

## Supplementary Material

### Reactions of trivalent iodine reagents with classic iridium and rhodium complexes

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#### Experimental

##### *Materials and Methods*

NMR solvents were purchased from Cambridge Isotope Laboratories and dried by stirring for 3 days over CaH<sub>2</sub>, which was distilled and stored over 3 Å molecular sieves in the glovebox. CH<sub>2</sub>Cl<sub>2</sub>, CH<sub>3</sub>CN, n-hexane, toluene and chloroform, which were purchased from Caledon Laboratories, were dried using an Innovative Technologies solvent purification system. All solvents were stored over 3 Å molecular sieves in nitrogen filled glovebox. All remaining reagents were ordered from Sigma-Aldrich and used as received. [Ir(PPh<sub>3</sub>)<sub>2</sub>(CO)Cl],<sup>[1]</sup> [Rh(PPh<sub>3</sub>)<sub>3</sub>Cl],<sup>[2]</sup> [Rh(dppe)<sub>2</sub>]Cl,<sup>[3]</sup> [PhI(Pyrr)<sub>2</sub>][OTf]<sub>2</sub> and [PhI(4-DMAP)<sub>2</sub>][OTf]<sub>2</sub><sup>[4]</sup> were prepared following published procedures.

The metathesis to produce  $[\text{Rh}(\text{dppe})_2]\text{OTf}$  was carried out by the addition of 1:1 stoichiometric ratio of TMS-Triflate to  $\text{CH}_2\text{Cl}_2$  solution of  $[\text{Rh}(\text{dppe})_2]\text{Cl}$  followed by solvent removal at reduced pressure.

#### *X-ray Crystallography Details*

Single crystals were selected under n-paratone oil, mounted on nylon loops and placed into a cold stream (172 K) of  $\text{N}_2$  on an Oxford CCD diffractometer using  $\text{Mo K}\alpha$  radiation. Structure solution and refinement were performed using the SHELXTL suite of software.

#### *Reaction of 2OAc with 9.*

A solution of **2OAc** (42 mg, 0.13 mmol) in 2 mL  $\text{CDCl}_3$  was added drop wise to a solution of **9** (100 mg, 0.13 mmol) in 5 mL  $\text{CDCl}_3$  and stirred for 3 hours at room temperature. A color change from bright yellow to light yellow was observed within 10 minutes. The solvent was reduced to half under reduced pressure and followed by addition of 10 mL n-hexane, which resulted in precipitation of a pale yellow solid. The solid was filtered, washed with n-hexane (2 x 10mL) and dried under reduced pressure.  $^{31}\text{P}$  NMR (162 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm): 2.0 (s), -1.6 (s), -9.6 (s), -10.9 (s), -12.6, -13.7 (s). See Figure 2 for the  $^1\text{H}$  NMR spectrum. The ESI-MS of the identified compounds  $[\text{M}]^{n+}$  :  $m/z$  876.1  $[\text{Ir}(\text{PPh}_3)_2(\text{CO})(\text{Cl})_2(\text{OAc})]$ , 894.1  $[\text{Ir}(\text{PPh}_3)_2(\text{OAc})_3]$ .

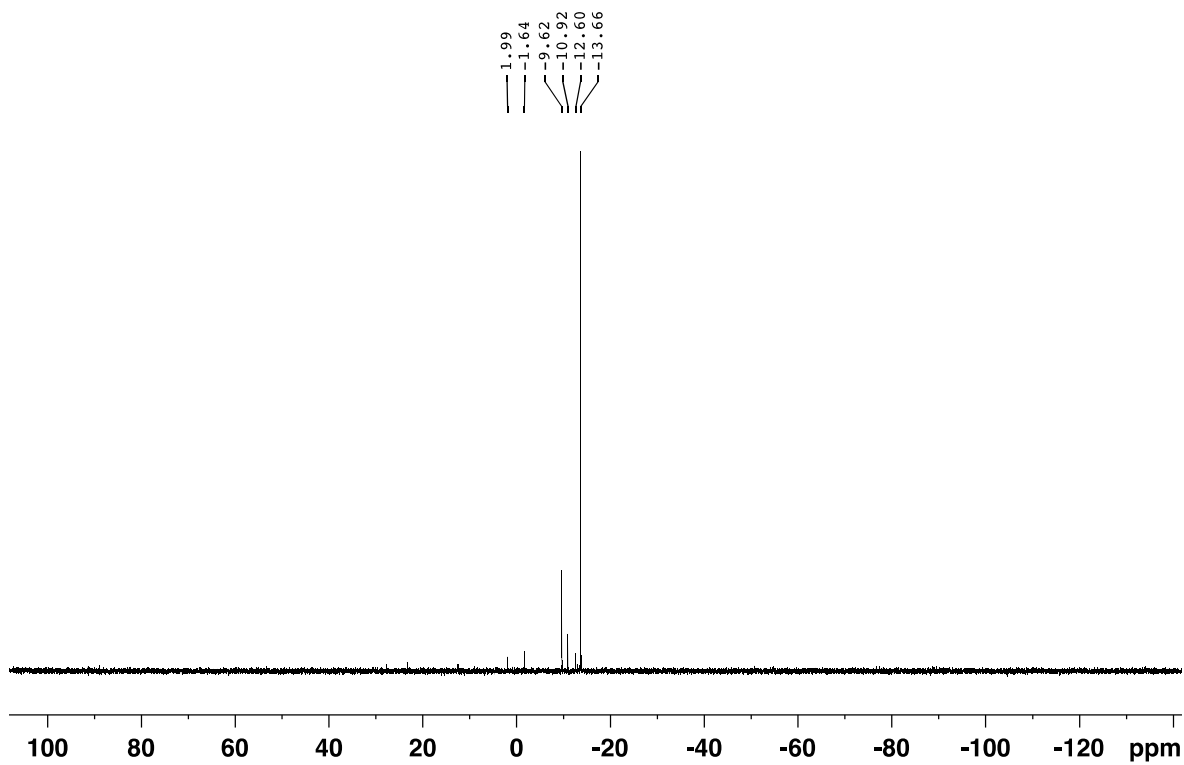


Figure 1. Reaction of **2OAc** with **9**  $^{31}\text{P}$  NMR.

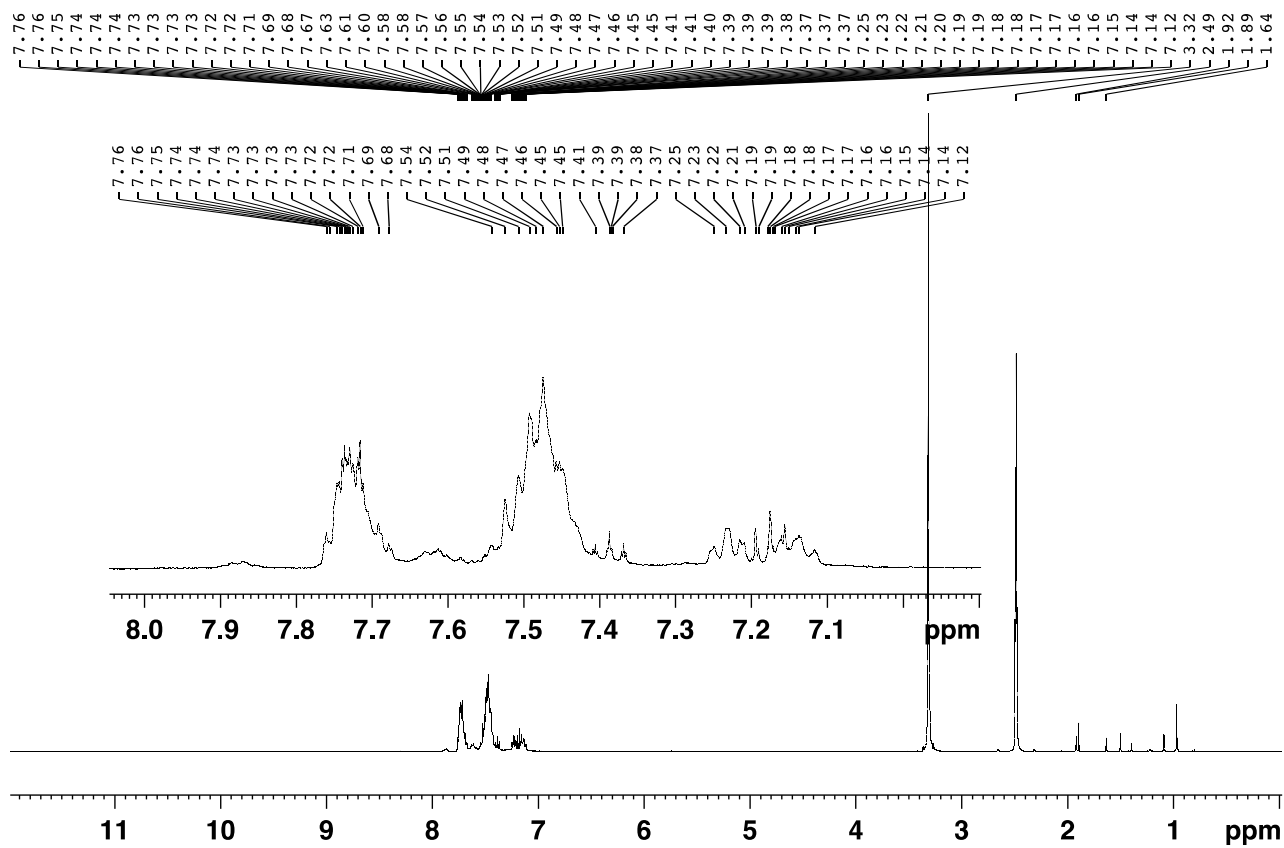


Figure 2. Reaction of **2OAc** with **9**  $^1\text{H}$  NMR.

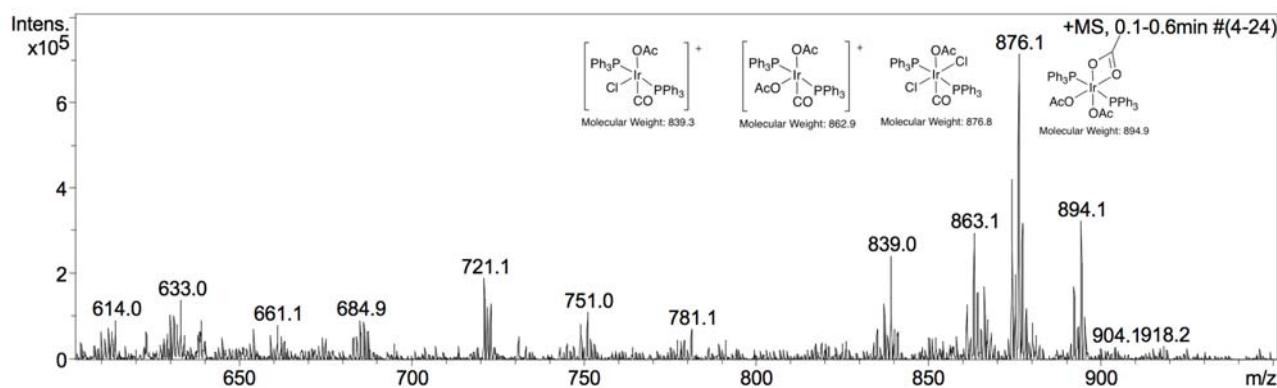


Figure 3. Reaction of **2OAc** with **9** ESI mass spectrum.

