CORRIGENDA

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"*Calculation of the Enthalpy of Formation of the Hydrogen Bond in Solutions of Alcohols in Non-polar Solvents from Thermodynamic Data.*" By R. H. Stokes. pp. 1343–8.

(a) In equation (3) the leading term on the right-hand side should be $(qh - q_0 h_0)x_1$.

(b) In lines 1 and 2 of p. 1345, read n_1 for n_2 and $n_1(1 - q)$ for $n_2(1 - q)$.

(c) p. 1345, line 10: For "solute" read "solvent".

(d) In equation (11) there is a sign error: the equation should be

$$L - L_0 = -AV_1\phi_1 + qh$$

This carries through to subsequent equations (12 and 18 in particular). Since A is a parameter determined from the data, this sign error does not affect the *fit* of equation (12) to the data, provided that the signs of A in Table 1 are reversed. The values for h, the enthalpy of formation of the hydrogen bond, are unaffected. However, the conclusion of the final paragraph, that the Flory–Huggins statistics give more physically satisfactory values of A, is no longer valid. The author is indebted to Mr R. P. Tomlins for detecting these errors.

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Page 815, line 11: *For* benzylic *read* benzilic.