

Supplementary Material

Photochemical Properties of Rare Earth Diclofenac Complexes in the Solid State

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4 Figures (Fig. S1-S4)

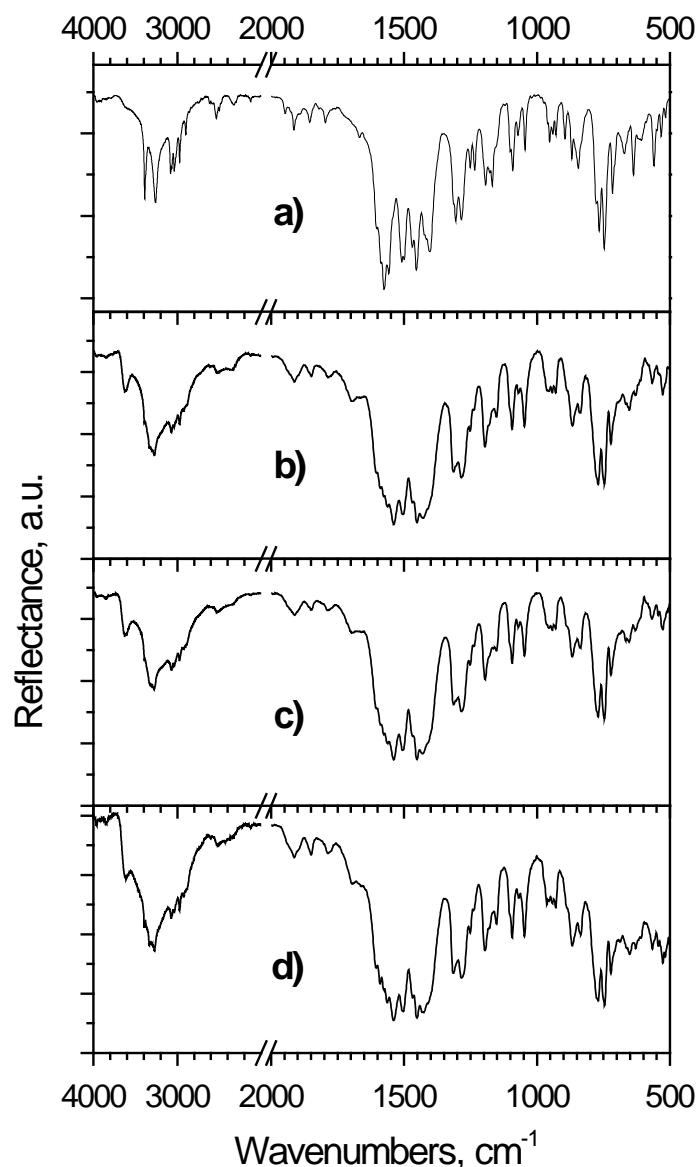


Fig. S1. FTIR spectra of a) Na(diclo), b) Eu(diclo)₃, b) Tb(diclo)₃, b) Gd(diclo)₃.

