

*Supplementary material for***Links between the size fractionation, chemical speciation of dissolved copper and chemical speciation of dissolved organic matter in the Loire estuary**Gabriel Dulaquais,<sup>A,C</sup> Matthieu Waeles,<sup>A</sup> Johann Breitenstein,<sup>A</sup> Joël Knoery<sup>B</sup> and Ricardo Riso<sup>A</sup><sup>A</sup>Univ Brest, CNRS, Ifremer, IRD, LEMAR, F-29280, Plouzane, France.<sup>B</sup>Laboratoire de biogéochimie des contaminants métalliques, Ifremer centre de Nantes, 44311 Nantes, France.<sup>C</sup>Corresponding author. Email: gabriel.dulaquais@univ-brest.fr**Table S1:** Copper speciation data and associated standard errors (in italics) used in this study.

<b>Salinity</b>	<b>dCu</b> (nmol.kg <sup>-1</sup> )	<b>dLCu</b> (nmol.kg <sup>-1</sup> )	<b>logK<sub>dCu2</sub></b> +	<b>pCu</b>	<b>sCu</b> (nmol.kg <sup>-1</sup> )	<b>sLCu</b> (nmol.kg <sup>-1</sup> )	<b>logK<sub>sCu2</sub></b> +	<b>cCu*</b> (nmol.kg <sup>-1</sup> )	<b>cLCu**</b> (nmol.kg <sup>-1</sup> )	<b>logK<sub>cCu2+***</sub></b>
<b>0.1</b>	<b>80.3</b> <i>1.2</i>	<b>147.1</b> <i>2.4</i>	<b>13.0</b> <i>0.1</i>	<b>12.9</b>	<b>76.6</b> <i>1.4</i>	<b>120.2</b> <i>4.1</i>	<b>10.6</b> <i>0.1</i>	<b>3.7</b> <i>2.4</i>	<b>26.9</b> <i>6.5</i>	<b>13.7</b> <i>0.1</i>
<b>0.2</b>	<b>25.6</b> <i>0.7</i>	<b>71.3</b> <i>1.1</i>	<b>12.1</b> <i>0.2</i>	<b>12.3</b>	<b>20.6</b> <i>0.5</i>	<b>62.5</b> <i>4.3</i>	<b>11.0</b> <i>0.2</i>	<b>4.9</b> <i>1.2</i>	<b>8.8</b> <i>5.4</i>	<b>13.0</b> <i>0.3</i>
<b>4.48</b>	<b>36.4</b> <i>1.1</i>	<b>102.2</b> <i>7</i>	<b>12.3</b> <i>0.1</i>	<b>12.7</b>	<b>17.5</b> <b>0.9</b>	<b>60.4</b> <i>3.6</i>	<b>11.0</b> <i>0.1</i>	<b>18.8</b> <i>2</i>	<b>41.8</b> <i>10.6</i>	<b>12.9</b> <i>0.2</i>
<b>7.11</b>	<b>28.5</b> <i>1.2</i>	<b>83.8</b> <i>9</i>	<b>12.7</b> <i>0.1</i>	<b>13.0</b>	<b>16.1</b> <i>2</i>	<b>47.9</b> <i>3.6</i>	<b>11.1</b> <i>0.1</i>	<b>12.4</b> <i>3.2</i>	<b>35.9</b> <i>12.6</i>	<b>13.1</b> <i>0.1</i>
<b>17.19</b>	<b>13.1</b> <i>0.3</i>	<b>23.8</b> <i>0.9</i>	<b>11.8</b> <i>0.1</i>	<b>11.7</b>	<b>8.3</b> <i>0.5</i>	<b>17.1</b> <i>0.9</i>	<b>11.2</b> <i>0.3</i>	<b>4.8</b> <i>0.8</i>	<b>6.7</b> <i>1.8</i>	<b>12.2</b> <i>0.4</i>
<b>29.6</b>	<b>5.0</b> <i>0.4</i>	<b>8.4</b> <i>0.1</i>	<b>11.4</b> <i>0.1</i>	<b>11.2</b>	<b>3.4</b> <i>0.5</i>	<b>6.0</b> <i>0.4</i>	<b>10.5</b> <i>0.2</i>	<b>1.6</b> <i>0.9</i>	<b>2.4</b> <i>0.5</i>	<b>11.9</b> <i>0.3</i>

