

10.071/CH15630\_AC

©The Authors 2016

Australian Journal of Chemistry 2016, 69(8), 836-845

## **Supplementary Material**

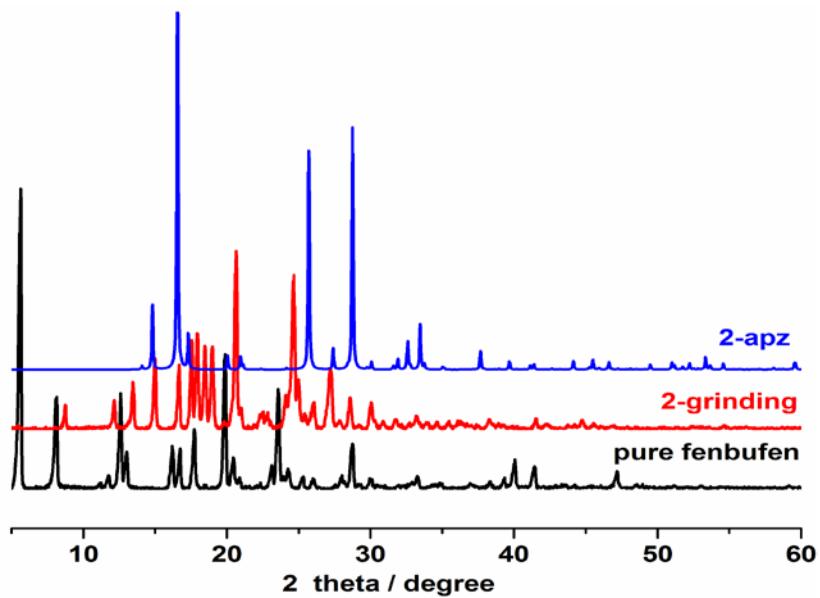
# **Effect of co-crystal formers on the supramolecular patterns and luminescent properties of co-crystals comprising Fenbufen and diverse *N*-heterocycles**

*Fa-Yuan Ge,<sup>A</sup> Li-Hua Huo,<sup>A</sup> Shu-Nan Zhao,<sup>A</sup> Zhao-Peng Deng,<sup>A,C</sup> Zhu-Yan Zhang<sup>B</sup> and Shan Gao<sup>A,C</sup>*

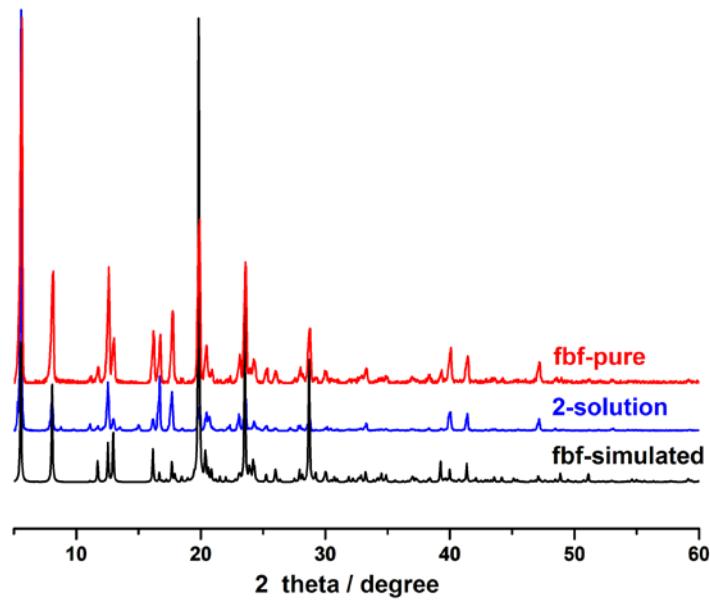
<sup>A</sup> Key Laboratory of Functional Inorganic Material Chemistry, Ministry of Education, School of Chemistry and Materials Science, Heilongjiang University, Harbin 150080, People's Republic of China.

<sup>B</sup> Laboratory Centre of Pharmacy, College of Pharmacy Harbin Medical University, Harbin 150081, People's Republic of China

