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## ACCESSORY PUBLICATION

Synthesis, Structure and Theoretical Investigations of an Alkaline Earth Vanadate Oxide Compound (Ca<sub>4</sub>V<sub>4</sub>O<sub>14</sub>): Electronic, Optical and Chemical Bond Properties

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V1-O1 <sup>A</sup>	2.053(3)	Ca1-O1 <sup>A</sup>	2.467(3)	Ca2-O2 <sup>C</sup>	2.938(3)
V1-O2	1.864(3)	Ca1-O1	3.289(3)	Ca2-O3 <sup>E</sup>	2.392(3)
V1-O2 <sup>H</sup>	1.883(3)	Ca1-O2	2.333(3)	Ca2-O3 <sup>D</sup>	2.413(3)
V1-O6	1.651(3)	Ca1-O3	3.453(3)	Ca2-O4	3.166(3)
V1-07	1.672(3)	Cal-O4	2.455(3)	Ca2-O5	2.423(3)
V2-O1	1.740(3)	Ca1-O4 <sup>C</sup>	2.407(3)	Ca2-O5 <sup>G</sup>	2.532(3)
V2-O3	1.695(3)	Ca1-O5 <sup>B</sup>	2.391(3)	Ca2-O6 <sup>F</sup>	2.406(3)
V2-O4	1.689(3)	Cal-O6 <sup>D</sup>	2.462(3)	Ca2-O6	3.629(3)
V2-05	1.730(3)	Cal-O7 <sup>A</sup>	2.310(3)	Ca2-O7 <sup>D</sup>	2.317(3)
O6-V1-O7	112.40(14)	O2-V1-O1 <sup>A</sup>	80.56(12)	O4-V2-O5	106.90(15)
O6-V1-O2	117.46(13)	02 <sup>H</sup> -V1-O1 <sup>A</sup>	158.43(12)	O3-V2-O5	109.12(14)
07-V1-O2	129.98(14)	O2-V1-O2 <sup>H</sup>	80.13(13)	O4-V2-O1	108.37(14)

Table S1. Selected bond lenghts (Å) and bond angles (°) for 1

06-V1-O2 <sup>H</sup>	101.58(14)	06-V1-01 <sup>A</sup>	95.79(13)	O3-V2-O1	112.09(14)
07-V1-O2 <sup>H</sup>	94.09(13)	O4-V2-O3	109.60(14)	05-V2-01	110.62(14)
07-V1-01 <sup>A</sup>	90.97(13)				

Symmetry transformations used to generate equivalent atoms: <sup>A</sup> -x + 2, -y + 1, -z + 1; <sup>B</sup> x, y - 1, z; <sup>C</sup> -x + 1, -y + 1, -z + 2; <sup>D</sup> x, y, z + 1; <sup>E</sup> -x + 1, -y + 2, -z + 2; <sup>F</sup> x, y + 1, z + 1; <sup>G</sup> -x + 2, -y + 2, -z + 2; <sup>H</sup> -x + 1, -y + 1, -z + 1.



**Fig. S1.** Perspective view of the stacking structure in crystalline **1**.  $(Ca^{2+}: purple, V^{5+}: green, and O^{2-}: red)$ 



Fig. S2. Calculated energy band structure of 1.



Fig. S3. Calculated dielectric function in different polarization directions of 1.



Fig. S4. Calculated real and imaginary part of dielectric functions of polycrystalline 1.