

## Supplementary Material

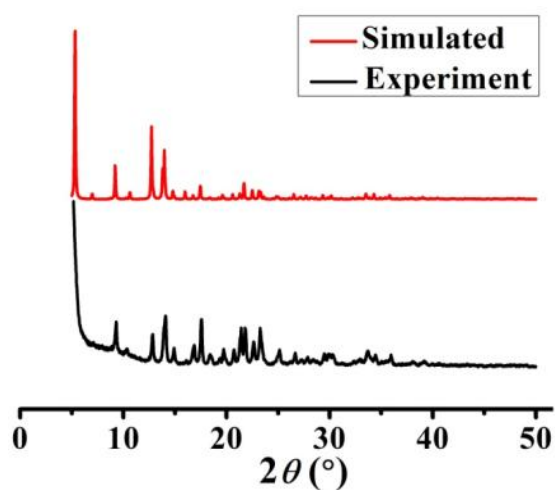
### Specific Recognition of Methanol Using a Symmetric Tetramethylcucurbit[6]uril-Based Porous Supramolecular Assembly Incorporating Adsorbed Dyes

Fei Yang Tian,<sup>A</sup> Rui Xue Cheng,<sup>A</sup> Yun Qian Zhang,<sup>A</sup> Zhu Tao,<sup>A,B</sup> and Qian Jiang Zhu<sup>A</sup>

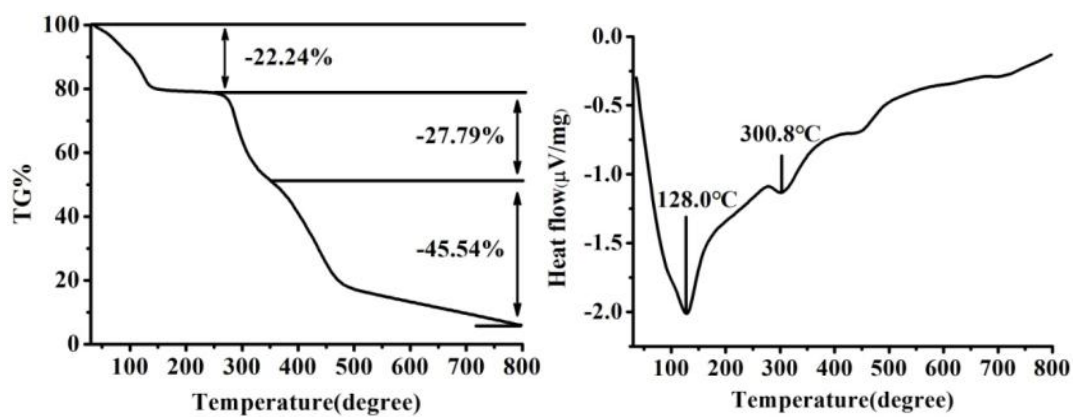
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<sup>B</sup>Corresponding author. Email: gzutao@263.net

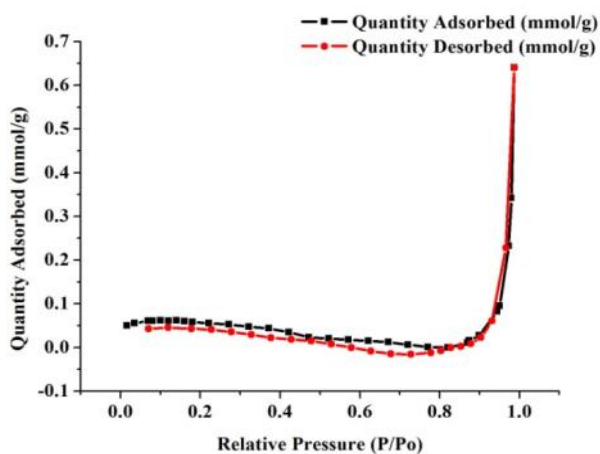
#### EXPERIMENTAL SECTION



**Fig. S1** Powder X-ray diffraction analysis of A (a) simulation and (b) experiment.

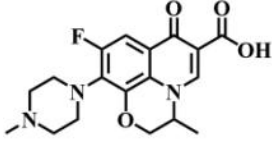
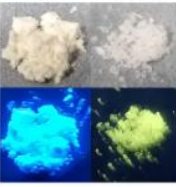
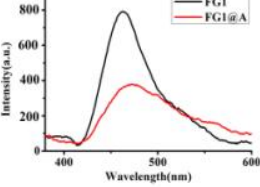
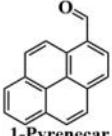
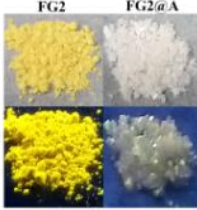
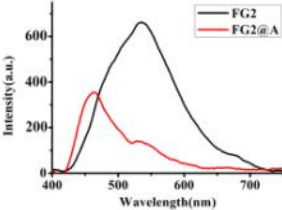
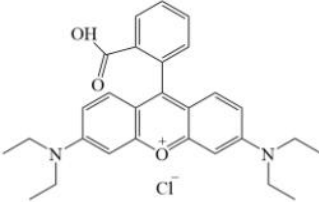
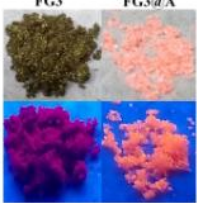
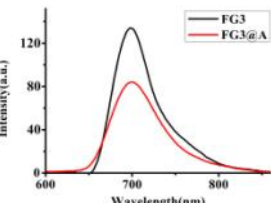
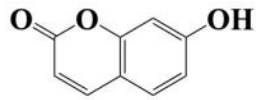
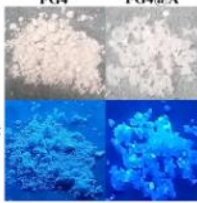
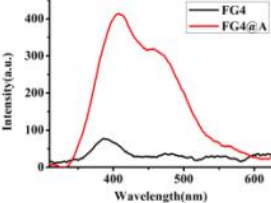
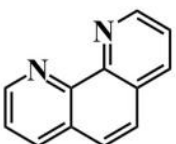
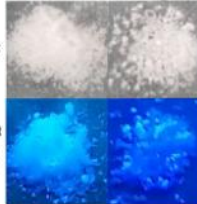
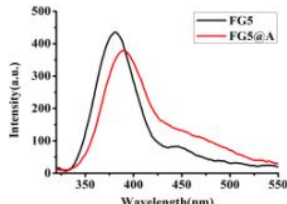
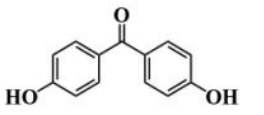
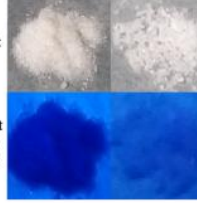
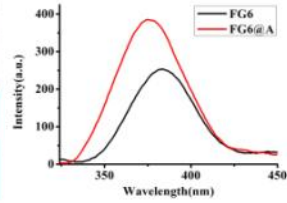


**Fig. S2** TG (top) and DTA (bottom) curves obtained for A under N<sub>2</sub>.

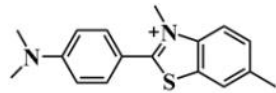


**Fig. S3** The sorption isotherms of N<sub>2</sub> at 77 K using a micrometrics ASAP2020HD88 automated sorption analyzer. (■ symbol=adsorption and, ●=desorption of A.)

**Table S1** A general survey of loading **A** with 15 **FGs** to form luminescent assemblies **FG@As**: (first column) **FGs** and (second column) the fluorescence spectra of the solid **FGs**, **FG@As**.

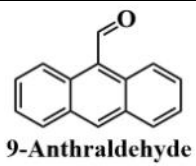
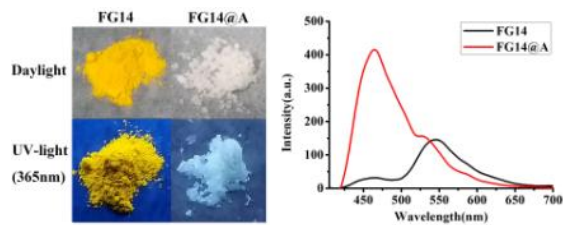
<p style="text-align: center;"><b>FGs</b></p>	<p style="text-align: center;">Survey of loading <b>A</b> with <b>FGs</b> (saturated absorption)</p>
<div style="text-align: center;">  <p><b>Levofloxacin</b></p> <p><b>FG1</b></p> </div>	<div style="display: flex; justify-content: space-around;"> <div style="text-align: center;"> <p><b>FG1</b>   <b>FG1@A</b></p>  </div> <div style="text-align: center;">  </div> </div>
<div style="text-align: center;">  <p><b>1-Pyrene-carboxaldehyde</b></p> <p><b>FG2</b></p> </div>	<div style="display: flex; justify-content: space-around;"> <div style="text-align: center;"> <p><b>FG2</b>   <b>FG2@A</b></p>  </div> <div style="text-align: center;">  </div> </div>
<div style="text-align: center;">  <p><b>FG3</b></p> </div>	<div style="display: flex; justify-content: space-around;"> <div style="text-align: center;"> <p><b>FG3</b>   <b>FG3@A</b></p>  </div> <div style="text-align: center;">  </div> </div>
<div style="text-align: center;">  <p><b>7-Hydroxycoumarin</b></p> <p><b>FG4</b></p> </div>	<div style="display: flex; justify-content: space-around;"> <div style="text-align: center;"> <p><b>FG4</b>   <b>FG4@A</b></p>  </div> <div style="text-align: center;">  </div> </div>
<div style="text-align: center;">  <p><b>1,10-Phenanthroline</b></p> <p><b>FG5</b></p> </div>	<div style="display: flex; justify-content: space-around;"> <div style="text-align: center;"> <p><b>FG5</b>   <b>FG5@A</b></p>  </div> <div style="text-align: center;">  </div> </div>
<div style="text-align: center;">  <p><b>4,4'-Dihydroxybenzophenone</b></p> <p><b>FG6</b></p> </div>	<div style="display: flex; justify-content: space-around;"> <div style="text-align: center;"> <p><b>FG6</b>   <b>FG6@A</b></p>  </div> <div style="text-align: center;">  </div> </div>

<p style="text-align: center;"> <chem>NCC1=CC=C2C=C1C=C3C=CC=CC23</chem>  <b>1-Pyrenemethylamine hydrochloride</b>  <b>FG7</b> </p>	<p style="text-align: center;"> <b>FG7</b>    <b>FG7@A</b> </p>
<p style="text-align: center;"> <chem>Oc1ccc(cc1)/N=N/c2ccccc2</chem>  <b>4-Phenylazophenol</b>  <b>FG8</b> </p>	<p style="text-align: center;"> <b>FG8</b>    <b>FG8@A</b> </p>
<p style="text-align: center;"> <chem>CCN(CC)c1ccc(cc1)/N=N/c2ccccc2C(=O)O</chem>  <b>Ethyl Red</b>  <b>FG9</b> </p>	<p style="text-align: center;"> <b>FG9</b>    <b>FG9@A</b> </p>
<p style="text-align: center;"> <chem>C1=CC=C2C=CC=CC1C2</chem>  <b>Pyrene</b>  <b>FG10</b> </p>	<p style="text-align: center;"> <b>FG10</b>    <b>FG10@A</b> </p>
<p style="text-align: center;"> <chem>Oc1ccc2c(c1)ncn2</chem>  <b>8-Hydroxyquinoline</b>  <b>FG11</b> </p>	<p style="text-align: center;"> <b>FG11</b>    <b>FG11@A</b> </p>
<p style="text-align: center;"> <chem>Oc1ccc2ccccc12</chem>  <b>1-Naphthol</b>  <b>FG12</b> </p>	<p style="text-align: center;"> <b>FG12</b>    <b>FG12@A</b> </p>
<p style="text-align: center;"> <chem>CN(C)c1ccc(cc1)S(=O)(=O)Cl</chem>  <b>Dansyl Chloride</b>  <b>FG13</b> </p>	<p style="text-align: center;"> <b>FG13</b>    <b>FG13@A</b> </p>



