Alkaloids of *Ochrosia poweri* Bail. The Stereochemistry of Poweridine and the Identity of Elliptamine and Reserpiline

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Abstract

The alkaloid poweridine from *Ochrosia poweri* Bail. has been shown to have the same relative configuration as yohimbine. Elliptamine, previously obtained from *O. poweri*, has been identified as reserpiline.

A study of the proton and ¹³C nuclear magnetic resonance spectra of the alkaloid poweridine, isolated previously from the leaves of *Ochrosia poweri* Bail.,¹ has shown that this alkaloid has the same relative stereochemistry as yohimbine, and subject only to the assumption that C15–H has the α -configuration usual for this type of alkaloid, the absolute structure can be given as (1). The main structural features of poweridine have already been established.¹ It has been shown to give 7-methoxyyobyrine on selenium dehydrogenation, and the acid formed by alkaline hydrolysis reacts with ethyl chloroformate to give a β -lactone in the same way as yohimbic acid is converted into yohimbic acid lactone.² The formation of a β -lactone under these conditions indicates that the carboxyl group and the hydroxy group are *cis*, and that the hydroxy group is axial.

The p.m.r. spectrum of poweridine (CDCl₃ solution) shows a narrow multiplet at δ 5.42 for C17–H, similar to that found for C17–H (δ 5.40) in the spectrum of *O*-acetylyohimbine (2). The narrow multiplet for C17–H indicates that all couplings to C17–H are small and specifically excludes any possibility that C17–H is axial because there is no large *trans*-diaxial coupling. The signal from C16–H is a doublet of doublets (δ 2.35; *J* 11, 2 Hz), and irradiation at the C17–H signal collapses the C16–H signal to a doublet (*J* 11 Hz). The large coupling between C15–H and C16–H indicates that these protons have a *trans*-diaxial configuration.

Confirmation for these stereochemical assignments and for the close relationship of poweridine (1) and O-acetylyohimbine (2) has been obtained by a comparison of their ¹³C n.m.r. spectra and also those of O-acetyl- α -yohimbine (O-acetylrauwolscine) (3), yohimbine and reserpine (4). Chemical shift data for these alkaloids are listed in Table 1, together with probable assignments based on single frequency off-resonance proton decoupled (SFORD) spectra, and on chemical shift data for indole alkaloids

² Diassi, P. A., and Dylion, C. M., J. Amer. Chem. Soc., 1958, 80, 3746.

¹ Doy, F. A., and Moore, B. P., Aust. J. Chem., 1962, 15, 548.

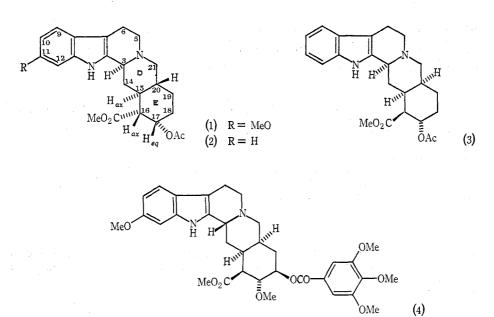


Table 1.	Chemical	shift o	lata fro	m 13C	n.m.r.	spectra
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Carbon atom	Multi- plicity ^A	Power- idine	<i>O</i> -Acetyl- yohimbine	Yohim- bine	O-Acetyl- α-yohimbine	Reser- pine
C2	s	129.5	134.4	134.7	134.2	130.7
C3	d	60.0	59.9	60.0	60.2	54.0
C5	t	52.9	52.8	52.9	53.4	49.3
C6	t	$21 \cdot 8$	21.6	21.9	21.6	17.0
C7	s	108.0	108.1	108.5	108.3	108.3
C 8	s	121.5	127.6	127.6	127.4	122.5
С9	d	118.6	118.2	118.2	118.1	118.6
C10	d	108.9	121.4	121.4	121 • 4	109.2
C11	đ	$152 \cdot 2$ (s)	119.4	119.5	119.4	156 · 5 (s)
C12	d	95.4	110.9	110.8	111.0	95.6
C13	s	133.2	136.2	136.2	136.4	136.6
C14	t	30.0	29.9	31.7	27.4	29.9
C15	đ	37.0	36.9	36.9	36.6	34.3
C16	d	$51 \cdot 2$	51.0	52-6	52.0	52.0
C17	d	69·7	69.7	69.1	69.2	78.0
C18	t.	34.1	33.9	34.5	31.0	78 · 2 (d)
C19	t	24.1	23.9	23.4	24.8	24.5
C 20	d	40.2	39.9	40.9	38.4	32.5
C 21	t	61 · 2	61.0	61.5	60.4	52.0
C22	S	172.3	172.4	175.5	172.8	172.9
CO ₂ CH ₃	q	51 · 8	51.9			
OCOCH ₃	S	170.0	170.1		170.0	
OCOCH ₃	q	21.0	21.0		21.0	
OCH ₃	q	55.8	2			56·0