#### Reaction of **2R** with **9**.

A solution of **2R** (0.13 mmol) in 5 mL  $\text{CDCl}_3$  was added drop wise to a solution of **9** (100 mg, 0.13 mmol) in 5 mL  $\text{CDCl}_3$  and stirred for 3 hours at room temperature. A color change from bright yellow to light yellow was observed within 30 minutes. The solvent was reduced to half under reduced pressure and followed by addition of 10 mL n-hexane, which resulted in precipitation of a pale yellow solid. The solid was filtered, washed with n-hexane (2 x 10mL) and dried under reduced pressure. **R = NMe<sub>2</sub>**:  $^{31}\text{P}$  NMR (162 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm): -2.4 (s), -8.4 (s), -15.4 (s). See Figure 5 for the  $^1\text{H}$  NMR. ESI-MS  $[\text{M}]^{n+}$  :  $m/z$  797.1  $[\text{Ir}(\text{PPh}_3)(\text{CO})(\text{Cl})_2(\text{DMAP})_2]^+$ , 937.0  $[\text{Ir}(\text{PPh}_3)_2(\text{CO})(\text{Cl})_2(\text{DMAP})]^+$ . **R = H**:  $^{31}\text{P}$  NMR (162 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm): 65.9 (s), 26.9 (s), -3.9 (s), -14.6 (s), -15.2 (s), -16.4 (s), -23.1 (s). ESI-MS  $[\text{M}]^{n+}$  :  $m/z$  711.1  $[\text{Ir}(\text{PPh}_3)(\text{CO})(\text{Cl})_2(\text{Pyr})_2]^+$ , 894.1  $[\text{Ir}(\text{PPh}_3)_2(\text{CO})(\text{Cl})_2(\text{Pyr})]^+$ .

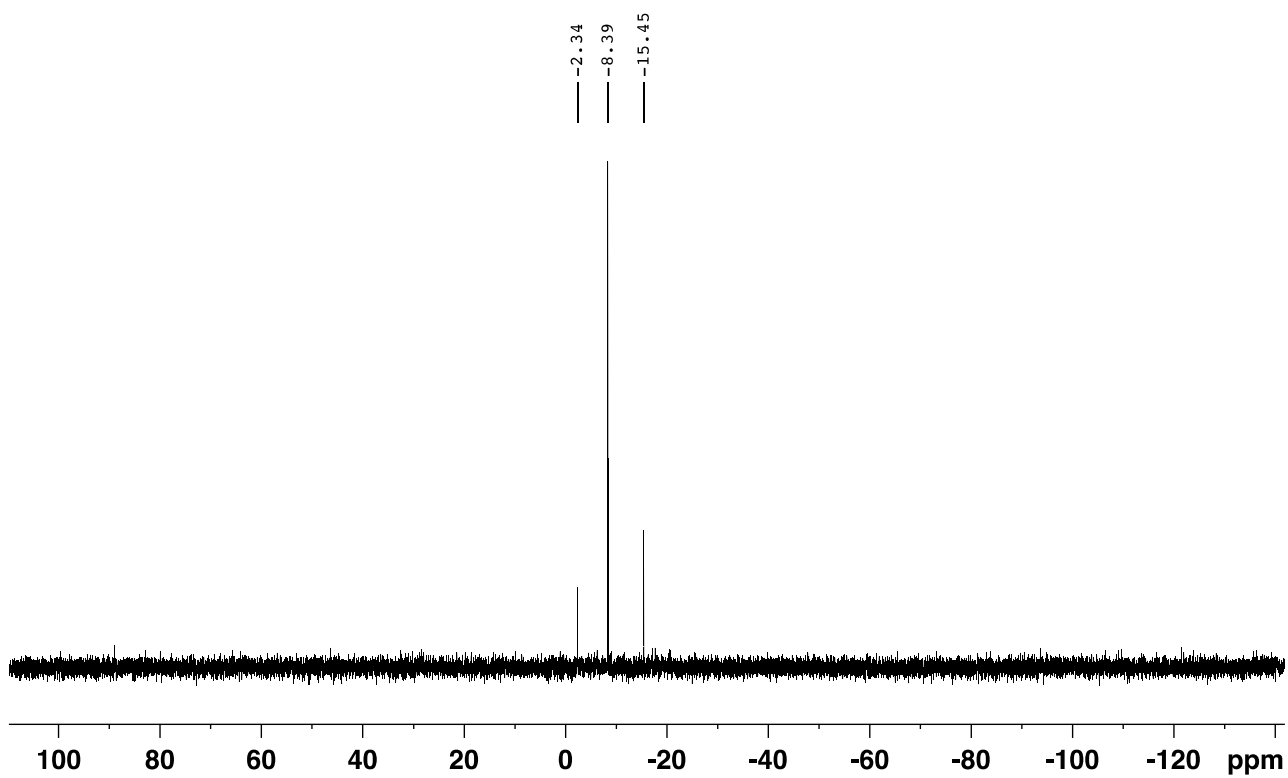


Figure 4. Reaction of  $2\text{NMe}_2$  with  $9$   $^{31}\text{P}$  NMR.

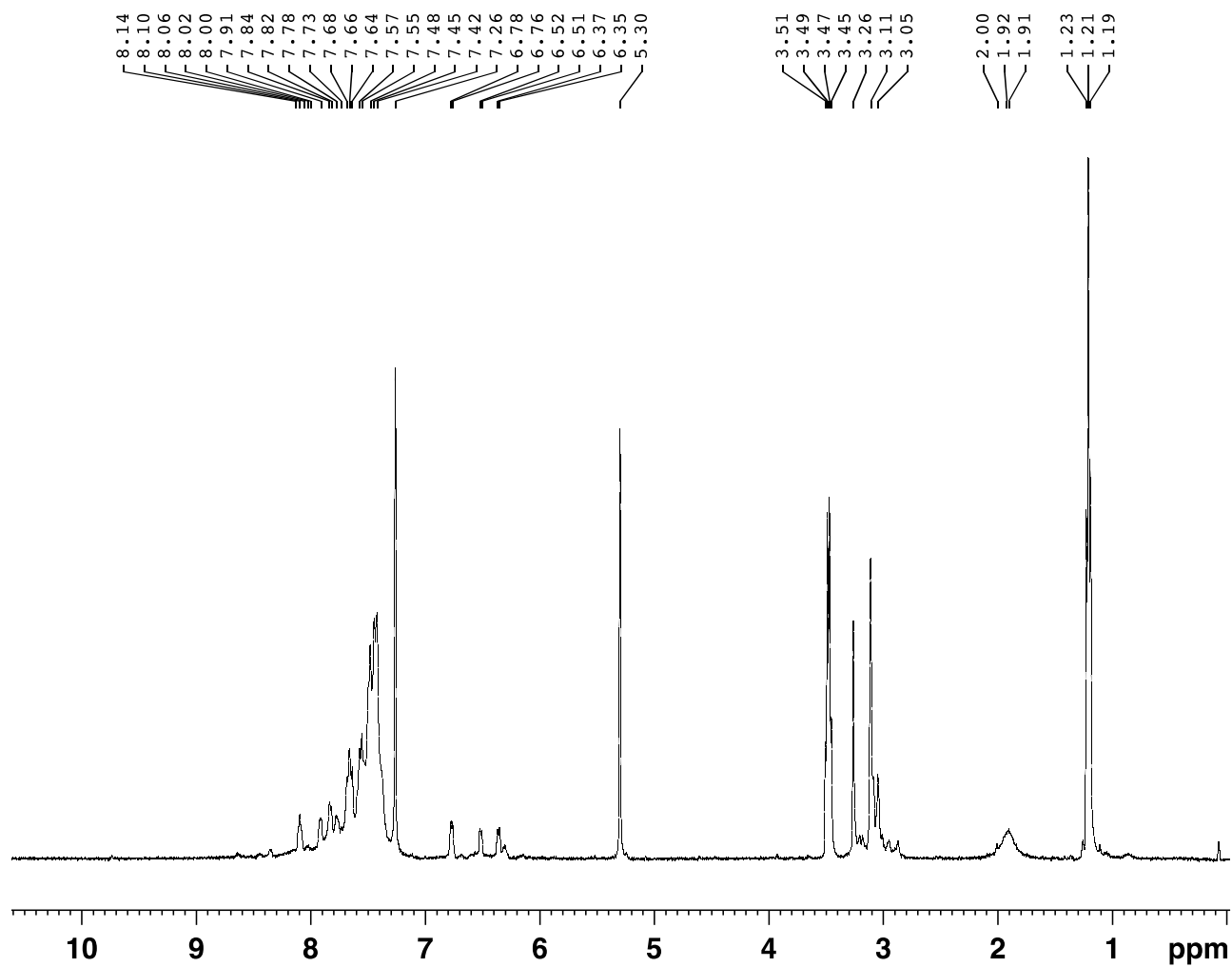


Figure 5. Reaction of 2NMe<sub>2</sub> with 9 <sup>1</sup>H NMR.

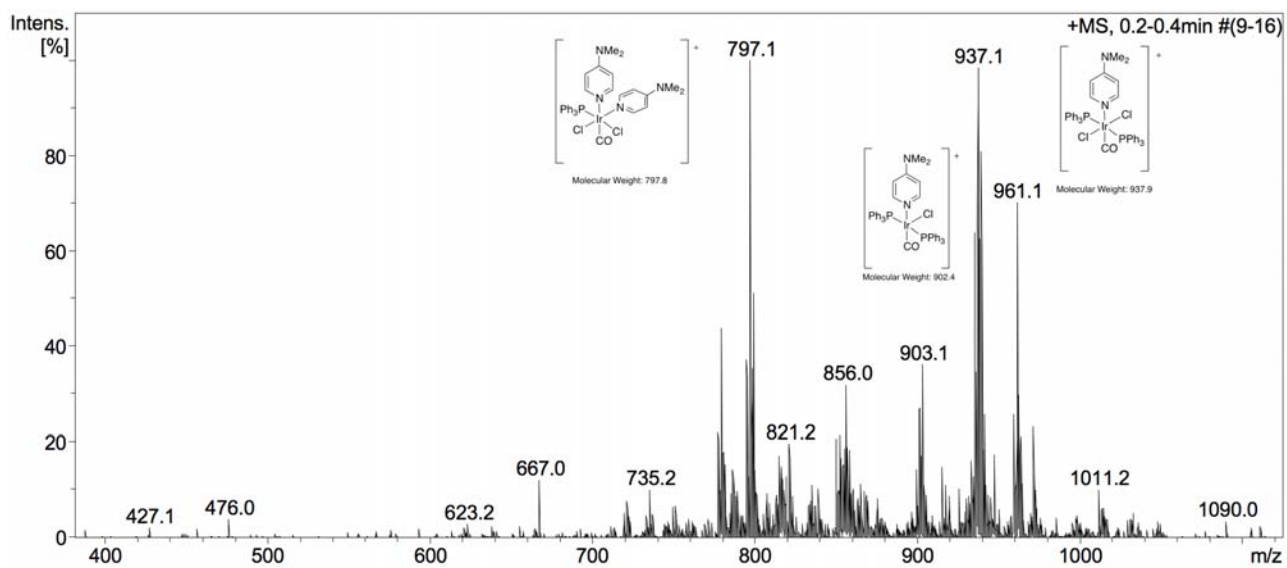


Figure 6. Reaction of 2NMe<sub>2</sub> with 9 ESI mass spectrum.

