

## Supplementary Material

**CCC-NHC pincer Zr diamido complexes: Synthesis, characterization, and catalytic activity in hydroamination/cyclization of unactivated amino-alkenes, -alkynes, and allenes**

*Henry U. Valle,<sup>a</sup> Gopalkrishna Akurathi,<sup>a</sup> Joon Cho<sup>b</sup>, Wesley D. Clark<sup>a</sup>, Amarraj Chakraborty,<sup>a</sup>*

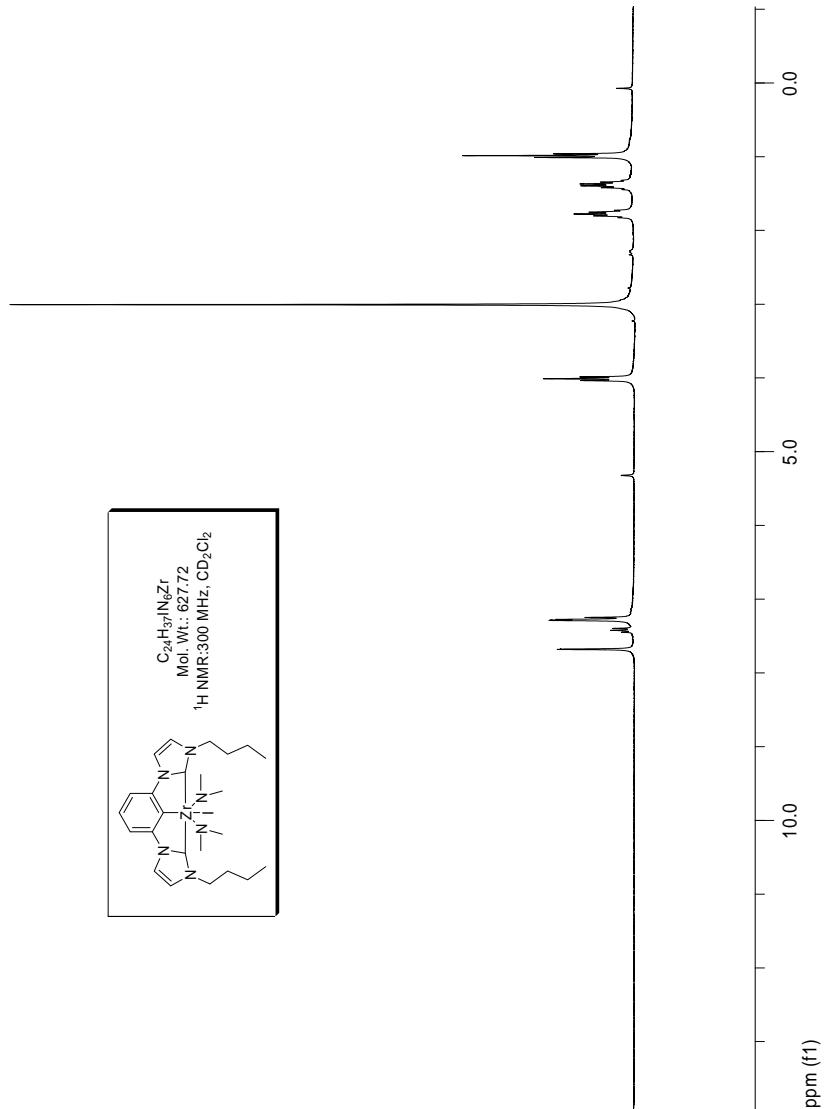
*T. Keith Hollis\*<sup>a,b</sup>*

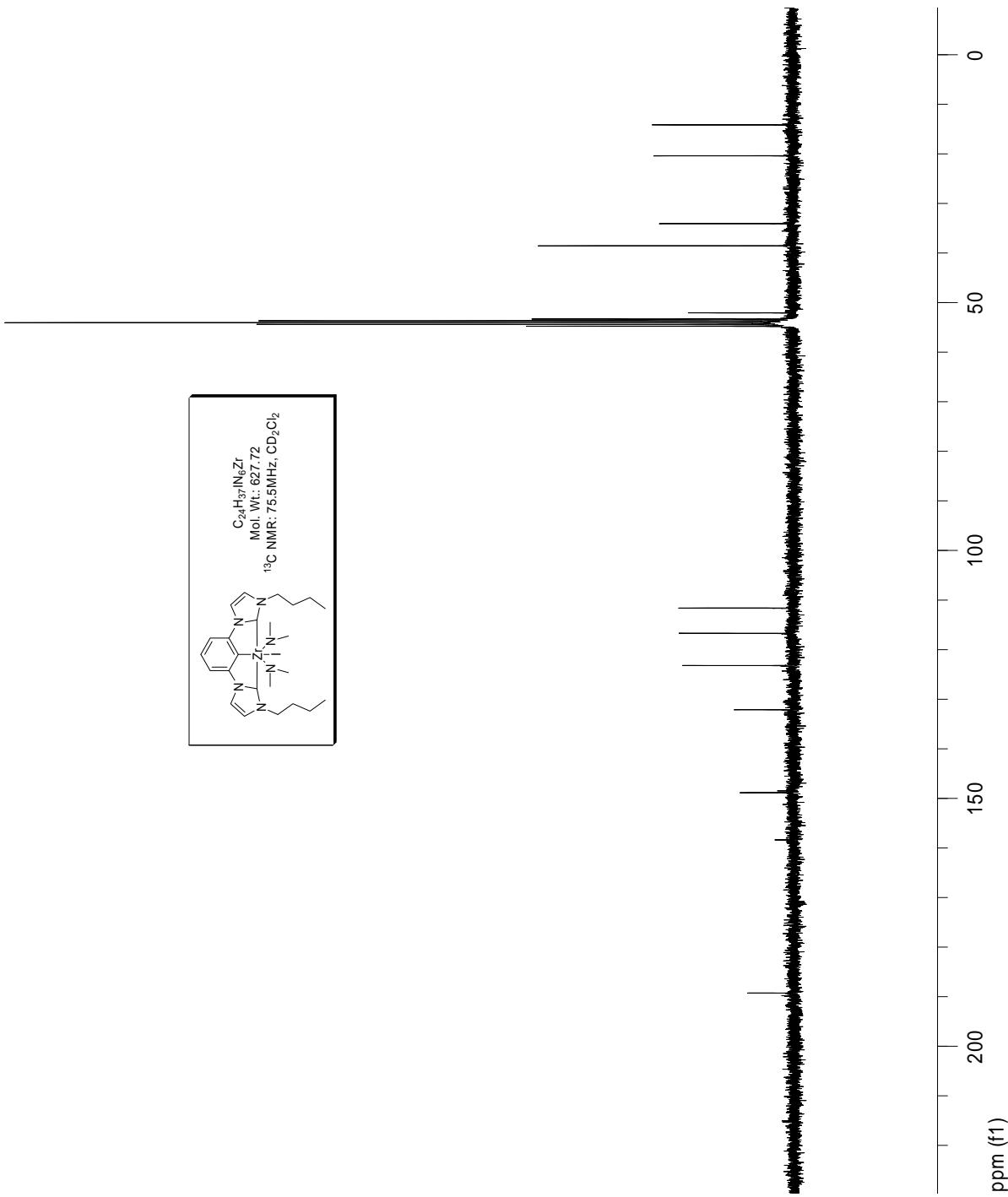
<sup>a</sup> Department of Chemistry, Mississippi State University, Mississippi State, MS 39762, United States

<sup>b</sup> Department of Chemistry and Biochemistry, The University of Mississippi, University, MS, 38677, United States