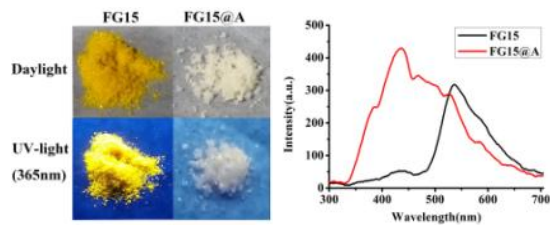
Thioflavin T

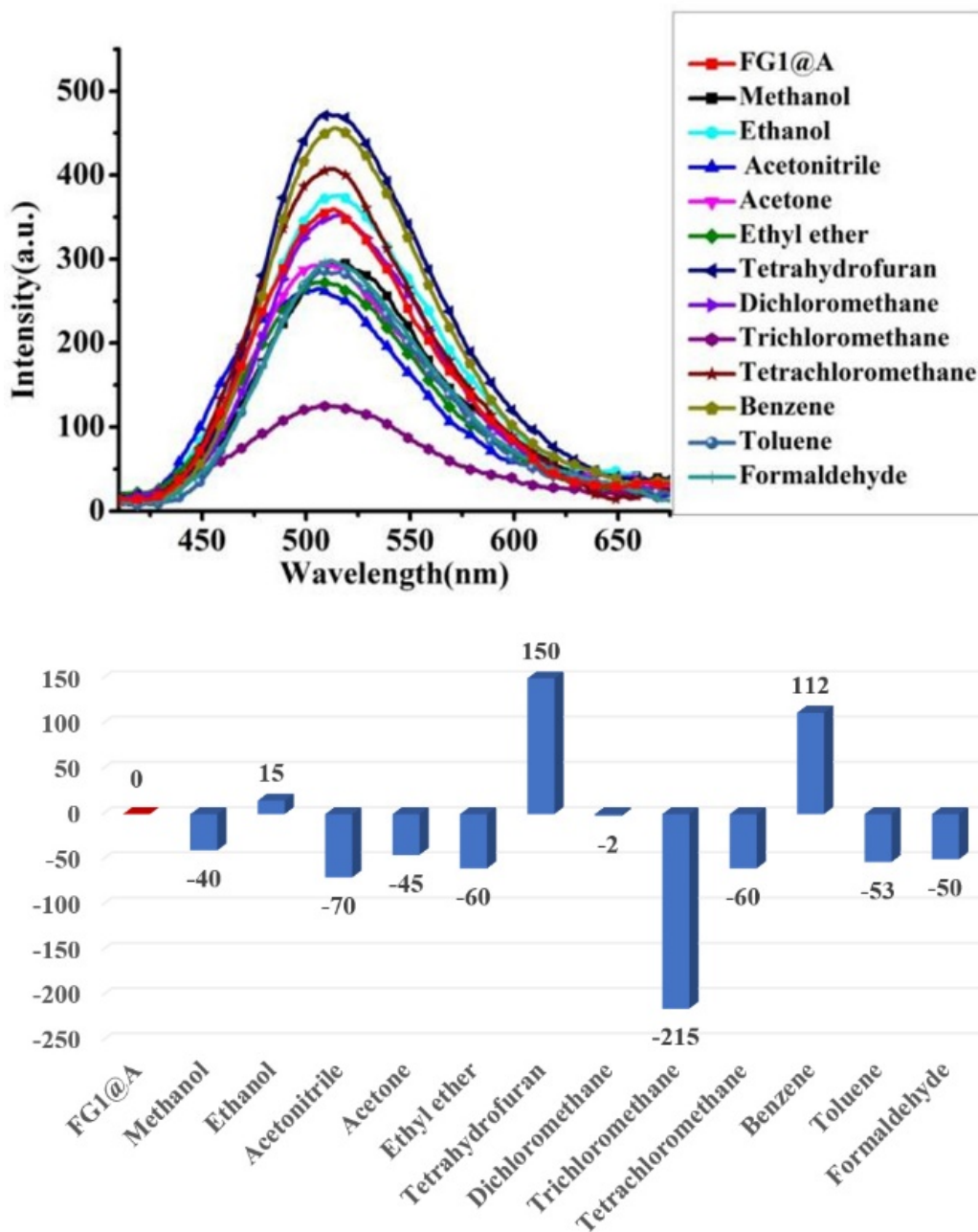
FG14



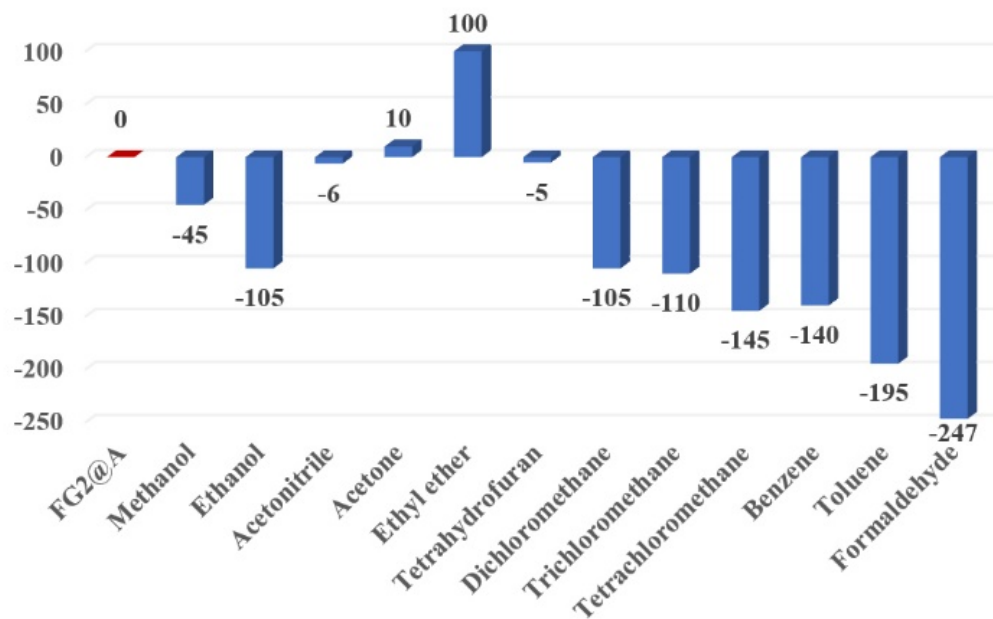
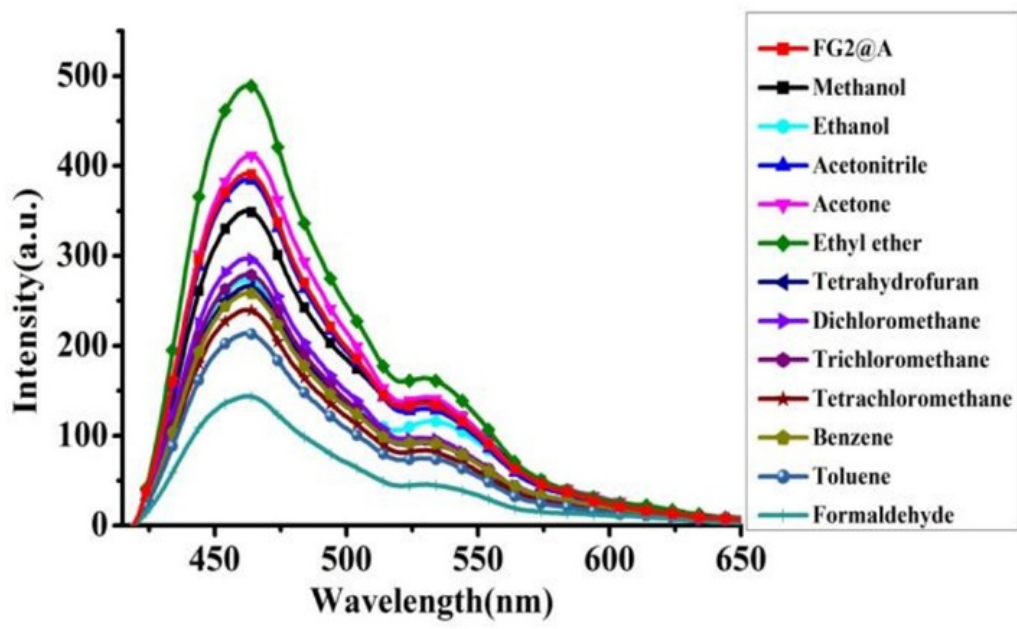
9-Anthraldehyde

FG15

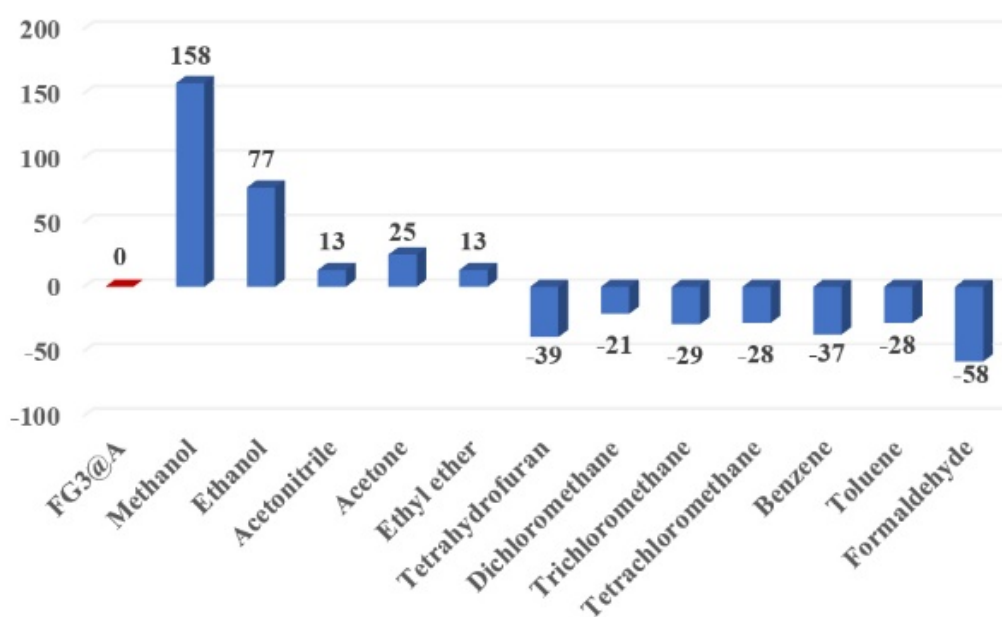
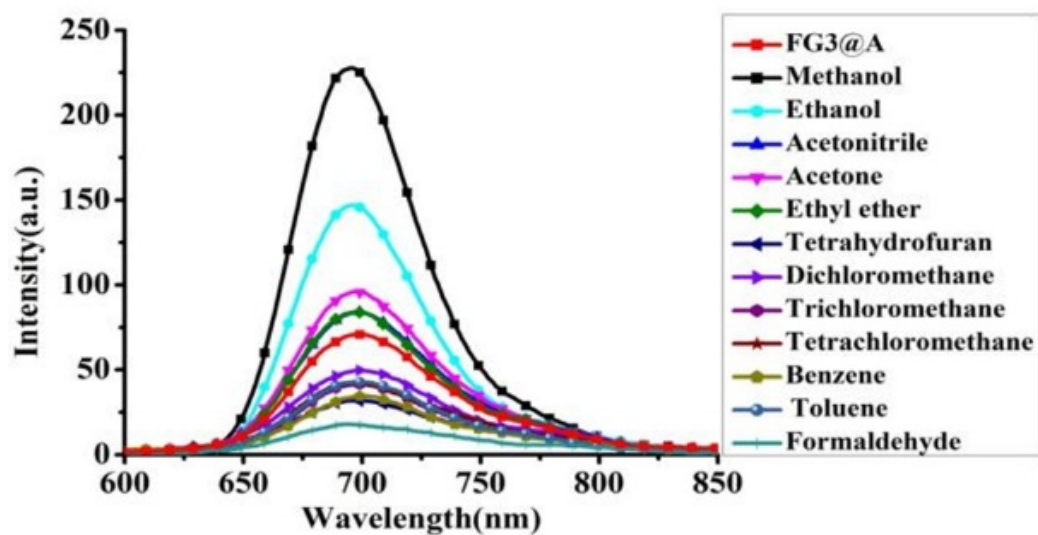