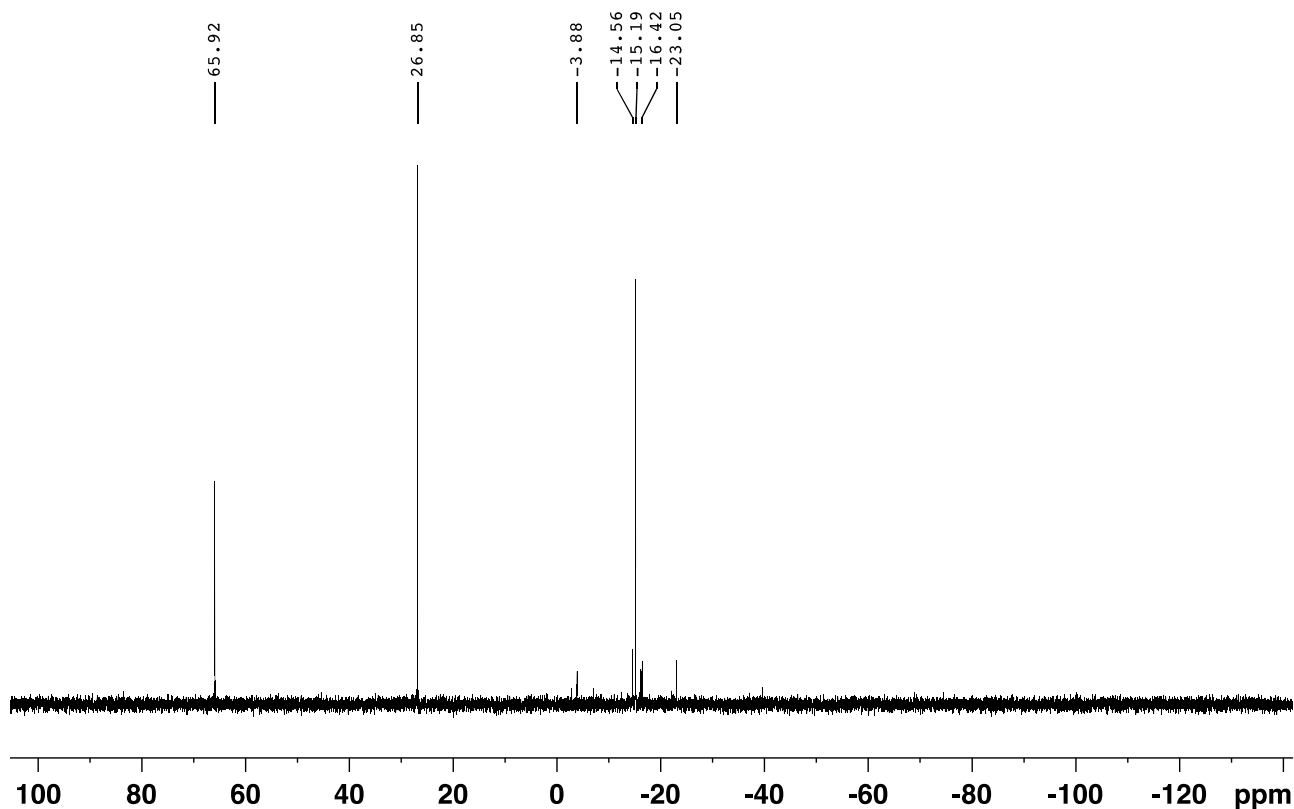


Figure 7. Reaction of **2H** with **9**  $^{31}\text{P}$  NMR.

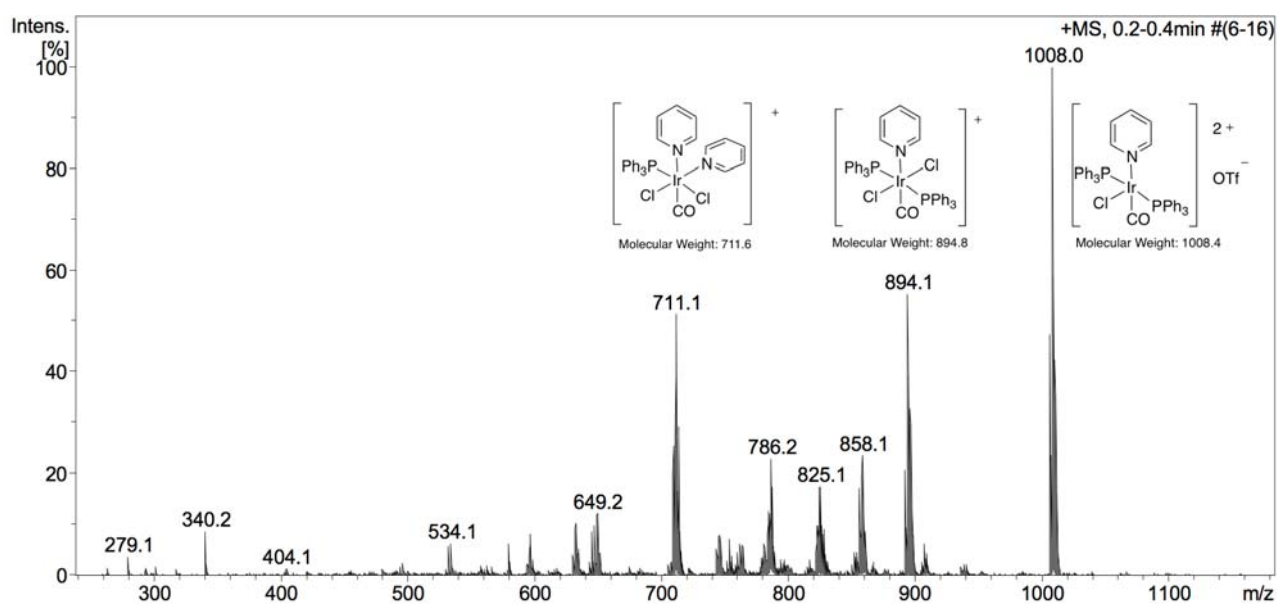


Figure 8. Reaction of **2H** with **9** ESI mass spectrum.

*Reaction of 2OAc.OTf with 9.*

A mixture of **2OAc** (20.6 mg, 0.064 mmol) and TMS-OTf (23.3  $\mu\text{L}$ , 0.128 mmol) in 2 mL  $\text{CDCl}_3$  was added drop wise to a solution of **9** (50 mg, 0.064 mmol) in 2 mL  $\text{CDCl}_3$ . A color change from

bright yellow to brown was observed in 10 min. Aliquot was removed for NMR and mass spectrometry analysis.  $^{31}\text{P}$  NMR (162 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm): 14.4 (s), 8.0 (s), 5.3 (s), 2.6 (s), 0.4 (s), -3.6 (s), -8.1 (s), -15.6 (s), -21.4 (s). ESI-MS  $[\text{M}]^{\text{n}+}$  :  $m/z$  745.2  $[\text{Ir}(\text{PPh}_3)_2\text{CO}]^+$ , 753.2  $[\text{Ir}(\text{PPh}_3)_2\text{Cl}]^+$ , 786.1  $[\text{Ir}(\text{PPh}_3)_2\text{Cl}_2]^+$ , 803.1  $[\text{Ir}(\text{PPh}_3)_2\text{CO}(\text{OAc})]^+$ , 839.1  $[\text{Ir}(\text{PPh}_3)_2\text{CO}(\text{OAc})\text{Cl}]^+$ .

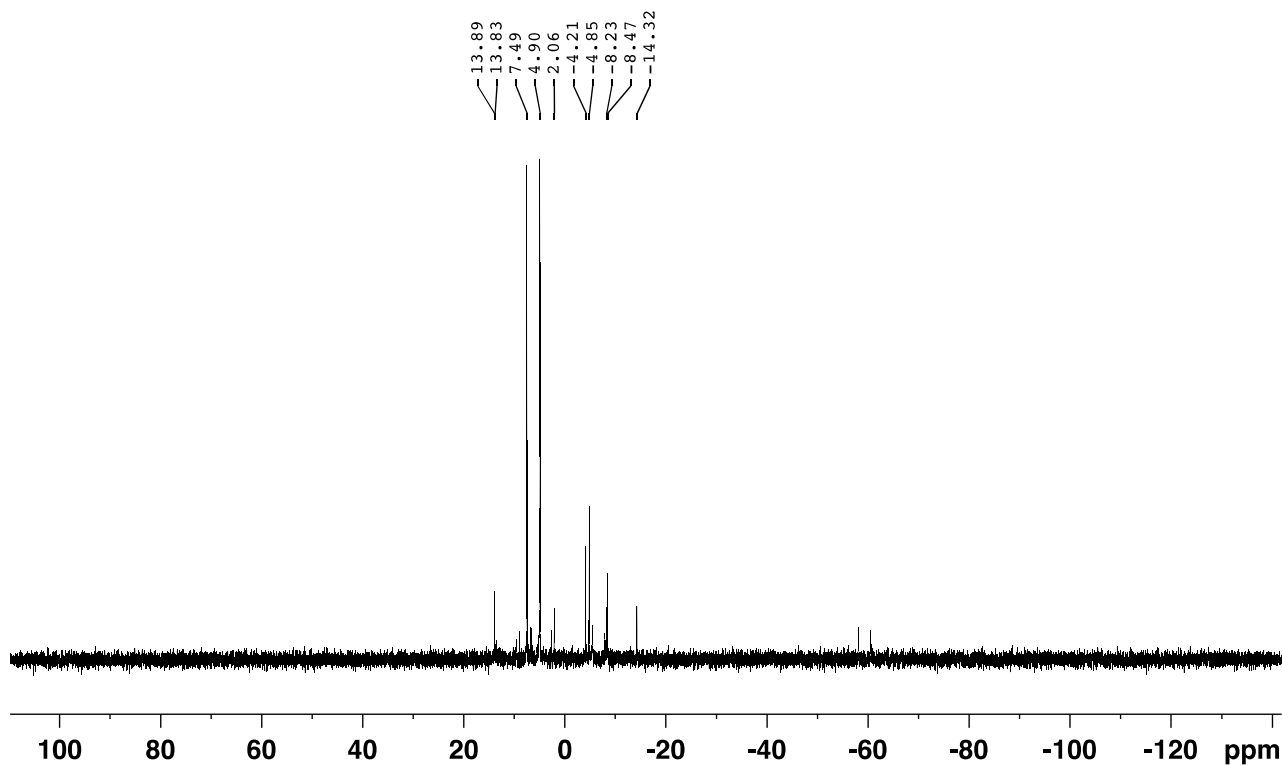


Figure 9. Reaction of **2OAc.OTf** with **9**  $^{31}\text{P}$  NMR.

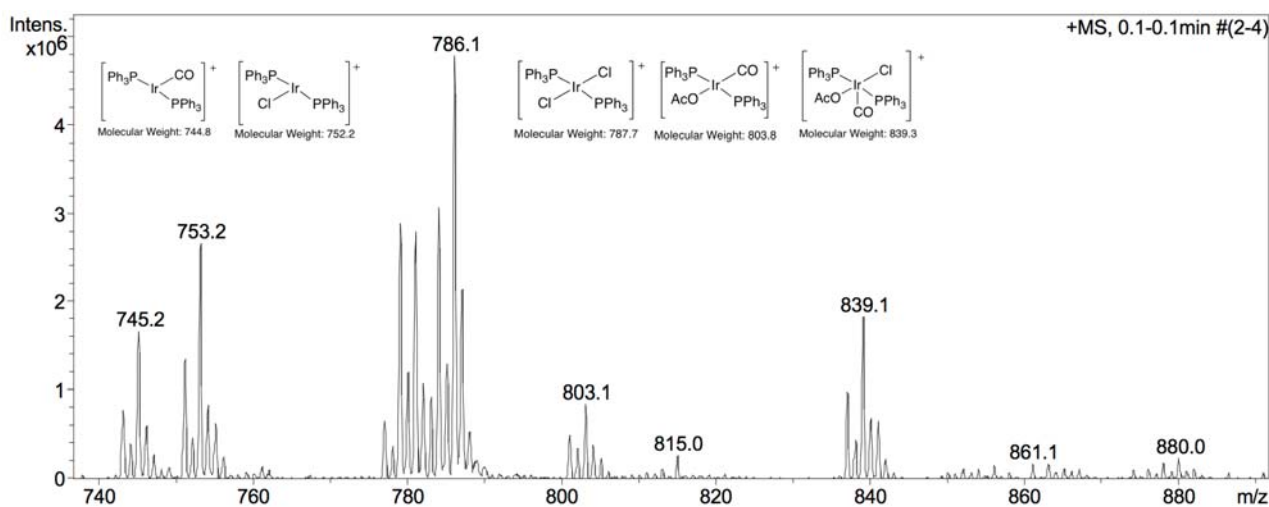


Figure 10. Reaction of **2OAc.OTf** with **9** ESI mass spectrum.