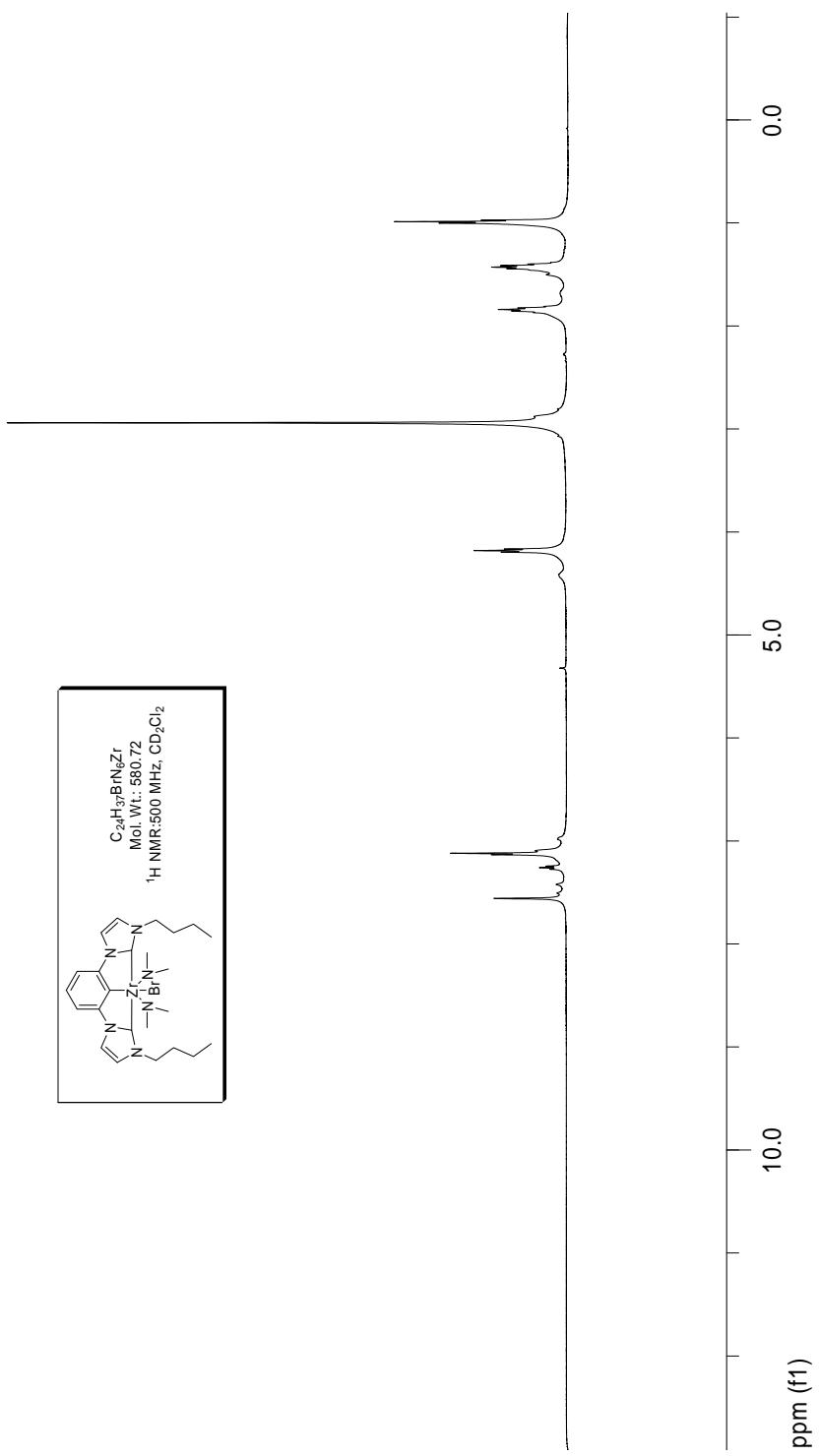
\*Email: khollis@chemistry.msstate.edu

NMR spectroscopic data

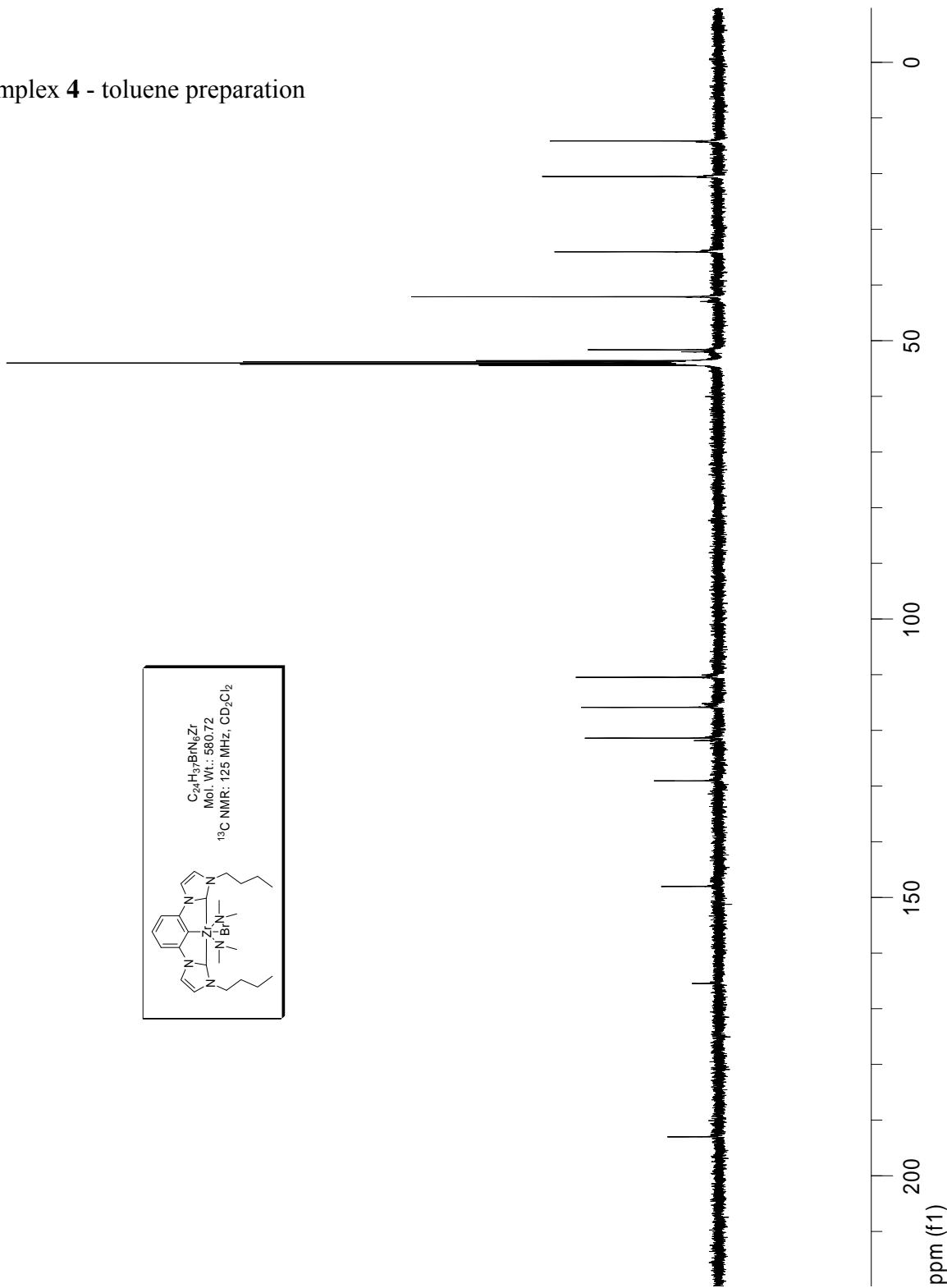




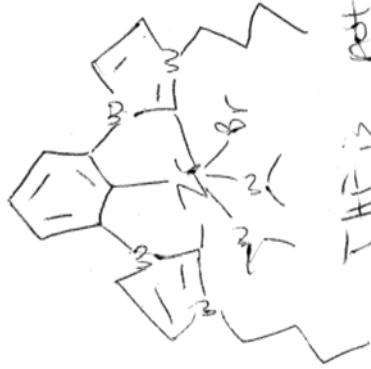
Complex 4 - toluene preparation



Complex 4 - toluene preparation



WDC-2-168-3-proton  
After drying overnight  
Exptno 2



## *<sup>1</sup>H* Preparation



### Current Data Parameters

NAME WDC-2-168-3-proton

EXPNO 2

PROCNO 1

### F2 - Acquisition Parameters

Date 20140118

Time 23:42

INSTRUM spect

PROBHD 5 mm TBI 1H/13

PULPROG zg30

TD 65536

SOLVENT Tol

NS 16

DS 0

SWH 6009.615 Hz

P1DRES 0.091699 Hz

AQ 5.4525952 sec

RG 1820

DW 83.200 usec

DE 6.50 usec

TB 298.2 K

D1 1.0000000 sec

TD0 1

### ===== CHANNEL f1 =====

SP01 300.1118533 MHz

NUC1 1H

P1 8.00 usec

P1W1 3.54809999 W

### F2 - Processing parameters

SI 65536

SP 300.1100105 MHz

WDW EM

SSB 0

LB 0.30 Hz

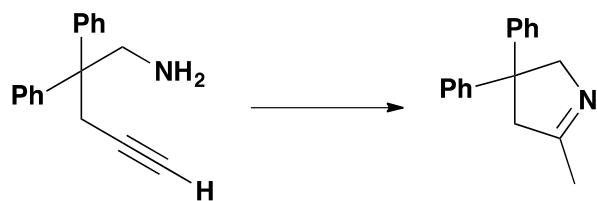
GB 0

PC 1.00



**Table 2** entry 1

**5-Methyl-3,3-diphenyl-3,4-dihydro-2*H*-pyrrole:**



In a screw cap NMR tube 2,2-diphenyl-4-pentyn-1-amine (0.025 g, 0.106 mmol), 2-(1,3-Bis(N-butyl-imidazol-2-ylidene) phenylene)bis(dimethylamido)(iodo)zirconium (IV) (0.004 g, 0.0054 mmol) and 0.4 ml of tol-d<sub>8</sub> was added in glove box and heated the reaction mixture at 160 °C. The progress of the reaction was monitored by using <sup>1</sup>H NMR spectroscopy. <sup>1</sup>H NMR (C<sub>6</sub>D<sub>6</sub>, 300MHz): δ 7.08-6.95 (m, 10H), 4.38-4.36 (q, J= 1.5 Hz, 2H), 2.68 (s, 2H), 1.75-1.74 (t, J = 1.8 Hz, 3H).

The proton NMR of this compound showed coupling not reported previously in the literature. The spectrum is included here for reference.

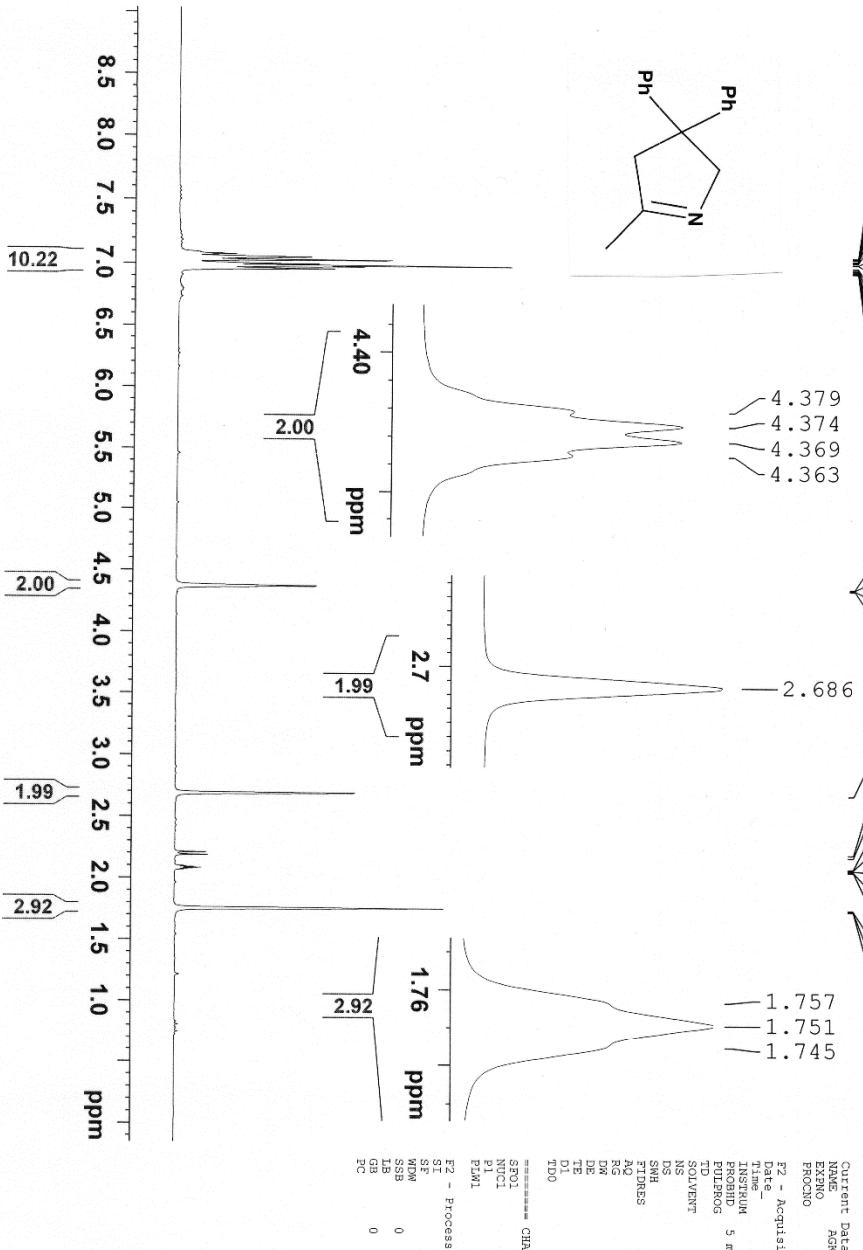
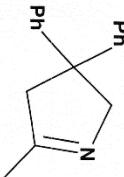
Standard 1H

7.081  
7.077  
7.071  
7.057  
7.052  
7.050  
7.040  
7.029  
7.027  
7.000  
6.997  
6.983  
6.975  
6.965  
6.959  
6.954

4.379  
4.374  
4.369  
4.363

2.686  
2.686  
2.206  
2.185  
2.095  
2.088  
2.080  
2.073  
2.066  
1.757  
1.751  
1.745