**Fig. S4** A general survey of the fluorescence spectra of FG1@A loaded with the 12 selected VOCs. The relative fluorescence intensities of FG1@A in response to the 12 selected VOCs.

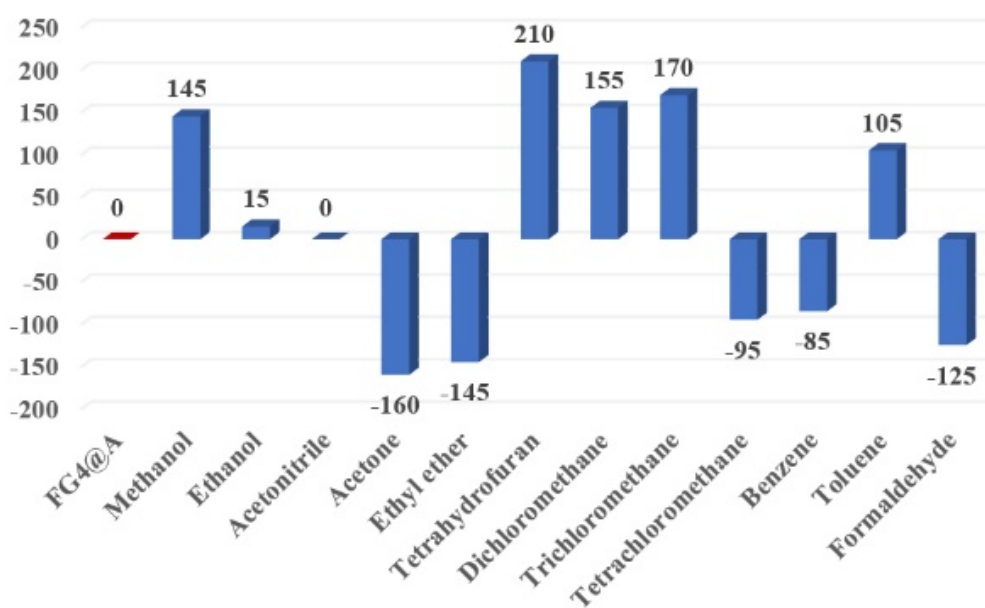
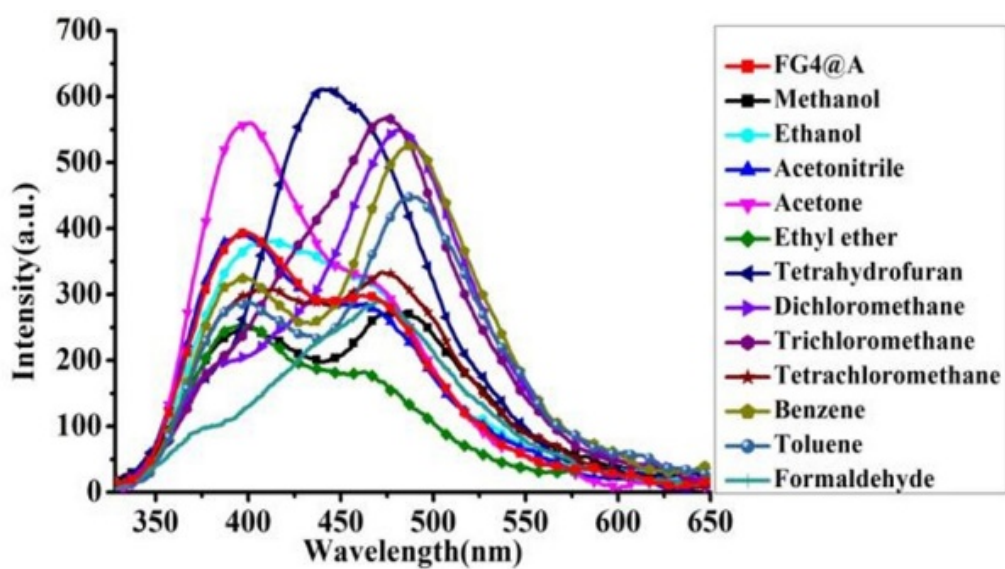


**Fig. S5** A general survey of the fluorescence spectra of FG2@A loaded with the 12 selected VOCs. The relative fluorescence intensities of FG2@A in response to the 12 selected VOCs.

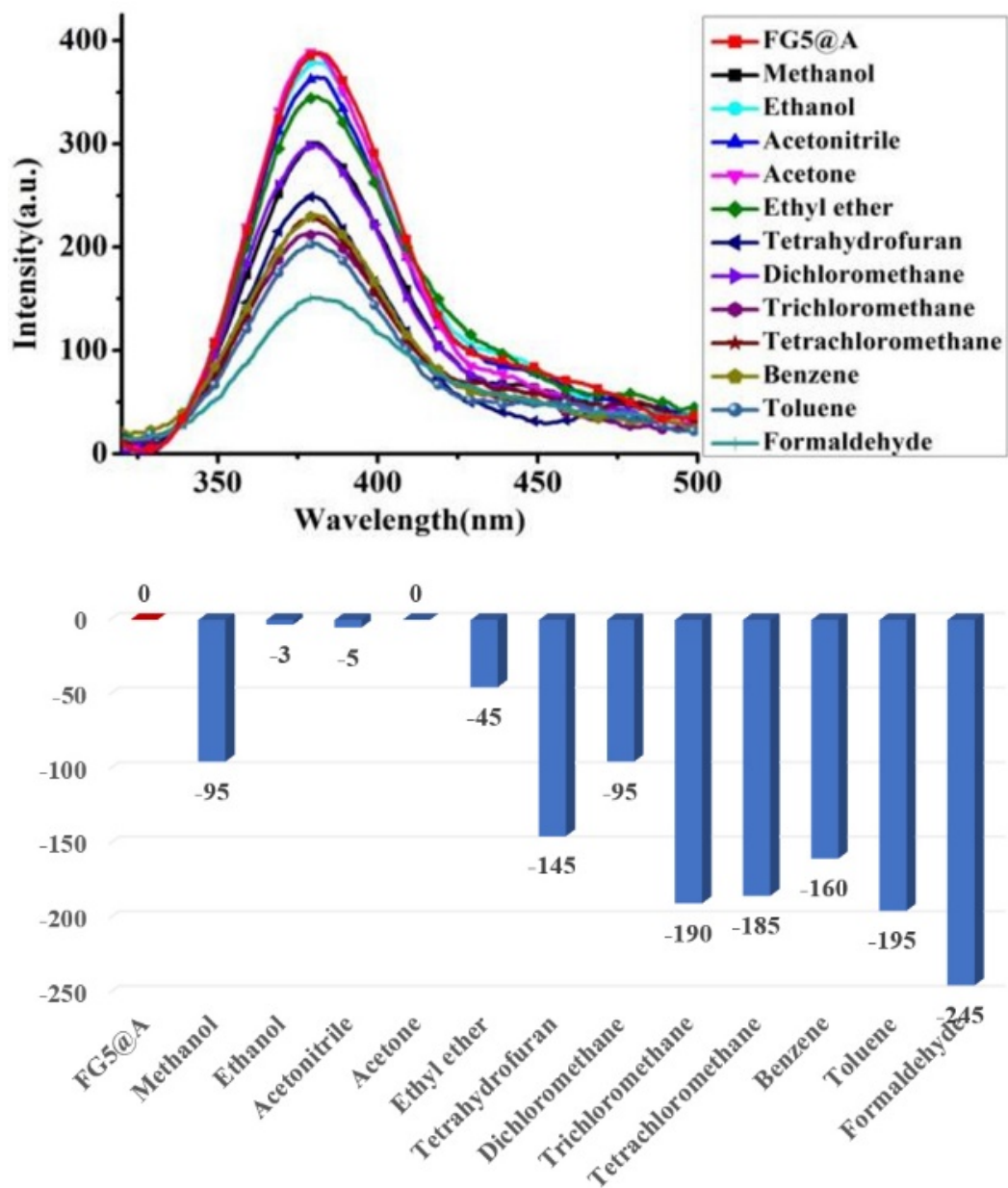


**Fig. S6** A general survey of the fluorescence spectra of FG3@A loaded with the 12 selected VOCs. The relative fluorescence intensities of FG3@A in response to the 12 selected VOCs.