*Reaction of 2OAc with 10.*

A solution of **2OAc** (35 mg, 0.11 mmol) in 2 mL CDCl<sub>3</sub> was added drop wise to a solution of **10** (100 mg, 0.11 mmol) in 5 mL CDCl<sub>3</sub> and stirred for 3 hours at room temperature. A color change from burgundy to light red was observed. The solvent was reduced to half under reduced pressure and followed by addition of 10 mL n-hexane, which resulted in precipitation of a light orange solid. The solid was filtered, washed with n-hexane (2 x 10mL) and dried under reduced pressure. <sup>31</sup>P NMR (162 MHz, CDCl<sub>3</sub>) δ (ppm): 24.5 (d, J= 122 Hz). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ (ppm): 7.43-7.38 (m, 6H), 7.37-7.32 (m, 12H), 7.20-7.17 (m, 12H), 2.08 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ (ppm): 191.24, 135.61, 130.94, 129.28, 127.71, 25.70. ESI-MS [M]<sup>n+</sup> : m/z 721.1 [Rh(PPh<sub>3</sub>)<sub>2</sub>(Cl)(OAc)]<sup>+</sup>.

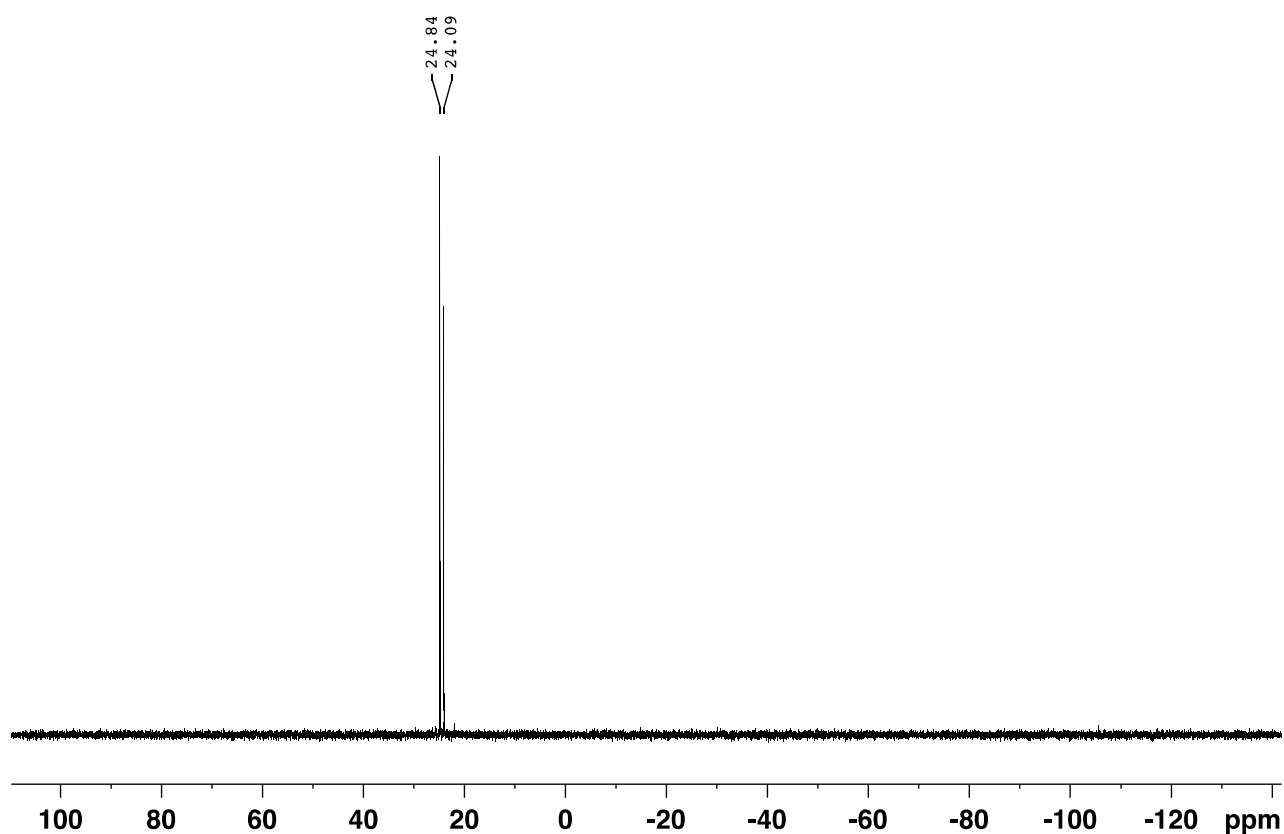


Figure 11. Isolated solid from the reaction of **2OAc** with **10** <sup>31</sup>P NMR.

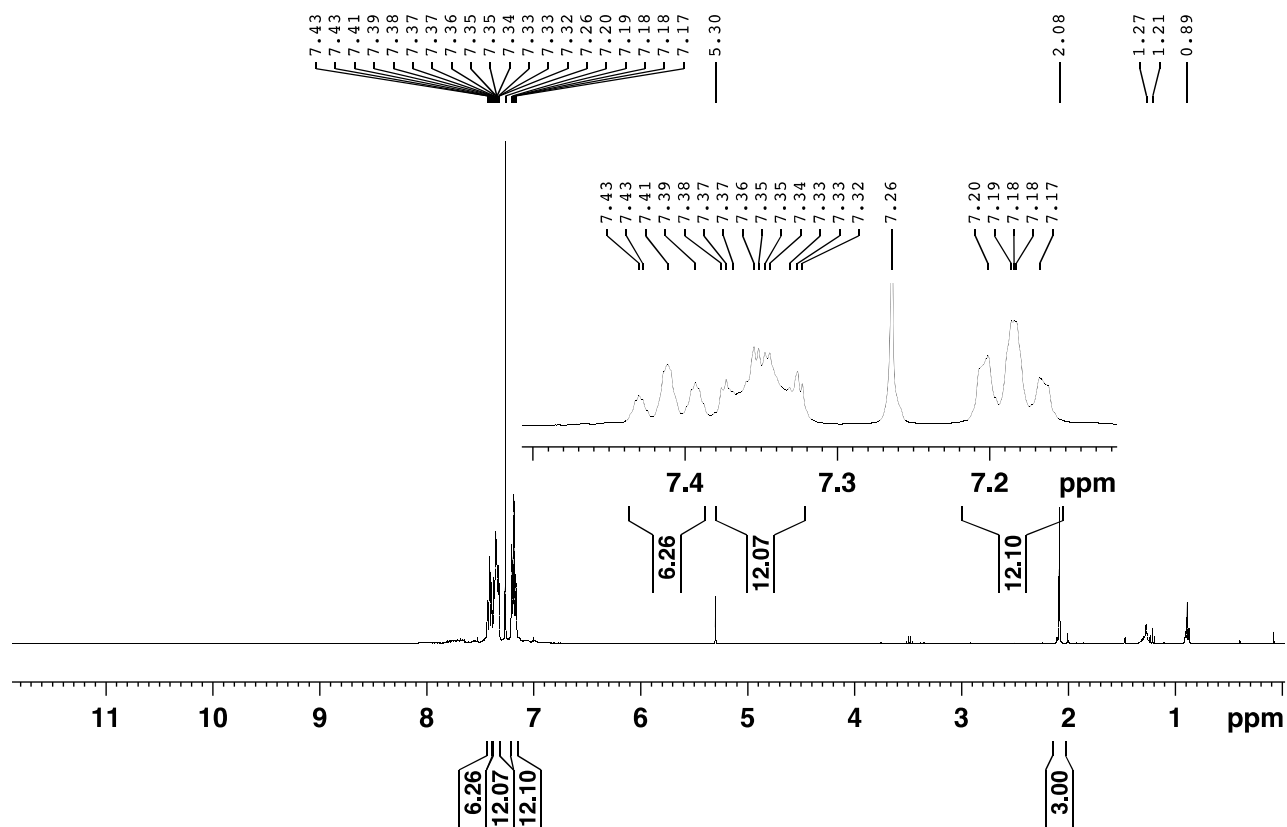


Figure 12. Reaction of **2OAc** with **10** <sup>1</sup>H NMR.

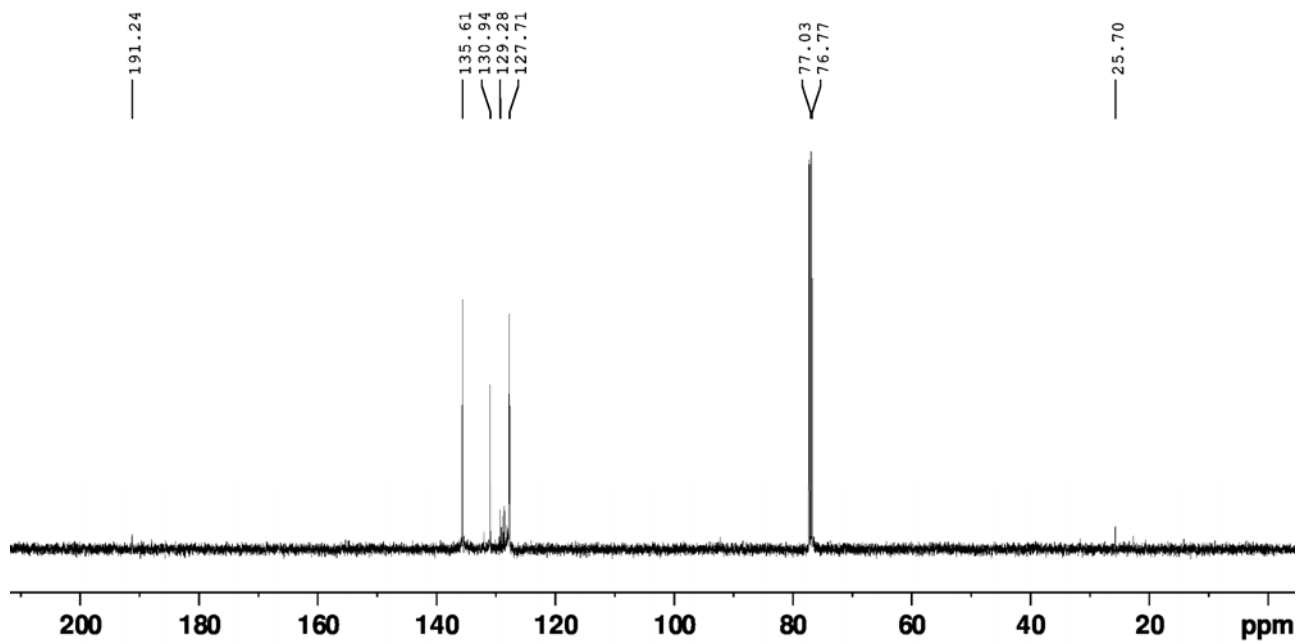


Figure 13. Reaction of **2OAc** with **10** <sup>13</sup>C NMR.

