

10.1071/EN17208_AC

©CSIRO 2018

Environmental Chemistry 2018, 15(1 & 2), 100-112

Supplementary Material

Effective adsorptive removal of 2, 4, 6-trinitrotoluene and hexahydro-1, 3, 5-trinitro-1, 3, 5-triazine by pseudo graphitic carbon: kinetics, equilibrium and thermodynamics

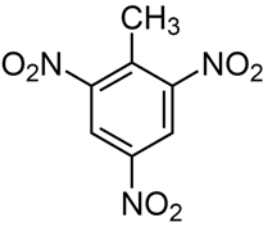
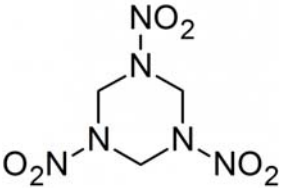
Jong-Soo Choi,^A Janardhan Reddy Koduru,^{A,C} Lakshmi Prasanna Lingamdinne,^A Jae-Kyu Yang^B and Yoon-Young Chang^{A,C}

^ADepartment of Environmental Engineering, Kwangwoon University, Seoul 01897, Republic of Korea.

^BIngenium College of Liberal Arts, Kwangwoon University, Seoul 01897, Republic of Korea.

^CCorresponding authors. Email: yychang@kw.ac.kr; redyjchem@gmail.com

Table S1. Physical properties and molecular structures of TNT and RDX

Parameter	TNT	RDX
Molecular weight (g mol ⁻¹)	227.13	222.26
Density (g cm ⁻³)	1.654	1.82
Solubility in water (mg L ⁻¹ in 20 °C)	130	39
Octanol-water partition coefficient (logK _{ow})	1.6	0.87
Molecular structure		

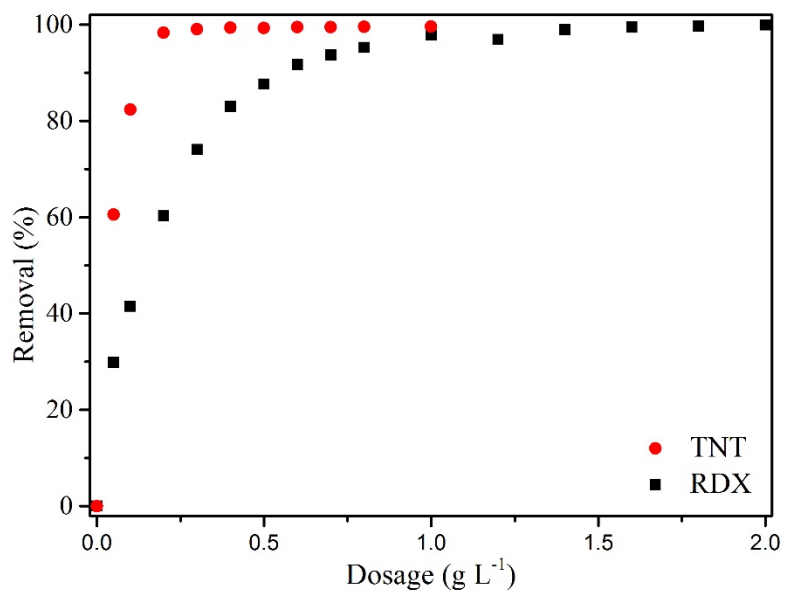


Fig. S1. The effect of dosage on adsorptive removal of TNT and RDX (10 mg L⁻¹) onto PGC at pH 6.0, 24 h equilibrium time at room temperature (293 K).

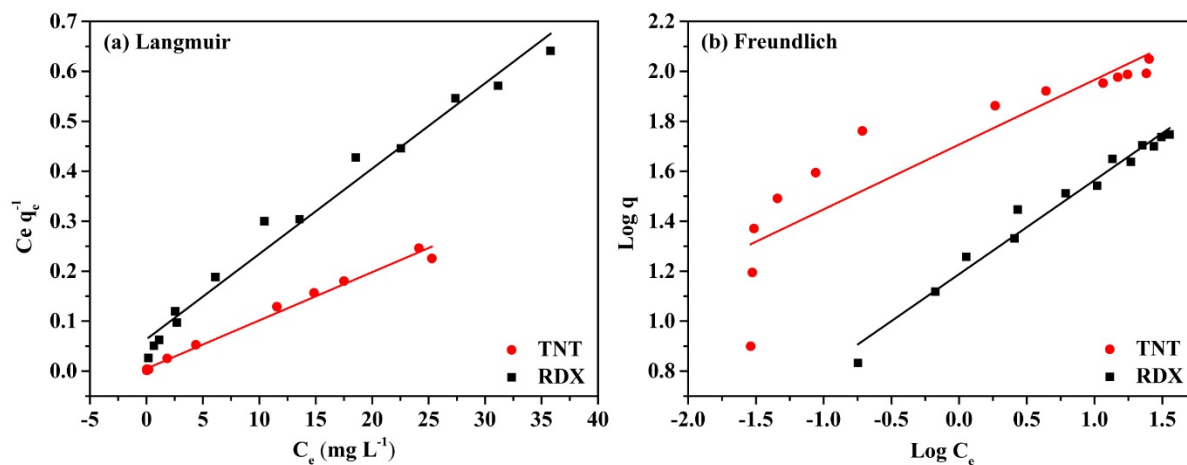


Fig. S2. Adsorption isotherm Langmuir (a) and Freundlich (b) plots in linear regression analysis for TNT and RDX (initial concentration: 10 mg L^{-1}) onto PGC (0.3 g L^{-1}) at pH 6.0, 24 h equilibrium time and room temperature (293 K).

