

## Supplementary Material

### **Incorporation of Vanadium and Molybdenum into Yttrium-Arsenotungstates Supported by Amino Acid Ligands**

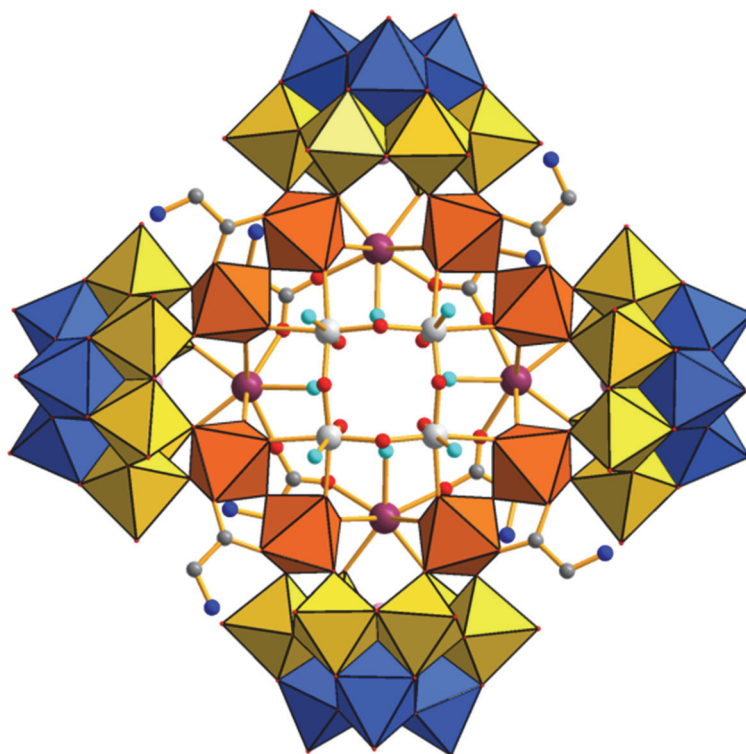
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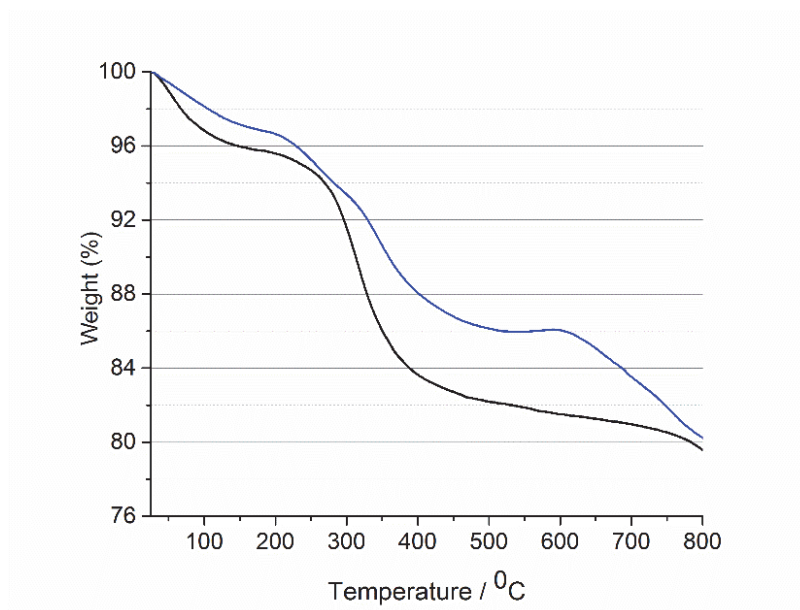


**Fig. S1.** Structural representation of idealised hybrid polyoxometalate  $[\text{As}_4(\text{M}_4)\text{W}_{44}\text{Y}_4\text{O}_{160}(\text{AA})_8(\text{H}_2\text{O})_{12}]^{n-}$  indicating the different metal sites:  $\{\text{M}_4\}$  "core" (grey),  $\{\text{Y}_4\text{M}_8\text{L}_8\}$  "ring" (orange),  $\{\text{AsM}_9\}$  lacunary Keggin "triad" (blue) and  $\{\text{AsM}_9\}$  lacunary Keggin "belt" (yellow).