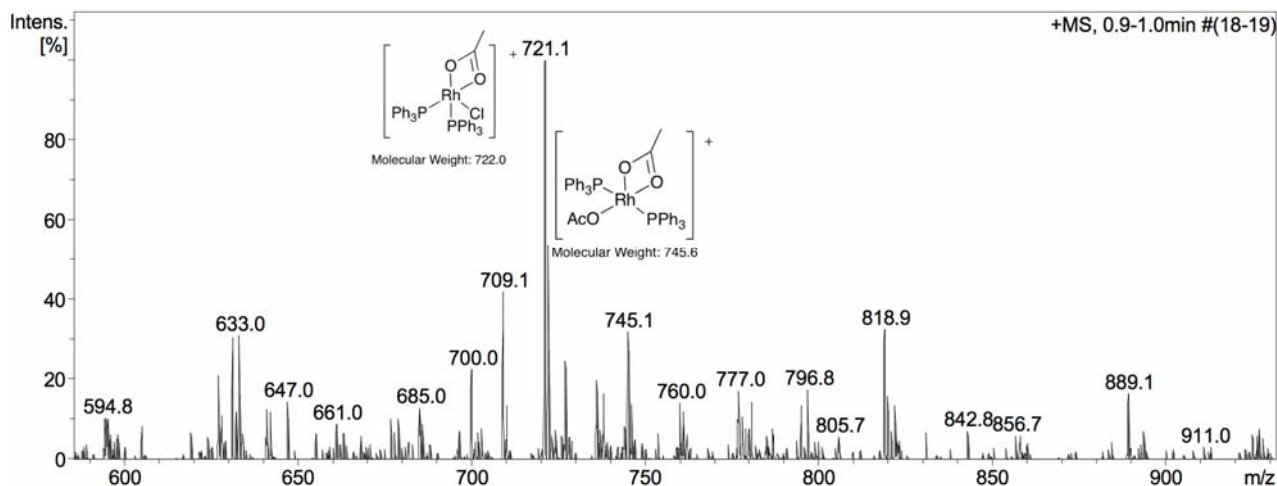


Figure 14. Reaction of **2OAc** with **10** ESI mass spectrum.

*Reaction of **2R** with Wilkinson's catalyst (**10**).*

A solution of **2R** (0.11 mmol) in 5 mL  $\text{CDCl}_3$  was added drop wise to a solution of **10** (100 mg, 0.11 mmol) in 5 mL  $\text{CDCl}_3$  and stirred for 3 hours at room temperature. A color change from burgundy to light red-brown was observed. The solvent was reduced to half under reduced pressure and followed by addition of 10 mL n-hexane, which resulted in precipitation of a light brown solid. The solid was filtered, washed with n-hexane (2 x 10mL) and dried under reduced pressure. **R = NMe<sub>2</sub>**:  $^{31}\text{P}$  NMR (162 MHz,  $\text{CH}_2\text{Cl}_2$ )  $\delta$  (ppm): 28.2 (s), 14.1 (d,  $J= 88 \text{ Hz}$ ), 11.9 (d,  $J=88 \text{ Hz}$ ). ESI-MS  $[\text{M}]^{n+}$ :  $m/z$  803.0  $[\text{Rh}(\text{PPh}_3)(\text{Cl})_2(\text{DMAP})_3]^+$ , 942.2  $[\text{Rh}(\text{PPh}_3)_2(\text{Cl})_2(\text{DMAP})_2]^+$ . **R = H**:  $^{31}\text{P}$  NMR (162 MHz,  $\text{CH}_2\text{Cl}_2$ )  $\delta$  (ppm): 65.0 (s), 14.0 (d,  $J= 84 \text{ Hz}$ ), 9.7 (d,  $J= 104 \text{ Hz}$ ). ESI-MS  $[\text{M}]^{n+}$ :  $m/z$  673.6  $[\text{Rh}(\text{PPh}_3)(\text{Cl})_2(\text{Pyr})_3]^+$ , 854.9  $[\text{Rh}(\text{PPh}_3)_2(\text{Cl})_2(\text{Pyr})_2]^+$ .

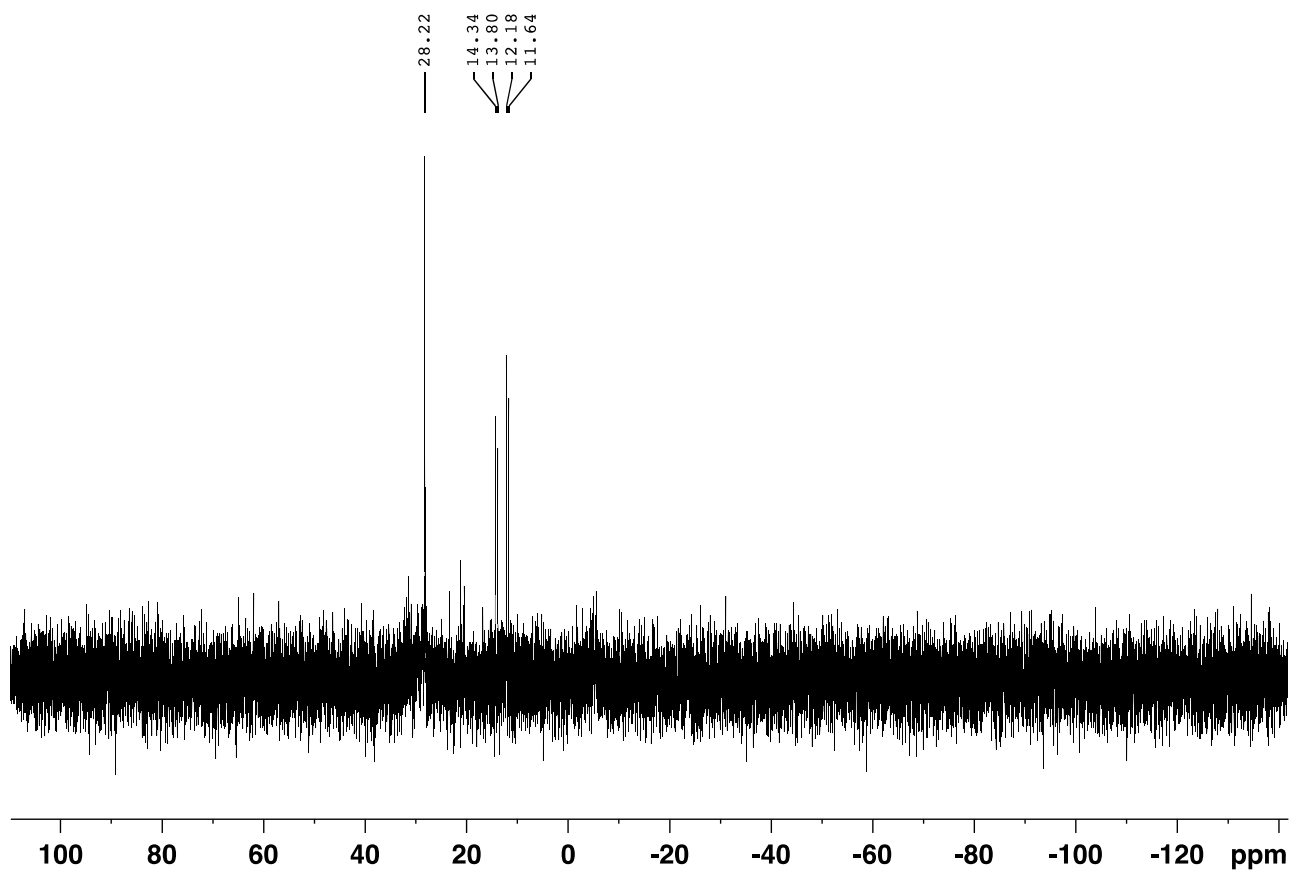


Figure 15. Reaction of  $2\text{NMe}_2$  with **10**  $^{31}\text{P}$  NMR.

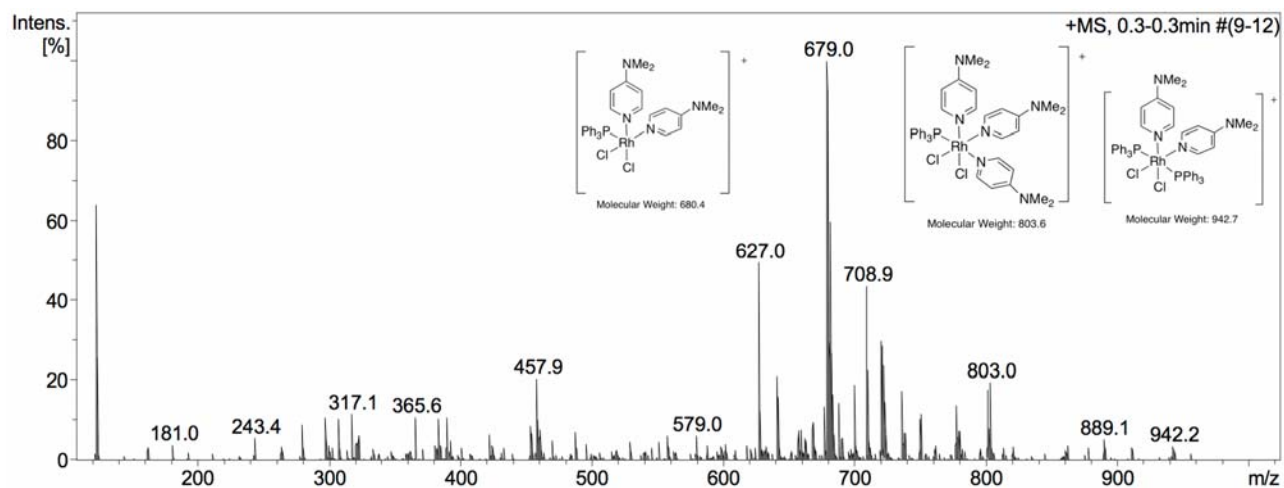


Figure 16. Reaction of  $2\text{NMe}_2$  with **10** ESI mass spectrum.

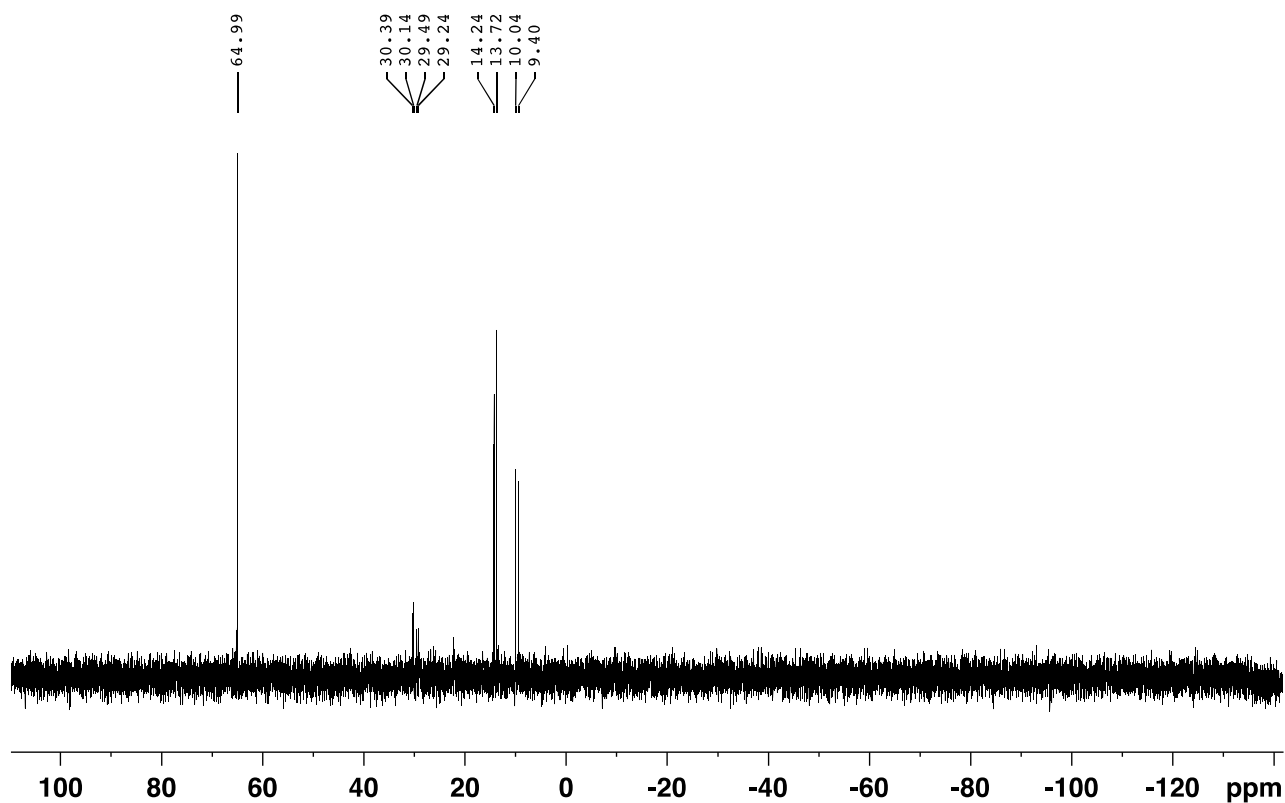


Figure 17. Reaction of **2H** with **10**  $^{31}\text{P}$  NMR.