1.757  
1.751  
1.745

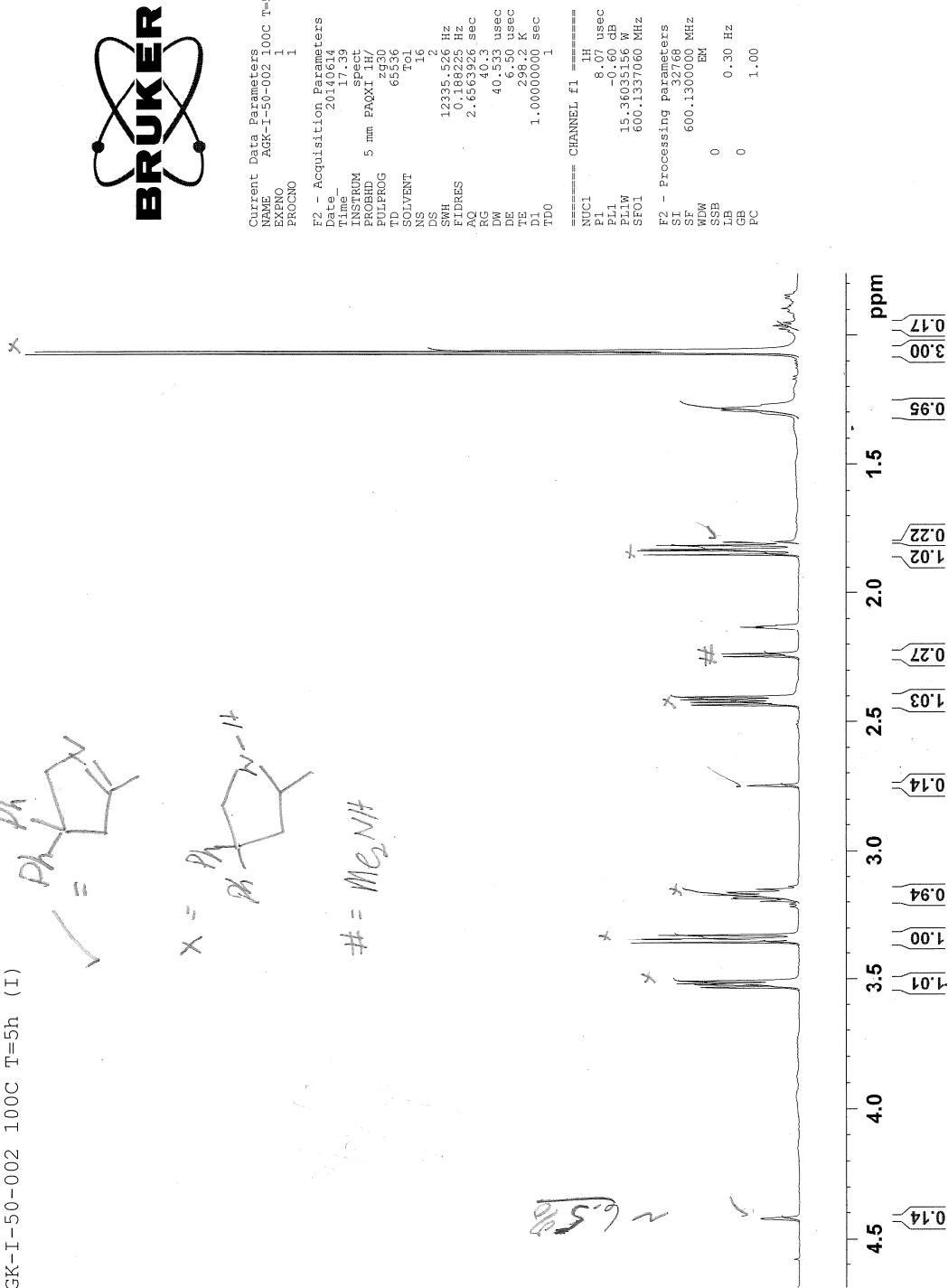
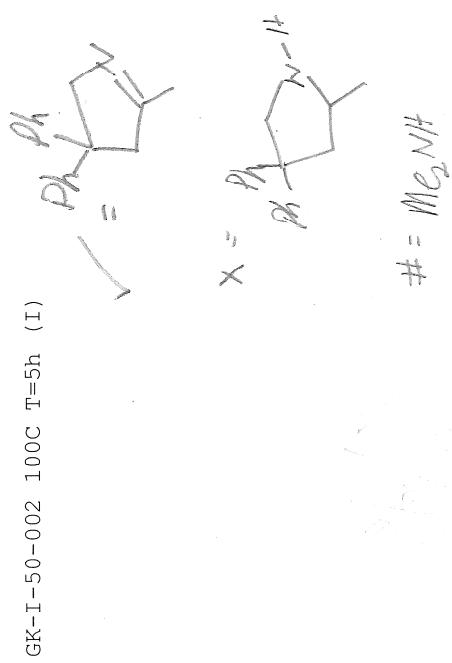


Current Date Parameters  
NAME = 1-49-001, 1  
EXNO =  
PROCNO = 1  
F2 - Acquisition Parameters  
Date = 2014/6/10  
Time = 11:38  
INSTRUM spect  
PROBID 5 mm PABBO BB  
PULPROG zg30  
TD 6536  
T1 7.02  
SW1 1.2  
DS 2  
SF1 6009.615 Hz  
SF2 6009.615 Hz  
FIDRES 5.422992 sec  
AQ 5.422992 sec  
RG 90.5  
DW 83.200 usec  
DE 6.00 usec  
TM 292.1 K  
D1 1.000000 sec  
TDO 1.000000 sec

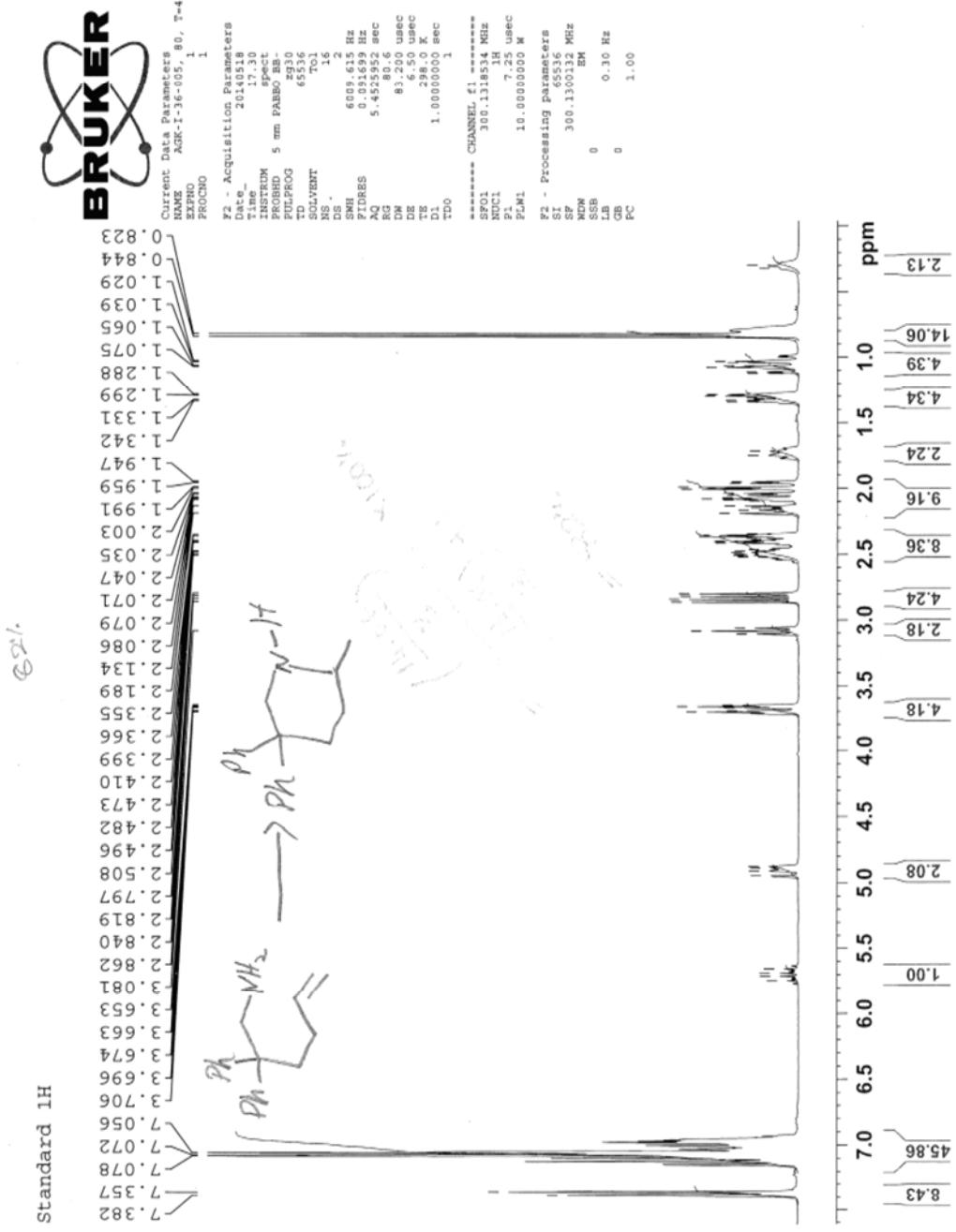
===== CHANNEL f1 ======  
SP01 300.1318154 MHz  
NUC1 <sup>1</sup>H  
P1 7.25 usec  
PL1 10.0000000 W  
F2 - Processing parameters  
SI 65536  
SF 300.1300117 MHz  
RMW 0  
SSB 0  
LB 0.30 Hz  
PC 1.00



Partial  $^1\text{H}$  NMR spectrum from Table 1 entry 1 catalytic trial



Partial <sup>1</sup>H NMR spectrum from Table 1 entry 5 catalytic trial



## Xray data

Iodo complex **3**

### CRYSTAL SUMMARY

Crystal data for C<sub>24</sub>H<sub>37</sub>I N<sub>6</sub>Zr; M<sub>r</sub> = 627.71; Monoclinic; space group P2<sub>1</sub>/c;  $a = 10.3811(2)$  Å;  $b = 11.8205(2)$  Å;  $c = 21.6790(4)$  Å;  $\alpha = 90^\circ$ ;  $\beta = 97.8822(8)^\circ$ ;  $\gamma = 90^\circ$ ;  $V = 2635.09(8)$  Å<sup>3</sup>; Z = 4; T = 100(2) K;  $\lambda(\text{Mo-K}\alpha) = 0.71073$  Å;  $\mu(\text{Mo-K}\alpha) = 1.612$  mm<sup>-1</sup>;  $d_{\text{calc}} = 1.582$  g.cm<sup>-3</sup>; 21565 reflections collected; 4908 unique ( $R_{\text{int}} = 0.0295$ ); giving  $R_1 = 0.0264$ ,  $wR_2 = 0.0666$  for 4475 data with [ $I > 2\sigma(I)$ ] and  $R_1 = 0.0297$ ,  $wR_2 = 0.0689$  for all 4908 data. Residual electron density (e<sup>-</sup>.Å<sup>-3</sup>) max/min: 0.794/-0.738.

An arbitrary sphere of data were collected on a yellow plate-like-like crystal, having approximate dimensions of 0.150 × 0.150 × 0.100 mm, on a Bruker APEX-II diffractometer using a combination of  $\omega$ - and  $\varphi$ -scans of 0.5° [1]. Data were corrected for absorption and polarization effects and analyzed for space group determination. The structure was solved by direct methods and expanded routinely [2]. The model was refined by full-matrix least-squares analysis of F<sup>2</sup> against all reflections. All non-hydrogen atoms were refined with anisotropic thermal displacement parameters. Unless otherwise noted, hydrogen atoms were included in calculated positions. Thermal parameters for the hydrogens were tied to the isotropic thermal parameter of the atom to which they are bonded (1.5 × for methyl, 1.2 × for all others).

### REFERENCES

- [1] Bruker AXS. (2008). *APEX-2*. Bruker-Nonius AXS, Madison, Wisconsin, USA.
- [2] G. M. Sheldrick, *Acta Cryst.*, **2008**, A64, 112.
- [3] H. D. Flack, *Acta Cryst.*, **1983**, A39, 876.
- [4] R. W. W. Hooft, L. H. Straver & A. L. Spek, *J. Appl. Cryst.*, **2008**, 41, 96.
- [5] L. Yang, D. R. Powell & R. P. Houser, *Dalton Trans.*, **2007**, 955.
- [6] A. W. Addison, T. N. Rao, J. Reedijk, J. van Rijn & G. C. Verschoor, *J. Chem. Soc. Dalton Trans.*, **1984**, 1349.

Table S1. Crystal data and structure refinement for **3**.