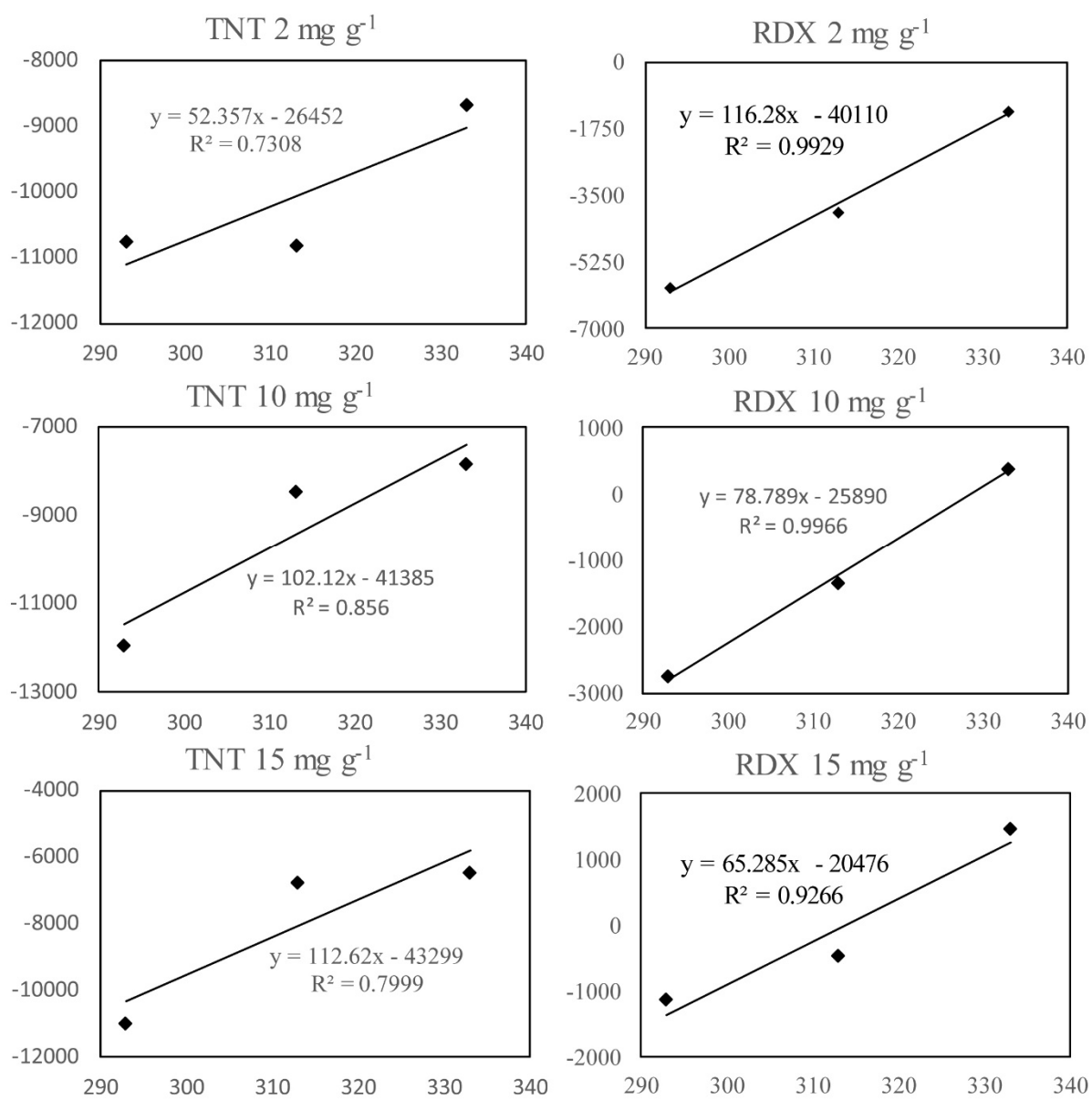


Fig. S3. ΔG vs T plots for TNT and RDX (initial concentration: 2, 10 and 15 mg L⁻¹) removal by using PGC (0.3 g L⁻¹) at pH 6.0, 24 h equilibrium time and room temperature (293 K).