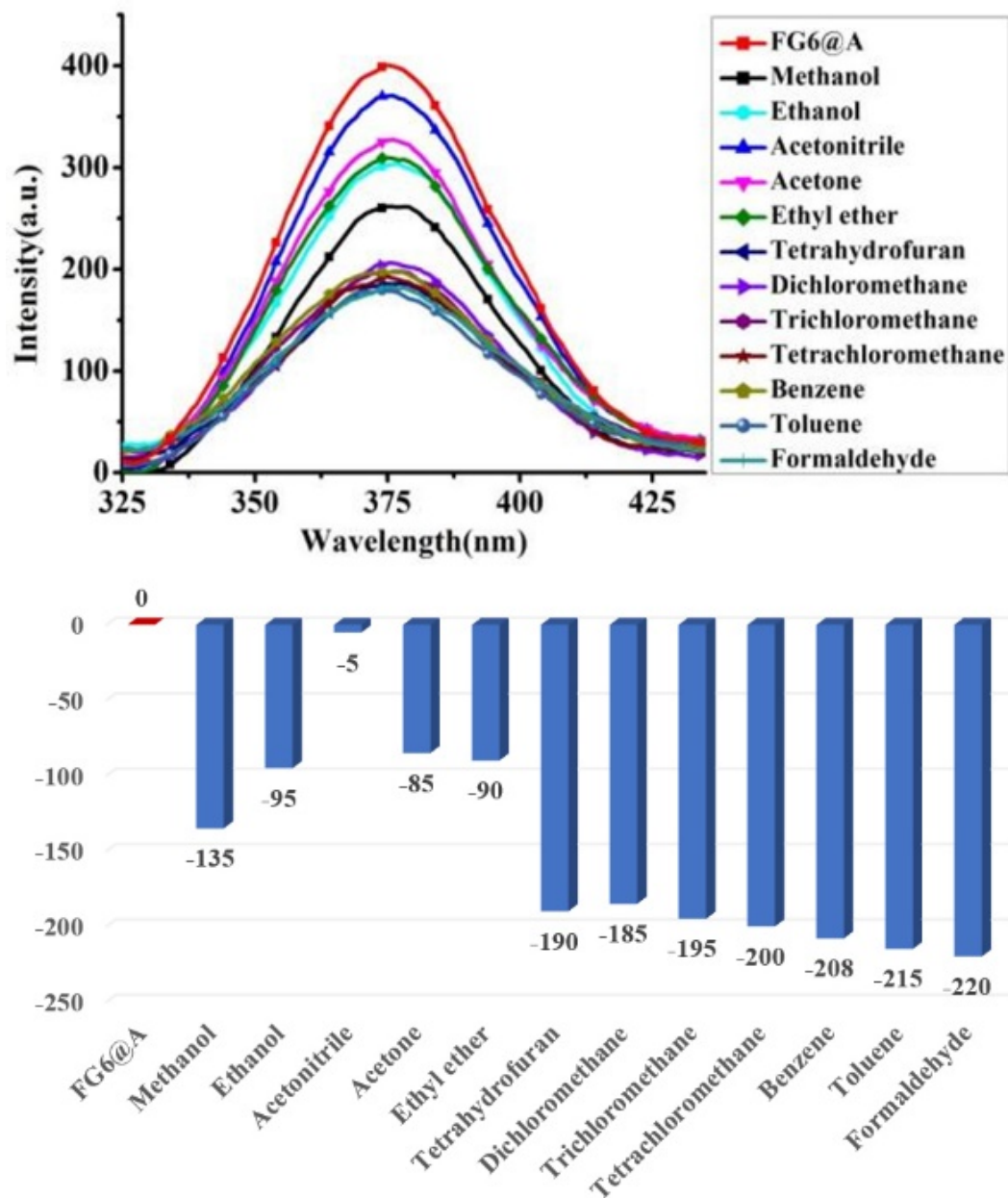




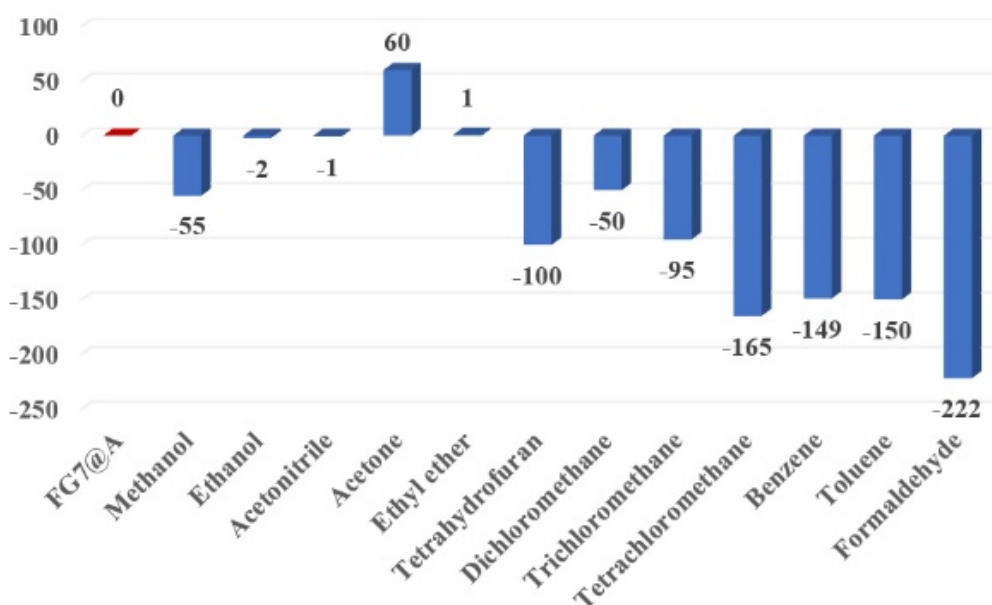
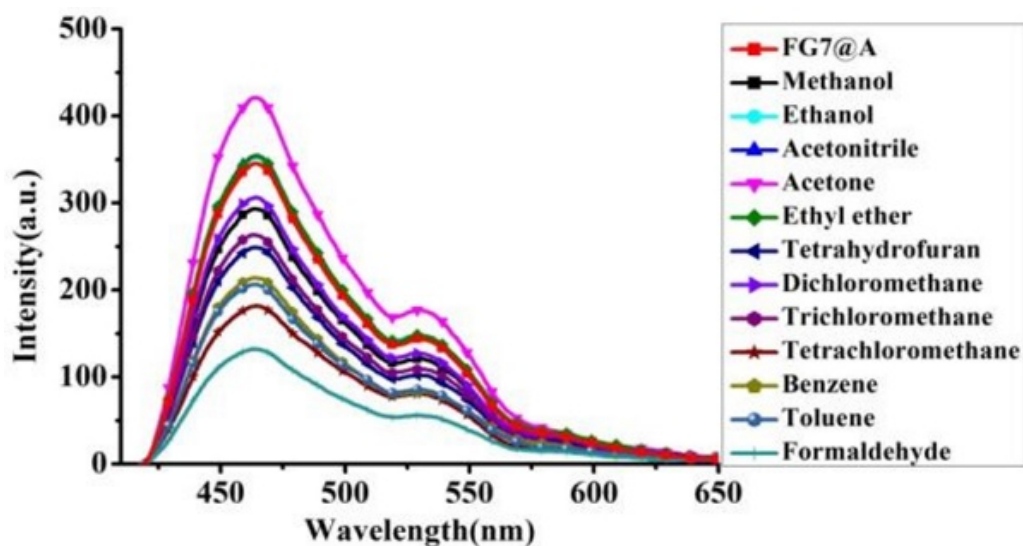
**Fig. S7** A general survey of the fluorescence spectra of FG4@A loaded with the 12 selected VOCs. The relative fluorescence intensities of FG4@A in response to the 12 selected VOCs.



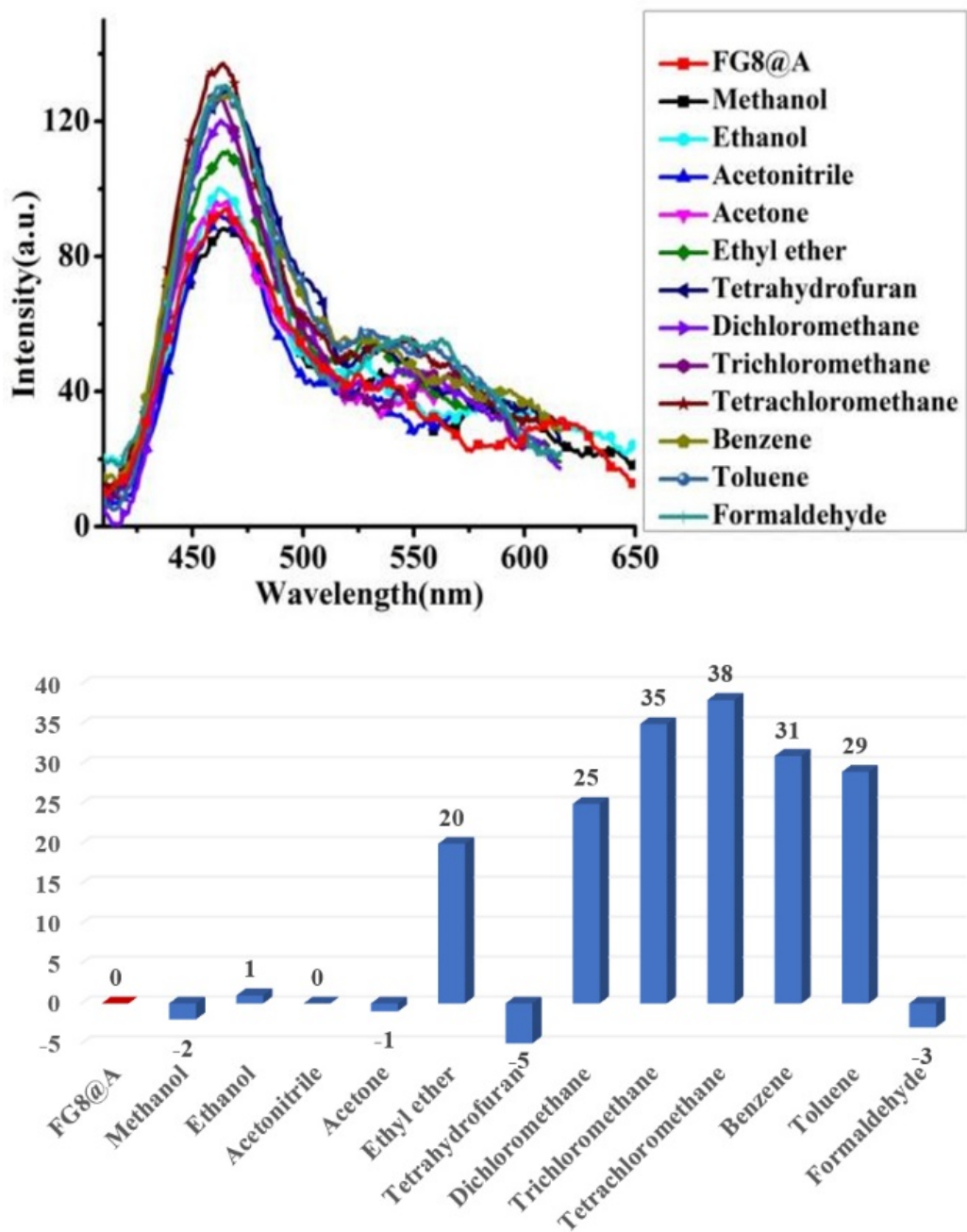
**Fig. S8** A general survey of the fluorescence spectra of FG5@A loaded with the 12 selected VOCs. The relative fluorescence intensities of FG5@A in response to the 12 selected VOCs.



