

## Supplementary Material

### Construction of a New Zn(II) Coordination Polymer for Selective Fluorescence Sensing of CCl<sub>4</sub>

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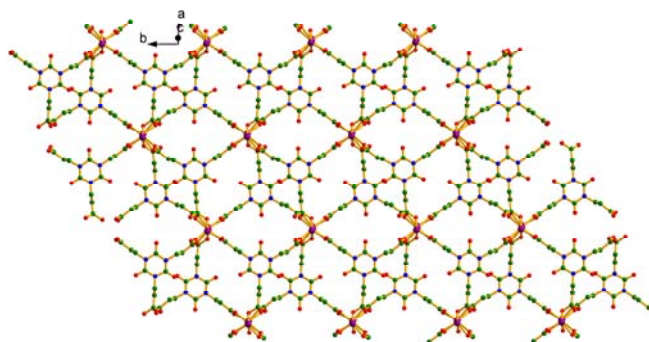
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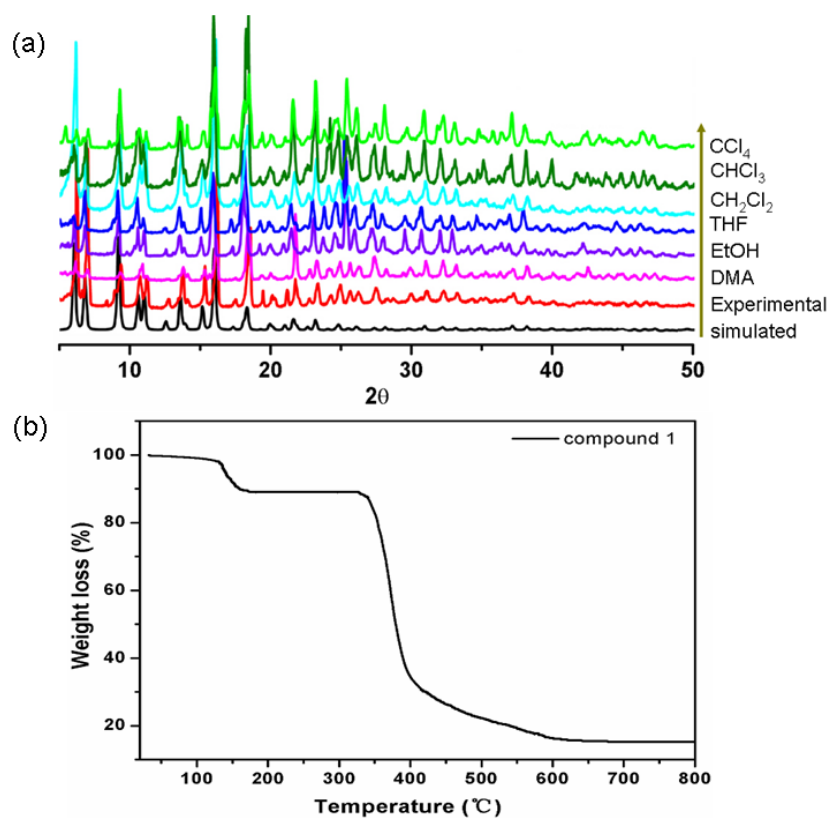
**Table S1.** Selected bond lengths (Å) and angles (°) for compound **1**

Zn(1)-O(6) <sup>a</sup>	1.936(4)	Zn(1)-O(2)	1.952(4)
Zn(1)-N(1)	1.988(5)	Zn(1)-O(4) <sup>b</sup>	1.993(5)
Zn(2)-O(1) <sup>c</sup>	2.051(4)	Zn(2)-O(1)	2.051(4)
Zn(2)-O(5) <sup>a</sup>	2.106(4)	Zn(2)-O(5) <sup>c</sup>	2.106(4)
Zn(2)-O(4) <sup>d</sup>	2.143(4)	Zn(2)-O(4) <sup>b</sup>	2.143(4)
O(6) <sup>a</sup> -Zn(1)-O(2)	102.7(2)	O(6) <sup>a</sup> -Zn(1)-N(1)	116.3(2)
O(2)-Zn(1)-N(1)	102.39(19)	O(6) <sup>a</sup> -Zn(1)-O(4) <sup>b</sup>	113.85(18)
O(2)-Zn(1)-O(4) <sup>b</sup>	104.42(18)	N(1)-Zn(1)-O(4) <sup>b</sup>	114.8(2)
O(1) <sup>c</sup> -Zn(2)-O(1)	88.2(3)	O(1) <sup>c</sup> -Zn(2)-O(5) <sup>a</sup>	179.94(18)
O(1)-Zn(2)-O(5) <sup>a</sup>	91.75(17)	O(1) <sup>c</sup> -Zn(2)-O(5) <sup>c</sup>	91.75(17)
O(1)-Zn(2)-O(5) <sup>c</sup>	179.94(18)	O(5) <sup>a</sup> -Zn(2)-O(5) <sup>c</sup>	88.3(2)
O(1) <sup>c</sup> -Zn(2)-O(4) <sup>d</sup>	94.39(16)	O(1)-Zn(2)-O(4) <sup>d</sup>	90.88(16)
O(5) <sup>a</sup> -Zn(2)-O(4) <sup>d</sup>	85.65(16)	O(5) <sup>c</sup> -Zn(2)-O(4) <sup>d</sup>	89.08(16)
O(1) <sup>c</sup> -Zn(2)-O(4) <sup>b</sup>	90.88(16)	O(1)-Zn(2)-O(4) <sup>b</sup>	94.39(16)
O(5) <sup>a</sup> -Zn(2)-O(4) <sup>b</sup>	89.08(16)	O(5) <sup>c</sup> -Zn(2)-O(4) <sup>b</sup>	85.65(16)
O(4) <sup>d</sup> -Zn(2)-O(4) <sup>b</sup>	172.7(2)		

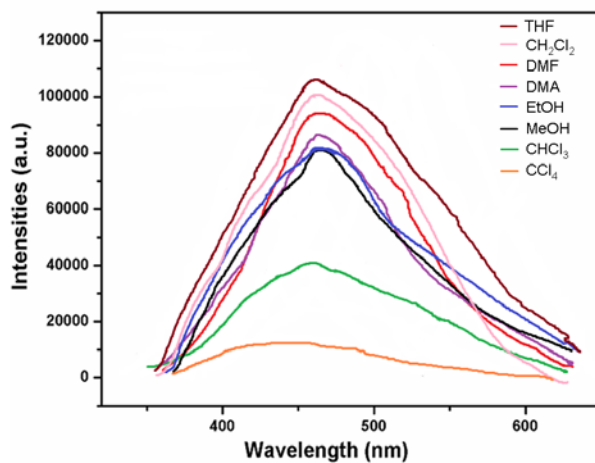
Symmetry codes: (a)  $x, y + 1, z$ ; (b)  $x - 1, -y - 1, z - 1/2$ ; (c)  $-x - 1, y + 1, -z + 1/2$ ; (d)  $-x, -y - 1, -z + 1$ ; (e)  $-x - 1, y, -z + 1/2$ .



**Fig. S1.** The 2D layer constructed by the trinuclear  $Zn_3$  clusters and  $tci^{3-}$  ligands.



**Fig.S2** (a) PXRD patterns for compound **1**. (b) TGA curve for compound **1**.



**Fig.S3** The emission spectra of the samples immersed in different solvents.