

# Steroid Alkaloids of *Marsdenia rostrata*. III\* Rostratine and Dihydrorostratine

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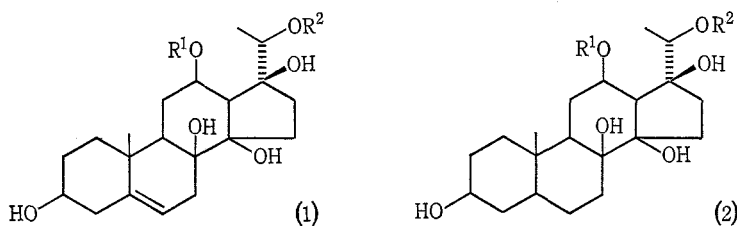
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## Abstract

The ester alkaloids rostratine and dihydrorostratine were isolated and characterized. On the basis of the mass spectral fragmentation pattern their acetate ester groups were assigned to the C 12 and the nicotinate ester groups to the C 20 hydroxyl groups of the steroid skeleton.

We have previously assigned part-structures (1 and 2;  $R^1$  or  $R^2$  = acetyl,  $R^2$  or  $R^1$  = nicotinoyl) to rostratine and dihydrorostratine, the constituents of the unseparated alkaloid mixture obtained from the Toonumbar collection of *M. rostrata*.<sup>1</sup> This paper describes the isolation, characterization and structure elucidation of the two alkaloids.



Rostratine (1;  $R^1$  = acetyl,  $R^2$  = nicotinoyl),  $C_{29}H_{39}NO_8$ , was separated from dihydrorostratine (2;  $R^1$  = acetyl;  $R^2$  = nicotinoyl),  $C_{29}H_{41}NO_8$ , by thin-layer chromatography on silver nitrate impregnated silica gel. The spectra of the individual alkaloids confirmed the deductions made from the u.v., i.r., mass and n.m.r.<sup>2,3</sup> spectra obtained from the original alkaloid mixture.

Detailed examination of the mass spectral fragmentation pattern of rostratine, where high resolution measurements were made on an MS-9 spectrometer with computerized data acquisition and analysis system, allows us to assign the nicotinoyl group to the C 20 hydroxyl and, consequently, the acetyl group to the C 12 hydroxyl group of the steroid molecule. The base peak at  $m/e$  124 ( $C_6H_6NO_2$ , protonated nicotinic acid) in the mass spectrum of rostratine is probably formed by hydrogen

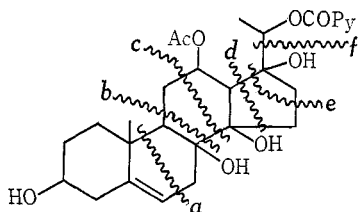
\* Part II, *Aust. J. Chem.*, 1973, 26, 1835.

<sup>1</sup> Summons, R. E., Ellis, J., and Gellert, E., *Phytochemistry*, 1972, 11, 3335.

<sup>2</sup> Sawlewicz, L., Weiss, Ek., and Reichstein, T., *Helv. Chim. Acta*, 1967, 50, 504, 530.

<sup>3</sup> Schaub, F., Kaufmann, H., Stöcklin, W., and Reichstein, T., *Helv. Chim. Acta*, 1968, 51, 738.

transfer often observed with esters of aliphatic alcohols. Other prominent peaks at  $m/e$  123 ( $C_6H_5NO_2$ , nicotinic acid), 106 ( $C_6H_4NO$ , nicotinoyl), 105 ( $C_6H_3NO$ , nicotinoyl-H), and 78 ( $C_5H_4N$ , pyridyl), which are also present in the spectrum of nicotinic acid itself, could arise either directly from the molecular ion,  $m/e$  529, or from the  $m/e$  124 fragment. The same peaks are also present in the low-resolution spectrum of dihydrorostratine. Peaks such as  $m/e$  511 ( $M^+ - H_2O$ ), 493 ( $M^+ - 2 \times H_2O$ ), 469 ( $M^+ - AcOH$ ), 433 ( $M^+ - AcOH - 2 \times H_2O$ ), 346 ( $M^+ - AcOH - \text{nicotinic acid}$ ), 328 ( $M^+ - AcOH - \text{nicotinic acid} - H_2O$ ) and 292 ( $M^+ - AcOH - \text{nicotinic acid} - 3 \times H_2O$ ) are due to loss of combinations of water, acetic and nicotinic acid molecules, while the additional loss of a methyl group gives rise to peaks at  $m/e$  454 ( $469 - Me^\cdot$ ), 418 ( $433 - Me^\cdot$ ), 313 ( $328 - Me^\cdot$ ). Similar peaks appear, at two mass units higher, in the spectrum of dihydrorostratine. Retro Diels-Alder fission of ring B<sup>4,5</sup> (which is of course considerably less significant with dihydrorostratine) as shown by dissection *a* generates peaks at  $m/e$  138 ( $C_9H_{14}O$ ) and 120 ( $C_9H_{12}$ ;  $138 - H_2O$ ) together with  $m/e$  373 ( $C_{20}H_{23}NO_6$ ) and 313 ( $373 - AcOH$ ). This shows clearly that the hydroxyl group at C 3 is not esterified and confirms the partial formulae (1 and 2;  $R^1$  or  $R^2 = AcO$ ,  $R^2$  or  $R^1 = \text{nicotinoyl}$ ) proposed earlier.