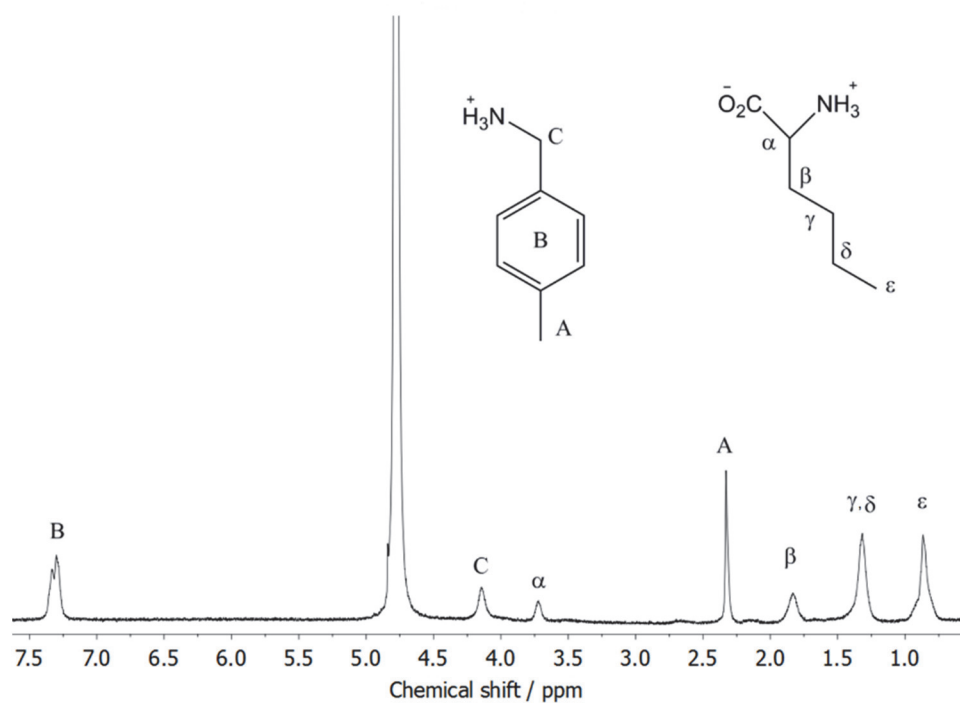
**Table S1.** V, Mo and W site occupancies (%) for disordered V/W and V/Mo/W sites and number of V and Mo per POM from crystallography and elemental analysis for 1·72H<sub>2</sub>O, 2·87H<sub>2</sub>O and 3·96H<sub>2</sub>O.

	1·72H <sub>2</sub> O	2·87H <sub>2</sub> O	3·96H <sub>2</sub> O
<b>V occupancies (%)</b>			
{M <sub>4</sub> } core	43(2), 54(2), 48(2), 55(2)	50 <sup>c</sup>	50 <sup>c</sup>
{Y <sub>4</sub> M <sub>8</sub> L <sub>8</sub> } ring <sup>b</sup>	0	0	0
{AsM <sub>9</sub> } triad <sup>b</sup>	0	0	0
{AsM <sub>9</sub> } belt <sup>b</sup>	0	0	0
<b>W occupancies (%)</b>			
{M <sub>4</sub> } core	57(2), 46(2), 52(2), 45(2)	50 <sup>c</sup>	50 <sup>c</sup>
{Y <sub>4</sub> M <sub>8</sub> L <sub>8</sub> } ring <sup>b</sup>	100	100	96(2), 84(1) <sup>c</sup>
{AsM <sub>9</sub> } triad <sup>b</sup>	100	100	77(2), 93(2) <sup>c</sup>
{AsM <sub>9</sub> } belt <sup>b</sup>	100	100	100
<b>Mo occupancies (%)</b>			
{M <sub>4</sub> } core	-	-	0
{Y <sub>4</sub> M <sub>8</sub> L <sub>8</sub> } ring <sup>b</sup>	-	-	4(2), 16(1) <sup>c</sup>
{AsM <sub>9</sub> } triad <sup>b</sup>	-	-	23(2), 7(2) <sup>c</sup>
{AsM <sub>9</sub> } belt <sup>b</sup>	-	-	0
<b>Total V per POM</b>			
crystallography (sum of V occupancies)	2	2	2
elemental analysis for V (bulk)	2	2	2
<b>Total Mo per POM</b>			
crystallography (sum of Mo occupancies)	-	-	2
elemental analysis for Mo (bulk)	-	-	2