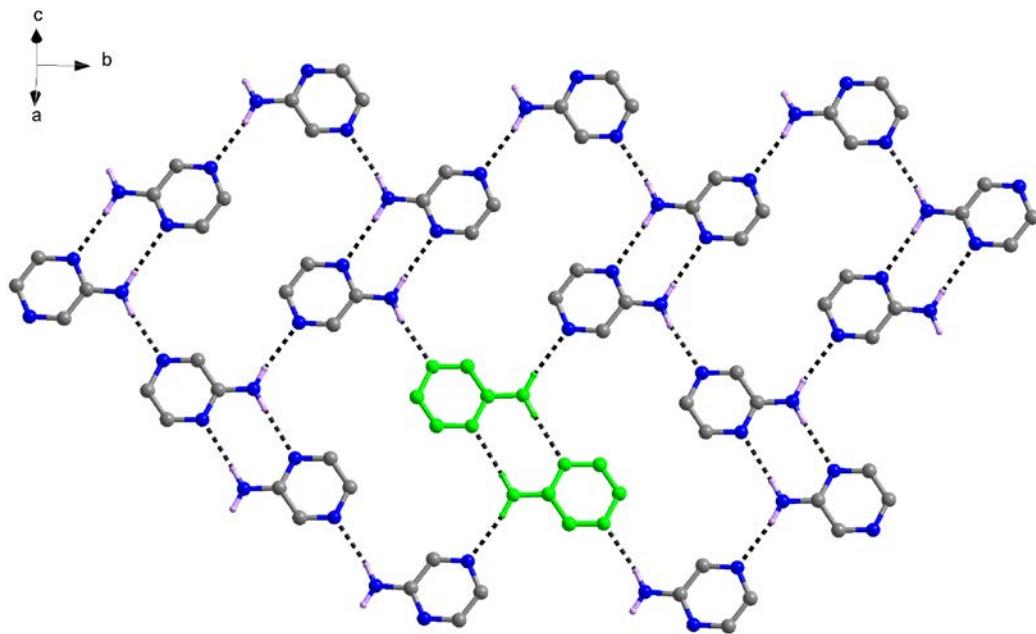
<sup>C</sup> Corresponding authors. E-mail: dengzhaopeng@hlju.edu.cn; shangao67@yahoo.com



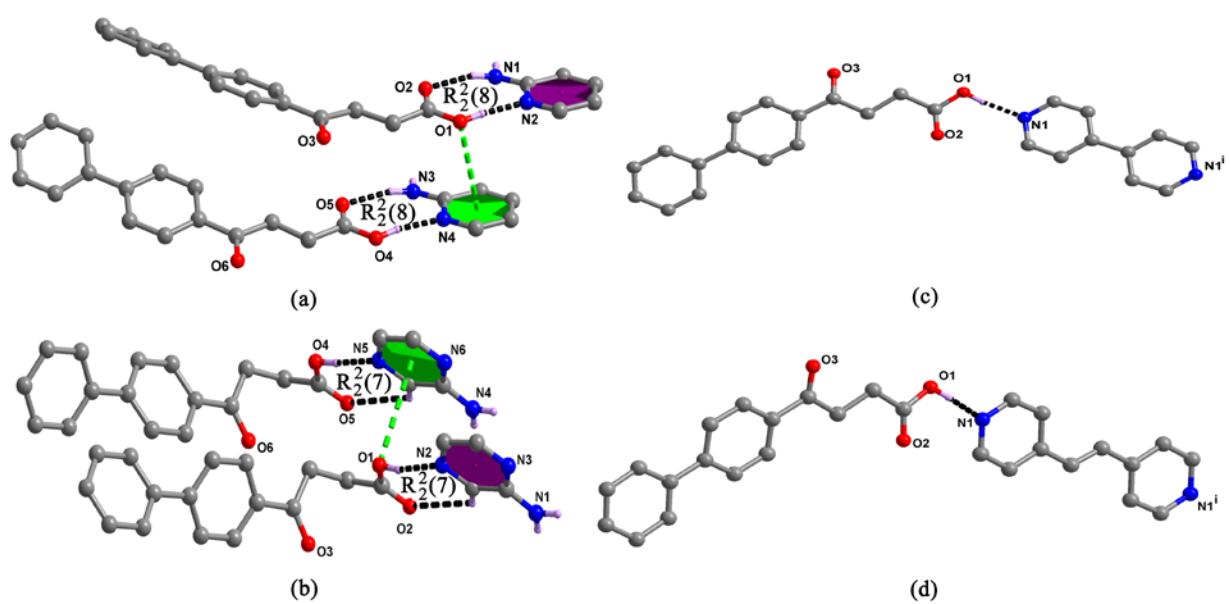
**Fig. S1** Comparison of the PXRD patterns among the pure Fbf, 2-apz and new phase obtained from grinding method.