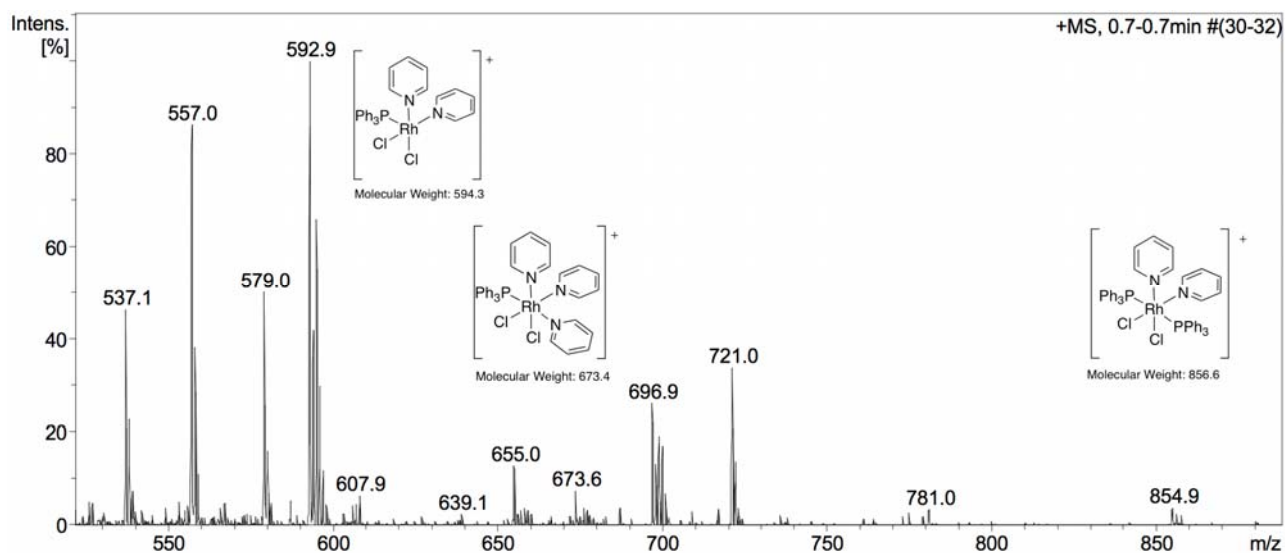


Figure 18. Reaction of **2H** with **10** ESI mass spectrum.

#### Reaction of **2OAc.OTf** with **10**.

A mixture of **2OAc** (17.4 mg, 0.054 mmol) and TMS-OTf (19.7  $\mu\text{L}$ , 0.108 mmol) in 2 mL  $\text{CDCl}_3$  was added drop wise to a solution of **10** (50 mg, 0.054 mmol) in 2 mL  $\text{CDCl}_3$ . A color change from

burgundy to brown was observed in 10 min. Aliquot was removed for NMR and mass spectrometry analysis.  $^{31}\text{P}$  NMR (162 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm): 61.9 (s), 45.2 (dt,  $J=135\text{ Hz}$ ), 23.4 (s), 19.8 (dt,  $J=100\text{ Hz}$ ). ESI-MS  $[\text{M}]^{n+}$  :  $m/z$  297.1  $[\text{PPh}_3\text{-Cl}]^+$ , 307.1  $[\text{Rh-I-Ph}]^+$ , 406.0  $[\text{Rh-PPh}_3\text{-NCCH}_3]^+$ , 477.0  $[\text{Rh-PPh}_3\text{-Cl}_2\text{-NCCH}_3]^+$ , 568.9  $[\text{PPh}_3\text{-Rh-I-Ph}]^+$ , 627.0  $[\text{Rh}(\text{PPh}_3)_2]^+$ , 697.0  $[\text{Rh}(\text{PPh}_3)_2\text{Cl}_2]^+$ .

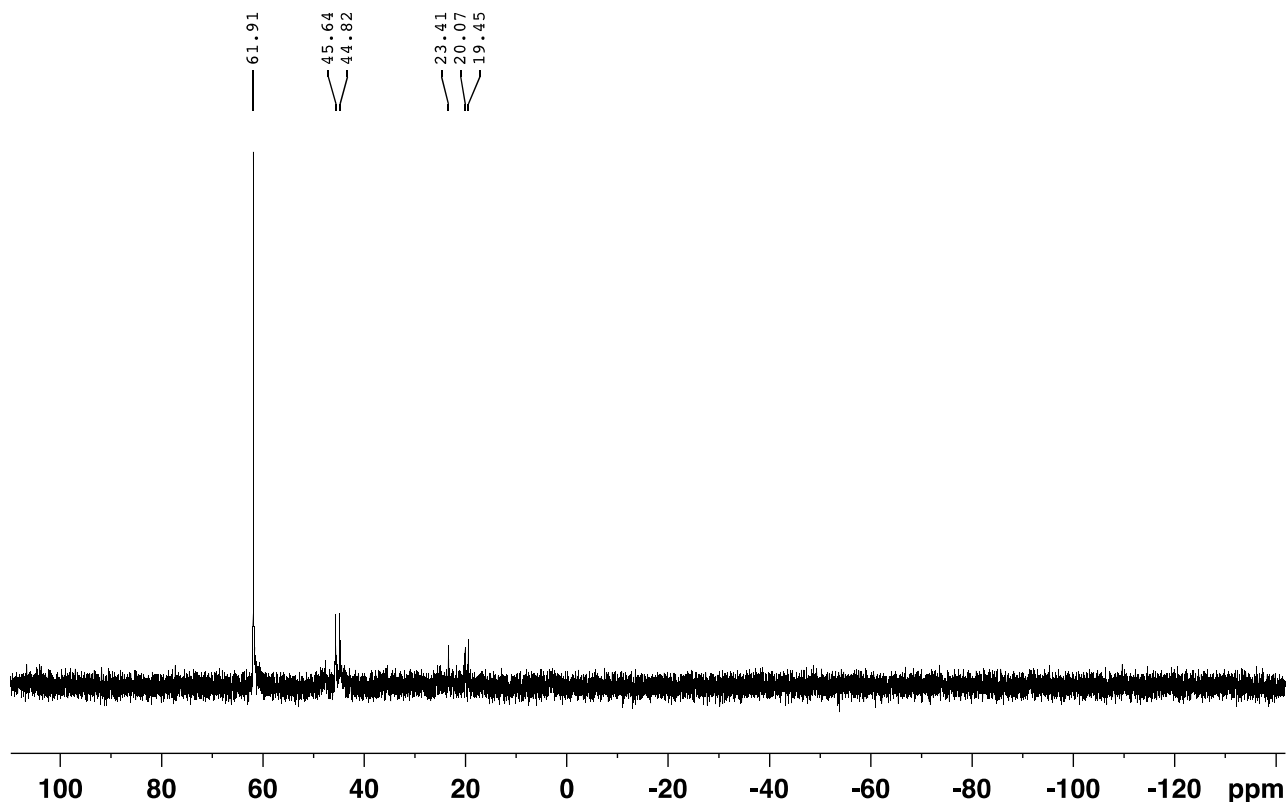


Figure 19. Reaction of **2OAc.OTf** with **10**  $^{31}\text{P}$  NMR.

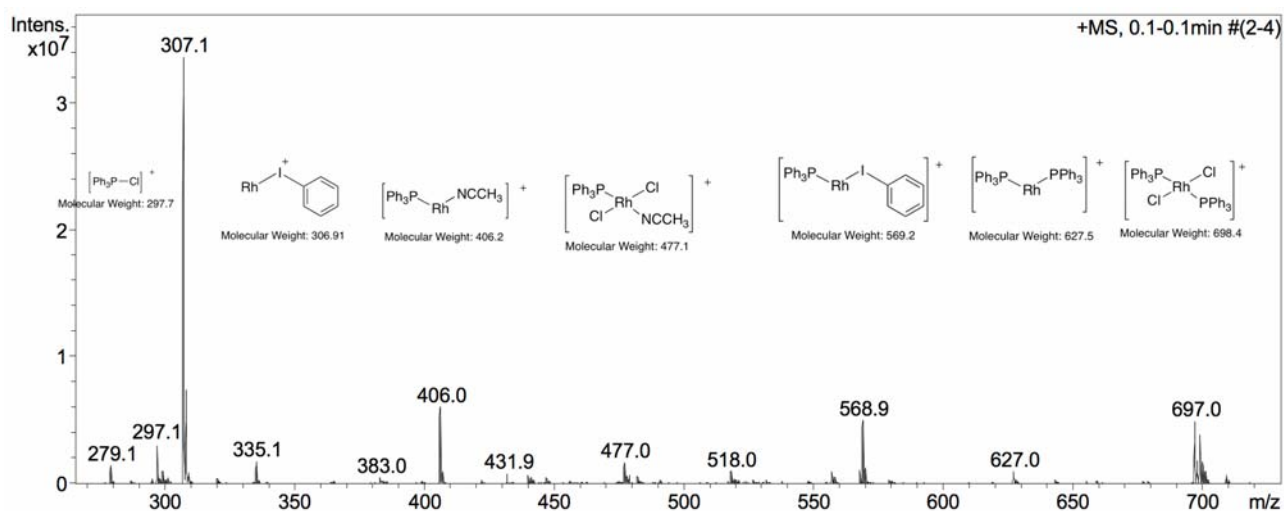


Figure 20. Reaction of **2OAc.OTf** with **10** ESI mass spectrum.

*Reaction of 2OAc.OTf with 11.*

A mixture of **2OAc** (31 mg, 0.095 mmol) and TMS-OTf (35  $\mu$ L, 0.19 mmol) in 5 mL CH<sub>2</sub>Cl<sub>2</sub> was added drop wise to a solution of **11** (100 mg, 0.095 mmol) in 5 mL CH<sub>2</sub>Cl<sub>2</sub> and stirred for one hour at room temperature. A color change from bright yellow to yellow was observed within 5 minutes. The solvent was reduced to half under reduced pressure and followed by addition of 10 mL n-hexane, which resulted in precipitation of a yellow solid. The solid was filtered, washed with n-hexane (2 x 10mL) and dried under reduced pressure (84 mg, 70% yield). <sup>31</sup>P NMR (162 MHz, CH<sub>2</sub>Cl<sub>2</sub>):  $\delta$  (ppm) 58.6 (dt, J= 11, 84 Hz), 42.5 (dt, J= 11, 115 Hz). <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>CN)  $\delta$  (ppm): 7.81-7.76 (m, 8H), 7.68-7.60 (m, 8H), 7.58-7.56 (m, 12H), 7.39-7.34 (m, 4H), 7.24-7.20 (m, 4H), 7.18-7.15 (m, 4H), 2.68-2.56 (m, 8H), 1.96 (s, 3H). <sup>13</sup>C NMR (100 MHz, CD<sub>3</sub>CN):  $\delta$  (ppm): 172.07, 134.55, 134.08, 133.55, 133.25, 130.45, 130.24, 126.05, 125.51, 24.37, 20.22, 16.68. ESI-MS [M]<sup>nt</sup> :  $m/z$  479.1 [Rh(dppe)<sub>2</sub>(OAc)]<sup>2+</sup>.