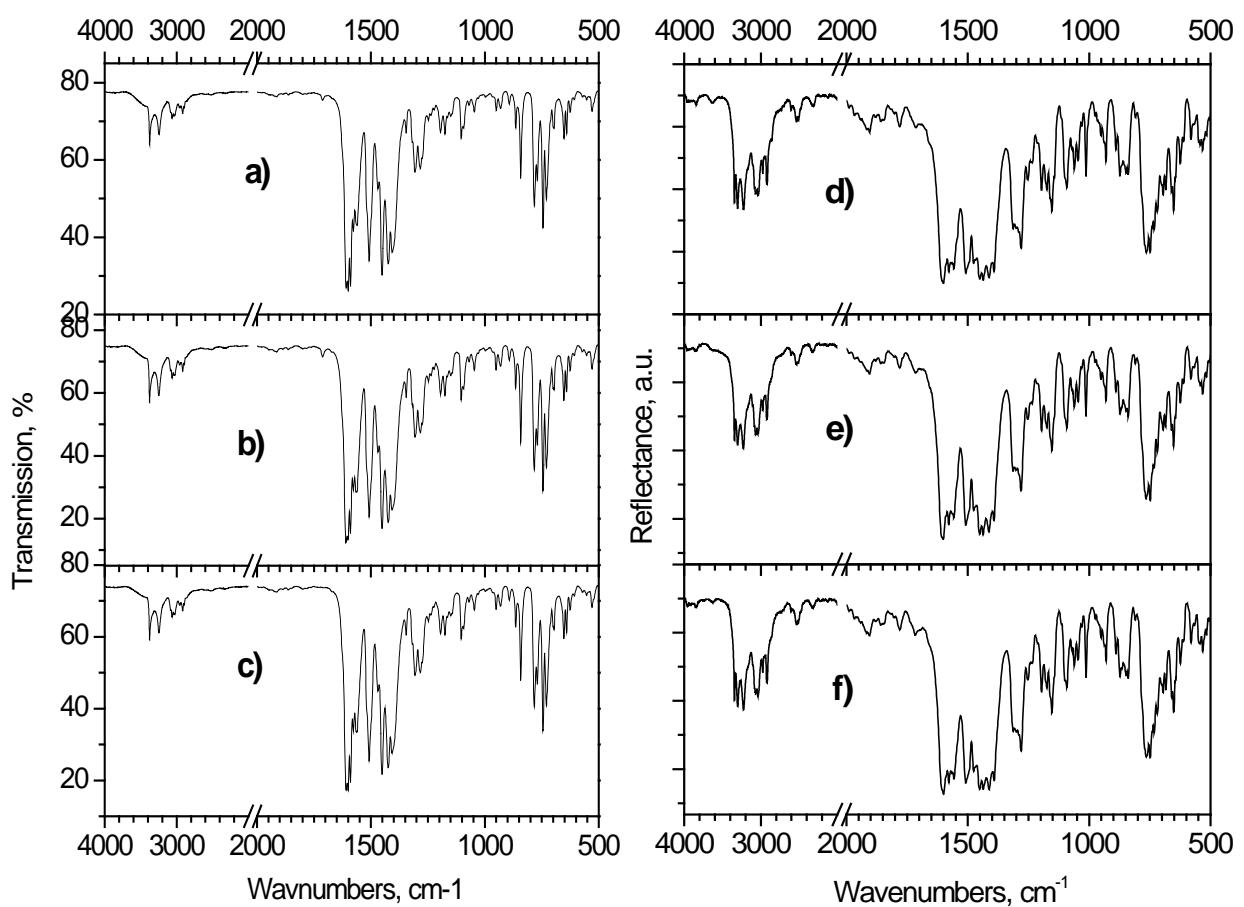


Fig. S2. FTIR spectra of a) Eu(diclo)₃phen, b) Tb(diclo)₃phen, c) Gd(diclo)₃phen, d) Eu(diclo)₃bipy, e) Tb(diclo)₃bipy, f) Gd(diclo)₃bipy.

