

## Supplementary Material

# Heterometallic tetranuclear $\{\text{Mn}^{\text{III}}_2\text{Ln}^{\text{III}}_2\}_n$ 1-D coordination polymers: Employing sulfonate ligands as connecting groups

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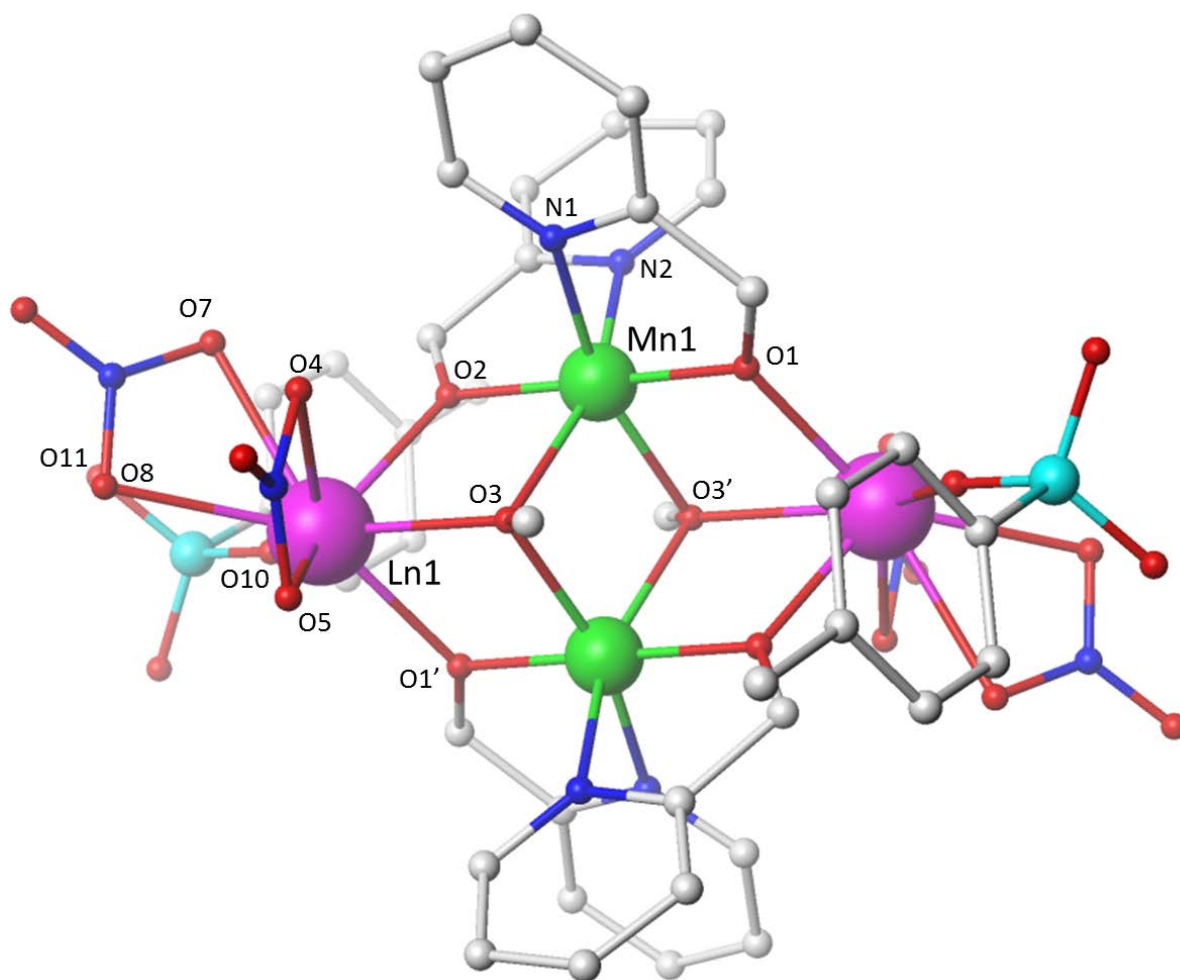
**Table S1.** Crystallographic data for compounds **1 - 5**.

