Supplementary Material

Structures and photoluminescence properties of four cadmium(II) coordination polymers synthesized by rigid ligands and N-Donor Ligands

Qing-Guo Meng^{A,B}, Lin-Tong Wang^A, Ji-Tao Lu^A, Xin Wang^A, Wei Lu^A, Zi-Yu Song^A

^A Chemistry & Chemical and Environmental Engineering College, Weifang University, Weifang Shandong 261061, P. R. China

^B Corresponding author. email: <u>mengqg@wfu.edu.cn</u>

Compound 1					
Cd1—O2	2.206 (11)	Cd1—O8 ⁱ	2.231 (10)	Cd1—N2	2.308 (12)
Cd1—O3	2.314 (10)	Cd1—N1	2.367 (10)	Cd1—O1	2.552 (9)
Cd2—O4	2.215 (10)	Cd2—O5 ⁱ	2.270 (10)	Cd2—N3	2.333 (12)
Cd2—O7 ⁱ	2.342 (9)	Cd2—N4	2.393 (11)	Cd2—O6 ⁱ	2.564 (10)
O8 ⁱ —Cd1—N2	87.6 (4)	O2—Cd1—O3	80.6 (4)	O8 ⁱ —Cd1—O3	105.4 (4)
N2—Cd1—O3	79.2 (4)	O2-Cd1-N1	98.6 (4)	O8 ⁱ —Cd1—N1	108.5 (4)
N2—Cd1—N1	69.2 (4)	O3—Cd1—N1	131.8 (4)	O2-Cd1-O1	53.1 (4)
$O8^{i}$ —Cd1—O1	90.7 (3)	N1-Cd1-O1	88.5 (4)	O4—Cd2—N3	107.5 (4)
O5 ⁱ —Cd2—N3	95.8 (4)	$O5^{i}$ —Cd2— $O7^{i}$	82.2 (3)	O4—Cd2—N4	89.0 (4)
N3—Cd2—N4	69.1 (4)	O7 ⁱ —Cd2—N4	80.2 (4)	$O4$ — $Cd2$ — $O6^{i}$	89.4 (4)
$O5^{i}$ —Cd2— O^{i}	54.3 (3)	N3—Cd2—O6 ⁱ	85.8 (3)	$O7^{i}$ —Cd2— $O6^{i}$	126.0 (3)
Symmetry codes: (i) $x+1$, y, z; (ii) $x-1$, y, z.					
Compound 2					
Cd1—N1	2.188 (4)	$Cd1-O3^{1}$	2.281 (3)	$Cd1-O4^{n}$	2.345 (3)
Cd1—O2	2.367 (3)	Cd1—O1	2.403 (3)	$Cd1-O1^{m}$	2.429 (3)
$N1$ — $Cd1$ — $O3^{i}$	88.70 (13)	N1—Cd1—O4 ⁱⁱ	92.96 (13)	$O3^{i}$ —Cd1—O4 ⁱⁱ	90.90 (6)
N1—Cd1—O2	165.31 (13)	$O3^{1}$ —Cd1—O2	85.30 (12)	$O4^{11}$ —Cd1—O2	94.61 (12)
N1-Cd1-01	138.69 (13)	O3 ¹ —Cd1—O1	94.10 (12)	$O4^{n}$ —Cd1—O1	80.20 (11)
O2-Cd1-O1	55.33 (10)	N1—Cd1—O1 ⁱⁱⁱ	89.22 (13)	O3 ⁱ —Cd1—O1 ⁱⁱⁱ	101.19 (13)
O4 ⁱⁱ —Cd1—O1 ⁱⁱⁱ	85.60 (11)	02—Cd1—O1 ⁱⁱⁱ	78.85 (10)	O1—Cd1—O1 ⁱⁱⁱ	130.16 (9)
Symmetry codes: (i) x, $-y+1/2$, $z-1/2$; (ii) $-x$, $-y+1$, $-z+1$; (iii) $-x$, $y+1/2$, $-z+1/2$; (iv) $-x$, $y-1/2$, $-z+1/2$; (v) x, $-y+1/2$, $z+1/2$.					