^A Observed multiplicity of signal in SFORD spectrum; exceptions are indicated within brackets in chemical shift columns.

cited in a recent review of ¹³C n.m.r. spectroscopy.³ The very close coincidence in chemical shifts in the spectra of poweridine and O-acetylyohimbine, apart from the shifts of the carbon atoms associated with the C11 methoxy group, indicate the close similarity of the two compounds. In particular, the C3 configuration (axial H) is the same in poweridine, O-acetylyohimbine and O-acetyl- α -vohimbine as shown by the C3 chemical shift at 60.0 + 0.2 p.p.m. whereas in reservine, with C3-H equatorial, the C3 resonance is at 54.0 p.p.m. The change in configuration at C3 in reserpine, in which a bulky axial substituent replaces the axial hydrogen of the other compounds, is reflected in the upfield shifts of the sterically perturbed γ -carbon atoms, viz, C 5, C 15 and C 21. Similarly the configuration at the D/E ring junction is the same in poweridine and O-acetylyohimbine with the chemical shifts of C 20 (40.0 ± 0.2) p.p.m.) and C15 (37.0+0.1 p.p.m.) being identical. A change in configuration at C 20 from an axial hydrogen in poweridine to an equatorial hydrogen and hence bulky axial substituent in Q-acetyl- α -vohimbine is again reflected in the shift of the C 20 resonance and the two sterically perturbed γ -carbons (C14 and C18). The similarity in shifts of the A ring signals of poweridine with those of reserpine confirm the position of the methoxyl group as C11.

Re-examination of the alkaloid elliptamine,^{1,4} isolated as a crystalline picrate from O. poweri and O. elliptica, has shown that this alkaloid is identical with reservation.

Experimental

General

N.m.r. spectra were measured in $CDCl_3$ solutions and chemical shifts are referred to $SiMe_4$ ($\delta 0.00$). The proton spectra were measured on a Varian HA100 spectrometer operating at 100 MHz and the carbon spectra on a Varian CFT 20 spectrometer operating at 20 MHz and locked to the deuterium signal in the solvent, $CDCl_3$. The single frequency off-resonance spectra were recorded with the decoupling frequency approximately 1000 Hz upfield from the SiMe₄ signal.

Poweridine

P.m.r. spectrum: δ 3·80, 3H, s, OMe; 3·68, 3H, s, CO₂Me; 2·07, 3H, s, OCOMe; 7·32, 1H, d, J 8 Hz, C9–H; 6·73, doublet of doublets, J 8, 2 Hz, C10–H; 6·79, 1H, d, J 2 Hz, C12–H; 5·42, multiplet, all couplings small, C17–H; 2·35, doublet of doublets, J 11, 2 Hz, C16–H. Mass spectrum: 426 (M⁺ and base peak, 100%), 368 (10), 366 (10), 214 (14), 200 (16), 199 (24), 176 (12).

Elliptamine

Elliptamine picrate⁴ was obtained as orange-red solvated crystals, m.p. 170°, from aqueous methanol, and non-solvated crystals, m.p. 215–216° from anhydrous methanol. (Literature m.p. reserpiline picrate, 173°.) The n.m.r. spectrum indicated a close relationship to isoreserpiline, and comparison of the mass spectra confirmed that the two bases were isomeric (M⁺ at m/e 412) and very clearly similar in structure. The molecular ion peak at m/e 412 was consistent with the molecular formula C₂₃H₂₈N₂O₅ previously deduced from microanalytical data. The free base yielded a crystalline oxalate, m.p. 244–245° (dec.) (literature m.p. reserpiline oxalate, 244–245°), which had an i.r. spectrum identical with that recorded for reserpiline oxalate.⁵

Manuscript received 9 April 1975

³ Wenkert, E., Bindra, J. S., Chong, C.-J., Cochran, D. W., and Schell, F. M., Accounts Chem. Res., 1974, 7, 46.

⁴ Douglas, B., Kirkpatrick, J. L., Moore, B. P., and Weisbach, J. A., *Aust. J. Chem.*, 1964, **17**, 246. ⁵ 'Physical Data of Indole and Dihydroindole Alkaloids' 4th Edn (Eli Lilly & Co.: Indianapolis 1960).