	<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>
Formula <sup>A</sup>	Mn <sub>2</sub> Pr <sub>2</sub> C <sub>42</sub> H <sub>51</sub> O <sub>26</sub> N <sub>9</sub> S <sub>2</sub>	Mn <sub>2</sub> Nd <sub>2</sub> C <sub>42</sub> H <sub>51</sub> O <sub>26</sub> N <sub>9</sub> S <sub>2</sub>	Mn <sub>2</sub> Gd <sub>2</sub> C <sub>42</sub> H <sub>51</sub> O <sub>26</sub> N <sub>9</sub> S <sub>2</sub>	Mn <sub>2</sub> Tb <sub>2</sub> C <sub>42</sub> H <sub>51</sub> O <sub>26</sub> N <sub>9</sub> S <sub>2</sub>
M, g mol <sup>-1</sup>	1553.71	1560.37	1586.39	1589.73
Crystal system	monoclinic	Monoclinic	Monoclinic	Monoclinic
Space group	<i>P2<sub>1</sub>/c</i>	<i>P2<sub>1</sub>/c</i>	<i>P2<sub>1</sub>/c</i>	<i>P2<sub>1</sub>/c</i>
a/[Å]	10.935(2)	10.906(2)	10.845(2)	10850(2)
b/[Å]	23.709(5)	23.926(5)	23.848(5)	23.920(5)
c/[Å]	12.183(2)	12.189(2)	12.151(2)	12.130(2)
α/[°]	90	90	90	90
β/[°]	113.69(3)	113.68(3)	113.35(3)	113.36(3)
γ/[°]	90	90	90	90
V/[Å <sup>3</sup> ]	2892.4(11)	2912.8(12)	2885.2(11)	2890.2(10)
T/K	100(2)	100(2)	100(2)	100(2)
Z	2	2	2	2
ρ <sub>calc</sub> [g cm <sup>-3</sup> ]	1.781	1.776	1.823	1.823
Λ <sup>B</sup> /[Å]	0.71079	0.71079	0.71079	0.71079
Data Measured	17915	36088	36431	18892
Ind. Reflns	4941	5074	4922	5022
R <sub>int</sub>	0.0700	0.0427	0.0888	0.0370
Reflns with I > 2σ(I)	4044	4899	4771	4746
Parameters	408	420	421	420
Restraints	95	64	64	64
R <sub>1</sub> <sup>C</sup> (obs), wR <sub>2</sub> <sup>C</sup> (all)	0.0924, 0.2365	0.0657, 0.1469	0.0516, 0.1307	0.0610, 0.1373
goodness of fit	1.081	1.195	1.131	1.222
Largest residuals/[e Å <sup>-3</sup> ]	2.522, -2.003	2.657, -1.866	1.384, -1.667	1.328, -2.056
	<b>5</b>			
Formula <sup>A</sup>	Mn <sub>2</sub> Dy <sub>2</sub> C <sub>42</sub> H <sub>51</sub> O <sub>26</sub> N <sub>9</sub> S <sub>2</sub>			
M, g mol <sup>-1</sup>	1596.89			
Crystal system	monoclinic			
Space group	<i>P2<sub>1</sub>/c</i>			
a/[Å]	10.842(2)			
b/[Å]	23.762(5)			
c/[Å]	12.108(2)			
α/[°]	90			
β/[°]	113.12(3)			
γ/[°]	90			
V/[Å <sup>3</sup> ]	2868.7(11)			
T/K	100(2)			
Z	2			
ρ <sub>calc</sub> [g cm <sup>-3</sup> ]	1.845			
Λ <sup>B</sup> /[Å]	0.71079			
Data Measured	18492			
Ind. Reflns	4732			
R <sub>int</sub>	0.0319			
Reflns with I > 2σ(I)	4626			
Parameters	421			

Restraints 64  
 $R_1^C(\text{obs})$ , 0.0570, 0.1319  
 $wR_2^C(\text{all})$   
 goodness of fit 1.230  
 Largest  
 residuals/[ $e \text{ \AA}^{-3}$ ] 1.959, -1.730

<sup>A</sup> Including solvate molecules. <sup>B</sup> Graphite monochromator.