\* Estimated as cCu = dCu – sCu

\*\* Estimated as  $cLCu = dLCu - sLCu$

\*\*\* Estimated using equation 8 of the manuscript

**Table S2:** Dissolved organic matter speciation data and associated standard errors (in italics) used in this study.

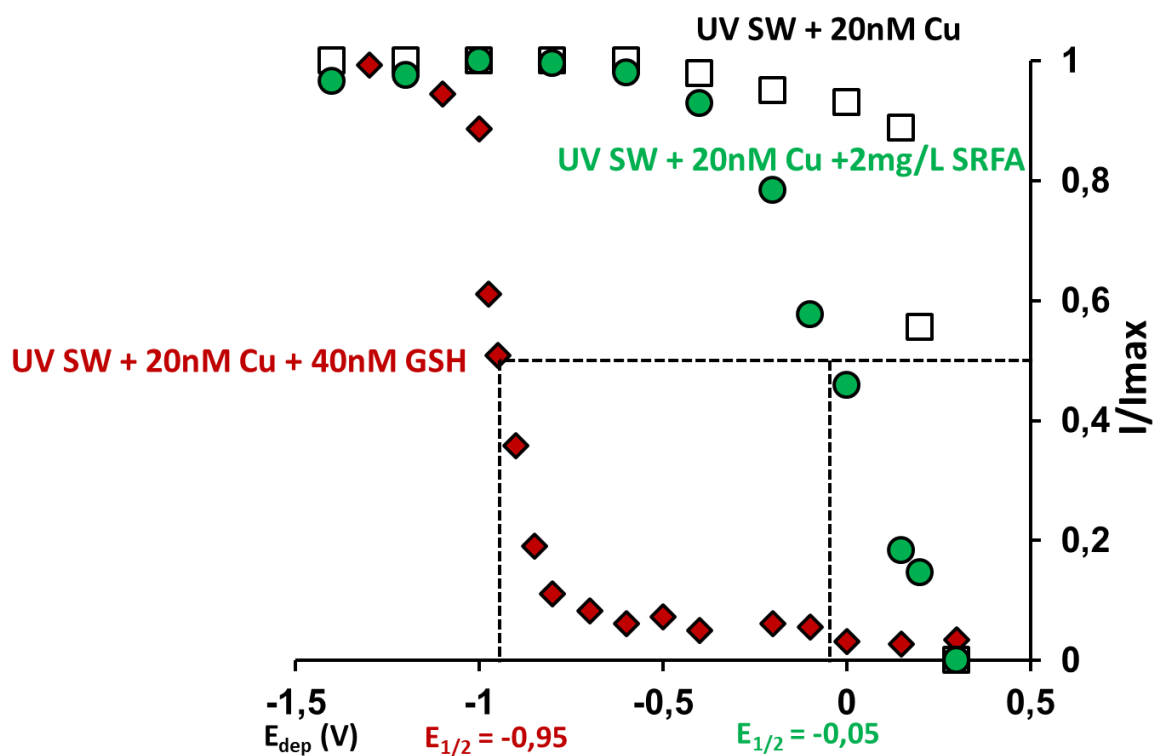
<b>Salinity</b>	<b>Dissolved eHS</b> <b>mg eq SRFA.kg<sup>-1</sup></b>	<b>Soluble eHS</b> <b>mg eq SRFA.kg<sup>-1</sup></b>	<b>DOC</b> <b>mg C .kg<sup>-1</sup></b>	<b>sOC*</b> <b>mg C.kg<sup>-1</sup></b>	<b>cOC*</b> <b>mg C.kg<sup>-1</sup></b>
<b>0.1</b>	<b>4.1</b> <i>0.12</i>	<b>3.4</b> <i>0.10</i>	<b>3.7</b> <i>0.11</i>	<b>3.2</b> <i>0.10</i>	<b>0.5</b> <i>0.21</i>
<b>0.2</b>	<b>2.6</b> <i>0.08</i>	<b>2.7</b> <i>0.08</i>	<b>2.3</b> <i>0.07</i>	<b>2.2</b> <i>0.07</i>	<b>0.1</b> <i>0.14</i>
<b>4.48</b>	<b>2.6</b> <i>0.08</i>	<b>2.4</b> <i>0.07</i>	<b>3.5</b> <i>0.11</i>	<b>2.6</b> <i>0.08</i>	<b>0.9</b> <i>0.19</i>
<b>7.11</b>	<b>2.2</b> <i>0.07</i>	<b>1.9</b> <i>0.06</i>	<b>3.2</b> <i>0.10</i>	<b>2.4</b> <i>0.07</i>	<b>0.8</b> <i>0.17</i>
<b>17.19</b>	<b>1.0</b> <i>0.03</i>	<b>0.7</b> <i>0.02</i>	<b>2.0</b> <i>0.06</i>	<b>1.7</b> <i>0.05</i>	<b>0.3</b> <i>0.11</i>
<b>29.6</b>	<b>0.6</b> <i>0.02</i>	<b>0.5</b> <i>0.01</i>	<b>1.2</b> <i>0.04</i>	<b>1.0</b> <i>0.03</i>	<b>0.2</b> <i>0.07</i>

