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Supplementary Material

Magnesium alkoxide complexes of (benzimidazolylmethyl)amino ligands: Synthesis and applications in ring-opening polymerization reactions of εcaprolactone and lactides

Ekemini D. Akpan,^a Bernard Omondi,^a Stephen O. Ojwach^{b*}

^aSchool of Chemistry and Physics, Westville Campus, University of KwaZulu-Natal, Private Bag X54001, Durban, 4000, South Africa.

^bSchool of Chemistry and Physics, Pietermaritzburg Campus, University of KwaZulu-Natal, Private Bag X01, Scottsville, 3209, South Africa.

*Email: Ojwach@ukzn.ac.za

Typical procedure for the synthesis of (benzimidazolylmethyl) amine ligands



Scheme S1 : Synthesis of (benzimidazolylmethyl)amine ligands L1–L3.

The starting materials, 2-(chloromethyl)benzimidazole (1 mole equivalent), and equimolar quantities of KI and the respective amines were dissolved in ethanol (20 mL) and the resulting mixture was heated to reflux for 6 h at 80 °C. An equimolar amount of KOH was then added to the mixture and further refluxed for 2 h, after which the reaction mixture was cooled to room temperature and poured into ice-cold water to give a precipitate. The resulting precipitate was filtered, washed with water and dried in *vacuo* to give the respective compounds.

N-((1*H*-benzo[s]imidazol-2-yl)methyl)-2,6-dimethylaniline (L1)

This compound was prepared using 2-(chloromethyl)benzimidazole (0.50 g, 3.00 mmol), 2,6-dimethylaniline (0.36 g, 0.37 mL, 3.00 mmol), KI (0.50 g, 3.00 mmol) and KOH (0.17 g, 3.00 mmol) and was obtained as a pale-yellow solid (0.61 g, 81%). ¹H NMR (CDCl₃, 400 MHz): δ (ppm) 2.31 (s, 6H, CH₃), 4.45 (s, 2H, CH₂), 6.95 (t, ³*J* = 7.5 Hz, 1H, Ar), 7.06 (d, ³*J* = 7.5 Hz, 2H, Ar), 7.29 (m, 2H, Ar), 7.61 (m, 2H, Ar). ¹³C NMR (CDCl₃, 100 MHz) δ (ppm) 153.3, 144.8, 130.2, 129.1, 123.3, 122.6, 46.7, 18.3. IR (Nujol): ν = 2927 (w), 1630 (m), 1533 (s), 1428 (s).

ESI-TOF MS: *m/z* calculated for C₁₆H₁₇N₃ [M + H⁺] 252.33; found 252.15. Anal. Calcd. For C₁₆H₁₇N₃: C, 76.46; H, 6.83; N, 16.72. Found: C, 76.40; H, 6.79; N, 16.69.

N-((1*H*-benzo[*d*]*imidazo*l-2-*y*l)*methyl*)-2,6-*diisopropylaniline* (*L*2)

A mixture of 2-(chloromethyl)benzimidazole (0.50 g, 3.00 mmol), 2,6-diisopropylaniline (0.53 g, 0.56 mL, 3.00 mmol), KI (0.50 g, 3.00 mmol) and KOH (0.17 g, 3.00 mmol) a Pale-yellow solid. (0.82 g, 88%). ¹H NMR (CDCl₃, 400 MHz): δ (ppm) 1.26 (d, ³*J* = 5.3 Hz, 12H, CH₃), 3.34 (m, 2H, CH), 4.39 (s, 2H, CH₂), 7.17 (s, 3H, Ar), 7.30 (m, 3H, Ar), 7.49 (t, ³*J* = 7.5 Hz, 1H, Ar), 7.78 (t, ³*J* = 7.5 Hz, 1H, Ar), 9.73 (1H, NH). ¹³C NMR (CDCl₃, 100 MHz) δ (ppm) 152.9, 142.9, 141.6, 125.0, 123.9, 49.9, 27.8, 24.2. IR (Nujol): ν = 3354 (w), 2960 (s), 2869 (m), 1630 (m), 1544 (w), 1457 (s), 1432 (s). ESI-TOF MS: *m/z* calculated for C₂₀H₂₅N₃ [M + H⁺] 308.43; found 308.58 Anal. Calcd. For C₂₀H₂₅N₃: C, 78.14; H, 8.20; N, 13.67. Found: C, 78.30; H, 8.29; N, 13.69.