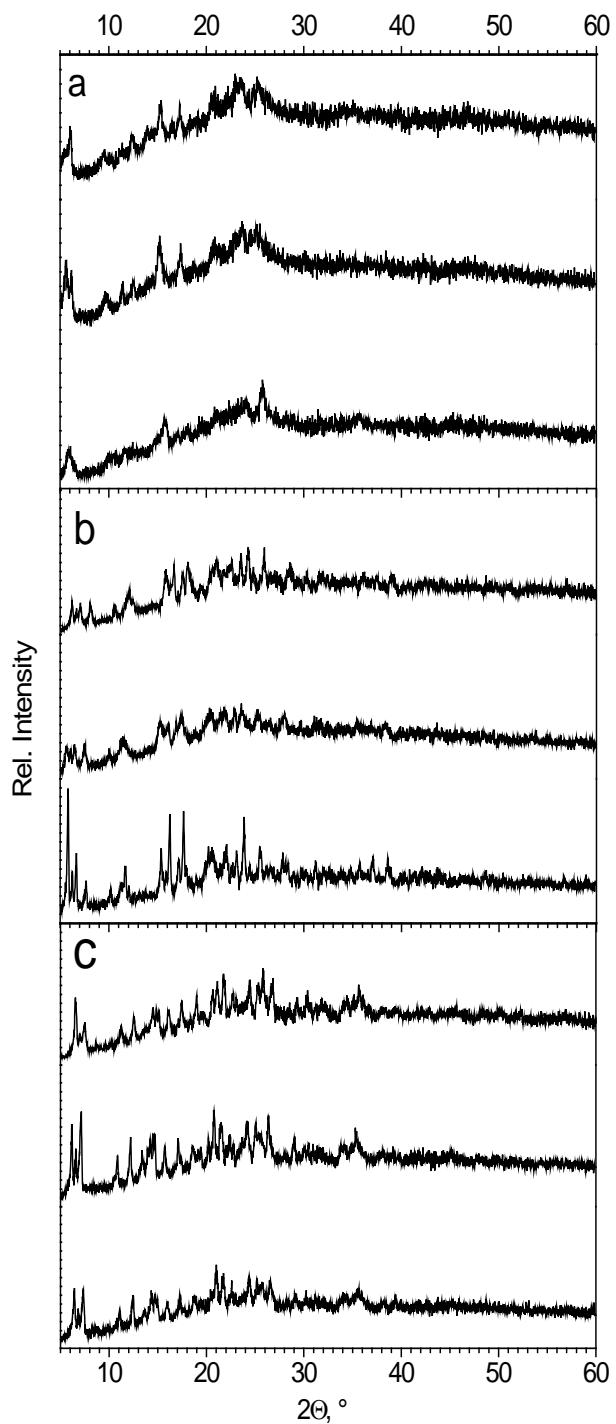


Fig. S3. X-ray powder diffraction of **a**) Eu(diclo)₃(top), Tb(diclo)₃ (center), Gd(diclo)₃ (bottom); **b**) Eu(diclo)₃phen, Tb(diclo)₃phen, Gd(diclo)₃phen; **c**) Eu(diclo)₃bipy, Tb(diclo)₃bipy, Gd(diclo)₃bipy.

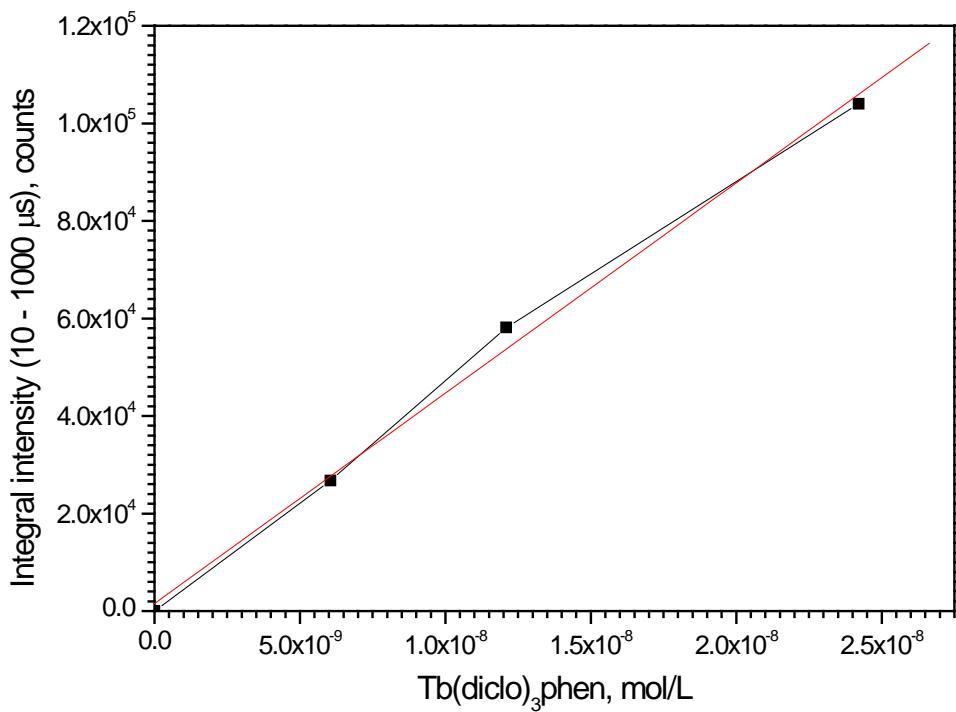


Fig. S4. Linear range for time integrated luminescence determination of $\text{Tb}(\text{diclo})_3\text{phen}$, measurements in ethanol. Beyond ca. 2.5×10^{-8} mol/L, agglomerates and ultimately nanoparticles form, which lead to a lower emission increase in this experimental setup (Perkin-Elmer Victor IV), standard plate reader and vials).