The characteristic fission of ring c of 8,14-steroid diols<sup>5-7</sup> provides peaks significant for the assignment of the ester groups. Peaks at  $m/e$  161 (A/B ring fragment -  $H_2O - H$ ) and  $m/e$  289 (c/D ring fragment -  $AcOH$ ) are formed according to dissection *b*, while cleavage at dissection *c* yields  $m/e$  175 (A/B ring fragment -  $H_2O - H$ ) and  $m/e$  276 (c/D ring fragment -  $AcO$ ). The peaks at  $m/e$  166 ( $289 - 123$ ) and  $m/e$  153 ( $276 - 123$ ) are due to the additional loss of a molecule of nicotinic acid. However, the most important peaks for the assignment of the ester peaks originate from the fission of ring d, i.e.  $m/e$  206 (dissection *d*) and  $m/e$  193 (dissection *e*), and from the side chain,  $m/e$  150 (dissection *f*). Focused metastable measurements from the first field-free region show that  $m/e$  206 is a parent of both  $m/e$  193 and 150 peaks and is a daughter of  $m/e$  469, 289 and 276 peaks confirming the evidence for assigning structure (1;  $R^1 = Ac$ ,  $R^2 = \text{nicotinoyl}$ ) to rostratine and, consequently, structure (2;  $R^1 = Ac$ ,  $R^2 = \text{nicotinoyl}$ ) to dihydrorostratine.

## Experimental

The alkaloid mixture from the Toonumbar collection of *M. rostrata*<sup>1</sup> was chromatographed on silver nitrate impregnated silica gel as described for rostratamine.<sup>8</sup> The  $R_F$  values quoted refer to the same system. Extraction of the appropriate zones on the chromatogram gave:

<sup>4</sup> Kapur, B. M., Allgeier, H., and Reichstein, T., *Helv. Chim. Acta*, 1967, **50**, 2147.

<sup>5</sup> Meister, L., Stöcklin, W., and Reichstein, T., *Helv. Chim. Acta*, 1970, **53**, 2044.

<sup>6</sup> Saner, A., Stöckel, K., and Reichstein, T., *Helv. Chim. Acta*, 1972, **55**, 1221.

<sup>7</sup> Duff, A. G., Gellert, E., and Rudzats, R., *Phytochemistry*, 1973, **12**, 2943.

<sup>8</sup> Gellert, E., and Summons, R. E., *Aust. J. Chem.*, 1973, **26**, 1835.

(i) Rostratine,  $C_{29}H_{39}NO_8$ , m.p. 259–260° from MeOH–Et<sub>2</sub>O,  $[\alpha]_D^{22} + 49 \pm 1^\circ$  (c, 1.0 in MeOH),  $R_F$  0.45. Relevant and major peaks in the mass spectrum (calc. and found values agree within  $\pm 6$  p.p.m.):  $m/e$  529 (4.6%)  $M^+$ ; 511 (0.5)  $C_{29}H_{37}NO_7$ ; 493 (0.8)  $C_{29}H_{35}NO_6$ ; 469 (7.5)  $C_{27}H_{35}NO_6$ ; 454 (1.9)  $C_{26}H_{32}NO_6$ ; 433 (1.2)  $C_{27}H_{31}NO_4$ ; 418 (5.6)  $C_{26}H_{28}NO_4$ ; 400 (1.9)  $C_{26}H_{26}NO_3$ ; 373 (27)  $C_{20}H_{23}NO_6$ ; 346 (11)  $C_{21}H_{30}O_4$ ; 328 (5.9)  $C_{21}H_{28}O_3$ ; 313 (4.4)  $C_{18}H_{19}NO_4$ ; 313 (3.2)  $C_{20}H_{25}O_3$ ; 292 (1.3)  $C_{21}H_{24}O$ ; 289 (5.1)  $C_{16}H_{19}NO_4$ ; 276 (4.9)  $C_{15}H_{18}NO_4$ ; 206 (1.1)  $C_{11}H_{12}NO_3$ ; 193 (8.9)  $C_{10}H_{11}NO_3$ ; 175 (2.5)  $C_{12}H_{15}O$ ; 166 (5.8)  $C_{10}H_{14}O_2$ ; 161 (10)  $C_{11}H_{13}O$ ; 153 (3.4)  $C_9H_{13}O_2$ ; 150 (2.3)  $C_8H_8NO_2$ ; 138 (8.2)  $C_9H_{14}O$ ; 124 (100)  $C_6H_6NO_2$ ; 123 (38)  $C_6H_5NO_2$ ; 120 (28)  $C_9H_{12}$ ; 107 (15)  $C_6H_5NO$ ; 106 (55)  $C_6H_4NO$ ; 105 (19)  $C_6H_3NO$ ; 105 (17)  $C_8H_9$ ; 91 (12)  $C_7H_7$ ; 85 (35)  $C_4H_5O_2$ ; 78 (31)  $C_5H_4N$ ; 43 (54)  $C_2H_3O$ .

(ii) Dihydrorostratine,  $C_{29}H_{41}NO_8$ , m.p. 257–258° from MeOH–Et<sub>2</sub>O,  $[\alpha]_D^{22} + 55 \pm 2^\circ$  (c, 0.15 in MeOH),  $R_F$  0.60. Relevant and major peaks in the mass spectrum:  $m/e$  531 (0.1%)  $M^+$ ; 513 (0.5); 495 (1.2); 471 (7.4); 456 (3.6); 438 (1.8); 435 (1.4); 420 (0.9); 348 (11); 330 (7.6); 289 (3.2); 276 (4.9); 206 (6.5); 167 (8.4); 150 (1.1); 124 (95); 123 (100); 106 (50); 105 (45); 78 (30); 43 (50).

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### Corrigendum

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p. 24, Fig. 3, upper trace: for 11°C read –11°C.