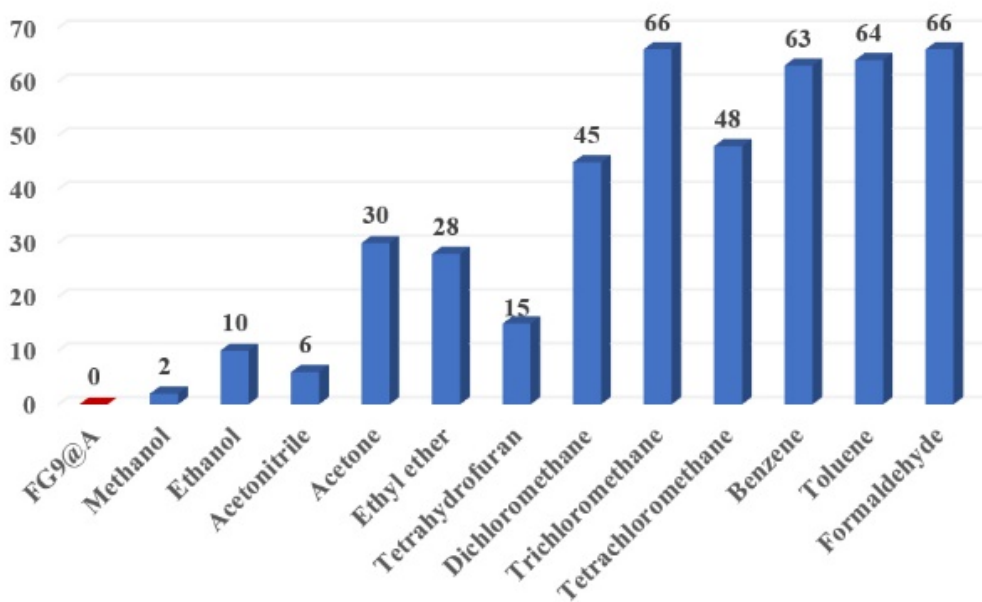
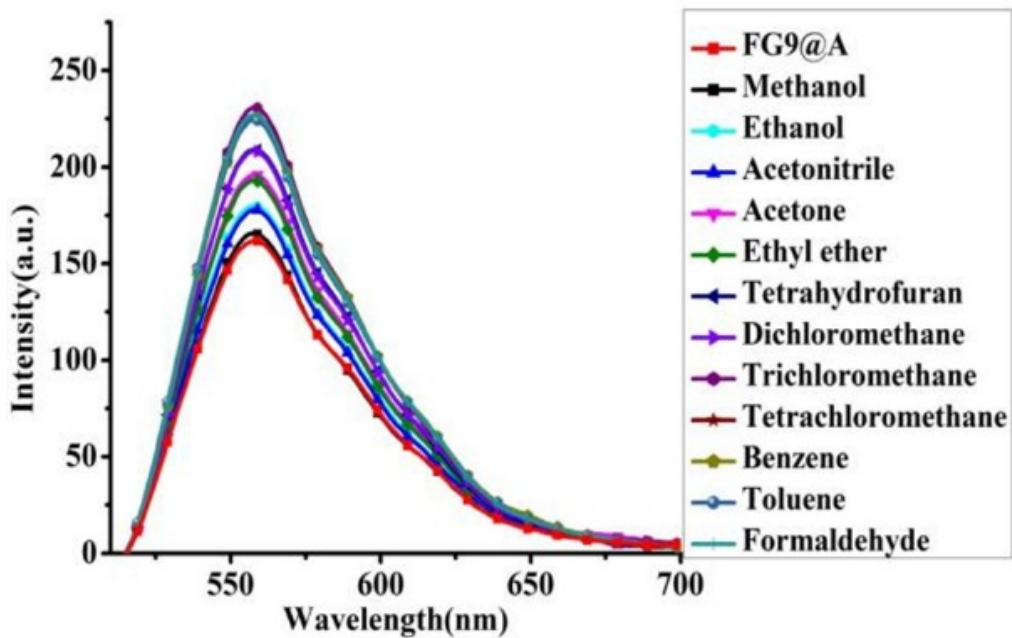
**Fig. S9** A general survey of the fluorescence spectra of FG6@A loaded with the 12 selected VOCs. The relative fluorescence intensities of FG6@A in response to the 12 selected VOCs.



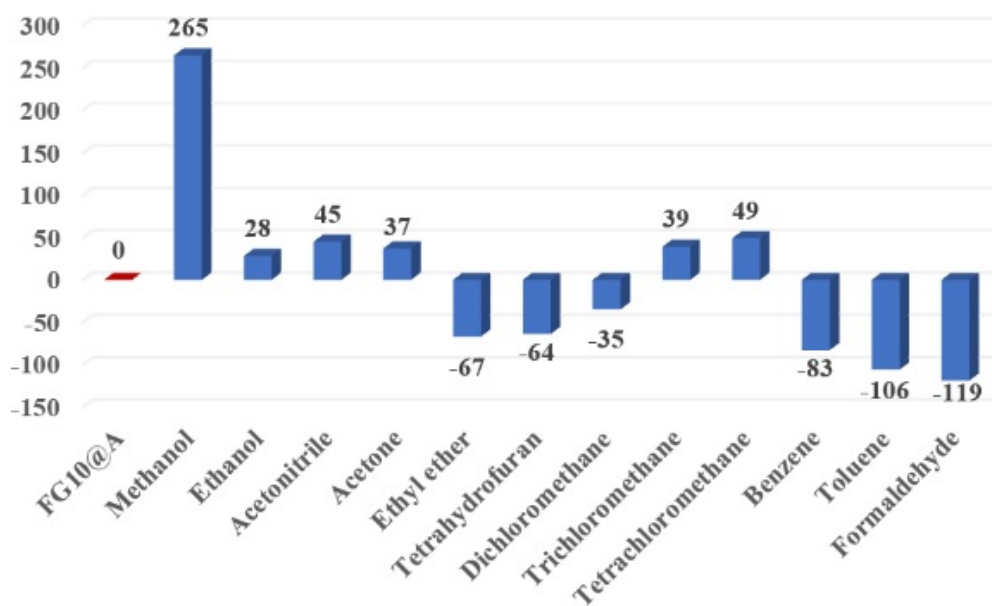
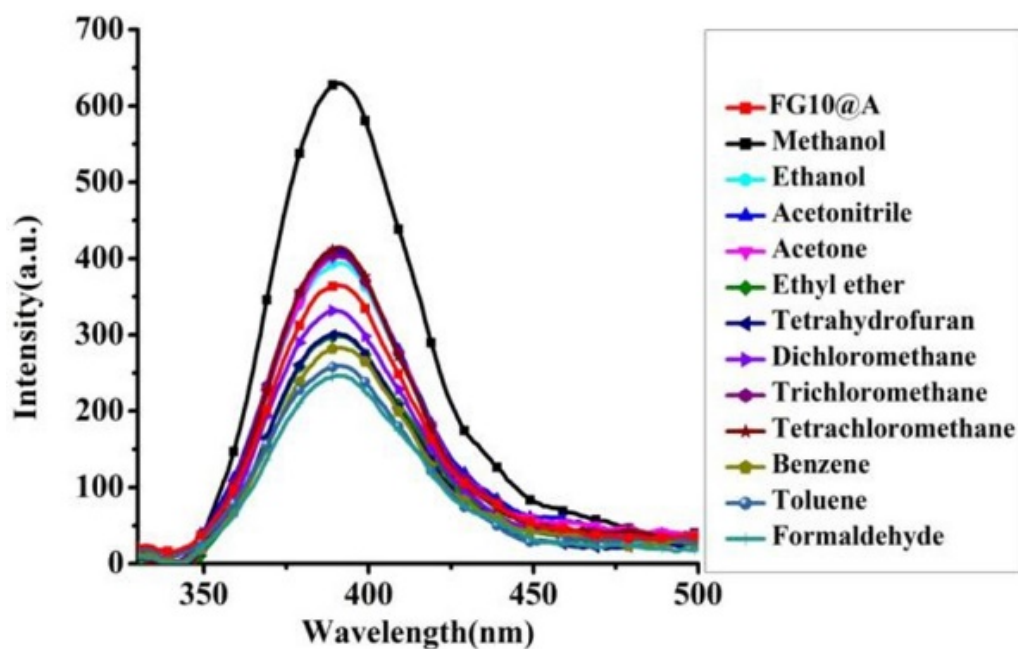
**Fig. S10** A general survey of the fluorescence spectra of FG7@A loaded with the 12 selected VOCs. The relative fluorescence intensities of FG7@A in response to the 12 selected VOCs.



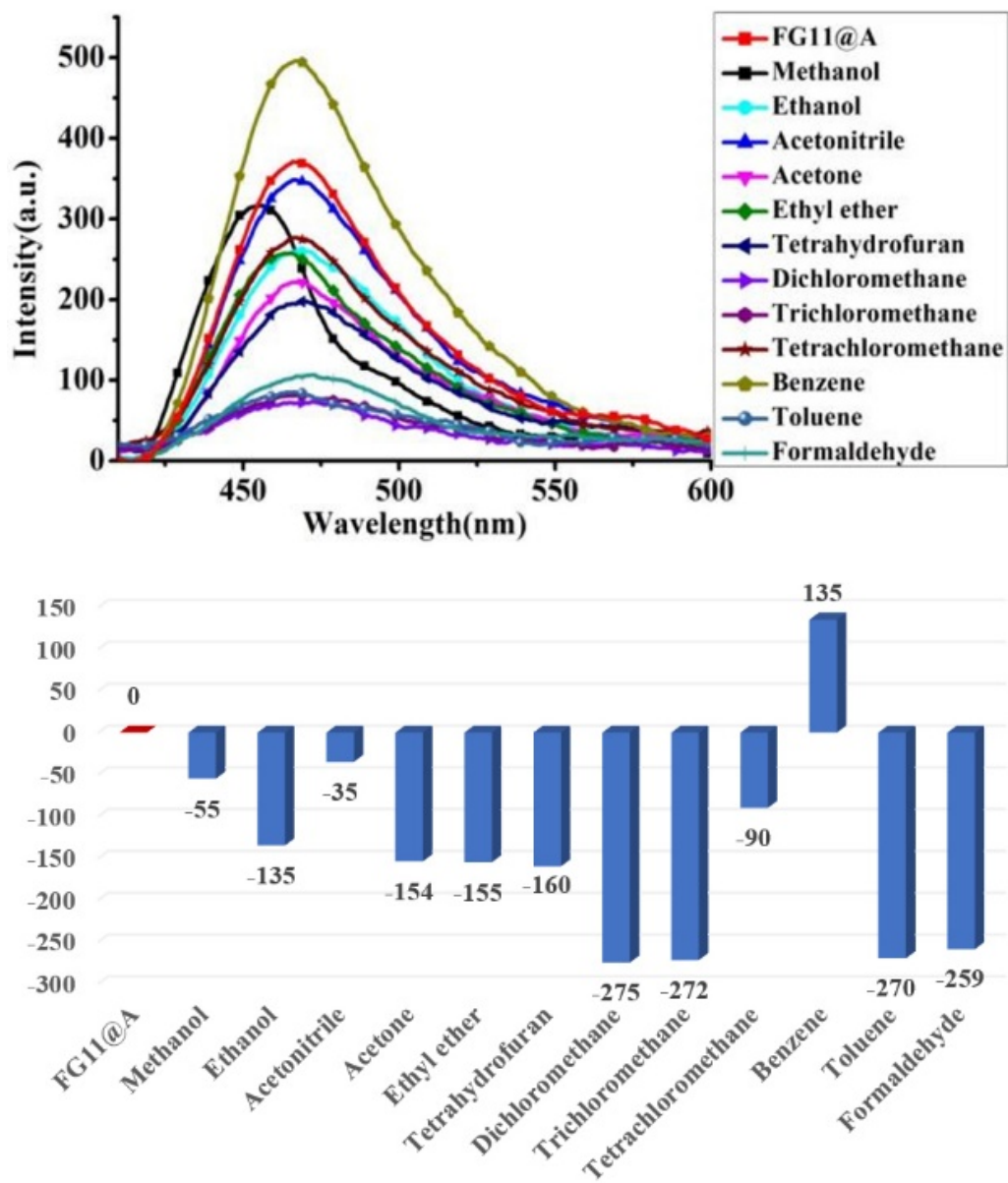
**Fig. S11** A general survey of the fluorescence spectra of FG8@A loaded with the 12 selected VOCs. The relative fluorescence intensities of FG8@A in response to the 12 selected VOCs.



**Fig. S12** A general survey of the fluorescence spectra of FG9@A loaded with the 12 selected VOCs. The relative fluorescence intensities of FG9@A in response to the 12 selected VOCs.