Identification code	wdc167
Empirical formula	C24 H37 I N6 Zr
Formula weight	627.71
Temperature	100(2) K
Wavelength	0.71073 Å
Crystal system	Monoclinic
Space group	P2 <sub>1</sub> /c
Unit cell dimensions	$a = 10.3811(2)$ Å $\alpha = 90^\circ$ $b = 11.8205(2)$ Å $\beta = 97.8822(8)^\circ$ $c = 21.6790(4)$ Å $\gamma = 90^\circ$
Volume	2635.09(8) Å <sup>3</sup>
Z	4
Density (calculated)	1.582 g.cm <sup>-3</sup>
Absorption coefficient ( $\mu$ )	1.612 mm <sup>-1</sup>
F(000)	1264
Crystal color, habit	yellow, plate-like
Crystal size	0.150 × 0.150 × 0.100 mm <sup>3</sup>
$\theta$ range for data collection	1.897 to 25.623°
Index ranges	-12 ≤ h ≤ 12, -12 ≤ k ≤ 14, -25 ≤ l ≤ 26
Reflections collected	21565
Independent reflections	4908 [R <sub>int</sub> = 0.0295]
Completeness to $\theta = 25.242^\circ$	99.6 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.7452 and 0.6261
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	4908 / 0 / 289
Goodness-of-fit on F <sup>2</sup>	1.080
Final R indices [I>2σ(I)]	R <sub>1</sub> = 0.0264, wR <sub>2</sub> = 0.0666
R indices (all data)	R <sub>1</sub> = 0.0297, wR <sub>2</sub> = 0.0689
Extinction coefficient	n/a
Largest diff. peak and hole	0.794 and -0.738 e <sup>-</sup> .Å <sup>-3</sup>

Table S2. Atomic coordinates and equivalent isotropic displacement parameters (Å<sup>2</sup>) for **3**. U(eq) is defined as one third of the trace of the orthogonalized U<sub>ij</sub> tensor.

	x	y	z	U(eq)
Zr(1)	0.26485(3)	0.49803(2)	0.36128(2)	0.010(1)
I(1)	0.18125(2)	0.47308(2)	0.21974(2)	0.016(1)
N(1)	0.3364(2)	0.2365(2)	0.32995(11)	0.013(1)
N(2)	0.1407(2)	0.2159(2)	0.34871(11)	0.013(1)
N(3)	0.5508(2)	0.5892(2)	0.33630(11)	0.015(1)

N(4)	0.4641(2)	0.7410(2)	0.36489(12)	0.016(1)
N(5)	0.2921(2)	0.4786(2)	0.45629(12)	0.015(1)
N(6)	0.1250(2)	0.6236(2)	0.34918(11)	0.014(1)
C(1)	0.4482(3)	0.4111(3)	0.33222(13)	0.013(1)
C(2)	0.5602(3)	0.4704(3)	0.32411(14)	0.015(1)
C(3)	0.6712(3)	0.4204(3)	0.30669(14)	0.019(1)
C(4)	0.6695(3)	0.3039(3)	0.29737(14)	0.020(1)
C(5)	0.5606(3)	0.2395(3)	0.30425(14)	0.017(1)
C(6)	0.4535(3)	0.2954(3)	0.32178(13)	0.014(1)
C(7)	0.2337(3)	0.2954(2)	0.34758(13)	0.012(1)
C(8)	0.3053(3)	0.1238(3)	0.31958(14)	0.016(1)
C(9)	0.1821(3)	0.1106(3)	0.33152(14)	0.016(1)
C(10)	0.4384(3)	0.6300(3)	0.35474(13)	0.013(1)
C(11)	0.6423(3)	0.6737(3)	0.33473(14)	0.019(1)
C(12)	0.5873(3)	0.7691(3)	0.35256(15)	0.020(1)
C(13)	0.3772(3)	0.8200(3)	0.39069(15)	0.018(1)
C(14)	0.4094(3)	0.8322(3)	0.46064(14)	0.019(1)
C(15)	0.3231(3)	0.9201(3)	0.48631(15)	0.023(1)
C(16)	0.3433(4)	0.9242(4)	0.55726(17)	0.034(1)
C(17)	0.0111(3)	0.2391(3)	0.36559(14)	0.015(1)
C(18)	0.0136(3)	0.2580(3)	0.43510(14)	0.017(1)
C(19)	0.0506(3)	0.1540(3)	0.47437(15)	0.024(1)
C(20)	0.0583(4)	0.1775(3)	0.54392(15)	0.027(1)
C(21)	0.0156(3)	0.5565(3)	0.36330(16)	0.021(1)
C(22)	0.0782(3)	0.7192(3)	0.31074(15)	0.019(1)
C(23)	0.3730(3)	0.3881(3)	0.48690(15)	0.020(1)
C(24)	0.2348(3)	0.5442(3)	0.50230(15)	0.024(1)
H(3A)	0.7439	0.4634	0.3015	0.023
H(4A)	0.7429	0.2685	0.2863	0.024
H(5A)	0.5593	0.1619	0.2974	0.020
H(8A)	0.3590	0.0679	0.3068	0.020
H(9A)	0.1342	0.0438	0.3287	0.020
H(11A)	0.7252	0.6661	0.3236	0.023
H(12A)	0.6251	0.8406	0.3560	0.024
H(13A)	0.2883	0.7936	0.3807	0.022
H(13B)	0.3832	0.8935	0.3714	0.022

H(14A)	0.4997	0.8544	0.4709	0.023
H(14B)	0.3982	0.7598	0.4802	0.023
H(15A)	0.3417	0.9940	0.4702	0.028
H(15B)	0.2328	0.9026	0.4718	0.028
H(16A)	0.2872	0.9806	0.5711	0.051
H(16B)	0.4322	0.9429	0.5719	0.051
H(16C)	0.3231	0.8517	0.5734	0.051
H(17A)	-0.0458	0.1759	0.3526	0.018
H(17B)	-0.0244	0.3057	0.3433	0.018
H(18A)	0.0750	0.3180	0.4483	0.021
H(18B)	-0.0717	0.2832	0.4427	0.021
H(19A)	0.1343	0.1265	0.4657	0.028
H(19B)	-0.0131	0.0950	0.4628	0.028
H(20A)	0.0820	0.1094	0.5668	0.040
H(20B)	-0.0248	0.2032	0.5530	0.040
H(20C)	0.1225	0.2347	0.5558	0.040
H(21A)	-0.0609	0.6030	0.3598	0.032
H(21B)	0.0340	0.5275	0.4049	0.032
H(21C)	0.0014	0.4947	0.3345	0.032
H(22A)	0.0039	0.7516	0.3261	0.028
H(22B)	0.0539	0.6944	0.2686	0.028
H(22C)	0.1457	0.7750	0.3120	0.028
H(23A)	0.3737	0.3932	0.5311	0.030
H(23B)	0.4601	0.3956	0.4772	0.030
H(23C)	0.3382	0.3162	0.4724	0.030
H(24A)	0.2653	0.5159	0.5432	0.035
H(24B)	0.1418	0.5380	0.4944	0.035
H(24C)	0.2595	0.6222	0.4997	0.035

Table S3. Anisotropic displacement parameters ( $\text{\AA}^2$ ) for **3**.

The anisotropic displacement factor exponent takes the form:

$$-2\pi^2[h^2a^{*2}U_{11} + \dots + 2hka^{*}b^{*}U_{12}]$$

	U <sub>11</sub>	U <sub>22</sub>	U <sub>33</sub>	U <sub>23</sub>	U <sub>13</sub>	U <sub>12</sub>
Zr(1)	0.0101(1)	0.0101(2)	0.0110(2)	-0.0003(1)	0.0023(1)	0.0005(1)

I(1)	0.0188(1)	0.0153(1)	0.0125(1)	-0.0014(1)	0.0006(1)	0.0037(1)
N(1)	0.0159(13)	0.0105(13)	0.0125(13)	0.0000(10)	0.0016(10)	0.0011(10)
N(2)	0.0142(12)	0.0124(13)	0.0114(13)	0.0002(10)	0.0008(10)	-0.0003(10)
N(3)	0.0144(12)	0.0157(14)	0.0142(13)	0.0009(11)	0.0023(10)	-0.0031(10)
N(4)	0.0174(13)	0.0144(14)	0.0144(13)	0.0011(11)	-0.0001(10)	-0.0030(10)
N(5)	0.0180(13)	0.0138(13)	0.0126(13)	-0.0010(10)	0.0029(10)	-0.0009(10)
N(6)	0.0138(12)	0.0131(13)	0.0142(13)	-0.0001(10)	0.0017(10)	0.0012(10)
C(1)	0.0125(14)	0.0158(16)	0.0090(14)	0.0011(12)	0.0010(11)	0.0018(11)
C(2)	0.0143(15)	0.0178(17)	0.0117(15)	0.0008(12)	0.0001(11)	0.0014(12)
C(3)	0.0128(15)	0.0283(19)	0.0161(16)	0.0013(14)	0.0016(12)	-0.0034(13)
C(4)	0.0153(15)	0.0278(19)	0.0166(16)	-0.0007(14)	0.0033(12)	0.0071(13)
C(5)	0.0179(15)	0.0164(16)	0.0155(16)	0.0011(13)	0.0013(12)	0.0060(12)
C(6)	0.0143(14)	0.0161(16)	0.0106(15)	0.0020(12)	0.0013(11)	0.0009(12)
C(7)	0.0133(14)	0.0121(15)	0.0107(14)	0.0002(12)	0.0004(11)	0.0008(11)
C(8)	0.0226(16)	0.0097(15)	0.0165(16)	-0.0008(12)	0.0023(12)	0.0027(12)
C(9)	0.0221(16)	0.0107(15)	0.0165(16)	-0.0007(12)	0.0027(12)	-0.0001(12)
C(10)	0.0148(14)	0.0135(16)	0.0106(15)	0.0001(12)	-0.0002(11)	0.0007(11)
C(11)	0.0169(15)	0.0222(18)	0.0181(16)	0.0006(13)	0.0038(12)	-0.0071(13)
C(12)	0.0225(17)	0.0187(17)	0.0196(17)	0.0015(13)	0.0027(13)	-0.0084(13)
C(13)	0.0227(16)	0.0103(16)	0.0215(17)	-0.0011(13)	-0.0001(13)	0.0008(12)
C(14)	0.0210(16)	0.0165(17)	0.0185(17)	0.0003(13)	0.0013(13)	-0.0009(13)
C(15)	0.0224(17)	0.0218(18)	0.0257(18)	-0.0028(15)	0.0046(13)	0.0010(14)
C(16)	0.035(2)	0.042(2)	0.027(2)	-0.0065(18)	0.0096(16)	0.0007(17)
C(17)	0.0124(14)	0.0127(15)	0.0203(16)	-0.0012(13)	0.0025(12)	-0.0018(11)
C(18)	0.0174(16)	0.0143(16)	0.0214(17)	-0.0016(13)	0.0063(12)	-0.0014(12)
C(19)	0.0333(19)	0.0197(18)	0.0192(17)	0.0000(14)	0.0086(14)	-0.0009(14)
C(20)	0.035(2)	0.0253(19)	0.0204(18)	0.0027(15)	0.0064(15)	-0.0021(15)
C(21)	0.0134(15)	0.0184(17)	0.0312(19)	0.0017(14)	0.0021(13)	0.0018(13)
C(22)	0.0207(16)	0.0168(17)	0.0181(17)	0.0003(13)	0.0017(12)	0.0057(13)
C(23)	0.0240(17)	0.0212(18)	0.0152(16)	0.0024(13)	0.0009(13)	0.0010(13)
C(24)	0.0310(19)	0.0235(18)	0.0180(17)	-0.0016(14)	0.0098(14)	0.0019(14)

Table S4. Bond lengths [ $\text{\AA}$ ] for **3**.