<sup>C</sup>  $R1 = \frac{\sum ||F_o| - |F_c||}{\sum |F_o|}$ ,  $wR2 = \frac{\{\sum [w(F_o^2 - F_c^2)^2]\}^{1/2}}{\{\sum [w(F_o^2)^2]\}^{1/2}}$ .



**Table S2.** Selected bond distances ( $\text{\AA}$ ) and angles ( $^\circ$ ) for complexes **1** – **5**, using the labelling scheme shown above.

	<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>	<b>5</b>
Ln1-O10	2.393(9)	2.393(6)	2.339(5)	2.327(6)	2.315(6)
Ln1-O11'	2.438(7) <sup>I</sup>	2.417(6) <sup>I</sup>	2.381(5) <sup>II</sup>	2.373(6) <sup>I</sup>	2.363(6) <sup>I</sup>
Ln1-O1'	2.445(9) <sup>I</sup>	2.428(6) <sup>I</sup>	2.373(4) <sup>II</sup>	2.356(6) <sup>I</sup>	2.339(6) <sup>I</sup>

Ln1-O2	2.446(8)	2.417(5)	2.377(5)	2.357(6)	2.332(5)
Ln1-O5	2.487(17)	2.489(9)	2.508(6)	2.449(8)	2.440(9)
Ln1-O3/O3x (*)	2.55(4)	2.51(4)	2.465(3)	2.440(5)	2.425(4)
Ln1-O7	2.551(10)	2.549(6)	2.500(5)	2.510(7)	2.506(7)
Ln1-O4	2.558(14)	2.563(8)	2.451(7)	2.513(8)	2.503(8)
Ln1-O8	2.575(8)	2.553(6)	2.513(5)	2.478(6)	2.474(6)
Mn1-O3'/O3x (*)	1.936(2) <sup>I</sup>	2.015(3) <sup>I</sup>	1.992(5) <sup>II</sup>	1.984(4) <sup>I</sup>	2.015(3) <sup>I</sup>
Mn1-O1	1.901(7)	1.898(6)	1.906(5)	1.906(6)	1.899(6)
Mn1-O2	1.909(8)	1.904(6)	1.889(5)	1.920(6)	1.922(6)
Mn1-N1	2.064(12)	2.064(8)	2.188(6)	2.059(8)	2.084(8)
Mn1-N2	2.174(11)	2.182(7)	2.062(6)	2.193(7)	2.172(8)
Mn1-O3/O3x (*)	2.2035(3)	2.190(3)	2.185(3)	2.215(5)	2.163(4)
Mn1...Mn1'	3.166(3)	3.147(4)	3.112(4)	3.133(3)	3.081(3)
Ln1...Mn1	3.632(2)	3.614(4)	3.481(4)	3.570(4)	3.530(4)
Ln1...Mn1'	3.571(2)	3.544(4)	3.567(4)	3.487(4)	3.468(4)
Mn1-O3/O3x- Mn1 (*)	98.395(2)	96.91(2)	96.72(2)	96.51(2)	95.42(2)

Symmetry transformation: (I) - x, - y, - z; (II) 1 - x, 1 - y, 1 - z.

(\*) It was found that a methoxide and hydroxide bridging ligand occupy the same site throughout the crystal, modelled at 50:50 occupancy. The asterix denotes an average bond distance and angle from the two ligands.