\* see section 3.3 for definition of sCO and cOC

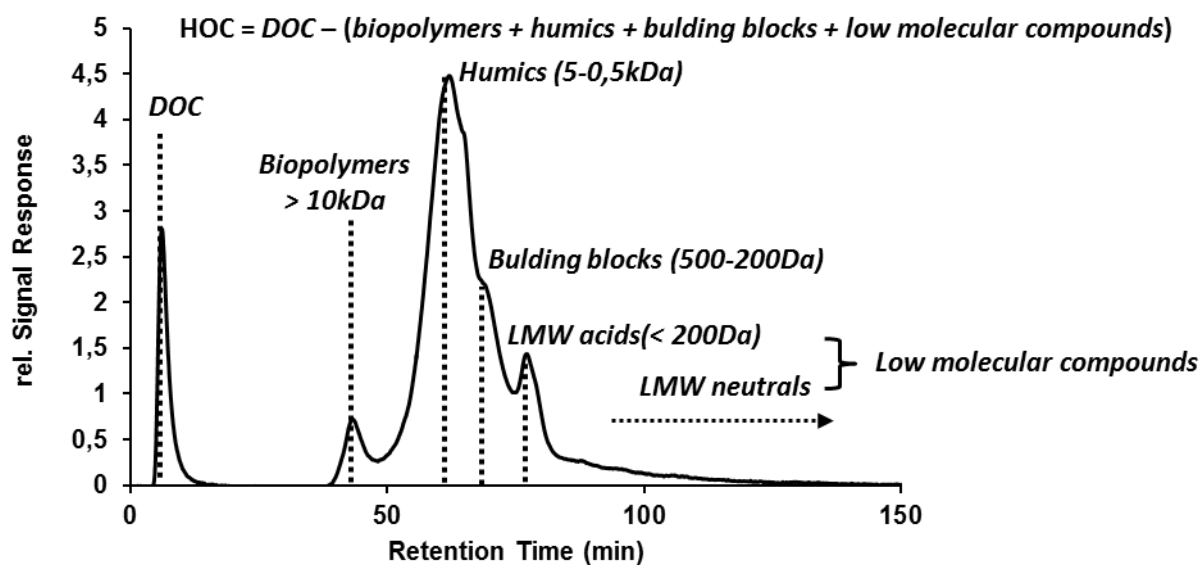
**Table S3:** Dissolved copper concentrations measured by Waeles et al., 2004 along the salinity gradient of the Loire Estuary during winter 2001

<b>Salinity</b>	<b>dCu (nmol.L-1) winter 2001</b>
<b>31.4</b>	<b>6.99</b>
<b>26.6</b>	<b>13.52</b>
<b>22.2</b>	<b>18.13</b>
<b>18</b>	<b>21.12</b>
<b>12</b>	<b>20.89</b>
<b>9.1</b>	<b>19.31</b>
<b>5.3</b>	<b>18.56</b>
<b>2.1</b>	<b>17,26</b>

<b>1.1</b>	<b>16.3</b>
<b>0.4</b>	<b>16.65</b>



**Figure S1:** Pseudopolarographic experiments conducted on a UV seawater (Bay of brest) spiked with only 20nM of copper (white squares), 20nM of copper and 2mg/L of Suwannee River Fulvic Acid (SRFA, green dots) and 20nM of copper and 40nM of glutathione (GSH, red losanges). Half wave potentials are indicated.



**Figure S2:** Typical chromatogram recorded by the organic carbon detector of the size exclusion chromatography coupled with organic carbon detection for an estuarine sample. The different fractions operationally defined by their retention time are indicated. In this study LMW acids and neutrals are combined as low molecular compounds. Hydrophobic DOC (HOC) is defined as the difference between DOC and the carbon content of the 4 other fractions.