyacetamide

С	1.260004	-0.123876	1.149298		
0	0.851706	-0.557041	2.172708		
С	2.702492	0.205553	0.872271		
Ν	0.392048	0.024404	0.041532		
Η	3.294297	-0.174788	1.691720		
Η	2.827782	1.278729	0.795260		
Η	3.031861	-0.234721	-0.060582		
0	-0.900679	0.207263	0.463823		
0	0.739679	1.068776	-0.758996		
С	-1.767563	-0.765054	0.097350		
0	-1.511539	-1.630739	-0.660448		
С	-3.078156	-0.545720	0.797544		
С	0.602116	0.729601	-2.123727		
Η	-2.933030	-0.662289	1.865156		
Η	-3.797784	-1.266818	0.440098		
Η	-3.434089	0.461804	0.618590		
Η	1.203904	-0.136776	-2.362318		
Η	0.949089	1.592945	-2.672296		
Η	-0.432139	0.528749	-2.366983		
Ene	ergy: HF/	6-31G(d)		-548.3834875 au	-17.02 kJmol <sup>-1</sup>
	B3L	2YP/6-31G(d)	//HF/6-31G(d)	-551.497474 au	
Ele	ctrostatic Grou	p Charges:			

Ν	+0.22
$OCH_3$	+0.03
OAc	-0.04
CH <sub>3</sub> CO	0.23

#### **Cartesian coordinates for Ammonia**

Н -0.931597 0.000000 -0.092570

N H H	0.0000 0.4657 0.4657	000 798 798	0.000000 -0.806787 0.806787	0.277710 -0.092571 -0.092571		
Energ	y:	HF/6- B3LY	-31G(d) 2P/6-31G(d)	//HF/6-31G(d)	-56.1843562 au -56.5472397au	-28.12 kJmol <sup>-1</sup>

### Cartesian coordinates for N-ammonium-N-methoxyacetamide

Η	1.172855	2.721849	-0.859750
С	1.122558	1.659770	-0.675712
С	-0.207264	1.361632	-0.060515
Η	1.916550	1.363594	-0.001911
Η	1.241197	1.116078	-1.604829
0	-1.029652	2.135447	0.294160
Ν	-0.542882	-0.040336	0.046665
0	0.485130	-0.822240	0.471525
Ν	-1.616476	-0.205866	0.969849
С	0.716221	-1.943477	-0.390496
Η	0.946850	-1.602720	-1.387802
Η	-0.147544	-2.594005	-0.402556
Η	1.560922	-2.452876	0.043215
Η	-2.179534	0.642855	0.939529
Η	-1.262444	-0.335657	1.911917
Η	-2.176486	-1.004049	0.706710

Energy:	HF/631G(d)	-377.1392218 -268.53 kJmol <sup>-1</sup>
	B3LYP/6-31G(d)//HF/6-31G(d)	-379.3433782

### **Cartesian coordinates for acetate**

С	0.082077	0.000000	-0.897205
С	-0.039357	0.000000	0.651982
Η	0.932092	0.000000	1.137179
Η	-0.602549	-0.873396	0.976526
Η	-0.602549	0.873396	0.976526
0	1.229227	0.000000	-1.350112
0	-0.998943	0.000000	-1.494894

Energy:	HF/631G(d)
	B3LYP/6-31G(d)//HF/6-31G(d)

-227.2250684 -298.61 kJmol<sup>-1</sup> -228.4958189

## $E_{\rm A}$ (kJmol<sup>-1</sup>):

	No solvation	Aqueou solvation
HF/6-31G(d)	218.92	148.79
B3LYP/6-31G(d)	127.42	57.30

# $\Delta E_{\text{react}} (\text{kJmol}^{-1})$ :

	No solvation	Aqueou solvation
HF/6-31G(d)	535.06	12.44
B3LYP/6-31G(d)	540.22	17.60

## Nuclear Magnetic Resonance Spectra

<sup>13</sup>C NMR *N*-(2-Butoxy)benzamide



# <sup>1</sup>H NMR *N*-(2-Butoxy)benzamide





<sup>13</sup>C NMR *N*-(2-Methylbenzyloxy)benzamide

# <sup>1</sup>H NMR *N*-(2-Methylbenzyloxy)benzamide







# <sup>1</sup>H NMR *N*-(3-Methylbenzyloxy)benzamide





<sup>13</sup>C NMR *N*-(2,6-Dimebenzyloxy)benzamide

<sup>1</sup>H NMR *N*-(2,6-Dimebenzyloxy)benzamide



# <sup>1</sup>H NMR *N*-(3,5-Dimethylbenzyloxy)benzamide



<sup>1</sup>H NMR *N*-tert-Butoxybenzamide





<sup>13</sup>C NMR *N*-Acetoxy-*N*-(2-butoxy)benzamide**11b** 





















<sup>13</sup>C NMR *N*-Acetoxy-*N*-(2,6-dimethylbenzyloxy)benzamide**12e** 





<sup>13</sup>C NMR *N*-Butanoyloxy-*N*-butoxybenzamide**13b** 













<sup>1</sup>H NMR *N*-Butoxy-*N*-(3,3-dimethylbutanoyloxy)benzamide**13e** 





<sup>1</sup>H NMR *N*-Butoxy-*N*-(2,2-dimethylpropanoyloxy)benzamide**13f** 



<sup>13</sup>C NMR *N*-(Adamantane-1-carboxoyloxy)-*N*-butoxybenzamide **13g** 







· \_\_\_\_

------

.