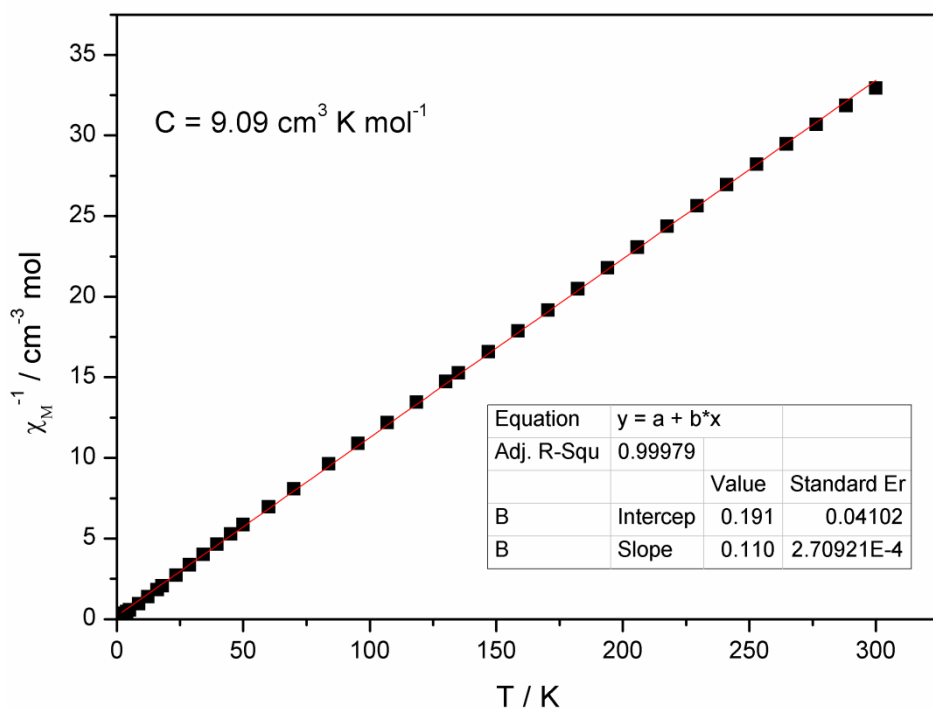


Figure S1.  $1/\chi_M$  versus  $T$  for 1.