atom-atom	distance	atom-atom	distance
Zr(1)-N(5)	2.053(3)	Zr(1)-N(6)	2.068(2)
Zr(1)-C(1)	2.325(3)	Zr(1)-C(10)	2.402(3)
Zr(1)-C(7)	2.429(3)	Zr(1)-C(21)	2.684(3)
Zr(1)-I(1)	3.0864(3)	N(1)-C(7)	1.371(4)
N(1)-C(8)	1.382(4)	N(1)-C(6)	1.433(4)
N(2)-C(7)	1.351(4)	N(2)-C(9)	1.384(4)
N(2)-C(17)	1.468(4)	N(3)-C(10)	1.371(4)
N(3)-C(11)	1.381(4)	N(3)-C(2)	1.435(4)
N(4)-C(10)	1.351(4)	N(4)-C(12)	1.383(4)
N(4)-C(13)	1.462(4)	N(5)-C(24)	1.453(4)
N(5)-C(23)	1.462(4)	N(6)-C(22)	1.448(4)
N(6)-C(21)	1.452(4)	C(1)-C(6)	1.388(4)
C(1)-C(2)	1.390(4)	C(2)-C(3)	1.392(4)
C(3)-C(4)	1.391(5)	C(3)-H(3A)	0.9300
C(4)-C(5)	1.388(4)	C(4)-H(4A)	0.9300
C(5)-C(6)	1.390(4)	C(5)-H(5A)	0.9300
C(8)-C(9)	1.349(4)	C(8)-H(8A)	0.9300
C(9)-H(9A)	0.9300	C(11)-C(12)	1.345(5)
C(11)-H(11A)	0.9300	C(12)-H(12A)	0.9300
C(13)-C(14)	1.514(4)	C(13)-H(13A)	0.9700
C(13)-H(13B)	0.9700	C(14)-C(15)	1.525(4)
C(14)-H(14A)	0.9700	C(14)-H(14B)	0.9700
C(15)-C(16)	1.525(5)	C(15)-H(15A)	0.9700
C(15)-H(15B)	0.9700	C(16)-H(16A)	0.9600
C(16)-H(16B)	0.9600	C(16)-H(16C)	0.9600
C(17)-C(18)	1.520(4)	C(17)-H(17A)	0.9700
C(17)-H(17B)	0.9700	C(18)-C(19)	1.514(5)
C(18)-H(18A)	0.9700	C(18)-H(18B)	0.9700
C(19)-C(20)	1.524(5)	C(19)-H(19A)	0.9700
C(19)-H(19B)	0.9700	C(20)-H(20A)	0.9600
C(20)-H(20B)	0.9600	C(20)-H(20C)	0.9600
C(21)-H(21A)	0.9600	C(21)-H(21B)	0.9600
C(21)-H(21C)	0.9600	C(22)-H(22A)	0.9600
C(22)-H(22B)	0.9600	C(22)-H(22C)	0.9600
C(23)-H(23A)	0.9600	C(23)-H(23B)	0.9600
C(23)-H(23C)	0.9600	C(24)-H(24A)	0.9600
C(24)-H(24B)	0.9600	C(24)-H(24C)	0.9600

Symmetry transformations used to generate equivalent atoms:

Table S5. Bond angles [°] for **3**.

atom-atom-atom	angle	atom-atom-atom	angle
N(5)-Zr(1)-N(6)	101.85(10)	N(5)-Zr(1)-C(1)	102.54(10)
N(6)-Zr(1)-C(1)	150.42(10)	N(5)-Zr(1)-C(10)	97.50(10)
N(6)-Zr(1)-C(10)	92.62(10)	C(1)-Zr(1)-C(10)	67.82(10)
N(5)-Zr(1)-C(7)	90.58(10)	N(6)-Zr(1)-C(7)	127.80(10)
C(1)-Zr(1)-C(7)	68.35(10)	C(10)-Zr(1)-C(7)	136.15(10)
N(5)-Zr(1)-C(21)	90.74(10)	N(6)-Zr(1)-C(21)	32.39(10)
C(1)-Zr(1)-C(21)	160.92(10)	C(10)-Zr(1)-C(21)	124.49(10)
C(7)-Zr(1)-C(21)	98.26(10)	N(5)-Zr(1)-I(1)	165.47(7)
N(6)-Zr(1)-I(1)	81.12(7)	C(1)-Zr(1)-I(1)	79.47(7)
C(10)-Zr(1)-I(1)	96.56(7)	C(7)-Zr(1)-I(1)	76.69(7)
C(21)-Zr(1)-I(1)	84.29(8)	C(7)-N(1)-C(8)	111.2(2)
C(7)-N(1)-C(6)	119.5(2)	C(8)-N(1)-C(6)	129.2(2)
C(7)-N(2)-C(9)	111.9(2)	C(7)-N(2)-C(17)	123.8(2)
C(9)-N(2)-C(17)	124.3(2)	C(10)-N(3)-C(11)	111.6(3)
C(10)-N(3)-C(2)	118.6(2)	C(11)-N(3)-C(2)	129.7(3)
C(10)-N(4)-C(12)	111.6(3)	C(10)-N(4)-C(13)	124.4(2)
C(12)-N(4)-C(13)	123.8(3)	C(24)-N(5)-C(23)	110.0(2)
C(24)-N(5)-Zr(1)	128.0(2)	C(23)-N(5)-Zr(1)	121.92(19)
C(22)-N(6)-C(21)	109.7(2)	C(22)-N(6)-Zr(1)	143.88(19)
C(21)-N(6)-Zr(1)	97.89(18)	C(6)-C(1)-C(2)	115.0(3)
C(6)-C(1)-Zr(1)	122.2(2)	C(2)-C(1)-Zr(1)	122.8(2)
C(1)-C(2)-C(3)	123.9(3)	C(1)-C(2)-N(3)	113.0(3)
C(3)-C(2)-N(3)	123.0(3)	C(4)-C(3)-C(2)	117.7(3)
C(4)-C(3)-H(3A)	121.2	C(2)-C(3)-H(3A)	121.2
C(5)-C(4)-C(3)	121.4(3)	C(5)-C(4)-H(4A)	119.3
C(3)-C(4)-H(4A)	119.3	C(4)-C(5)-C(6)	117.5(3)
C(4)-C(5)-H(5A)	121.2	C(6)-C(5)-H(5A)	121.2
C(1)-C(6)-C(5)	124.3(3)	C(1)-C(6)-N(1)	113.9(2)
C(5)-C(6)-N(1)	121.8(3)	N(2)-C(7)-N(1)	103.7(2)
N(2)-C(7)-Zr(1)	140.1(2)	N(1)-C(7)-Zr(1)	116.07(19)
C(9)-C(8)-N(1)	106.7(3)	C(9)-C(8)-H(8A)	126.7
N(1)-C(8)-H(8A)	126.7	C(8)-C(9)-N(2)	106.6(3)
C(8)-C(9)-H(9A)	126.7	N(2)-C(9)-H(9A)	126.7
N(4)-C(10)-N(3)	103.4(2)	N(4)-C(10)-Zr(1)	138.9(2)
N(3)-C(10)-Zr(1)	117.7(2)	C(12)-C(11)-N(3)	106.2(3)
C(12)-C(11)-H(11A)	126.9	N(3)-C(11)-H(11A)	126.9
C(11)-C(12)-N(4)	107.1(3)	C(11)-C(12)-H(12A)	126.4
N(4)-C(12)-H(12A)	126.4	N(4)-C(13)-C(14)	112.4(3)
N(4)-C(13)-H(13A)	109.1	C(14)-C(13)-H(13A)	109.1
N(4)-C(13)-H(13B)	109.1	C(14)-C(13)-H(13B)	109.1
H(13A)-C(13)-H(13B)	107.9	C(13)-C(14)-C(15)	111.7(3)
C(13)-C(14)-H(14A)	109.3	C(15)-C(14)-H(14A)	109.3

C(13)-C(14)-H(14B)	109.3	C(15)-C(14)-H(14B)	109.3
H(14A)-C(14)-H(14B)	107.9	C(16)-C(15)-C(14)	112.6(3)
C(16)-C(15)-H(15A)	109.1	C(14)-C(15)-H(15A)	109.1
C(16)-C(15)-H(15B)	109.1	C(14)-C(15)-H(15B)	109.1
H(15A)-C(15)-H(15B)	107.8	C(15)-C(16)-H(16A)	109.5
C(15)-C(16)-H(16B)	109.5	H(16A)-C(16)-H(16B)	109.5
C(15)-C(16)-H(16C)	109.5	H(16A)-C(16)-H(16C)	109.5
H(16B)-C(16)-H(16C)	109.5	N(2)-C(17)-C(18)	112.5(2)
N(2)-C(17)-H(17A)	109.1	C(18)-C(17)-H(17A)	109.1
N(2)-C(17)-H(17B)	109.1	C(18)-C(17)-H(17B)	109.1
H(17A)-C(17)-H(17B)	107.8	C(19)-C(18)-C(17)	114.0(3)
C(19)-C(18)-H(18A)	108.8	C(17)-C(18)-H(18A)	108.8
C(19)-C(18)-H(18B)	108.8	C(17)-C(18)-H(18B)	108.8
H(18A)-C(18)-H(18B)	107.7	C(18)-C(19)-C(20)	112.5(3)
C(18)-C(19)-H(19A)	109.1	C(20)-C(19)-H(19A)	109.1
C(18)-C(19)-H(19B)	109.1	C(20)-C(19)-H(19B)	109.1
H(19A)-C(19)-H(19B)	107.8	C(19)-C(20)-H(20A)	109.5
C(19)-C(20)-H(20B)	109.5	H(20A)-C(20)-H(20B)	109.5
C(19)-C(20)-H(20C)	109.5	H(20A)-C(20)-H(20C)	109.5
H(20B)-C(20)-H(20C)	109.5	N(6)-C(21)-Zr(1)	49.72(13)
N(6)-C(21)-H(21A)	109.5	Zr(1)-C(21)-H(21A)	159.2
N(6)-C(21)-H(21B)	109.5	Zr(1)-C(21)-H(21B)	81.7
H(21A)-C(21)-H(21B)	109.5	N(6)-C(21)-H(21C)	109.5
Zr(1)-C(21)-H(21C)	81.7	H(21A)-C(21)-H(21C)	109.5
H(21B)-C(21)-H(21C)	109.5	N(6)-C(22)-H(22A)	109.5
N(6)-C(22)-H(22B)	109.5	H(22A)-C(22)-H(22B)	109.5
N(6)-C(22)-H(22C)	109.5	H(22A)-C(22)-H(22C)	109.5
H(22B)-C(22)-H(22C)	109.5	N(5)-C(23)-H(23A)	109.5
N(5)-C(23)-H(23B)	109.5	H(23A)-C(23)-H(23B)	109.5
N(5)-C(23)-H(23C)	109.5	H(23A)-C(23)-H(23C)	109.5
H(23B)-C(23)-H(23C)	109.5	N(5)-C(24)-H(24A)	109.5
N(5)-C(24)-H(24B)	109.5	H(24A)-C(24)-H(24B)	109.5
N(5)-C(24)-H(24C)	109.5	H(24A)-C(24)-H(24C)	109.5
H(24B)-C(24)-H(24C)	109.5		

Symmetry transformations used to generate equivalent atoms:

Table S6. Torsion angles [°] for **3**.

atom-atom-atom-atom	angle	atom-atom-atom-atom	angle
C(6)-C(1)-C(2)-C(3)	0.2(4)	Zr(1)-C(1)-C(2)-C(3)	-179.9(2)
C(6)-C(1)-C(2)-N(3)	-179.1(2)	Zr(1)-C(1)-C(2)-N(3)	0.8(4)
C(10)-N(3)-C(2)-C(1)	0.9(4)	C(11)-N(3)-C(2)-C(1)	178.2(3)
C(10)-N(3)-C(2)-C(3)	-178.4(3)	C(11)-N(3)-C(2)-C(3)	-1.1(5)
C(1)-C(2)-C(3)-C(4)	-0.5(5)	N(3)-C(2)-C(3)-C(4)	178.7(3)