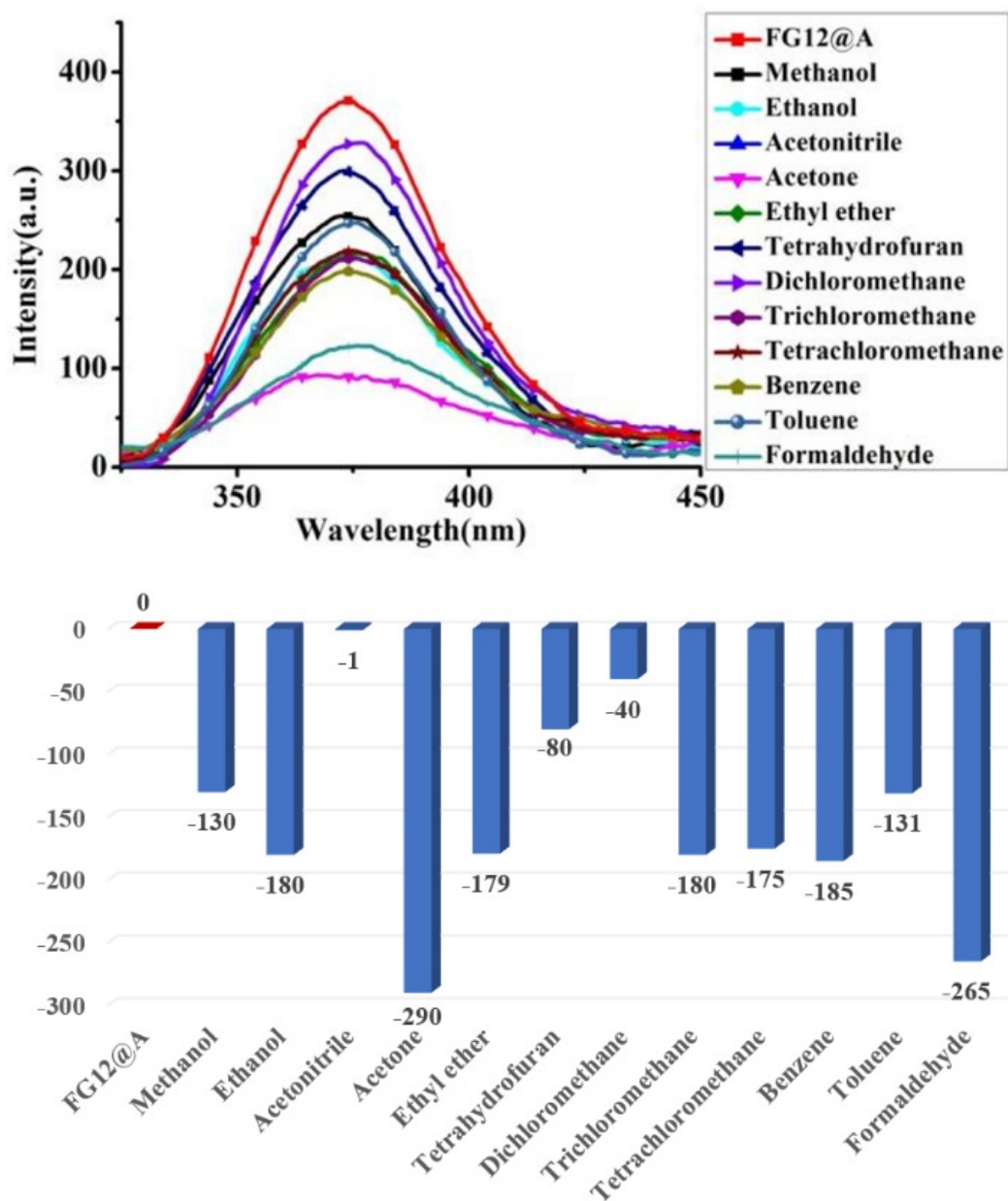


**Fig. S13** A general survey of the fluorescence spectra of FG10@A loaded with the 12 selected VOCs. The relative fluorescence intensities of FG10@A in response to the 12 selected VOCs.

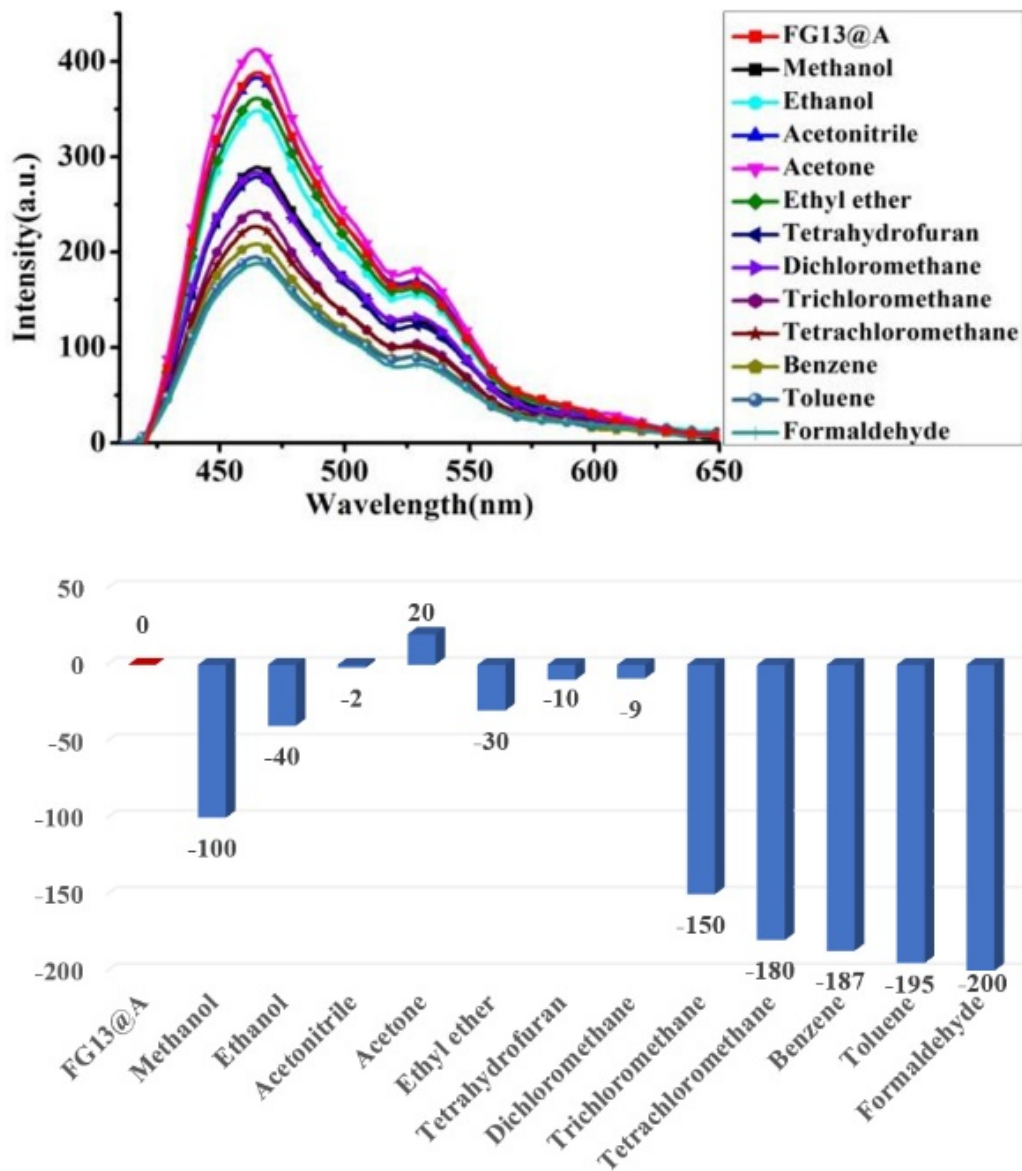


**Fig. S14** A general survey of the fluorescence spectra of FG11@A loaded with the 12 selected VOCs. The relative fluorescence intensities of FG11@A in response to the 12 selected VOCs.