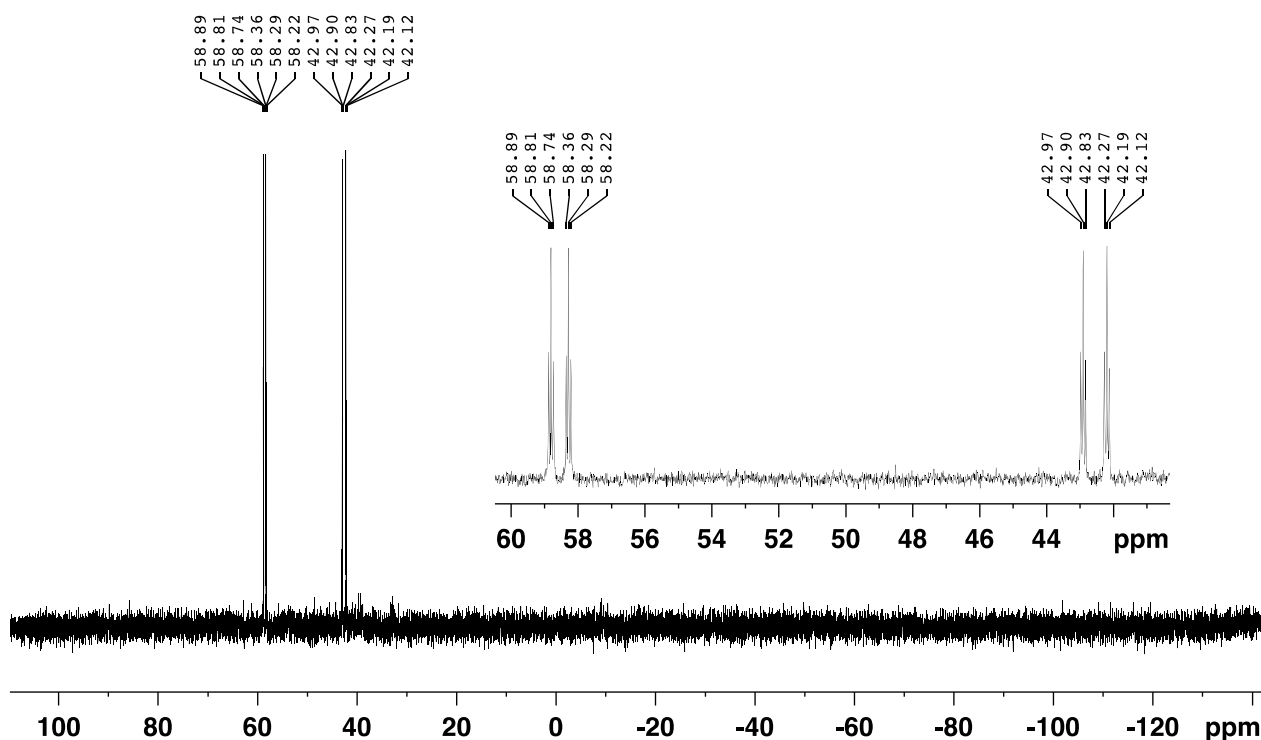


Figure 21. Reaction of **2OAc.OTf** with **11** <sup>31</sup>P NMR.

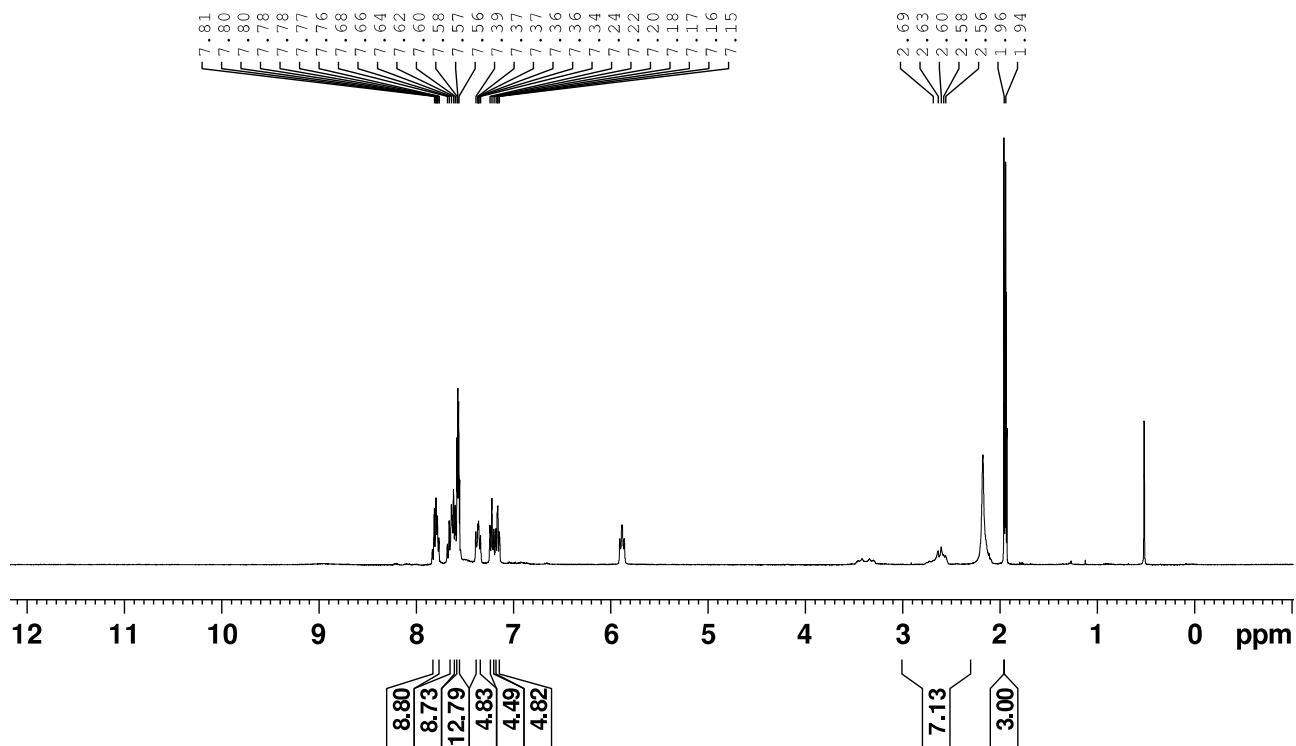


Figure 22. Reaction of **2OAc** with **11**  $^1\text{H}$  NMR.

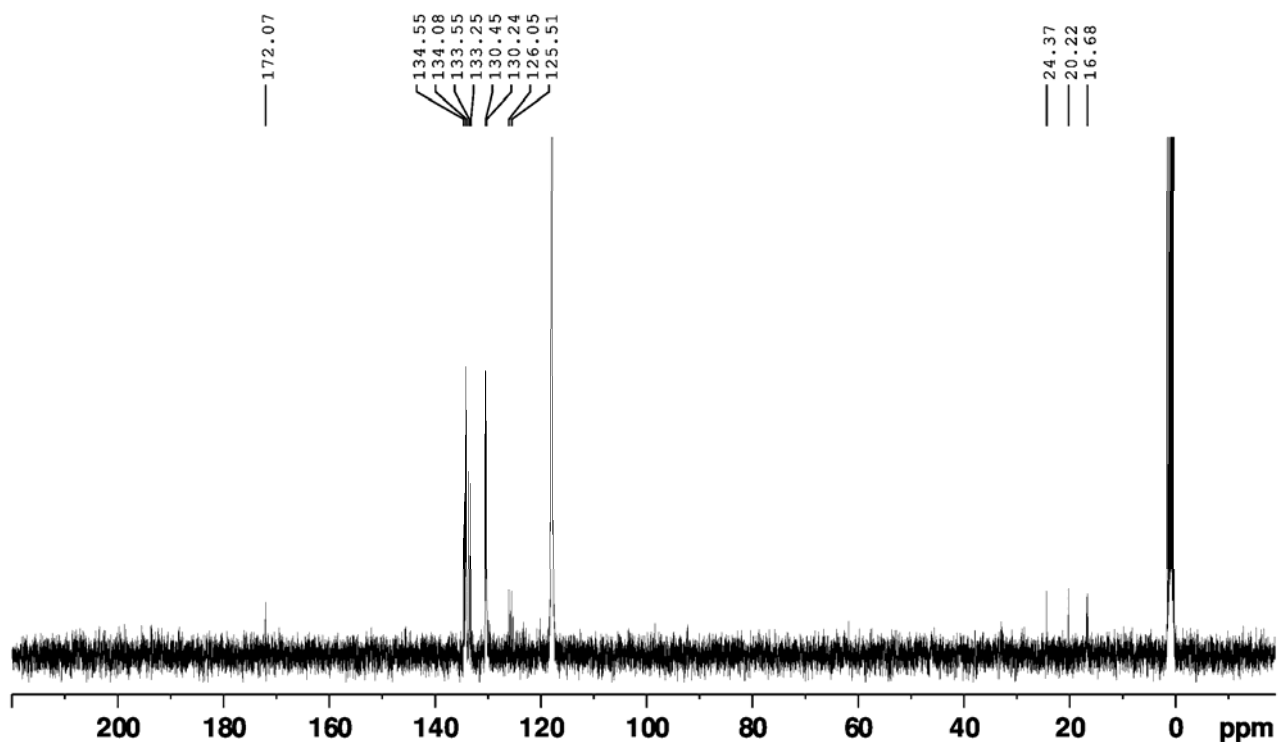


Figure 23. Reaction of **2OAc** with **11**  $^{13}\text{C}$  NMR.



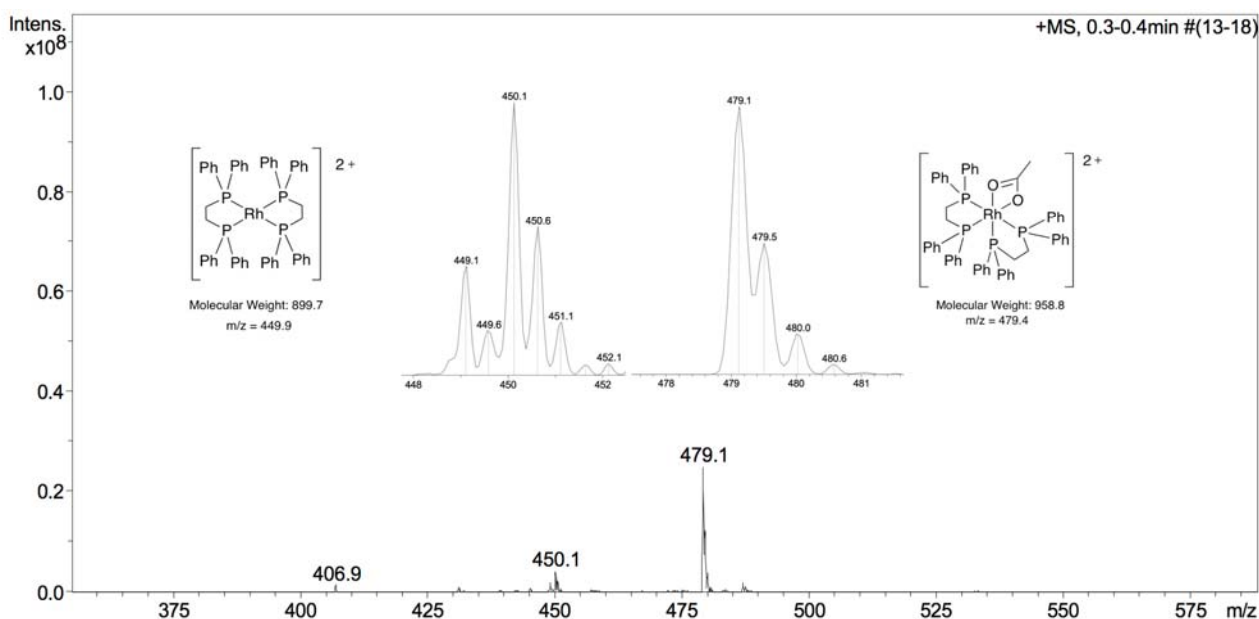


Figure 24. Reaction of **2OAc** with **11** ESI mass spectrum.