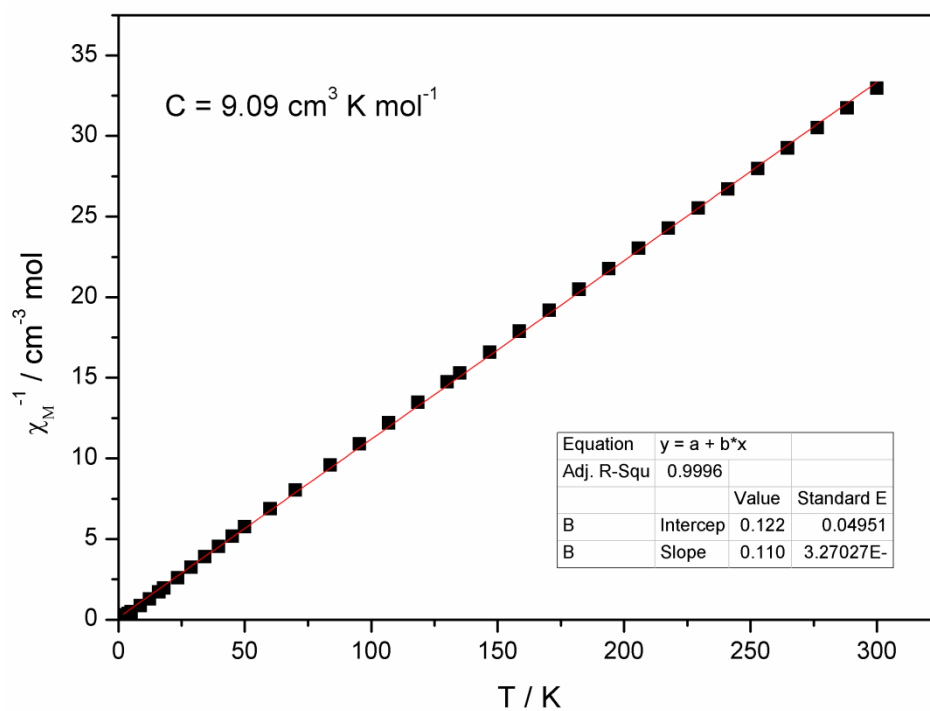
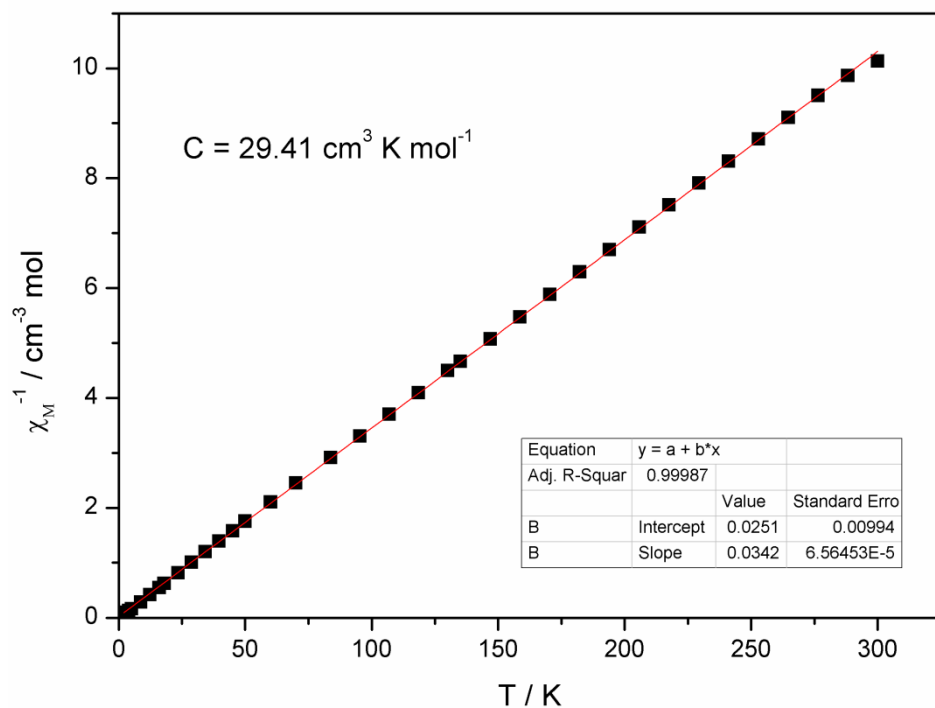
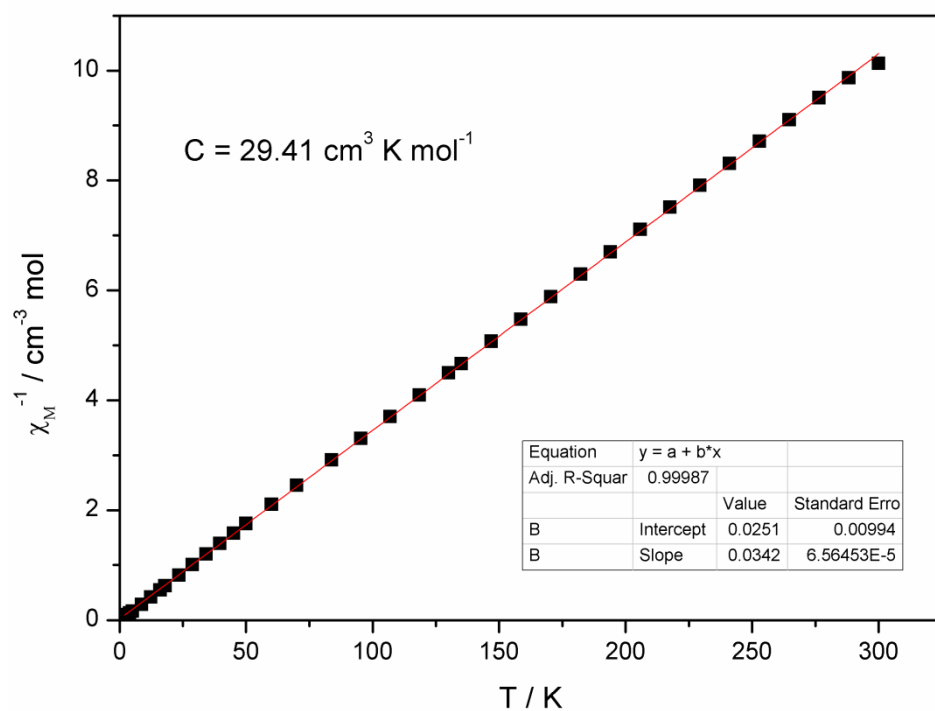