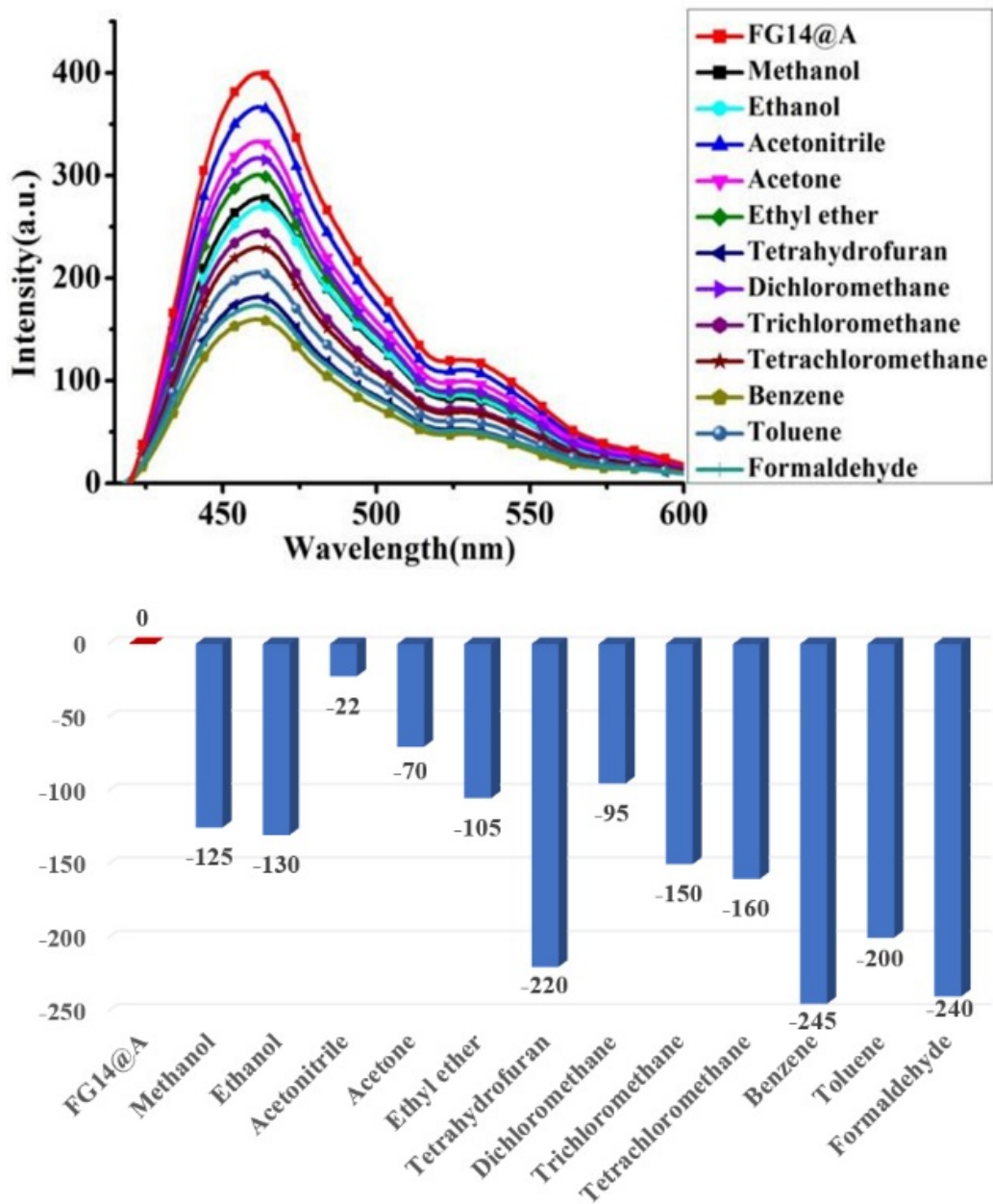




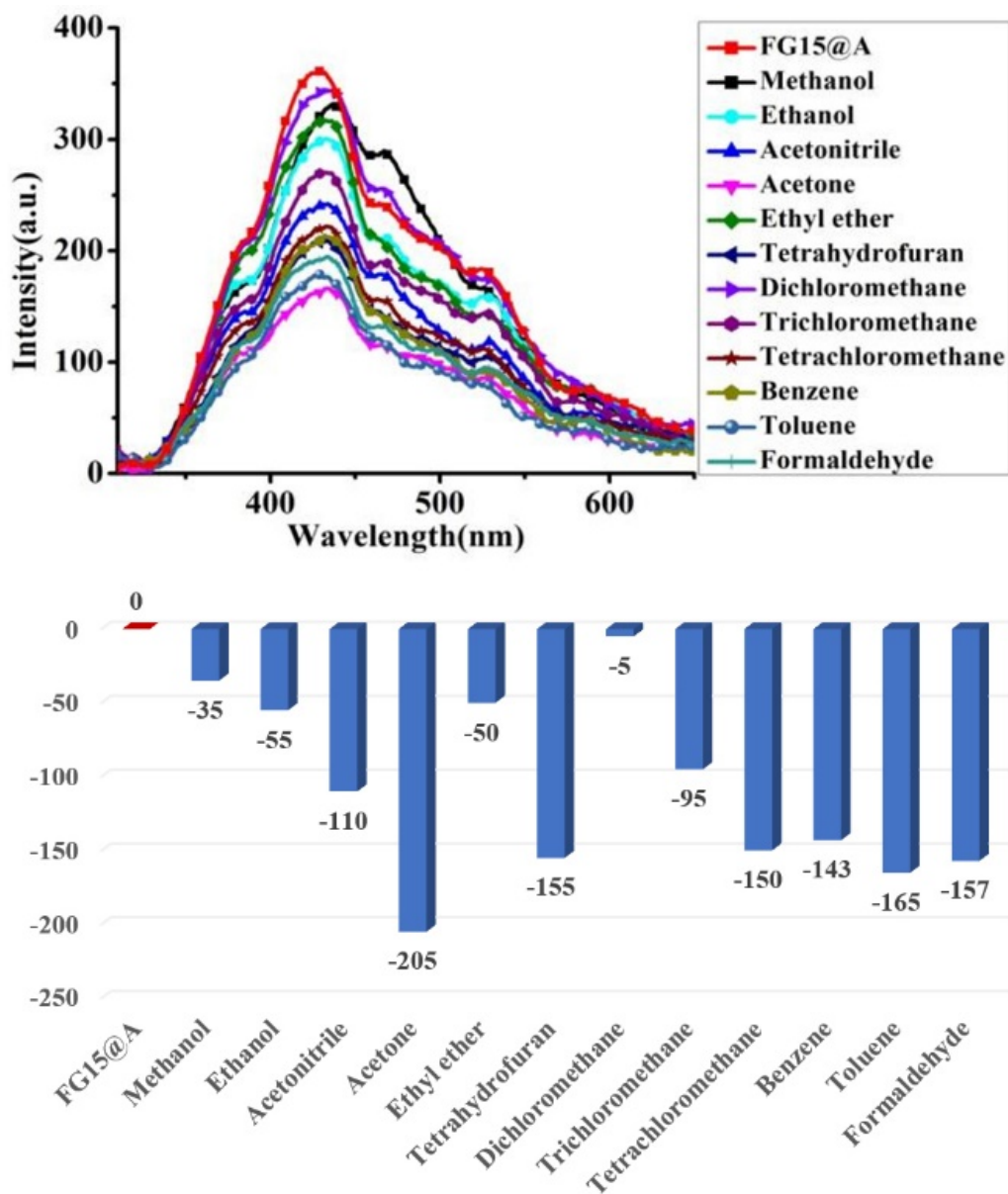
**Fig. S15** A general survey of the fluorescence spectra of FG12@A loaded with the 12 selected VOCs. The relative fluorescence intensities of FG12@A in response to the 12 selected VOCs.



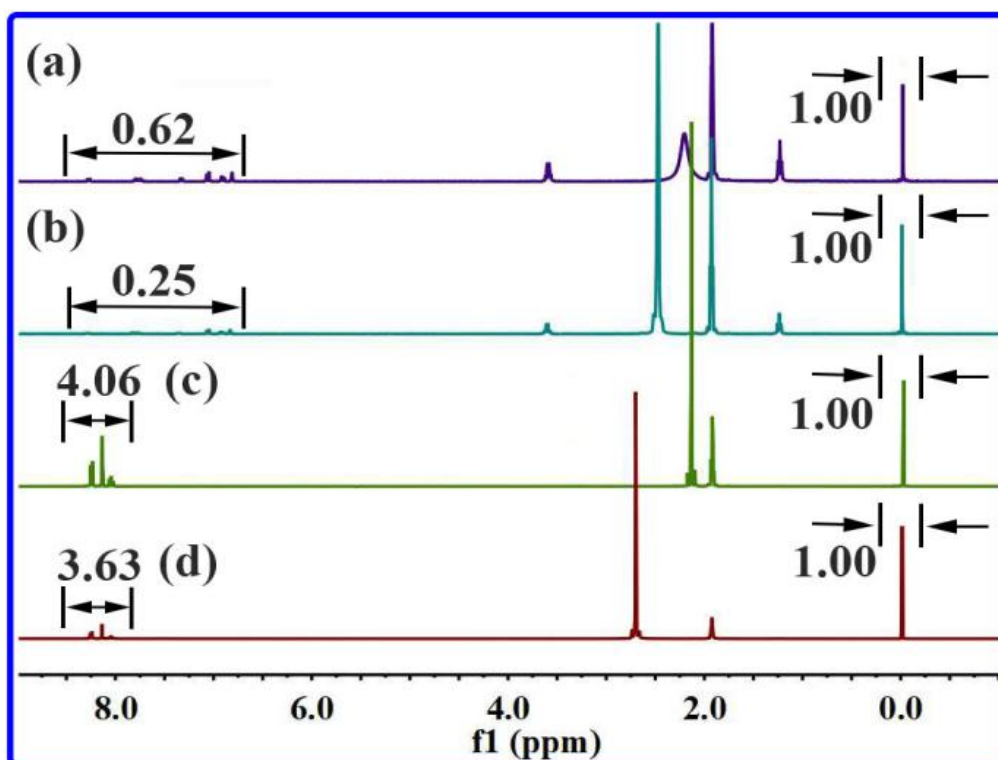
**Fig. S16** A general survey of the fluorescence spectra of FG13@A loaded with the 12 selected VOCs. The relative fluorescence intensities of FG13@A in response to the 12 selected VOCs.



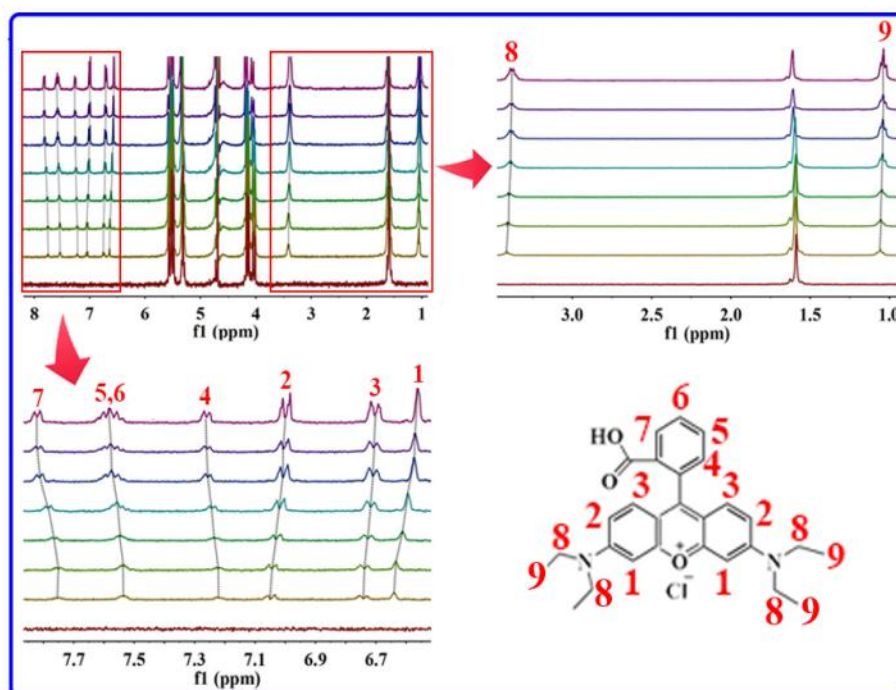
**Fig. S17** A general survey of the fluorescence spectra of FG14@A loaded with the 12 selected VOCs. The relative fluorescence intensities of FG14@A in response to the 12 selected VOCs.



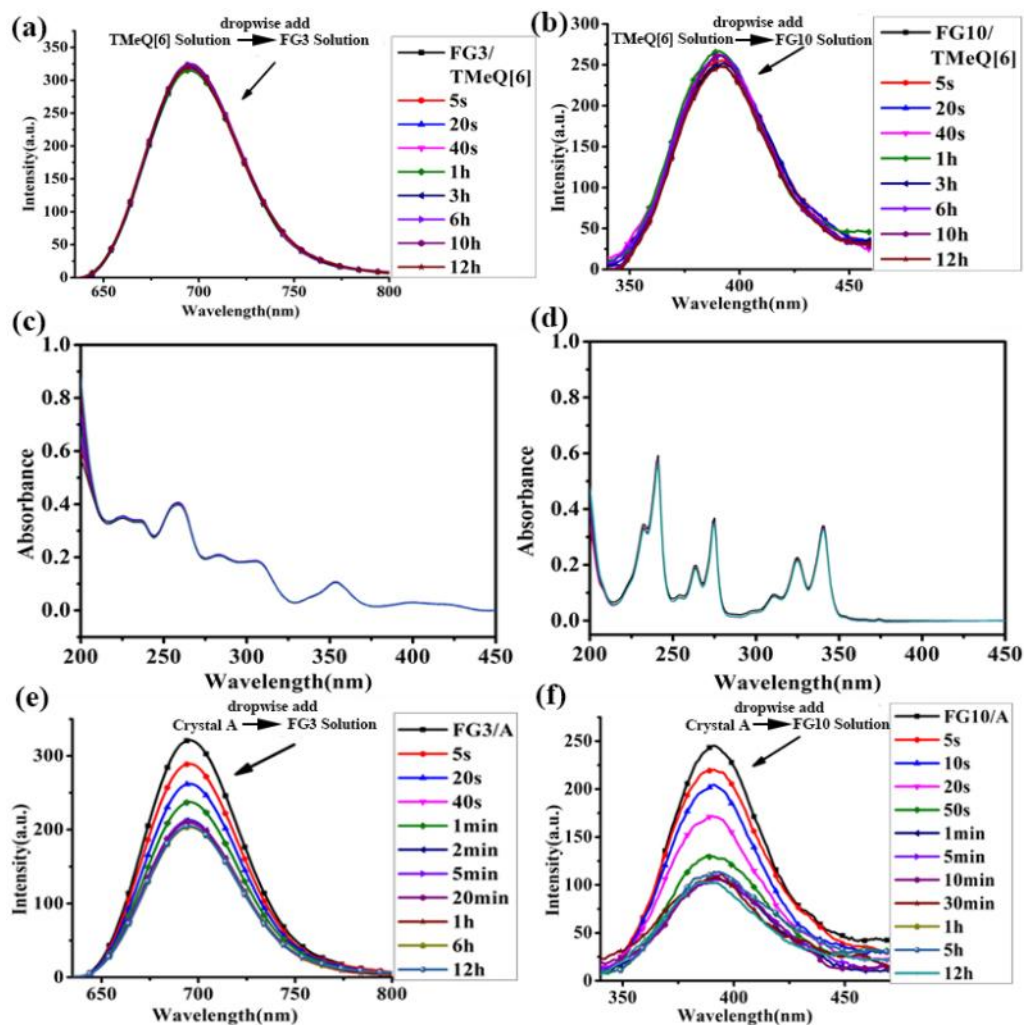
**Fig. S18** A general survey of the fluorescence spectra of FG15@A loaded with the 12 selected VOCs. The relative fluorescence intensities of FG15@A in response to the 12 selected VOCs.