#### *Reaction of 2NMe<sub>2</sub> with 11.*

A solution of **2NMe<sub>2</sub>** (72 mg, 0.095 mmol) in 5 mL CDCl<sub>3</sub> was added drop wise to a solution of **11** (100 mg, 0.095 mmol) in 5 mL CDCl<sub>3</sub> and stirred for 3 hours at room temperature. A color change from bright yellow to yellow was observed. The solvent was reduced to half under reduced pressure and followed by addition of 10 mL n-hexane, which resulted in precipitation of a pale orange solid. The solid was filtered, washed with n-hexane (2 x 10mL) and dried under reduced pressure. <sup>31</sup>P NMR (162 MHz, CH<sub>2</sub>Cl<sub>2</sub>) δ (ppm) 40.7 (dt, J= 15, 113 Hz), 34.5 (dt, 15, 86 Hz). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ (ppm): 7.98-6.65 (m, 50H), 3.19 (s, 12H). ESI-MS [M]<sup>n+</sup> : m/z 380.7 [Rh(dppe)<sub>2</sub>(DMAP)<sub>2</sub>]<sup>3+</sup>, 510.1 [Rh(dppe)<sub>2</sub>(DMAP)]<sup>2+</sup>.

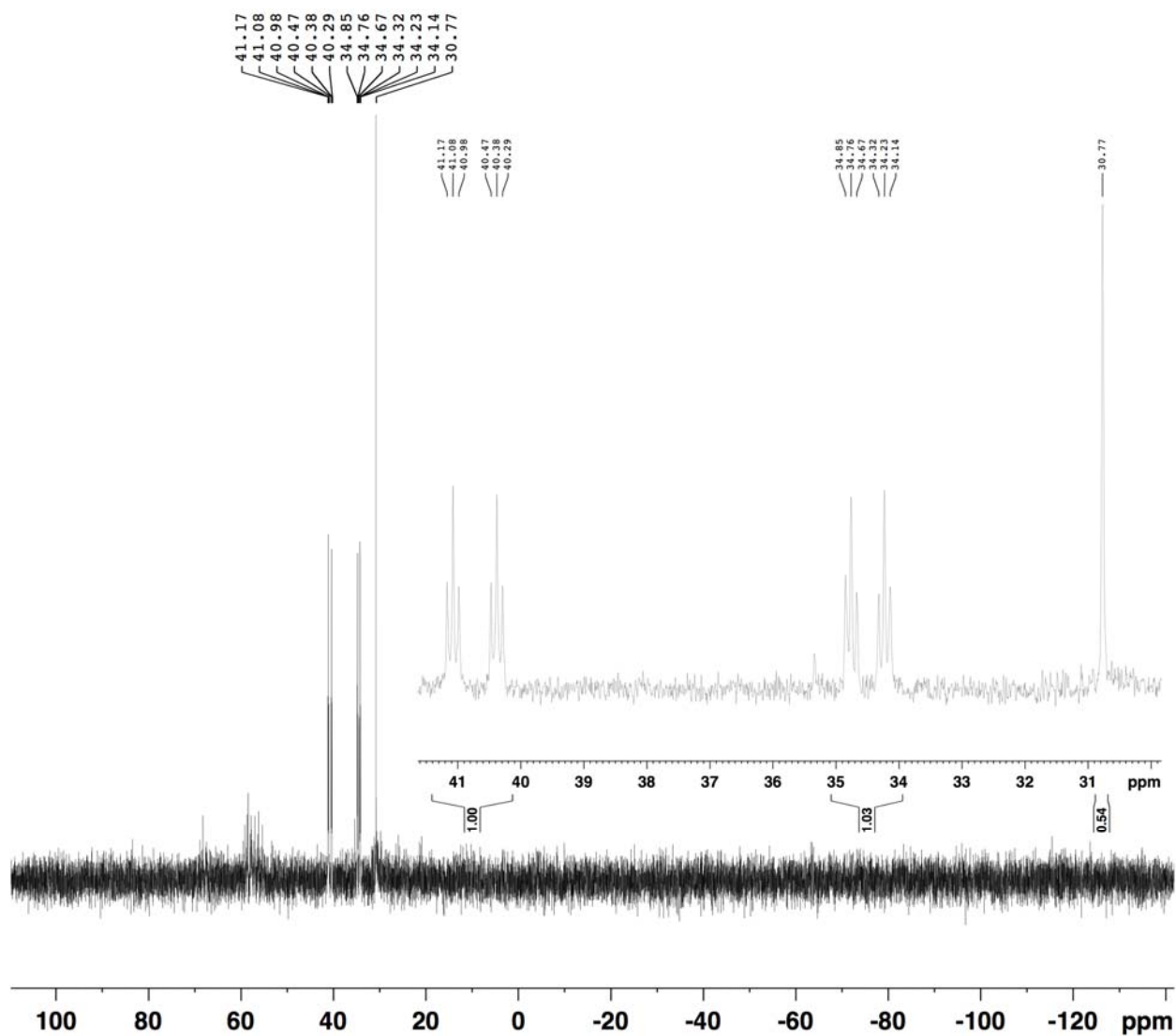


Figure 25. Reaction of  $2\text{NMe}_2$  with **11**  $^{31}\text{P}$  NMR.

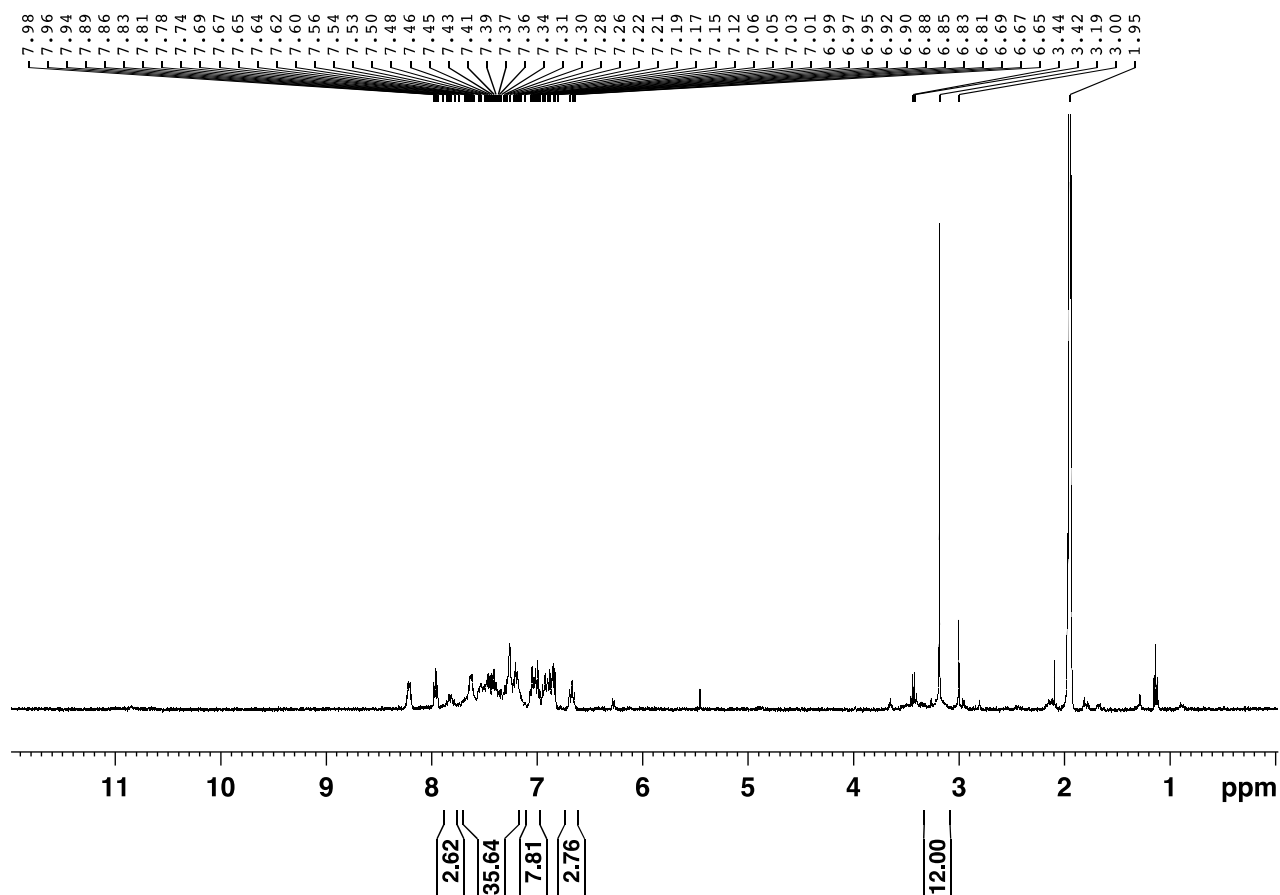


Figure 26. Reaction of  $2\text{NMe}_2$  with **11**  $^1\text{H}$  NMR.

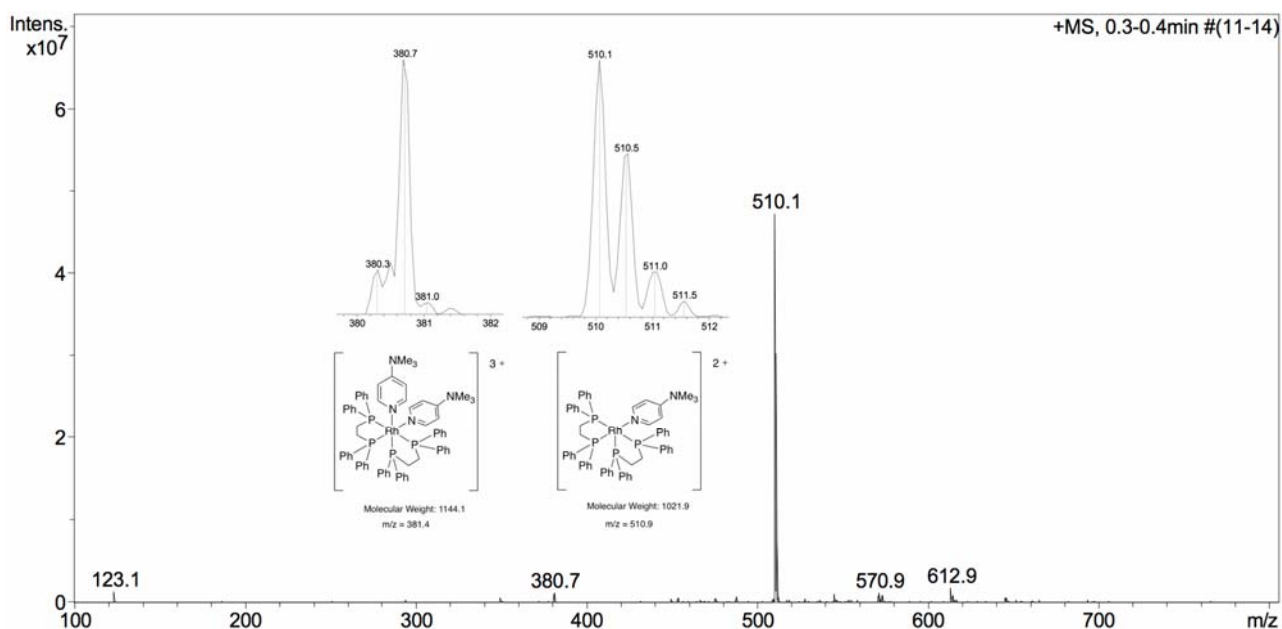


Figure 27. Reaction of  $2\text{NMe}_2$  with **11** ESI mass spectrum.

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