N-((1*H*-benzo[d]imidazol-2-yl)methyl)-2,4,6-trimethylaniline (L3)

2-(chloromethyl)benzimidazole (0.50 g, 3.00 mmol), 2,4,6-trimethylaniline (0.41 g, 0.43 mL, 3.00 mmol), KI (0.50 g, 3.00 mmol) and KOH (0.17 g, 3.00 mmol) Pale-yellow solid. (0.72 g, 90%). ¹H NMR (CDCl₃, 400 MHz): δ (ppm) 2.27 (s, 6H, CH₃), 4.40 (s, 2H, CH₂), 6.88 (s, 2H, Ar), 7.29 (m, 3H, Ar), 7.44 (b, 1H, Ar), 7.76 (1H, Ar), 9.83 (s, 1H, NH). ¹³C NMR (CDCl₃, 100 MHz) δ (ppm) 153.4, 143.4, 132.8, 130.5, 129.7, 122.9, 122.3, 119.4, 110.7, 46.9, 20.6, 18.1. IR (Nujol): v = 2908 (m), 1630 (s), 1481 (s), 1420 (s). ESI-TOF MS: *m/z* calculated for C₁₇H₁₉N₃ [M + H⁺] 266.35; found 266.25. Anal. Calcd. For C₁₇H₁₉N₃: C, 76.95; H, 7.22; N, 15.84. Found: C, 76.70; H, 7.29; N, 15.69.



Fig. S1: ¹H NMR spectra of ligand **L1** (top) and complex **1** (bottom) showing the appearance of resonance of methylene hydrogen of the bridging benzyl alkoxides at 4.73 ppm indicative of complex formation.



Fig. S2: ¹³C NMR spectrum of ligand L1 (top) and complex 1 (bottom) showing the appearance of resonance peak at 64.88 ppm for the methylene linkage carbon of $-OCH_2C_6H_5$ derivative indicating formation of complex.



Fig. S3: Variable temperature of complex 1 from -40 °C to 40 °C, showing no changes to the CH₂ linker protons.



Fig. S4: Mass spectra showing complex 1 showing its fragmentation pattern to give a base peak at 252 amu corresponding to L1 unit.



Fig. S5: (a) First order kinetic plots of $\ln[CL]_0/[CL]_t vs.$ time for complexes 1–4 in the ROP of ε -CL in toluene at 110 °C, $[CL]_0/[I] = 200$. (b) First order kinetic plots of $\ln[LA]_0/[LA]_t vs.$ time for complexes 1–3 in the ROP of _{D,L}-LA and _L-LA to PLAs in toluene at 110 °C, $[LA]_0/[I] = 200$.



Fig. S6: (a) First order kinetic plots of $\ln[LA]_{o}/[LA]_{t}$ vs. time for complex 2 in the ROP of L-LA in toluene at 110 °C at different [LA]_o/[I] ratios in the presence of 2 equiv BnOH. (b) Plot of $\ln k_{app}$ vs. $\ln[2]$ for the determination of order of reactions in the ROP of L-LA in the presence of BnOH.



Fig. S7: Stability studies showing the catalytic activities of catalyst **2** in the ROP of ε -CL in the first and second cycle. The *k*_{obs} of 0.101 h- was observed in the second cycle compared to *k*_{obs} of 0.177 h- reported in the first cycle, representing about 40% drop in activity.



Fig. S8: ¹H NMR spectrum of PCL in CDCl₃ obtained using complex **2**, revealing the presence of phenyl ring protons (7.37 ppm) and CH₂ (5.12 ppm) signal of $-OCH_2C_6H_5$ group. Reaction condition:[CL]₀:[I]₀ = 200:1 in toluene at 110 °C, 12 h.



Fig. S9: ¹H NMR spectrum of poly(L-LA) in CDCl₃ obtained using complex **2**, showing signals that the polymer chain is capped with one benzyl ester and one hydroxyl end. Reaction condition: $[LA]_0:[\mathbf{2}]_0 = 200:1$ in toluene at 110 °C, 20 h.



Fig. S10: ESI-MS spectrum of cyclic oligomer obtained by using catalyst 2. Reaction condition: $[CL]_0: [2]_0 = 200:1$ in toluene at 110 °C, 12 h.



Fig. S11: (a) ¹³C NMR spectra carbonyl region and (b) ¹³C NMR methine region of poly(L-LA) supporting isotactic microstructure of PLA prepared from L-LA using complex **2**. Reaction condition:[LA]₀:[**2**]₀ = 200:1 in toluene at 110 °C



Fig. S12: ¹H homonuclear decoupled NMR of the methine region of $poly(_{D,L}-LA)$ obtained from complex **2** showing the formation of atactic microstructure.