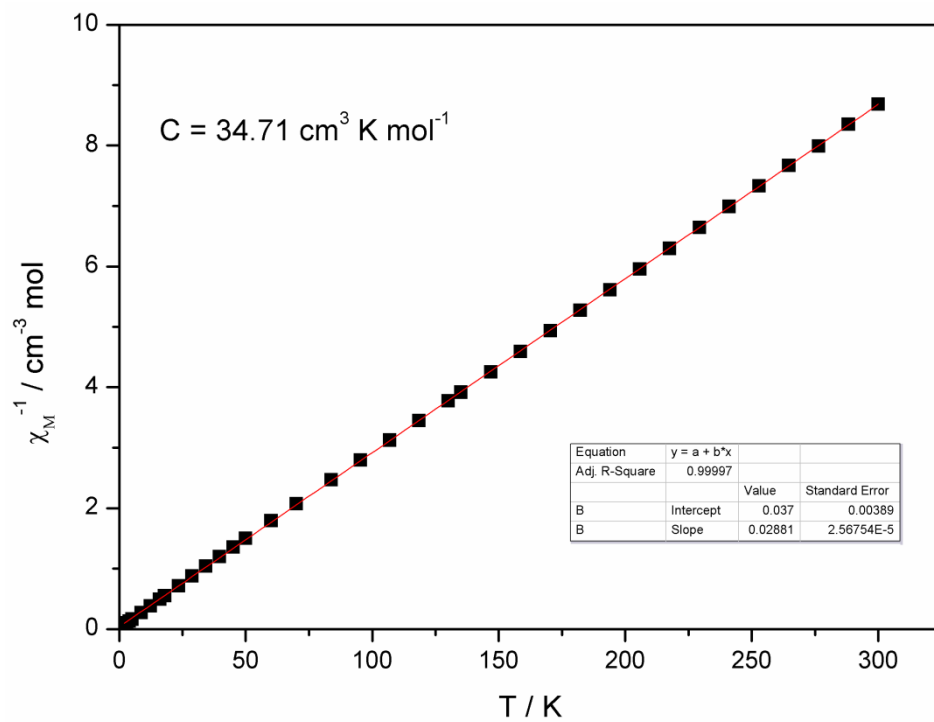
Figure S2.  $1/\chi_M$  versus  $T$  for 2.