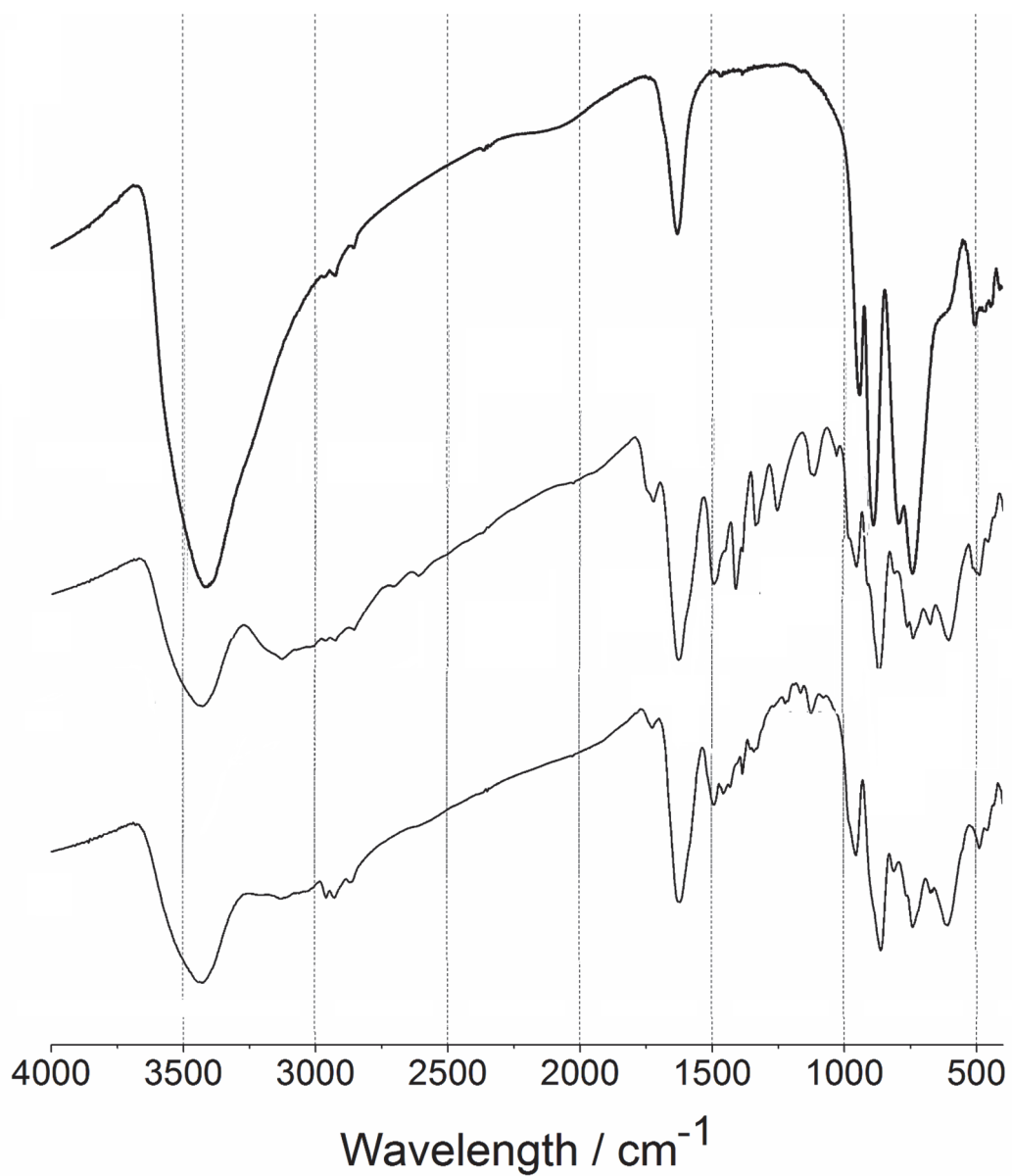
<sup>a</sup> restrained, <sup>b</sup> see Fig. S1; <sup>c</sup> per unit cell.



