10.1071/CH20197_AC

©CSIRO 2020

Australian Journal of Chemistry 2020, 73(12), 1250-1259

Supplementary Material

Preparation and Structures of Rare Earth 3-Benzoylpropanoates and 3-Phenylpropanoates

Nicholas C. Thomas,^A Owen A. Beaumont,^B Glen B. Deacon,^{B,E} Cornelius Gaertner,^B Craig M. Forsyth,^B Anthony E. Somers,^C and Peter C. Junk^{D,E}

- ^A College of Sciences, Chemistry Department, Auburn University at Montgomery, Montgomery, AL 36117, USA
- ^B School of Chemistry, Monash University, Clayton, Vic. 3800, Australia
- ^C Institute for Frontier Materials, Deakin University, Burwood, Vic. 3125, Australia
- ^D College of Science, Technology and Engineering, James Cook University, Townsville, Qld 4811, Australia
- ^E Corresponding authors. Email: glen.deacon@monash.edu; peter.junk@jcu.edu.au



Fig. S1. TGA plot of bulk La(bp)₃ sampled for microanalysis.



Fig. S2. Powder XRD of bulk La(bp)₃ before and after heating.



Fig. S3. Powder XRD of bulk La(bp)₃ before heating, simulated $[La(bp)_3(H_2O)_2] \cdot 1\frac{1}{3}H_2O$ and simulated $[Ce(bp)_3(H_2O)]$. Simulated patterns generated from single crystal data.



Fig. S4. Powder XRD of bulk $La(bp)_3$ after heating, simulated $[La(bp)_3(H_2O)_2] \cdot 1\frac{1}{3}H_2O$ and simulated $[Ce(bp)_3(H_2O)]$. Simulated patterns generated from single crystal data.



Fig. S5. ATR-IR of bulk La(bp)₃ before heating, after heating and [Ce(bp)₃(H₂O)].



Fig. S6. ATR-IR of $[RE(bp)_3(H_2O)_n]$ complexes (RE = La, n = 2; RE = Y, Ce, Pr, Nd, Yb, n = 1).



Fig. S7. ATR-IR of $[RE(pp)_3]$ (RE = Y, Yb) and $[Ln(pp)_3] \cdot nH_2O$ (Ln = La, n = 0.5; Ln = Ce, n = 1; Ln = Nd, n = 3)

[La(mbp)₃(H₂O)₂]·3H₂O Crystal and Structure Refinement Data:

C₃₃H₄₃LaO₁₄, M 802.58, triclinic, space group $P\overline{1}$ (No. 2), *a* 8.6610(17), *b* 9.941(2), *c* 21.902(4) Å, *α* 80.81(3)°, β 87.61°, γ 66.27(3)°, V 1703.6(7) Å³, Z 2, λ 0.71073 Å, T 100(2) K, μ 1.322 mm⁻¹, 2θ_{max} 51.364°, 24936 reflections collected, 6450 unique (R_{int} 0.1146), N_o 6062, 444 parameters, R_1 (I>2σ(I)) 0.0865, w R_2 (all data) 0.2118, GoF 1.098, $\Delta e_{min/max}$ -4.33/5.72 e[•] Å⁻³.

Table S1. Selected bond distances ((Å) for	$[La(mbp)_3(H_2O)_2] \cdot 3H_2O$
-------------------------------------	----	-------	-----------------------------------

Atoms	La(mbp) ₃	
M-O(1)	2.470(7)	
$M-O(2)^i$	2.586(7)	
M-O(4)	2.462(7)	
M-O(4) ⁱ	2.765(7)	
$M-O(5)^i$	2.628(7)	
M-O(7)	2.535(7)	
M-O(7) ⁱⁱ	2.678(7)	
M-O(8) ⁱⁱ	2.624(7)	
M-O(10)	2.600(7)	
M-O(11)	2.655(6)	

ⁱ Symmetry operator: 1-x,1-y,1-z

ⁱⁱ Symmetry operator: -x,1-y,1-z