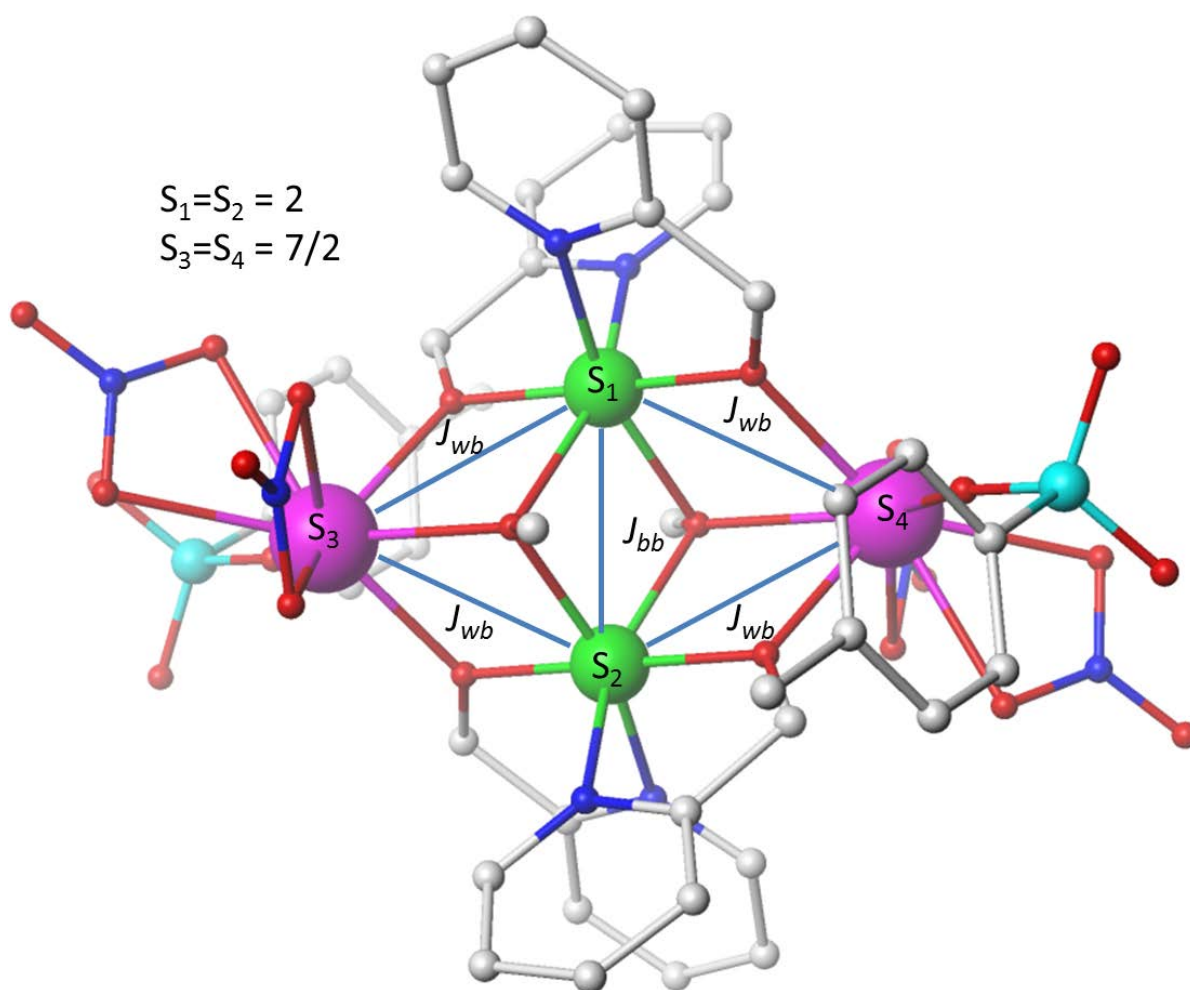
**Figure S3.**  $1/\chi_M$  versus  $T$  for 3.



**Figure S4.**  $1/\chi_M$  versus  $T$  for 4.



**Figure S5.**  $1/\chi_M$  versus  $T$  for **5**.



$$\hat{H} = -2J_{bb}(\hat{S}_1\hat{S}_2) - 2J_{wb}(\hat{S}_1\hat{S}_3 + \hat{S}_1\hat{S}_4 + \hat{S}_2\hat{S}_3 + \hat{S}_2\hat{S}_4)$$

**Figure S6.** Magnetic exchange coupling scheme used for fitting the magnetic data for complex 3.