Compound 3					
Cd1-08	2253(3)	Cd1—N2	2274(3)	$Cd1-O4^{i}$	2 302 (3)
Cd1 = 02	2.233(3) 2 320(2)	$Cd1-O2^{ii}$	2.271(3) 2 412(2)	$Cd1-O3^{i}$	2.502(3) 2 519(3)
Cd1 = 01	2.520(2) 2.629(3)	$Cd1 - C8^{i}$	2.112(2) 2.741(3)	N2—Cd1— 04^{i}	140.60(10)
08—Cd1—N2	$108\ 23\ (12)$	08 —Cd1— 04^{i}	87 59 (10)	08-Cd1-02	91 01 (10)
N2—Cd1—O2	124.54 (10)	$O4^{i}$ —Cd1—O2	89.81 (9)	$O8-Cd1-O2^{ii}$	167.19 (10)
N2—Cd1— $O2^{ii}$	83.90 (10)	$O4^{i}$ —Cd1— $O2^{ii}$	85.27 (9)	$O2$ — $Cd1$ — $O2^{ii}$	78.35 (9)
O_{2}^{0} Cd1 O_{2}^{i}	92 35 (11)	$N2$ Cd1 $O2^{i}$	01 55 (10)	$O4^{i}$ Cd1 $O2^{i}$	53 60 (0)
08-Cui-03	85.55 (11)	N2—Cu1—O5	91.33 (10)	04-03	55.09 (9)
$O2-Cd1-O3^{1}$	143.15 (9)	$O2^{n}$ —Cd1—O3 ¹	100.82 (9)	08—Cd1—O1	74.74 (10)
N2—Cd1—O1	82.48 (10)	$O4^{i}$ —Cd1—O1	136.93 (9)	O2—Cd1—O1	52.51 (8)
$O2^{n}$ —Cd1—O1	103.57 (8)	O3 ¹ —Cd1—O1	154.07 (10)		
Symmetry codes: (1) x, y-1, z; (11) $-x+1$, $-y$, $-z+1$; (111) x, y+1, z; (1v) $-x+2$, $-y$, $-z$.					
Compound 4					
Cd1—01	2.278 (4)	Cd1—O5	2.352 (4)	Cd1—O3	2.354 (4)
Cd1—N2	2.356 (4)	Cd1—N1	2.381 (5)	Cd1—O4	2.445 (4)
Cd1—O2	2.631 (4)	01—Cd1—O5	85.92 (15)	O1—Cd1—O3	136.01 (14)
O5—Cd1—O3	88.40 (14)	01—Cd1—N2	84.99 (16)	O5—Cd1—N2	94.67 (14)
O3—Cd1—N1	87.11 (16)	01—Cd1—N1	89.76 (16)	N2-Cd1-N1	95.86 (15)
O5-Cd1-O4	91.08 (14)	O3—Cd1—O4	53.99 (13)	N2-Cd1-O4	85.02 (15)
N1-Cd1-O4	95.07 (15)	O1-Cd1-O2	51.30 (14)	O5—Cd1—O2	85.46 (13)
O3—Cd1—O2	84.79 (12)	N1-Cd1-O2	83.30 (14)		
Symmetry codes: (i) x, -y, $z-1/2$; (ii) x, -y, $z+1/2$; (iii) $x+1/2$, $-y+1/2$, $z+1/2$; (iv) $x-1/2$, $-y+1/2$, $z-1/2$.					

(1) Table S1. Selected bond lengths (Å) and angles (°) for 1-4.

(2) Figure S1 The powder XRD patterns and the simulated ones from the singlecrystal diffraction data for compounds 1-4.



(3) Figure S2 Thermogravimetric analyses(TGA) of 1-4.