C(2)-C(3)-C(4)-C(5)	0.9(5)	C(3)-C(4)-C(5)-C(6)	-1.0(5)
C(2)-C(1)-C(6)-C(5)	-0.2(4)	Zr(1)-C(1)-C(6)-C(5)	179.9(2)
C(2)-C(1)-C(6)-N(1)	-179.5(2)	Zr(1)-C(1)-C(6)-N(1)	0.6(3)
C(4)-C(5)-C(6)-C(1)	0.6(5)	C(4)-C(5)-C(6)-N(1)	179.9(3)
C(7)-N(1)-C(6)-C(1)	-1.0(4)	C(8)-N(1)-C(6)-C(1)	175.4(3)
C(7)-N(1)-C(6)-C(5)	179.7(3)	C(8)-N(1)-C(6)-C(5)	-3.9(5)
C(9)-N(2)-C(7)-N(1)	-0.9(3)	C(17)-N(2)-C(7)-N(1)	179.7(2)
C(9)-N(2)-C(7)-Zr(1)	175.0(2)	C(17)-N(2)-C(7)-Zr(1)	-4.4(5)
C(8)-N(1)-C(7)-N(2)	1.0(3)	C(6)-N(1)-C(7)-N(2)	178.0(2)
C(8)-N(1)-C(7)-Zr(1)	-176.07(19)	C(6)-N(1)-C(7)-Zr(1)	1.0(3)
C(7)-N(1)-C(8)-C(9)	-0.8(3)	C(6)-N(1)-C(8)-C(9)	-177.4(3)
N(1)-C(8)-C(9)-N(2)	0.2(3)	C(7)-N(2)-C(9)-C(8)	0.4(3)
C(17)-N(2)-C(9)-C(8)	179.9(3)	C(12)-N(4)-C(10)-N(3)	0.7(3)
C(13)-N(4)-C(10)-N(3)	-174.5(3)	C(12)-N(4)-C(10)-Zr(1)	179.8(2)
C(13)-N(4)-C(10)-Zr(1)	4.6(5)	C(11)-N(3)-C(10)-N(4)	-0.5(3)
C(2)-N(3)-C(10)-N(4)	177.2(2)	C(11)-N(3)-C(10)-Zr(1)	-179.8(2)
C(2)-N(3)-C(10)-Zr(1)	-2.1(3)	C(10)-N(3)-C(11)-C(12)	0.2(4)
C(2)-N(3)-C(11)-C(12)	-177.2(3)	N(3)-C(11)-C(12)-N(4)	0.3(3)
C(10)-N(4)-C(12)-C(11)	-0.6(4)	C(13)-N(4)-C(12)-C(11)	174.6(3)
C(10)-N(4)-C(13)-C(14)	93.9(3)	C(12)-N(4)-C(13)-C(14)	-80.7(4)
N(4)-C(13)-C(14)-C(15)	176.6(3)	C(13)-C(14)-C(15)-C(16)	173.5(3)
C(7)-N(2)-C(17)-C(18)	-73.9(4)	C(9)-N(2)-C(17)-C(18)	106.7(3)
N(2)-C(17)-C(18)-C(19)	-66.4(3)	C(17)-C(18)-C(19)-C(20)	177.3(3)
C(22)-N(6)-C(21)-Zr(1)	156.3(3)		

Symmetry transformations used to generate equivalent atoms:

## Bromo complex 4

### CRYSTAL SUMMARY

Crystal data for C<sub>28</sub>H<sub>45</sub>BrN<sub>6</sub>OZr; M<sub>r</sub> = 652.83; Monoclinic; space group P2<sub>1</sub>/c;  $a = 9.0178(3)$  Å;  $b = 13.7634(4)$  Å;  $c = 24.1447(8)$  Å;  $\alpha = 90^\circ$ ;  $\beta = 92.5093(16)^\circ$ ;  $\gamma = 90^\circ$ ; V = 2993.86(17) Å<sup>3</sup>; Z = 4; T = 100(2) K;  $\lambda(\text{Mo-K}\alpha) = 0.71073$  Å;  $\mu(\text{Mo-K}\alpha) = 1.733$  mm<sup>-1</sup>; d<sub>calc</sub> = 1.448 g·cm<sup>-3</sup>; 24368 reflections collected; 5547 unique ( $R_{\text{int}} = 0.0284$ ); giving  $R_1 = 0.0390$ ,  $wR_2 = 0.0949$  for 5112 data with [ $I > 2\sigma(I)$ ] and  $R_1 = 0.0426$ ,  $wR_2 = 0.0963$  for all 5547 data. Residual electron density (e<sup>-</sup>·Å<sup>-3</sup>) max/min: 0.798/-0.771.

An arbitrary sphere of data were collected on a Yellow Needle-Like-like crystal, having approximate dimensions of 0.200 × 0.100 × 0.100 mm, on a Bruker APEX-II diffractometer using a combination of  $\omega$ - and  $\varphi$ -scans of 0.5° [1]. Data were corrected for absorption and polarization effects and analyzed for space group determination. The structure was solved by direct methods and expanded routinely [2]. The model was refined by full-matrix least-squares analysis of F<sup>2</sup> against all reflections. All non-hydrogen atoms were refined with anisotropic thermal displacement parameters. Unless otherwise noted, hydrogen atoms were included in calculated positions. Thermal parameters for the hydrogens were tied to the isotropic thermal parameter of the atom to which they are bonded (1.5 × for methyl, 1.2 × for all others).

### REFERENCES

- [1] Bruker AXS. (2008). *APEX-2*. Bruker-Nonius AXS, Madison, Wisconsin, USA.
- [2] G. M. Sheldrick, *Acta Cryst.*, **2008**, A64, 112.
- [3] H. D. Flack, *Acta Cryst.*, **1983**, A39, 876.
- [4] R. W. W. Hooft, L. H. Straver & A. L. Spek, *J. Appl. Cryst.*, **2008**, 41, 96.
- [5] L. Yang, D. R. Powell & R. P. Houser, *Dalton Trans.*, **2007**, 955.
- [6] A. W. Addison, T. N. Rao, J. Reedijk, J. van Rijn & G. C. Verschoor, *J. Chem. Soc. Dalton Trans.*, **1984**, 1349.

Table S7. Crystal data and structure refinement for **4**.

Identification code	WDC168
Empirical formula	C <sub>28</sub> H <sub>45</sub> BrN <sub>6</sub> OZr
Formula weight	652.83
Temperature	100(2) K
Wavelength	0.71073 Å
Crystal system	Monoclinic
Space group	P2 <sub>1</sub> /c
Unit cell dimensions	$a = 9.0178(3)$ Å $\alpha = 90^\circ$ $b = 13.7634(4)$ Å $\beta = 92.5093(16)^\circ$ $c = 24.1447(8)$ Å $\gamma = 90^\circ$
Volume	2993.86(17) Å <sup>3</sup>
Z	4
Density (calculated)	1.448 g.cm <sup>-3</sup>
Absorption coefficient ( $\mu$ )	1.733 mm <sup>-1</sup>
F(000)	1352
Crystal color, habit	Yellow, Needle-Like
Crystal size	0.200 × 0.100 × 0.100 mm <sup>3</sup>
θ range for data collection	1.703 to 25.540°
Index ranges	-10 ≤ h ≤ 10, -16 ≤ k ≤ 16, -28 ≤ l ≤ 29
Reflections collected	24368
Independent reflections	5547 [R <sub>int</sub> = 0.0284]
Completeness to $\theta = 25.242^\circ$	99.6 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.7452 and 0.5984
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	5547 / 0 / 340
Goodness-of-fit on F <sup>2</sup>	1.163
Final R indices [I>2σ(I)]	R <sub>1</sub> = 0.0390, wR <sub>2</sub> = 0.0949
R indices (all data)	R <sub>1</sub> = 0.0426, wR <sub>2</sub> = 0.0963
Extinction coefficient	0
Largest diff. peak and hole	0.798 and -0.771 e <sup>-</sup> .Å <sup>-3</sup>

Table S8. Atomic coordinates and equivalent isotropic displacement parameters ( $\text{\AA}^2$ ) for **4**. U(eq) is defined as one third of the trace of the orthogonalized  $U_{ij}$  tensor.

	x	y	z	U(eq)
Zr(1)	0.43694(4)	0.34668(2)	0.35257(2)	0.016(1)
Br(1)	0.29657(4)	0.35563(2)	0.24654(2)	0.021(1)
O(1)	0.1095(6)	0.8081(3)	0.4275(2)	0.082(1)
N(1)	0.7147(3)	0.2757(2)	0.28098(13)	0.020(1)
N(2)	0.6149(3)	0.1407(2)	0.30165(12)	0.019(1)
N(3)	0.5166(3)	0.5766(2)	0.33526(12)	0.019(1)
N(4)	0.3284(3)	0.5871(2)	0.38539(13)	0.020(1)
N(5)	0.2424(3)	0.2772(2)	0.37850(13)	0.024(1)
N(6)	0.5734(3)	0.3253(2)	0.42148(13)	0.021(1)
C(1)	0.6198(4)	0.4294(2)	0.30556(15)	0.019(1)
C(2)	0.6315(4)	0.5298(3)	0.30552(15)	0.019(1)
C(3)	0.7427(4)	0.5813(3)	0.28027(16)	0.022(1)
C(4)	0.8500(4)	0.5276(3)	0.25444(17)	0.025(1)
C(5)	0.8474(4)	0.4263(3)	0.25399(17)	0.025(1)
C(6)	0.7310(4)	0.3810(3)	0.27891(15)	0.020(1)
C(7)	0.5983(4)	0.2377(3)	0.30759(15)	0.019(1)
C(8)	0.8018(4)	0.2039(3)	0.25924(16)	0.022(1)
C(9)	0.7387(4)	0.1190(3)	0.27232(16)	0.022(1)
C(10)	0.4141(4)	0.5202(3)	0.36107(15)	0.019(1)
C(11)	0.4934(4)	0.6745(3)	0.34377(16)	0.023(1)
C(12)	0.3742(4)	0.6813(3)	0.37548(16)	0.023(1)
C(13)	0.1963(4)	0.5634(3)	0.41644(16)	0.022(1)
C(14)	0.0627(4)	0.5422(3)	0.37838(17)	0.029(1)
C(15)	-0.0732(5)	0.5137(3)	0.41072(19)	0.038(1)
C(16)	-0.1475(5)	0.5987(4)	0.4376(2)	0.051(1)
C(17)	0.5166(4)	0.0685(3)	0.32594(15)	0.021(1)
C(18)	0.5859(4)	0.0214(3)	0.37761(16)	0.026(1)
C(19)	0.4797(5)	-0.0513(3)	0.40191(17)	0.033(1)
C(20)	0.5486(6)	-0.1073(3)	0.4506(2)	0.045(1)
C(21)	0.2469(5)	0.2189(3)	0.42912(18)	0.036(1)
C(22)	0.0923(4)	0.2702(3)	0.3542(2)	0.038(1)

C(23)	0.7156(4)	0.2809(3)	0.43525(17)	0.027(1)
C(24)	0.5177(5)	0.3740(3)	0.46975(17)	0.029(1)
C(25)	0.0784(10)	0.8813(6)	0.4669(3)	0.099(3)
C(26)	-0.0085(8)	0.9601(6)	0.4364(3)	0.081(2)
C(27)	0.0175(7)	0.9422(5)	0.3763(3)	0.067(2)
C(28)	0.0381(12)	0.8330(6)	0.3757(3)	0.109(3)
H(3A)	0.7451	0.6503	0.2807	0.026
H(4A)	0.9271	0.5606	0.2366	0.030
H(5A)	0.9228	0.3899	0.2372	0.031
H(8A)	0.8889	0.2128	0.2391	0.026
H(9A)	0.7728	0.0560	0.2631	0.026
H(11A)	0.5501	0.7267	0.3300	0.027
H(12A)	0.3303	0.7394	0.3884	0.028
H(13A)	0.1733	0.6186	0.4409	0.026
H(13B)	0.2179	0.5060	0.4402	0.026
H(14A)	0.0870	0.4886	0.3529	0.035
H(14B)	0.0388	0.6004	0.3557	0.035
H(15A)	-0.1463	0.4812	0.3851	0.045
H(15B)	-0.0423	0.4663	0.4398	0.045
H(16A)	-0.2333	0.5756	0.4574	0.076
H(16B)	-0.1806	0.6454	0.4090	0.076
H(16C)	-0.0768	0.6303	0.4638	0.076
H(17A)	0.4923	0.0176	0.2981	0.026
H(17B)	0.4227	0.1006	0.3354	0.026
H(18A)	0.6115	0.0721	0.4055	0.032
H(18B)	0.6786	-0.0122	0.3682	0.032
H(19A)	0.3909	-0.0163	0.4141	0.039
H(19B)	0.4466	-0.0979	0.3726	0.039
H(20A)	0.4729	-0.1482	0.4669	0.067
H(20B)	0.5882	-0.0615	0.4786	0.067
H(20C)	0.6291	-0.1483	0.4379	0.067
H(21A)	0.2141	0.1526	0.4203	0.054
H(21B)	0.1810	0.2477	0.4558	0.054
H(21C)	0.3486	0.2174	0.4451	0.054
H(22A)	0.0220	0.2964	0.3802	0.056
H(22B)	0.0685	0.2019	0.3464	0.056