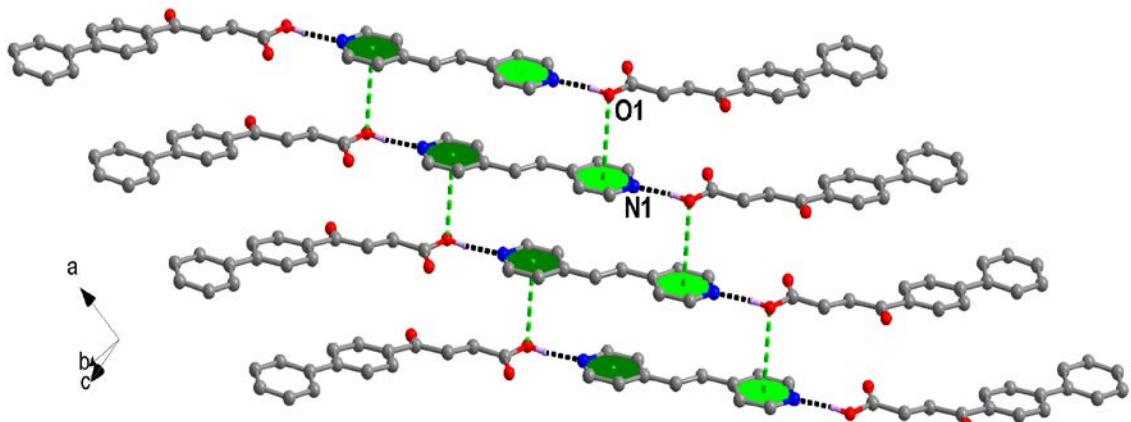
**Fig. S2** Comparison of the PXRD patterns among the pure Fbf and precipitates from the solution for preparing compound 2, as well as the simulated PXRD pattern from crystal structure of Fbf.



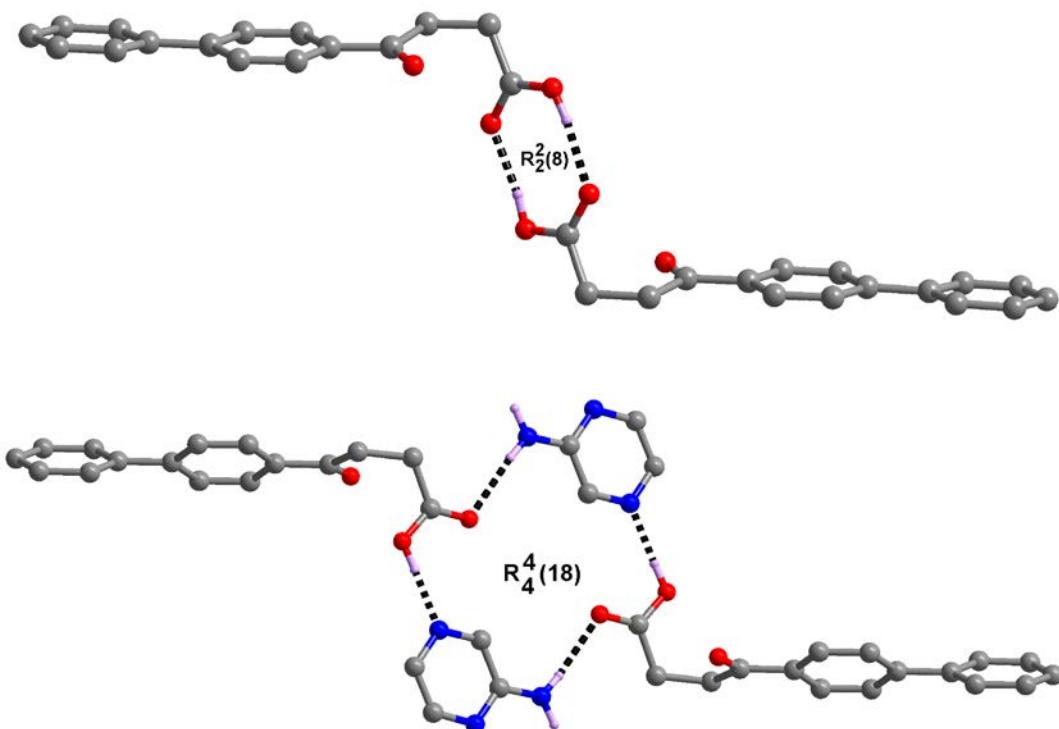
**Fig. S3** Infinite rigid layer structure assemble from pure 2-apz with the dimer highlighted by green color.