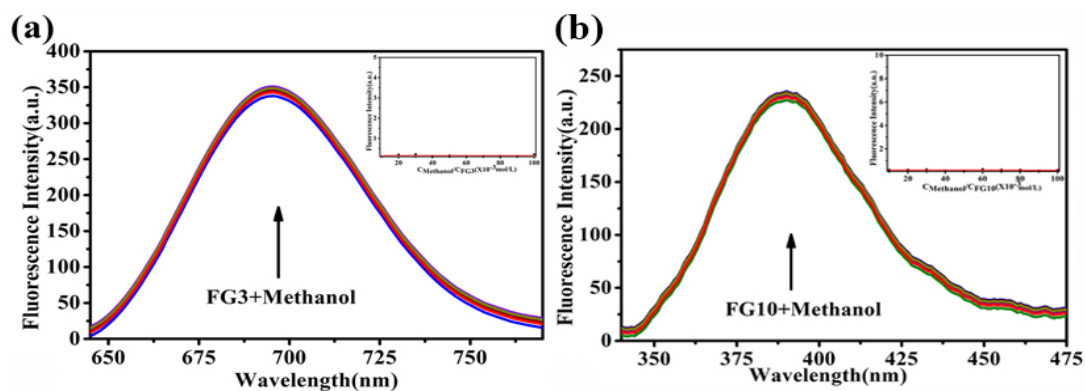
**Fig. S19** The  $^1\text{H}$  NMR spectra recorded in deuterated acetonitrile of: (a) 0.5 mL of 0.01 M **FG3**, (b) 0.5 mL of 0.01 M **FG3**+10 mg **A**; (c) 0.5 mL of 0.01 M **FG10** and (d) 0.5 mL of 0.01 M **FG10**+10 mg **A**.



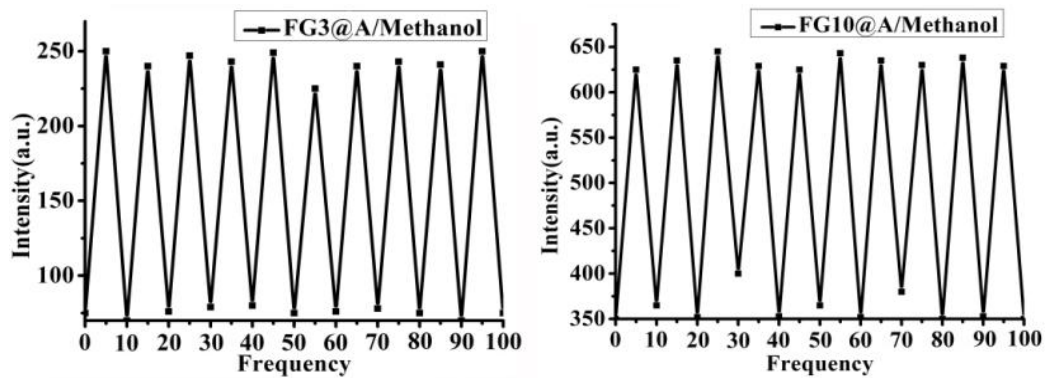
**Fig. S20** Titration  $^1\text{H}$  NMR spectra of **TMeQ[6]** with a gradual increase of **FG3**.



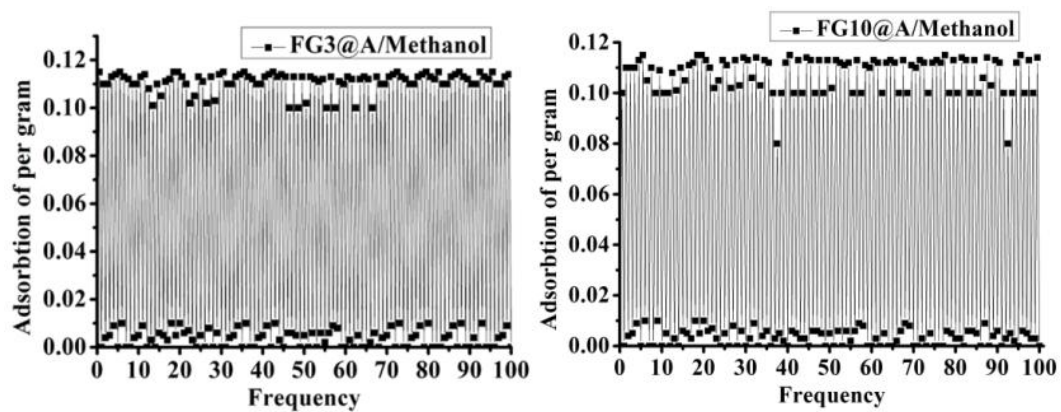
**Fig. S21** Titration fluorescence spectra of (a) FG3 and (b) FG10 with gradual increase of TMeQ[6]; titration absorption spectra of (c) FG3 and (d) FG10 with gradual increase of TMeQ[6]; titration fluorescence spectra of (e) FG3 and (f) FG10 with gradual increase of the solid assembly A, respectively.



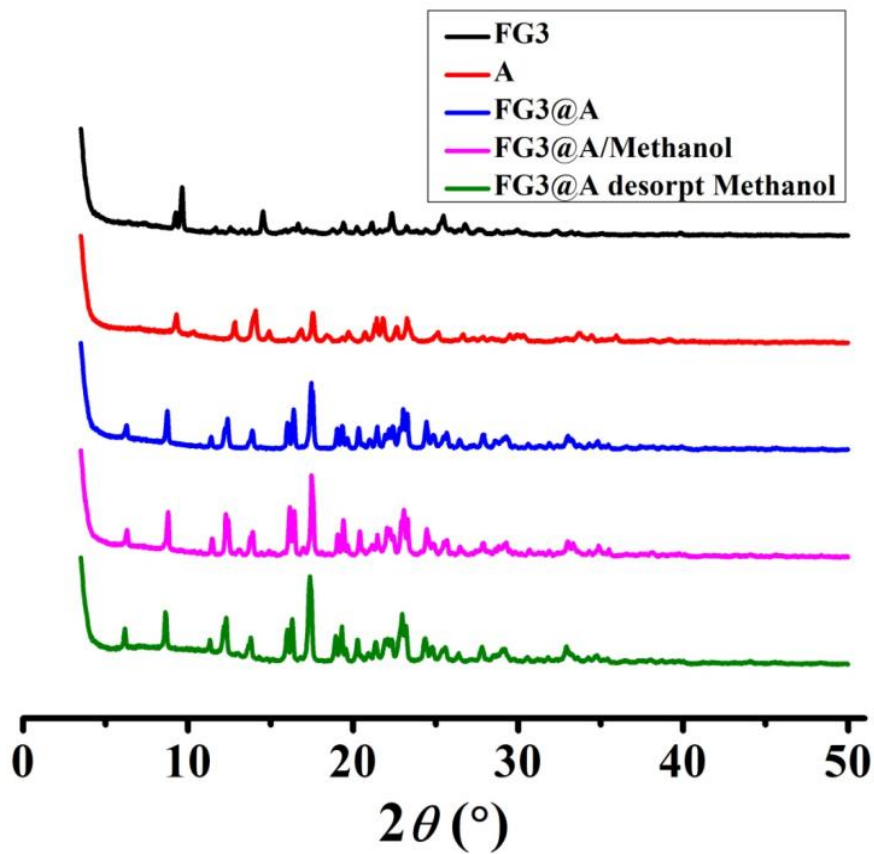
**Fig. S22** Titration fluorescence spectra of (a) FG3 and (b) FG10 with a gradual increase of methanol.



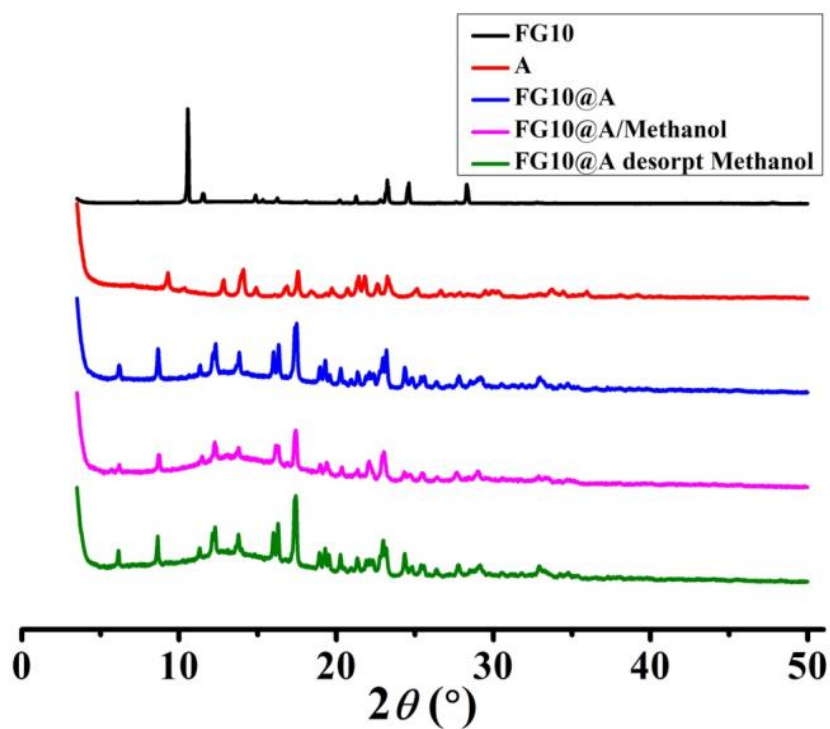
**Fig. S23** Lifetime experiments for the change in the fluorescence emission of (a) FG3@A and (b) FG10@A.



**Fig. S24** Lifetime experiments for methanol adsorption using (a) FG3@A and (b) FG10@A.



**Fig. S25** The XRD spectra of FG3, assembly A, FG3@A and FG3@A during the adsorption and desorption of methanol.



**Fig. S26** The XRD spectra of FG10, assembly A, FG10@A and FG10@A during the adsorption and desorption of methanol.