H(22C)	0.0852	0.3074	0.3196	0.056
H(23A)	0.7037	0.2314	0.4639	0.041
H(23B)	0.7856	0.3307	0.4491	0.041
H(23C)	0.7538	0.2504	0.4021	0.041
H(24A)	0.5131	0.3276	0.5004	0.043
H(24B)	0.4182	0.3997	0.4607	0.043
H(24C)	0.5843	0.4276	0.4807	0.043
H(25A)	0.1719	0.9080	0.4836	0.118
H(25B)	0.0195	0.8539	0.4968	0.118
H(26A)	-0.1155	0.9554	0.4436	0.097
H(26B)	0.0279	1.0252	0.4479	0.097
H(27A)	0.1074	0.9764	0.3645	0.081
H(27B)	-0.0689	0.9624	0.3524	0.081
H(28A)	-0.0592	0.7999	0.3712	0.131
H(28B)	0.1001	0.8136	0.3447	0.131

Table S9. Anisotropic displacement parameters ( $\text{\AA}^2$ ) for **4**.

The anisotropic displacement factor exponent takes the form:

$$-2\pi^2[h^2a^{*2}U_{11} + \dots + 2hka^{*}b^{*}U_{12}]$$

	U <sub>11</sub>	U <sub>22</sub>	U <sub>33</sub>	U <sub>23</sub>	U <sub>13</sub>	U <sub>12</sub>
Zr(1)	0.0168(2)	0.0115(2)	0.0204(2)	0.0007(1)	-0.0022(1)	-0.0005(1)
Br(1)	0.0248(2)	0.0140(2)	0.0233(2)	-0.0007(1)	-0.0055(1)	0.0010(1)
O(1)	0.103(4)	0.068(3)	0.075(3)	0.013(2)	0.003(3)	0.035(3)
N(1)	0.0184(15)	0.0136(15)	0.0263(17)	0.0022(12)	-0.0032(12)	0.0022(12)
N(2)	0.0213(15)	0.0147(15)	0.0199(16)	0.0017(12)	-0.0046(12)	0.0015(12)
N(3)	0.0196(15)	0.0123(14)	0.0233(16)	0.0000(12)	-0.0049(12)	-0.0012(11)
N(4)	0.0209(15)	0.0160(15)	0.0234(17)	-0.0005(12)	-0.0019(13)	-0.0016(12)
N(5)	0.0202(15)	0.0222(16)	0.0286(18)	0.0014(14)	-0.0010(13)	-0.0036(13)
N(6)	0.0237(16)	0.0160(15)	0.0215(16)	0.0004(12)	-0.0022(12)	-0.0018(12)
C(1)	0.0185(17)	0.0150(17)	0.0215(19)	0.0000(14)	-0.0054(14)	-0.0021(14)
C(2)	0.0187(17)	0.0185(18)	0.0201(19)	0.0003(14)	-0.0058(14)	0.0001(14)
C(3)	0.0212(18)	0.0141(17)	0.030(2)	0.0018(15)	-0.0066(15)	-0.0011(14)
C(4)	0.0181(18)	0.0202(19)	0.037(2)	0.0041(17)	0.0001(16)	-0.0051(15)

C(5)	0.0200(18)	0.0204(19)	0.036(2)	0.0001(17)	0.0002(16)	0.0028(15)
C(6)	0.0189(17)	0.0165(17)	0.0240(19)	-0.0009(15)	-0.0048(14)	-0.0031(14)
C(7)	0.0196(17)	0.0148(17)	0.0221(19)	-0.0001(14)	-0.0063(14)	-0.0003(14)
C(8)	0.0192(18)	0.0208(19)	0.026(2)	-0.0005(15)	-0.0007(15)	0.0063(15)
C(9)	0.0228(18)	0.0170(18)	0.026(2)	-0.0023(15)	-0.0023(15)	0.0061(14)
C(10)	0.0185(17)	0.0168(17)	0.0214(19)	0.0018(14)	-0.0050(14)	-0.0015(14)
C(11)	0.0276(19)	0.0100(16)	0.030(2)	0.0000(15)	-0.0045(16)	-0.0017(14)
C(12)	0.0274(19)	0.0117(17)	0.030(2)	-0.0026(15)	-0.0031(16)	0.0024(15)
C(13)	0.0229(18)	0.0194(18)	0.024(2)	-0.0002(15)	-0.0003(15)	-0.0003(15)
C(14)	0.026(2)	0.033(2)	0.028(2)	-0.0008(17)	-0.0027(16)	-0.0018(17)
C(15)	0.030(2)	0.045(3)	0.037(3)	0.008(2)	-0.0066(19)	-0.009(2)
C(16)	0.034(3)	0.073(4)	0.045(3)	0.002(3)	0.008(2)	0.005(2)
C(17)	0.0247(19)	0.0138(17)	0.025(2)	0.0020(15)	-0.0034(15)	-0.0028(14)
C(18)	0.034(2)	0.0200(19)	0.025(2)	0.0035(16)	-0.0071(17)	-0.0028(16)
C(19)	0.046(2)	0.022(2)	0.029(2)	0.0042(17)	-0.0027(19)	-0.0082(18)
C(20)	0.069(3)	0.032(2)	0.034(3)	0.012(2)	-0.002(2)	-0.005(2)
C(21)	0.039(2)	0.036(2)	0.033(2)	0.0098(19)	0.0001(19)	-0.016(2)
C(22)	0.025(2)	0.038(3)	0.049(3)	0.008(2)	-0.0010(19)	-0.0059(18)
C(23)	0.0237(19)	0.029(2)	0.029(2)	-0.0002(17)	-0.0065(16)	0.0001(16)
C(24)	0.035(2)	0.026(2)	0.026(2)	-0.0028(17)	-0.0062(17)	0.0046(17)
C(25)	0.113(6)	0.100(6)	0.085(6)	0.034(5)	0.032(5)	0.055(5)
C(26)	0.075(4)	0.078(5)	0.089(5)	0.020(4)	-0.003(4)	0.016(4)
C(27)	0.063(4)	0.071(4)	0.067(4)	0.018(3)	0.004(3)	0.014(3)
C(28)	0.170(9)	0.072(5)	0.081(5)	0.022(4)	-0.033(6)	0.035(5)

Table S10. Bond lengths [Å] for **4**.

atom-atom	distance	atom-atom	distance
Zr(1)-N(6)	2.047(3)	Zr(1)-N(5)	2.116(3)
Zr(1)-C(1)	2.339(4)	Zr(1)-C(7)	2.384(4)
Zr(1)-C(10)	2.407(4)	Zr(1)-Br(1)	2.8087(5)
O(1)-C(25)	1.422(9)	O(1)-C(28)	1.423(8)
N(1)-C(7)	1.359(5)	N(1)-C(8)	1.380(5)
N(1)-C(6)	1.457(4)	N(2)-C(7)	1.352(4)
N(2)-C(9)	1.380(5)	N(2)-C(17)	1.470(5)
N(3)-C(10)	1.377(5)	N(3)-C(11)	1.381(4)
N(3)-C(2)	1.438(5)	N(4)-C(10)	1.352(5)
N(4)-C(12)	1.385(5)	N(4)-C(13)	1.472(5)

N(5)-C(22)	1.455(5)	N(5)-C(21)	1.461(5)
N(6)-C(23)	1.446(5)	N(6)-C(24)	1.453(5)
C(1)-C(2)	1.385(5)	C(1)-C(6)	1.386(5)
C(2)-C(3)	1.391(5)	C(3)-C(4)	1.387(5)
C(3)-H(3A)	0.9500	C(4)-C(5)	1.394(5)
C(4)-H(4A)	0.9500	C(5)-C(6)	1.382(5)
C(5)-H(5A)	0.9500	C(8)-C(9)	1.344(5)
C(8)-H(8A)	0.9500	C(9)-H(9A)	0.9500
C(11)-C(12)	1.350(5)	C(11)-H(11A)	0.9500
C(12)-H(12A)	0.9500	C(13)-C(14)	1.511(5)
C(13)-H(13A)	0.9900	C(13)-H(13B)	0.9900
C(14)-C(15)	1.533(6)	C(14)-H(14A)	0.9900
C(14)-H(14B)	0.9900	C(15)-C(16)	1.508(7)
C(15)-H(15A)	0.9900	C(15)-H(15B)	0.9900
C(16)-H(16A)	0.9800	C(16)-H(16B)	0.9800
C(16)-H(16C)	0.9800	C(17)-C(18)	1.516(5)
C(17)-H(17A)	0.9900	C(17)-H(17B)	0.9900
C(18)-C(19)	1.521(5)	C(18)-H(18A)	0.9900
C(18)-H(18B)	0.9900	C(19)-C(20)	1.516(6)
C(19)-H(19A)	0.9900	C(19)-H(19B)	0.9900
C(20)-H(20A)	0.9800	C(20)-H(20B)	0.9800
C(20)-H(20C)	0.9800	C(21)-H(21A)	0.9800
C(21)-H(21B)	0.9800	C(21)-H(21C)	0.9800
C(22)-H(22A)	0.9800	C(22)-H(22B)	0.9800
C(22)-H(22C)	0.9800	C(23)-H(23A)	0.9800
C(23)-H(23B)	0.9800	C(23)-H(23C)	0.9800
C(24)-H(24A)	0.9800	C(24)-H(24B)	0.9800
C(24)-H(24C)	0.9800	C(25)-C(26)	1.511(9)
C(25)-H(25A)	0.9900	C(25)-H(25B)	0.9900
C(26)-C(27)	1.499(9)	C(26)-H(26A)	0.9900
C(26)-H(26B)	0.9900	C(27)-C(28)	1.514(10)
C(27)-H(27A)	0.9900	C(27)-H(27B)	0.9900
C(28)-H(28A)	0.9900	C(28)-H(28B)	0.9900

Symmetry transformations used to generate equivalent atoms:

Table S11. Bond angles [°] for **4**.

atom-atom-atom	angle	atom-atom-atom	angle
N(6)-Zr(1)-N(5)	99.88(12)	N(6)-Zr(1)-C(1)	93.02(12)
N(5)-Zr(1)-C(1)	167.10(12)	N(6)-Zr(1)-C(7)	85.57(12)
N(5)-Zr(1)-C(7)	112.39(12)	C(1)-Zr(1)-C(7)	68.16(12)
N(6)-Zr(1)-C(10)	97.11(12)	N(5)-Zr(1)-C(10)	110.35(12)
C(1)-Zr(1)-C(10)	67.84(12)	C(7)-Zr(1)-C(10)	136.00(12)