**Fig. S4** Crystal structures of compounds **1** (a), **2** (b), **3** (c) and **4** (d).



**Fig. S5** Tape structure in compound **4** extended by hydrogen bonding and lone pair... $\pi$  interactions.



**Fig. S6** Chair-shaped dimer in pure Fbf (top) and compound **2** (down) showing different hydrogen bonding rings.

**Table S1** Summary of the C-H···O interactions in Fbf and compounds **1-4**

| Compounds | C–H···O             | d(C···O) | d(H···O) | $\angle(\text{DHO})$ | Remarks |
|-----------|---------------------|----------|----------|----------------------|---------|
| Fbf       | C(3)-H(3A)···O(6)   | 3.412(5) | 2.595(8) | 141.9                | P2-P2   |
|           | C(19)-H(19B)···O(3) | 3.513(6) | 2.636(6) | 150.4                |         |
| <b>1</b>  | C(13)-H(13B)···O(5) | 3.375(2) | 2.693    | 127.7                | P2-P3   |
|           | C(12)-H(12B)···O(6) | 3.432(3) | 2.875    | 117.5                |         |
| <b>2</b>  | C(11)-H(11B)···O(6) | 3.390(8) | 2.428    | 171.8                | P2-P3   |
|           | C(27)-H(27B)···O(3) | 3.446(1) | 2.480    | 173.2                |         |
| <b>3</b>  | -                   | -        | -        | -                    | -       |
| <b>4</b>  | C(9)-H(9A)···O(2)   | 3.323(1) | 2.688    | 123.5                | P1-P2   |

**Table S2** Summary of the C-H···π interactions in compounds **1-4** and pure Fbf<sup>a</sup>

| Compounds | C–H···Cg <sub>n</sub>          | d(C···Cg <sub>n</sub> ) | d(H···Cg <sub>n</sub> ) | $\angle(\text{C-H-Cg}_n)$ | Remarks |
|-----------|--------------------------------|-------------------------|-------------------------|---------------------------|---------|
| Fbf       | C(7)-H(7A)···Cg <sub>1</sub>   | 3.634(1)                | 2.972                   | 129.5                     | P3-P4   |
|           | C(16)-H(16A)···Cg <sub>1</sub> | 3.733(6)                | 2.978                   | 139.5                     |         |
|           | C(10)-H(10A)···Cg <sub>2</sub> | 3.745(1)                | 3.053                   | 132.6                     |         |
|           | C(13)-H(13A)···Cg <sub>2</sub> | 3.704(7)                | 3.017                   | 132.1                     |         |
|           | C(22)-H(22A)···Cg <sub>3</sub> | 3.529(5)                | 2.841                   | 131.8                     |         |
|           | C(32)-H(32A)···Cg <sub>3</sub> | 3.733(3)                | 2.971                   | 140.2                     |         |
|           | C(26)-H(26A)···Cg <sub>4</sub> | 3.730(4)                | 3.036                   | 132.8                     |         |
|           | C(28)-H(28A)···Cg <sub>4</sub> | 3.755(9)                | 3.038                   | 135.3                     |         |
| <b>1</b>  | C(38)-H(38A)···Cg <sub>1</sub> | 3.627(8)                | 2.890                   | 137.3                     | P3-P4   |
|           | C(41)-H(41A)···Cg <sub>1</sub> | 3.652(6)                | 2.947                   | 133.7                     |         |
| <b>2</b>  | C(14)-H(14A)···Cg <sub>1</sub> | 3.563(3)                | 3.131                   | 110.4                     | P3-P4   |
|           | C(18)-H(18A)···Cg <sub>1</sub> | 3.553(8)                | 3.064                   | 114.6                     |         |
|           | C(36)-H(36A)···Cg <sub>2</sub> | 3.525(2)                | 3.000                   | 117.3                     |         |
|           | C(40)-H(40A)···Cg <sub>2</sub> | 3.539(7)                | 3.072                   | 113.0                     |         |
| <b>3</b>  | C(8)-H(8A)···Cg <sub>1</sub>   | 3.704(2)                | 2.880                   | 143.5                     | P2-P3   |
| <b>4</b>  | -                              | -                       | -                       | -                         | -       |

<sup>a</sup> Cg<sub>n</sub> (n=1, 2, 3 or 4) is used to distinguish different benzene rings.

**Table S3** Different dihedral angles of Pn-P(n+1) (n=1, 2, 3) in Fbf and compounds

**1-4**

| Complexes | P1-P2   | P2-P3   | P3-P4   |
|-----------|---------|---------|---------|
| Fbf       | 79.3(3) | 8.8(3)  | 5.0(4)  |
|           | 80.1(3) | 6.8(3)  | 2.6(4)  |
| <b>1</b>  | 7.3(3)  | 9.5(3)  | 14.0(2) |
|           | 8.6(2)  | 17.9(2) | 8.6(1)  |
| <b>2</b>  | 81.6(2) | 8.3(2)  | 40.3(2) |
|           | 86.0(2) | 24.9(2) | 36.5(2) |
| <b>3</b>  | 7.8(3)  | 4.8(3)  | 12.2(3) |
| <b>4</b>  | 11.8(6) | 3.2(5)  | 7.8(5)  |