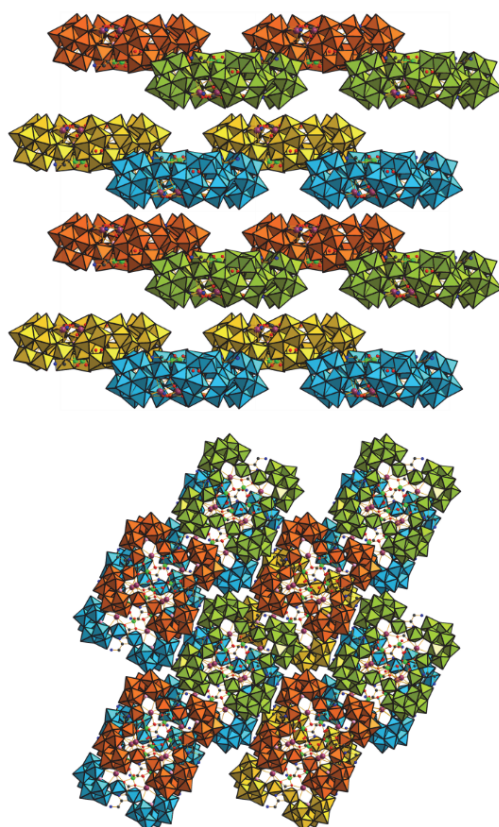
**Fig. S2.** Thermogravimetric analysis data for Gly-containing **1**·25H<sub>2</sub>O (blue) and Nle-containing **3**·20H<sub>2</sub>O (black). The calculated masses corresponding to loss of all organic cations and ligands are 11 and 15 % for **1** and **3**, respectively.



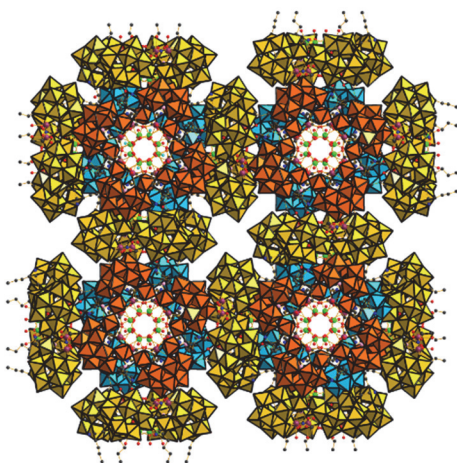
**Fig. S3.** Solution <sup>1</sup>H NMR spectrum of **3** in D<sub>2</sub>O with assignments.



**Fig. S4.** Infrared spectra (KBr) of (from top to bottom):  $K_{14}[As_2W_{19}O_{67}(H_2O)]$  precursor,  $1 \cdot 25H_2O$  and  $3 \cdot 20H_2O$ .



**Fig. S5.** Crystal packing diagram for  $1 \cdot 72\text{H}_2\text{O}$  looking down the: (top) crystallographic  $b$  axis, and (bottom) crystallographic  $a$  axis. Color code as per Fig. 1 except  $\text{WO}_6$  octahedra, yellow and blue is one double layer and green and orange in the other.



**Fig. S6.** Crystal packing diagram for  $2 \cdot 87\text{H}_2\text{O}$  and  $3 \cdot 96\text{H}_2\text{O}$  illustrating how the disk-shaped POMs are arranged on the six faces of the "cubic" cavities. Color code as per Fig. 1 except  $\text{WO}_6$  octahedra, orange for the POM on the "front" of the cubic cavities.

**Table S2.** XPS Binding Energies (eV) for **1**·25H<sub>2</sub>O and **3**·20H<sub>2</sub>O obtained from spectral deconvolution.

Compound	W(VI)		V(IV)
	4f <sub>7/2</sub>	4f <sub>5/2</sub>	2p <sub>3/2</sub>
	2		
<b>1</b>	35.	37.9	516.4
	9		
<b>3</b>	35.	37.7	516.3
	7		

**Table S3.** Spin Hamiltonian parameters obtained from simulating the 130 K X-band powder EPR spectra of **1**·25H<sub>2</sub>O and **3**·20H<sub>2</sub>O, assuming for three and two V(IV) environments, respectively

compound	$g_x$	$g_y$	$g_z$	$A_x^a$	$A_y^a$	$A_z^a$	environment	ratio
<b>1</b>	1.972	1.972	1.884	162	162	495	I	4
	1.972	1.972	1.942	162	162	520	II	2
	1.972	1.972	1.915	162	162	530	III	1
<b>3</b>	1.972	1.972	1.888	162	162	490	I	1
	1.972	1.972	1.900	162	162	510	II	1

<sup>a</sup> Hyperfine parameters are  $\times 10^{-4} \text{ cm}^{-1}$