N(6)-Zr(1)-Br(1)	168.01(9)	N(5)-Zr(1)-Br(1)	86.74(9)
C(1)-Zr(1)-Br(1)	80.53(8)	C(7)-Zr(1)-Br(1)	82.65(8)
C(10)-Zr(1)-Br(1)	89.84(8)	C(25)-O(1)-C(28)	108.8(5)
C(7)-N(1)-C(8)	111.6(3)	C(7)-N(1)-C(6)	118.7(3)
C(8)-N(1)-C(6)	129.7(3)	C(7)-N(2)-C(9)	111.3(3)
C(7)-N(2)-C(17)	123.6(3)	C(9)-N(2)-C(17)	125.0(3)
C(10)-N(3)-C(11)	111.9(3)	C(10)-N(3)-C(2)	119.1(3)
C(11)-N(3)-C(2)	129.0(3)	C(10)-N(4)-C(12)	112.5(3)
C(10)-N(4)-C(13)	124.1(3)	C(12)-N(4)-C(13)	123.3(3)
C(22)-N(5)-C(21)	107.0(3)	C(22)-N(5)-Zr(1)	132.8(3)
C(21)-N(5)-Zr(1)	120.2(2)	C(23)-N(6)-C(24)	110.2(3)
C(23)-N(6)-Zr(1)	138.0(3)	C(24)-N(6)-Zr(1)	111.7(2)
C(2)-C(1)-C(6)	115.0(3)	C(2)-C(1)-Zr(1)	122.8(3)
C(6)-C(1)-Zr(1)	122.1(2)	C(1)-C(2)-C(3)	124.4(3)
C(1)-C(2)-N(3)	113.0(3)	C(3)-C(2)-N(3)	122.6(3)
C(4)-C(3)-C(2)	117.1(3)	C(4)-C(3)-H(3A)	121.5
C(2)-C(3)-H(3A)	121.5	C(3)-C(4)-C(5)	121.7(4)
C(3)-C(4)-H(4A)	119.2	C(5)-C(4)-H(4A)	119.2
C(6)-C(5)-C(4)	117.4(4)	C(6)-C(5)-H(5A)	121.3
C(4)-C(5)-H(5A)	121.3	C(5)-C(6)-C(1)	124.3(3)
C(5)-C(6)-N(1)	122.9(3)	C(1)-C(6)-N(1)	112.7(3)
N(2)-C(7)-N(1)	103.8(3)	N(2)-C(7)-Zr(1)	137.9(3)
N(1)-C(7)-Zr(1)	118.2(2)	C(9)-C(8)-N(1)	106.2(3)
C(9)-C(8)-H(8A)	126.9	N(1)-C(8)-H(8A)	126.9
C(8)-C(9)-N(2)	107.0(3)	C(8)-C(9)-H(9A)	126.5
N(2)-C(9)-H(9A)	126.5	N(4)-C(10)-N(3)	102.8(3)
N(4)-C(10)-Zr(1)	139.9(3)	N(3)-C(10)-Zr(1)	117.3(2)
C(12)-C(11)-N(3)	106.4(3)	C(12)-C(11)-H(11A)	126.8
N(3)-C(11)-H(11A)	126.8	C(11)-C(12)-N(4)	106.5(3)
C(11)-C(12)-H(12A)	126.8	N(4)-C(12)-H(12A)	126.8
N(4)-C(13)-C(14)	112.0(3)	N(4)-C(13)-H(13A)	109.2
C(14)-C(13)-H(13A)	109.2	N(4)-C(13)-H(13B)	109.2
C(14)-C(13)-H(13B)	109.2	H(13A)-C(13)-H(13B)	107.9
C(13)-C(14)-C(15)	111.9(3)	C(13)-C(14)-H(14A)	109.2
C(15)-C(14)-H(14A)	109.2	C(13)-C(14)-H(14B)	109.2
C(15)-C(14)-H(14B)	109.2	H(14A)-C(14)-H(14B)	107.9
C(16)-C(15)-C(14)	113.6(4)	C(16)-C(15)-H(15A)	108.8
C(14)-C(15)-H(15A)	108.8	C(16)-C(15)-H(15B)	108.8
C(14)-C(15)-H(15B)	108.8	H(15A)-C(15)-H(15B)	107.7
C(15)-C(16)-H(16A)	109.5	C(15)-C(16)-H(16B)	109.5
H(16A)-C(16)-H(16B)	109.5	C(15)-C(16)-H(16C)	109.5
H(16A)-C(16)-H(16C)	109.5	H(16B)-C(16)-H(16C)	109.5
N(2)-C(17)-C(18)	112.5(3)	N(2)-C(17)-H(17A)	109.1
C(18)-C(17)-H(17A)	109.1	N(2)-C(17)-H(17B)	109.1
C(18)-C(17)-H(17B)	109.1	H(17A)-C(17)-H(17B)	107.8
C(17)-C(18)-C(19)	110.8(3)	C(17)-C(18)-H(18A)	109.5

C(19)-C(18)-H(18A)	109.5	C(17)-C(18)-H(18B)	109.5
C(19)-C(18)-H(18B)	109.5	H(18A)-C(18)-H(18B)	108.1
C(20)-C(19)-C(18)	112.9(4)	C(20)-C(19)-H(19A)	109.0
C(18)-C(19)-H(19A)	109.0	C(20)-C(19)-H(19B)	109.0
C(18)-C(19)-H(19B)	109.0	H(19A)-C(19)-H(19B)	107.8
C(19)-C(20)-H(20A)	109.5	C(19)-C(20)-H(20B)	109.5
H(20A)-C(20)-H(20B)	109.5	C(19)-C(20)-H(20C)	109.5
H(20A)-C(20)-H(20C)	109.5	H(20B)-C(20)-H(20C)	109.5
N(5)-C(21)-H(21A)	109.5	N(5)-C(21)-H(21B)	109.5
H(21A)-C(21)-H(21B)	109.5	N(5)-C(21)-H(21C)	109.5
H(21A)-C(21)-H(21C)	109.5	H(21B)-C(21)-H(21C)	109.5
N(5)-C(22)-H(22A)	109.5	N(5)-C(22)-H(22B)	109.5
H(22A)-C(22)-H(22B)	109.5	N(5)-C(22)-H(22C)	109.5
H(22A)-C(22)-H(22C)	109.5	H(22B)-C(22)-H(22C)	109.5
N(6)-C(23)-H(23A)	109.5	N(6)-C(23)-H(23B)	109.5
H(23A)-C(23)-H(23B)	109.5	N(6)-C(23)-H(23C)	109.5
H(23A)-C(23)-H(23C)	109.5	H(23B)-C(23)-H(23C)	109.5
N(6)-C(24)-H(24A)	109.5	N(6)-C(24)-H(24B)	109.5
H(24A)-C(24)-H(24B)	109.5	N(6)-C(24)-H(24C)	109.5
H(24A)-C(24)-H(24C)	109.5	H(24B)-C(24)-H(24C)	109.5
O(1)-C(25)-C(26)	107.2(6)	O(1)-C(25)-H(25A)	110.3
C(26)-C(25)-H(25A)	110.3	O(1)-C(25)-H(25B)	110.3
C(26)-C(25)-H(25B)	110.3	H(25A)-C(25)-H(25B)	108.5
C(27)-C(26)-C(25)	104.7(6)	C(27)-C(26)-H(26A)	110.8
C(25)-C(26)-H(26A)	110.8	C(27)-C(26)-H(26B)	110.8
C(25)-C(26)-H(26B)	110.8	H(26A)-C(26)-H(26B)	108.9
C(26)-C(27)-C(28)	101.4(6)	C(26)-C(27)-H(27A)	111.5
C(28)-C(27)-H(27A)	111.5	C(26)-C(27)-H(27B)	111.5
C(28)-C(27)-H(27B)	111.5	H(27A)-C(27)-H(27B)	109.3
O(1)-C(28)-C(27)	106.3(6)	O(1)-C(28)-H(28A)	110.5
C(27)-C(28)-H(28A)	110.5	O(1)-C(28)-H(28B)	110.5
C(27)-C(28)-H(28B)	110.5	H(28A)-C(28)-H(28B)	108.7

Symmetry transformations used to generate equivalent atoms:

Table S12. Torsion angles [°] for **4**.

atom-atom-atom-atom	angle	atom-atom-atom-atom	angle
C(6)-C(1)-C(2)-C(3)	1.2(5)	Zr(1)-C(1)-C(2)-C(3)	176.8(3)
C(6)-C(1)-C(2)-N(3)	-178.0(3)	Zr(1)-C(1)-C(2)-N(3)	-2.4(4)
C(10)-N(3)-C(2)-C(1)	1.9(4)	C(11)-N(3)-C(2)-C(1)	180.0(3)
C(10)-N(3)-C(2)-C(3)	-177.3(3)	C(11)-N(3)-C(2)-C(3)	0.8(6)
C(1)-C(2)-C(3)-C(4)	-1.2(6)	N(3)-C(2)-C(3)-C(4)	177.9(3)
C(2)-C(3)-C(4)-C(5)	-0.4(6)	C(3)-C(4)-C(5)-C(6)	1.9(6)
C(4)-C(5)-C(6)-C(1)	-2.0(6)	C(4)-C(5)-C(6)-N(1)	179.6(3)
C(2)-C(1)-C(6)-C(5)	0.5(5)	Zr(1)-C(1)-C(6)-C(5)	-175.2(3)
C(2)-C(1)-C(6)-N(1)	179.0(3)	Zr(1)-C(1)-C(6)-N(1)	3.4(4)
C(7)-N(1)-C(6)-C(5)	178.4(3)	C(8)-N(1)-C(6)-C(5)	-1.8(6)
C(7)-N(1)-C(6)-C(1)	-0.1(5)	C(8)-N(1)-C(6)-C(1)	179.7(3)
C(9)-N(2)-C(7)-N(1)	0.2(4)	C(17)-N(2)-C(7)-N(1)	177.5(3)
C(9)-N(2)-C(7)-Zr(1)	-176.4(3)	C(17)-N(2)-C(7)-Zr(1)	0.9(6)
C(8)-N(1)-C(7)-N(2)	-0.2(4)	C(6)-N(1)-C(7)-N(2)	179.6(3)
C(8)-N(1)-C(7)-Zr(1)	177.2(2)	C(6)-N(1)-C(7)-Zr(1)	-3.0(4)
C(7)-N(1)-C(8)-C(9)	0.1(4)	C(6)-N(1)-C(8)-C(9)	-179.7(3)
N(1)-C(8)-C(9)-N(2)	0.0(4)	C(7)-N(2)-C(9)-C(8)	-0.1(4)
C(17)-N(2)-C(9)-C(8)	-177.4(3)	C(12)-N(4)-C(10)-N(3)	0.1(4)
C(13)-N(4)-C(10)-N(3)	176.9(3)	C(12)-N(4)-C(10)-Zr(1)	178.8(3)
C(13)-N(4)-C(10)-Zr(1)	-4.3(6)	C(11)-N(3)-C(10)-N(4)	0.0(4)
C(2)-N(3)-C(10)-N(4)	178.4(3)	C(11)-N(3)-C(10)-Zr(1)	-179.1(2)
C(2)-N(3)-C(10)-Zr(1)	-0.7(4)	C(10)-N(3)-C(11)-C(12)	0.0(4)
C(2)-N(3)-C(11)-C(12)	-178.2(3)	N(3)-C(11)-C(12)-N(4)	0.1(4)
C(10)-N(4)-C(12)-C(11)	-0.1(4)	C(13)-N(4)-C(12)-C(11)	-177.0(3)
C(10)-N(4)-C(13)-C(14)	-78.7(4)	C(12)-N(4)-C(13)-C(14)	97.8(4)
N(4)-C(13)-C(14)-C(15)	177.8(3)	C(13)-C(14)-C(15)-C(16)	76.1(5)
C(7)-N(2)-C(17)-C(18)	-102.5(4)	C(9)-N(2)-C(17)-C(18)	74.4(4)
N(2)-C(17)-C(18)-C(19)	179.0(3)	C(17)-C(18)-C(19)-C(20)	174.8(4)
C(28)-O(1)-C(25)-C(26)	-4.4(9)	O(1)-C(25)-C(26)-C(27)	-17.2(9)
C(25)-C(26)-C(27)-C(28)	30.3(8)	C(25)-O(1)-C(28)-C(27)	24.3(10)
C(26)-C(27)-C(28)-O(1)	-33.8(9)		

Symmetry transformations used to generate equivalent atoms: