Supplementary Material

Improved Access to Linear Tetrameric Hydroxamic Acids with Potential as Radiochemical Ligands for Zirconium(IV)-89 PET Imaging

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Electronic Supplementary Material

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Figure S2b. ¹³C NMR spectrum (150 MHz, DMSO-*d*₆) for DFOB-PPH (**3**).



Figure S2c. ¹H-¹H COSY NMR spectrum (600 MHz, DMSO-*d*₆) for DFOB-PPH (**3**).



Figure S2d. ¹H-¹³C HSQC NMR spectrum (600 MHz, DMSO-*d*₆) for DFOB-PPH (3).



Figure S2e. Experimental (top) and calculated (bottom) isotope patterns for the $[M+H]^+$ adduct of DFOB-PPH (3).



Figure S3b. ¹³C NMR spectrum (150 MHz, DMSO- d_6) for DFOB-PPH^NO^CO (4).



Figure S3c. $^{1}H^{-1}H$ COSY NMR spectrum (600 MHz, DMSO- d_{6}) for DFOB-PPH^NO^CO (4).



Figure S3d. ¹H-¹³C HSQC NMR spectrum (600 MHz, DMSO-*d*₆) for DFOB-PPH^NO^CO (4).



Figure S3e. Experimental (top) and calculated (bottom) isotope patterns for the $[M+H]^+$ adduct of DFOB-PPH^NO^CO (4).



Figure S4b. ¹³C NMR spectrum (150 MHz, DMSO-*d*₆) for DFOB-PPH-*p*-Bn-SCN (5).



Figure S4c. ¹H-¹H COSY NMR spectrum (600 MHz, DMSO-*d*₆) for DFOB-PPH-*p*-Bn-SCN (**5**).



Figure S4d. ¹H-¹³C HSQC NMR spectrum (600 MHz, DMSO-*d*₆) for DFOB-PPH-*p*-Bn-SCN (5).



Figure S4e. Experimental (top) and calculated (bottom) isotope patterns for the $[M+H]^+$ adduct of DFOB-PPH-*p*-Bn-SCN (**5**).



Scheme S1. Complexes between Fe(III) and **1–4** as formed in a metal:ligand stoichiometry of (a) 1:1 or (b) 4:3.